

Nanoscale behaviour and properties of noble metals in electrochemical environment



NATIONAL INSTITUTE OF CHEMISTRY

D13 Department of Catalysis and Chemical Reaction Engineering

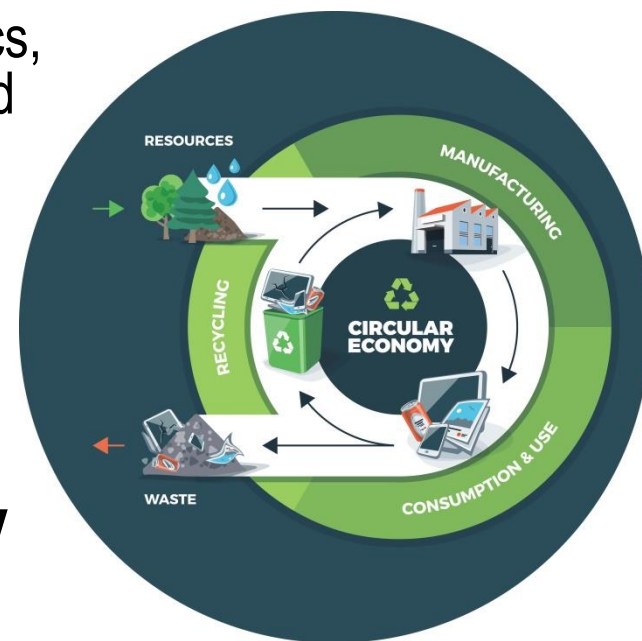
Dr. Nejc Hodnik

in close collaboration with D10 and D04

On one side our development and **life quality** is dependent on **raw materials** resources and economics, however on the other side it is limited by pollution and negative **climate changes**.

Strategies towards sustainable future:

- EU's transition to a **circular economy**
 - no supply risk!
- EU's transition to a **low-carbon economy**
 - clean and renewable energy!



<http://www.ies.be/node/3632>



<http://www.ft.lk/article/482633/Green-finance--Financing-the-transition-towards-a-low-carbon-economy>

Problem! Global Warming!



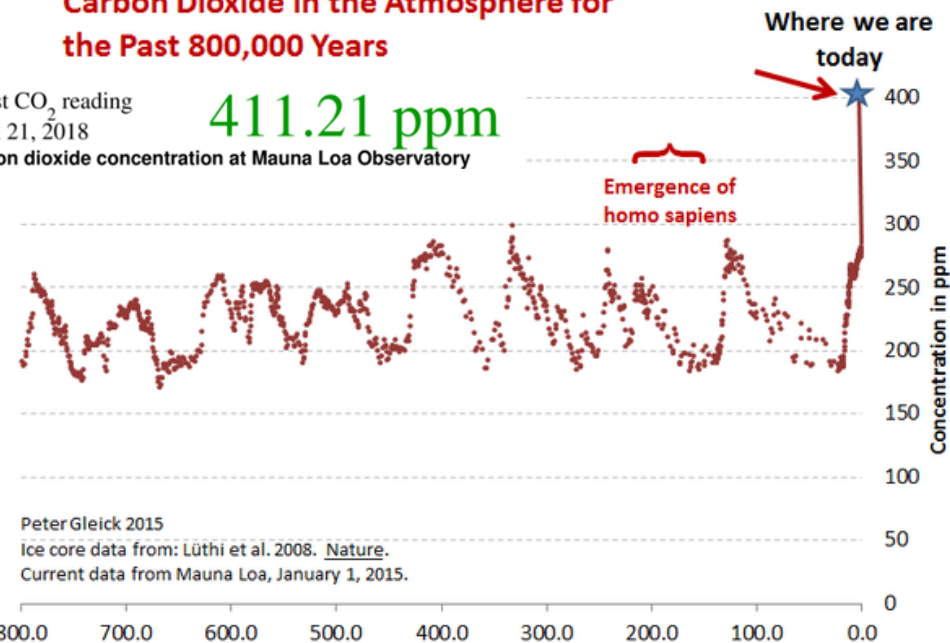
We must slow down global warming (below 2 °C) -> rise of greenhouse gas CO₂, that mostly comes from burning of fissile fuels (coal, oil and gas).

