

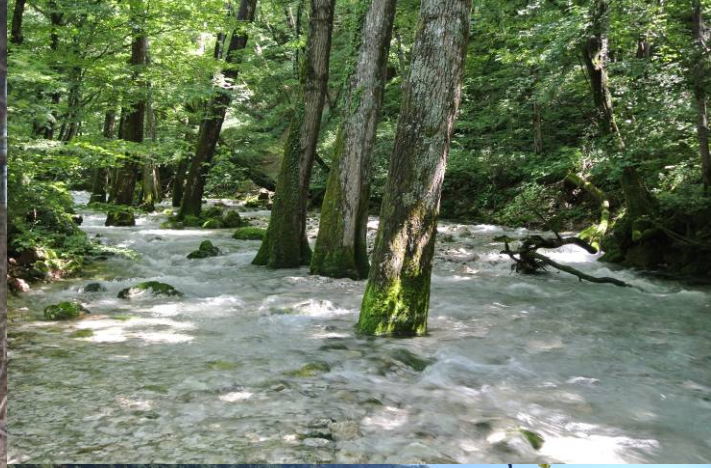


KAKO KROŽI VODA?

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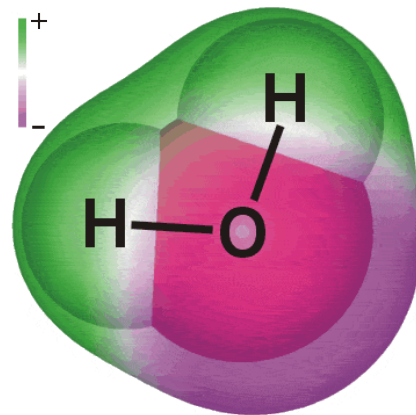
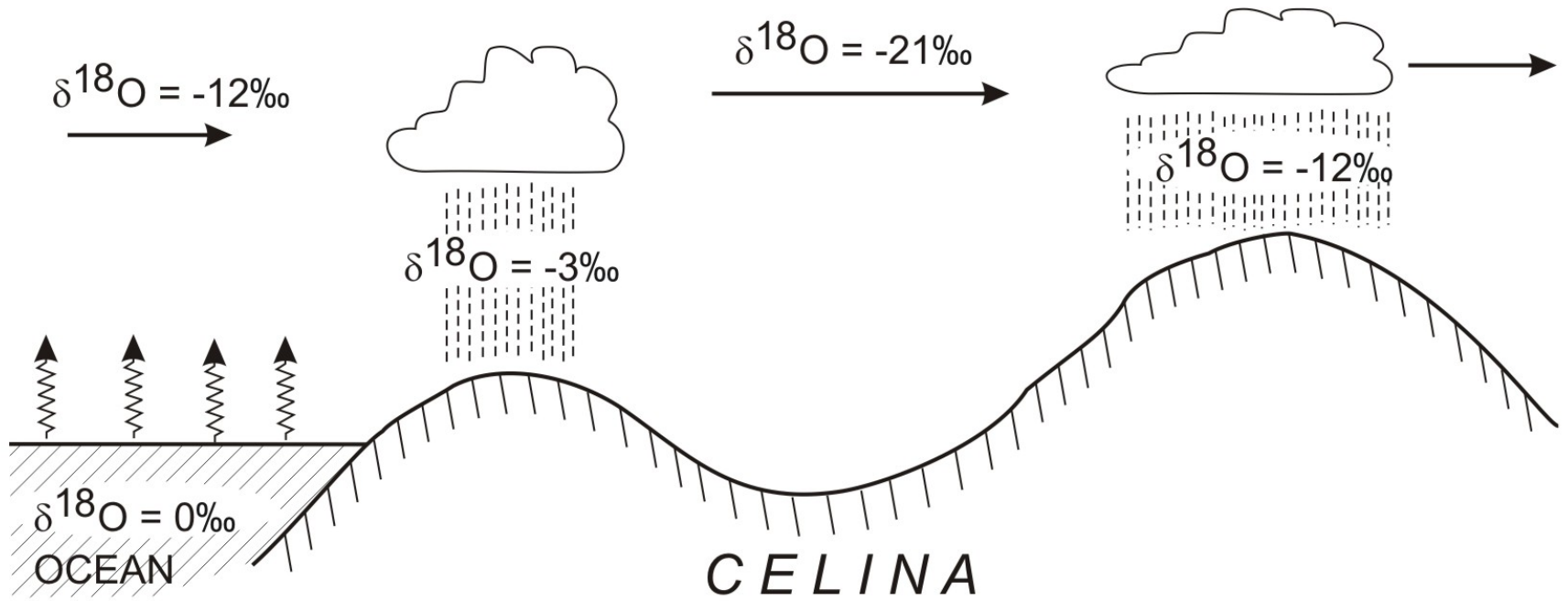