„KEELING CURVE“

Carbon Dioxide in the Atmosphere for the Past 800,000 Years

Latest CO₂ reading
April 21, 2018
Carbon dioxide concentration at Mauna Loa Observatory

411.21 ppm



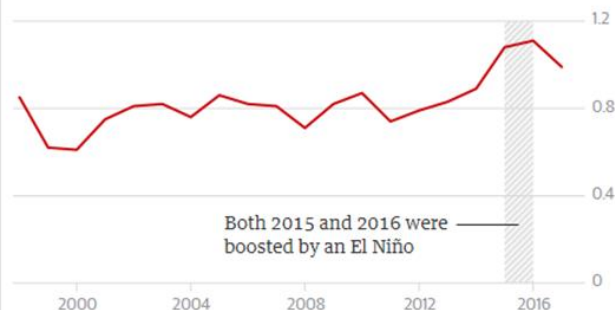
Peter Gleick 2015
Ice core data from: Lüthi et al. 2008. *Nature*.
Current data from Mauna Loa, January 1, 2015.

<https://scripps.ucsd.edu/programs/keelingcurve/>

-2016 was the warmest in history
-17 out of 18 warmest years were after 2001

Pollution

Anomaly, °C relative to 1850-1900



Guardian graphic. Source: Met Office

Problem! Global Warming!



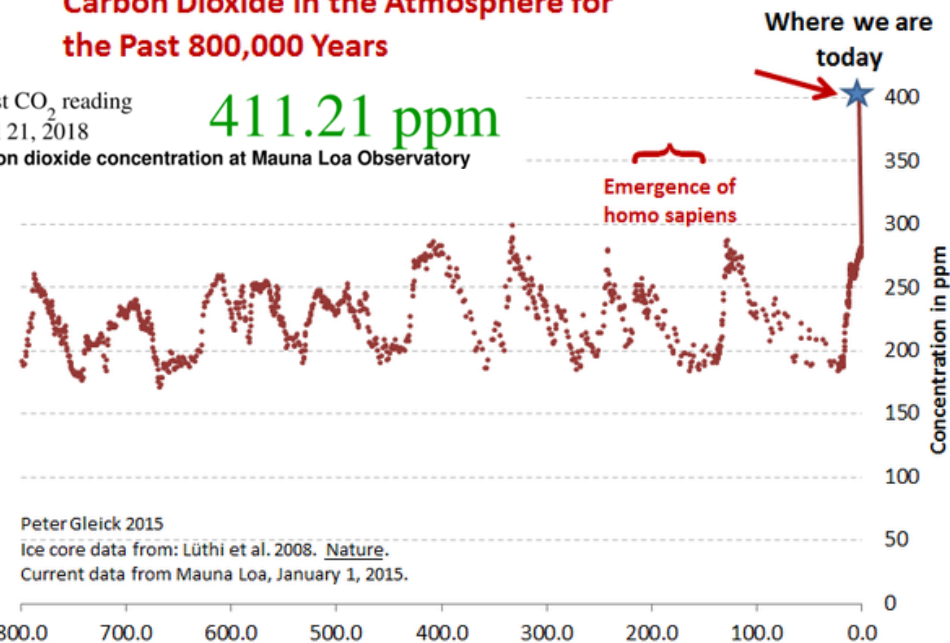
We must slow down global warming (below 2 °C) -> rise of greenhouse gas CO₂, that mostly comes from burning of fissile fuels (coal, oil and gas).

„KEELING CURVE“

Carbon Dioxide in the Atmosphere for the Past 800,000 Years

Latest CO₂ reading
April 21, 2018
Carbon dioxide concentration at Mauna Loa Observatory

411.21 ppm



Peter Gleick 2015
Ice core data from: Lüthi et al. 2008. *Nature*.
Current data from Mauna Loa, January 1, 2015.

<https://scripps.ucsd.edu/programs/keelingcurve/>

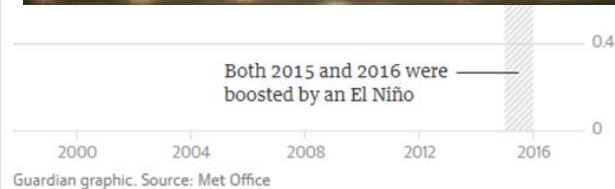
-2016 was the warmest in history
-17 out of 18 warmest years were after 2001

Pollution

-> to lower atmospheric pollution levels in big cities



Beijing
PM2.5 598 µg/m³
Max 25 µg/m³

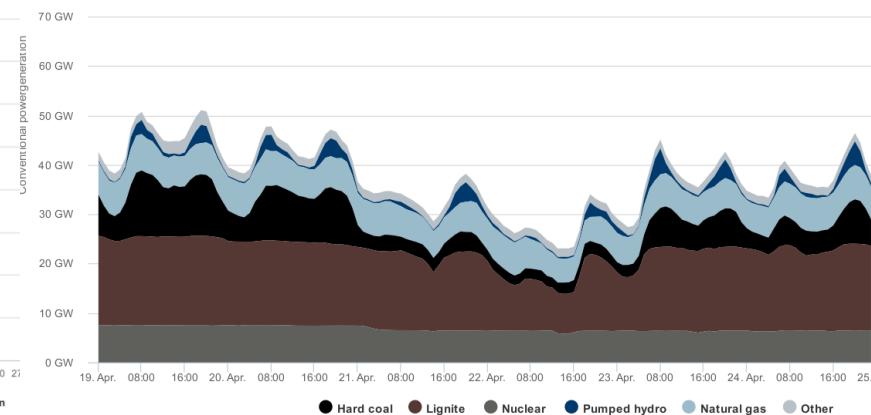
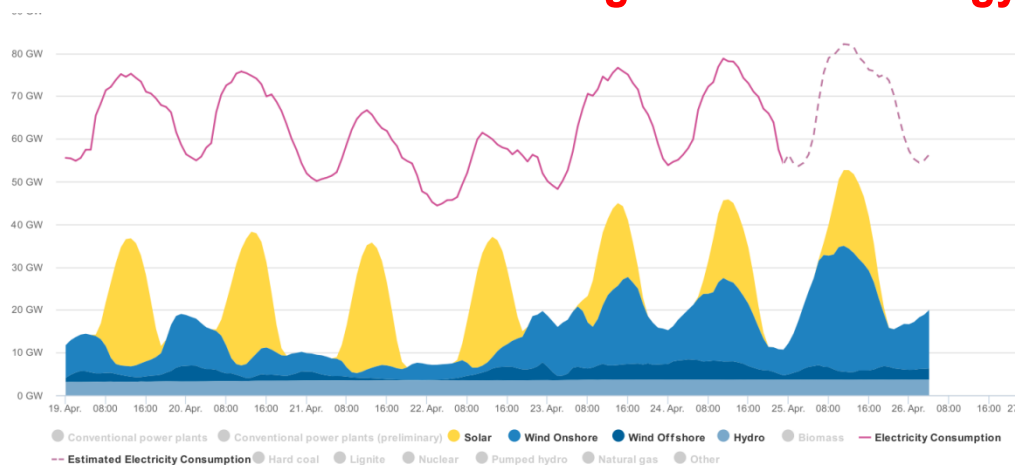


Guardian graphic. Source: Met Office

<http://www3.epa.gov/climatechange/ghgemissions/global.html>

What we can do?

- As scientists we can find new or more efficient ways for everyday tasks to be (more) sustainable.
- To lower the dependance from fossile fuels
 - Like turning CO₂ to chemicals or biomass to fuel
- Utilize intermittent energy sources like sun and wind (renewable sources)
- **Current Grand Challenges: to store energy & catalytic conversion processes**



The fluctuating Germany energy production from wind and solar including (mismatched) power demand in the last week.

Conventional Power Generation

<https://www.agora-energiewende.de/en/topics/-agothem-/Produkt/produkt/76/Agorameter/>

What we can do?