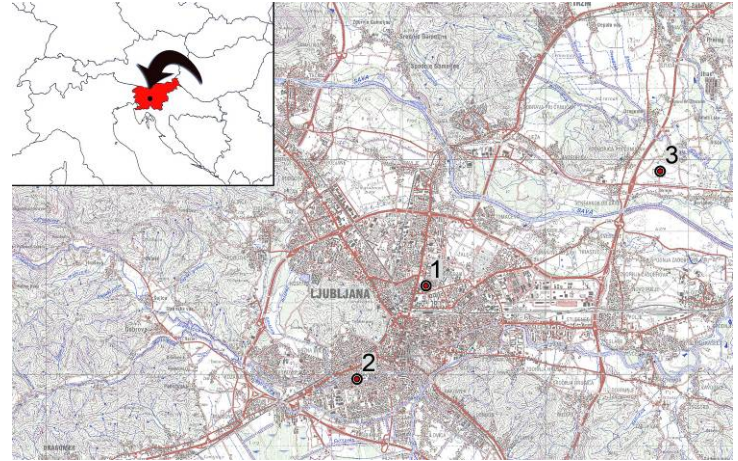
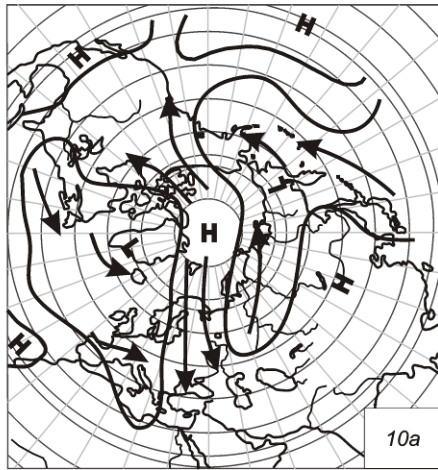
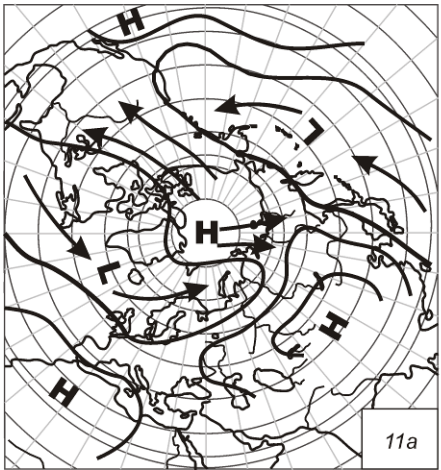
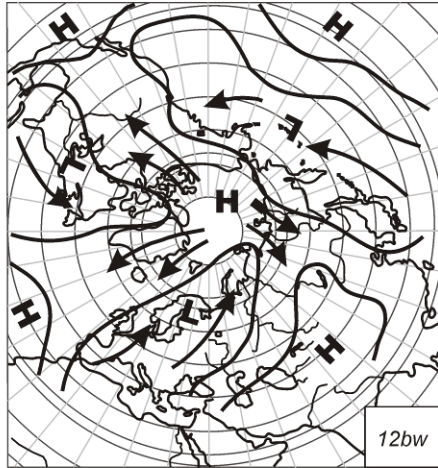
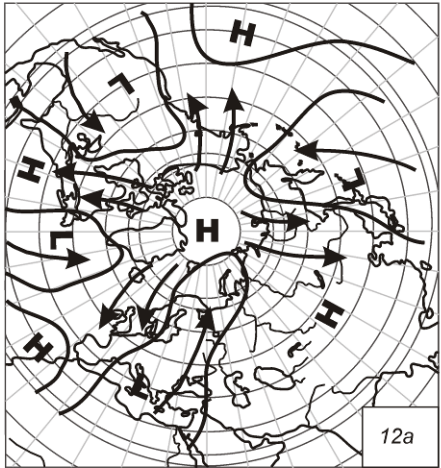
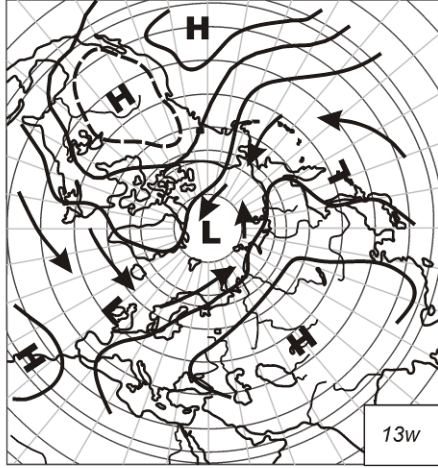
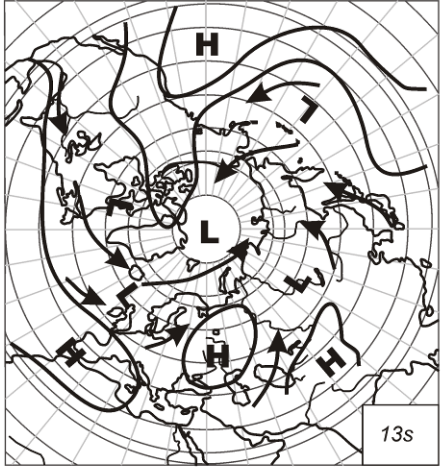
Kroženje vode