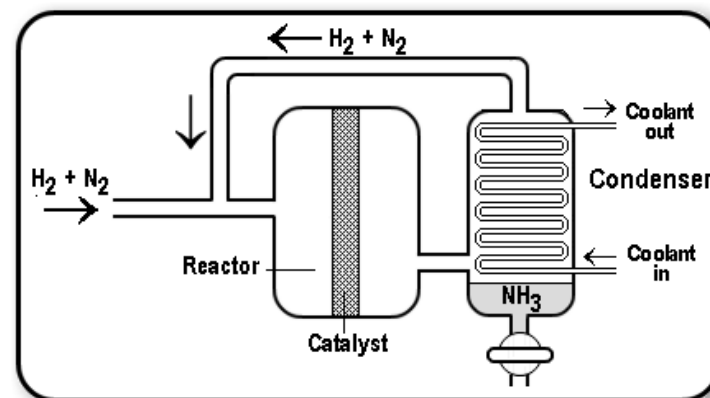
- Example from history: **Haber–Bosch process (1910):**
- is an artificial nitrogen fixation process and is the main industrial procedure for the production of ammonia today
- The process converts atmospheric **nitrogen** (N_2) to ammonia (NH_3) by a reaction with **hydrogen** (H_2) using a metal catalyst under high temperatures and pressures
- Before the Haber process, ammonia had been difficult to produce on an industrial scale; Grand Challenge.
- Although the Haber process is mainly used to produce fertilizer today, during World War I it provided Germany with a source of ammonia for the production of explosives.



Fritz Haber

1919 - Nobel Prize

https://en.m.wikipedia.org/wiki/Haber_process



<https://chemstuff.co.uk/academic-work/a-level/the-haber-process/>

What we can do?

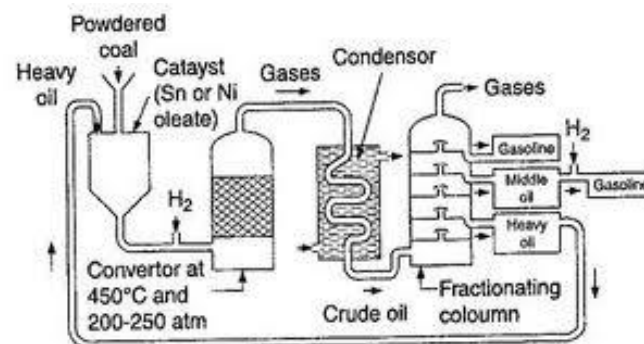
- Example from history: **Bergius process (1913)**:
- is a method of production of **liquid synthetic fuel** by catalytic hydrogenation of high-volatile bituminous coal at high temperature and pressure.
- Before the Bergius process, liquid fuel like diesel oil or petroleum was difficult to get, especially in Germany; Grand Challenge.



Fredrich Bergius
1931 - Nobel Prize

https://en.m.wikipedia.org/wiki/Haber_process

- The Fischer–Tropsch process (1925) is a collection of chemical reactions that converts a mixture of **carbon monoxide (CO)** and **hydrogen (H₂)** into **liquid hydrocarbons - fuel**.



<https://chemstuff.co.uk/academic-work/a-level/the-haber-process/>

What we can do?

- Example from history: **Bergius process (1913):**
- is a method of production of **liquid synthetic fuel** by catalytic hydrogenation of high-volatile bituminous coal at high temperature and pressure.
- Before the Bergius process, liquid fuel like diesel oil or petroleum was difficult to get, especially in Germany; Grand Challenge.
- The Fischer–Tropsch process (1925) is a collection of chemical reactions that converts a mixture of **carbon monoxide (CO)** and **hydrogen (H₂)** into **liquid hydrocarbons - fuel**.



Fredrich Bergius
1931 - Nobel Prize

https://en.m.wikipedia.org/wiki/Haber_process

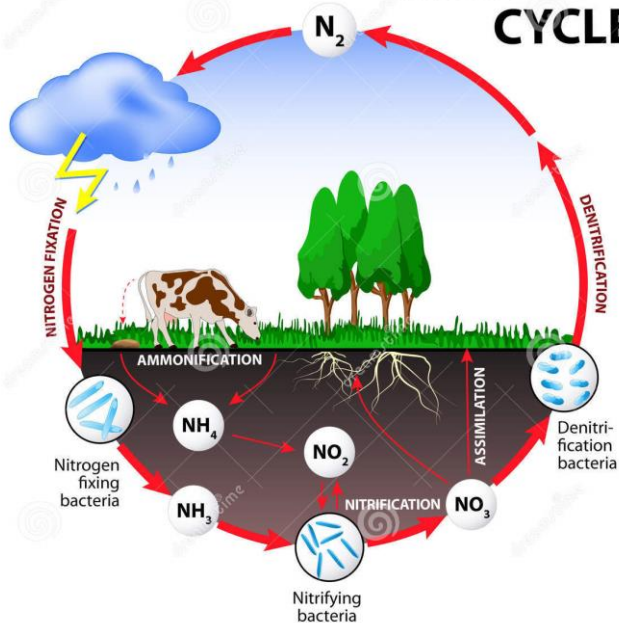


Carl Bosch
1931 - Nobel Prize

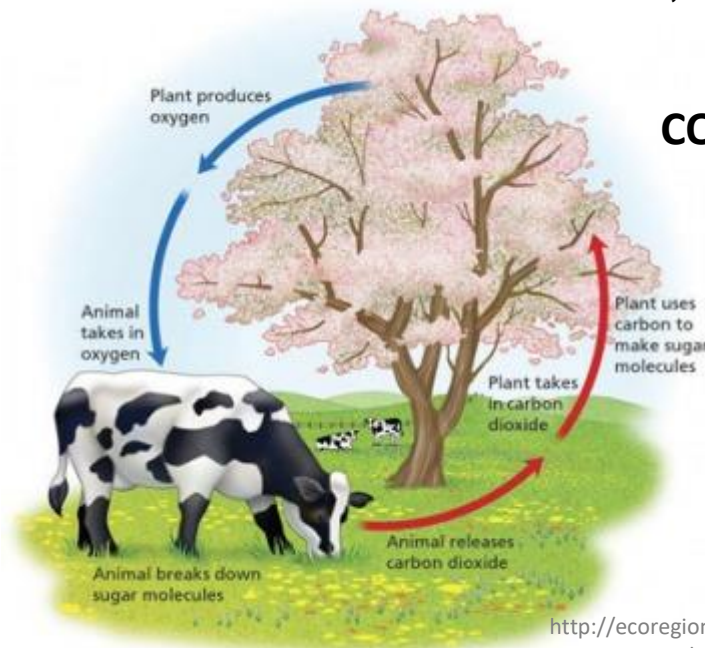
NEJC HODNIK

Forum40, Maj 2018, Ljubljana

NITROGEN CYCLE



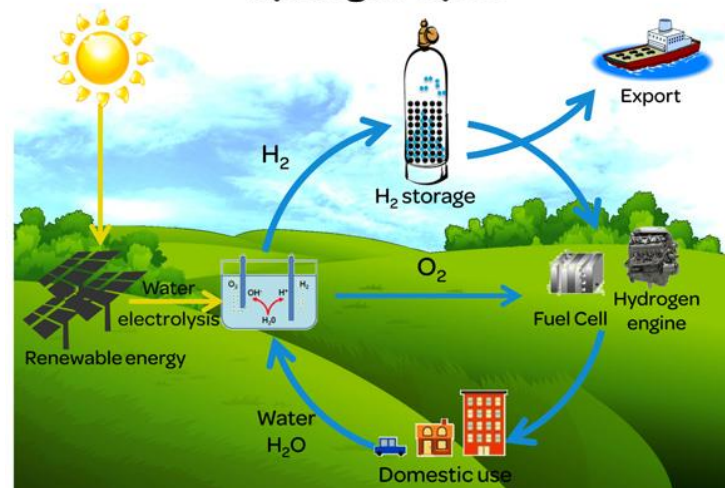
CO₂ cycle



Energy

<http://ecoregionproject.weebly.com/nitrogen-oxygen-and-carbon-cycles.html>

Hydrogen Cycle



<http://www.merlin.unsw.edu.au/energyh/about-hydrogen-energy/>

<https://www.dreamstime.com/royalty-free-stock-photos-nitrogen-cycle-image29063058>