U.S. Department of the Interior
U.S. Geological Survey
<http://ga.water.usgs.gov/edu/watercycle.html>









Relation between isotopic composition of precipitation and atmospheric circulation patterns



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ARTICLE INFO

Article history:

Received 15 July 2014

Received in revised form 12 August 2015

Accepted 18 August 2015

Available online 5 September 2015

This manuscript was handled by Geoff

Syme, Editor-in-Chief

Keywords:

Atmospheric circulation pattern

Elementary circulation mechanisms

Precipitation

Hydrogen isotopes

Oxygen isotopes

Deuterium excess

SUMMARY

Precipitation generating processes depend on atmospheric circulation patterns and consequently it is expected that its water stable isotopic composition of hydrogen and oxygen is related to them. Precipitation generated at similar atmospheric circulation patterns should have similar empirical distribution of $\delta^2\text{H}$ and $\delta^{18}\text{O}$ values. Mathematical model based on the linear combination of $\delta^2\text{H}$ and $\delta^{18}\text{O}$ values and on precipitation amount weighted average related to elementary air circulation mechanisms – ECM is proposed. The model enables estimation of average $\delta^2\text{H}$ and $\delta^{18}\text{O}$ values and their standard deviation for the precipitation generated at distinctive atmospheric circulation patterns. Approach in which atmospheric circulation patterns were classified as ECM based on the Dzerdzevskii classification was applied. Application of the model is illustrated on the long term precipitation record from Ljubljana GNIP station Slovenia. Estimated values of the parameters for empirical distributions of $\delta^2\text{H}$ and $\delta^{18}\text{O}$ of each ECM subtype have shown that calculated estimates are reasonable. Further applications of the proposed model enable new insight into the understanding of isotopes spatial and temporal distribution in precipitation important also for better understanding of climate proxies.

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1. Introduction

During studying isotopic composition of the precipitation frequently efforts have been made to elucidate the origin and transport path of water vapour resulting in precipitation at particular location (Liotta et al., 2006; Peng et al., 2005; Vreča et al., 2007) or over certain geographical region (Friedman et al., 2002; Rindsberger et al., 1983; Treble et al., 2005) or globally (Jouzel et al., 2000). It is widely accepted that isotopic composition of precipitation can help to understand the precipitation processes in the atmosphere (Craig and Gordon, 1965; Dansgaard, 1964; Gourcy et al., 2005). Several efforts can be found in the literature to understand also how the origin of the water vapour in the precipitation influences its isotopic composition (Araguas-Araguas et al., 2000; Craig, 1961; Gat, 1996) as well as what is the influence of other meteorological processes and variables on precipitation isotopic

characteristics (Rozanski et al., 1993; Liebminger et al., 2007; Siegenthaler and Oeschger, 1980). The interpretation of isotopic composition of precipitation is not straightforward; usually it consists of combination of physical approach based on the understanding of various fractionation processes (Gat, 2005, 2010) and on the interpretation of meteorological conditions resulting in precipitation event (Gedzelman and Arnold, 1994; Lykoudis et al., 2010).

Samples of precipitation for isotope analyses can be collected at various time scales. Most frequent time collection interval is one month. This is the case with International Atomic Energy Agency precipitation sampling program Global Network of Isotopes in Precipitation – GNIP which started in 1961 and resulted in large data base of precipitation isotopic characteristics from the meteorological stations around the globe (Gourcy et al., 2005; Schotter et al., 1996). Monthly samples represent a composite of all the precipitation events for the particular month. Interpretation of the isotopic data on the monthly basis become challenging when sample is a result of several precipitation events and different atmospheric circulation patterns especially at those places where precipitation genesis is a result of water vapour originating from different sources with distinctive isotopic signal. Isotopic composition of

Brenčič, M., Kononova, N.K., Vreča, P., 2015: Relation between isotopic composition of precipitation and atmospheric circulation patterns. Journal of Hydrology 529, 1422-1432.

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