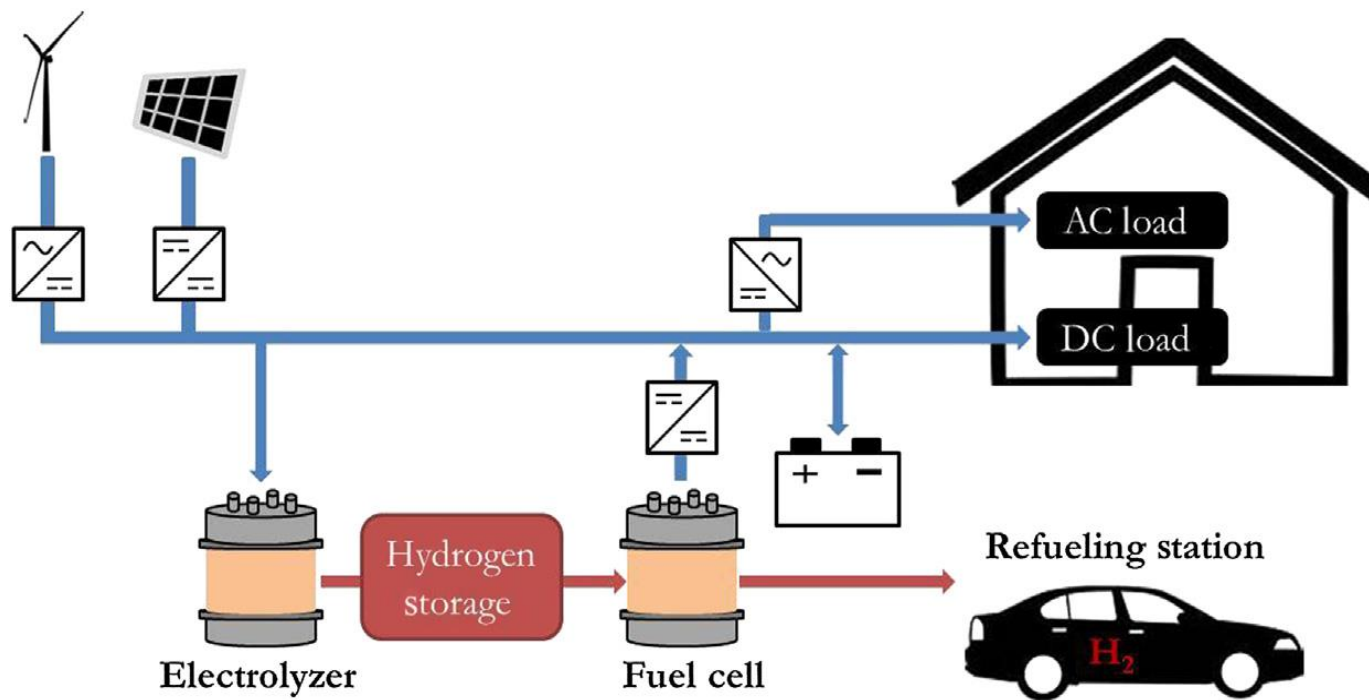
Fertilizer -> food

Common for all conversion processes -> catalysis

-90-95% of all industrial chemical processes

What we can do?

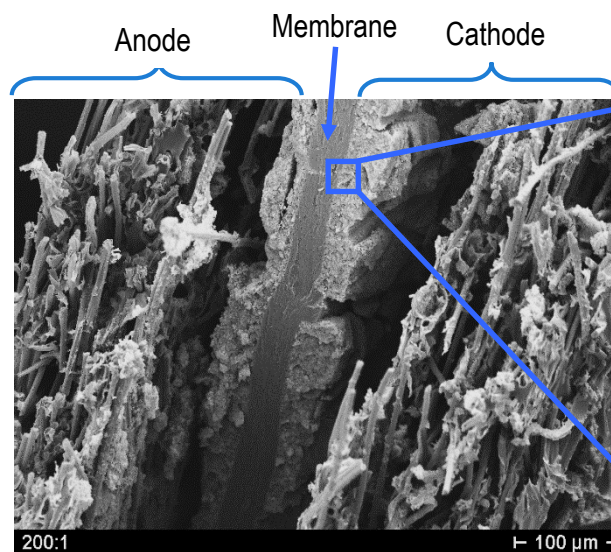
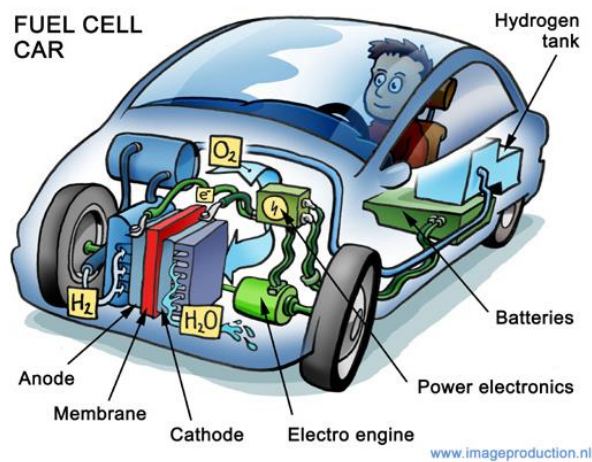
- **Hydrogen cycle at home:**
- self sustained hydrogen home



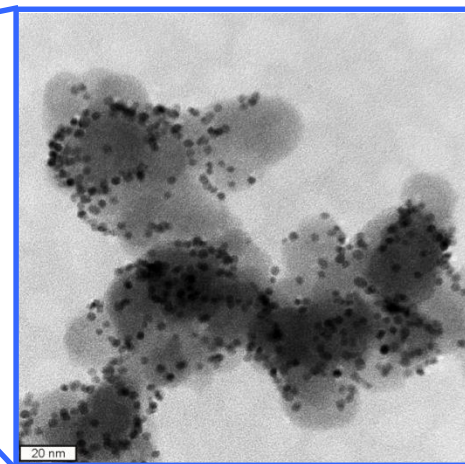
Current Grand Challenges: to store energy & better catalytic processes

Proton exchange membrane fuel cell

No emissions – no CO₂! Only 2H₂ + O₂ -> 2H₂O
Higher efficiency!

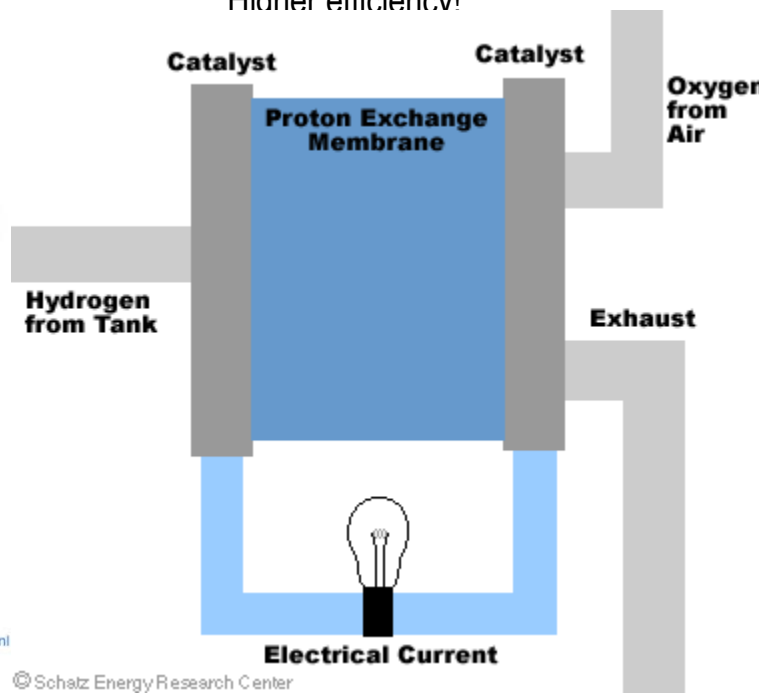
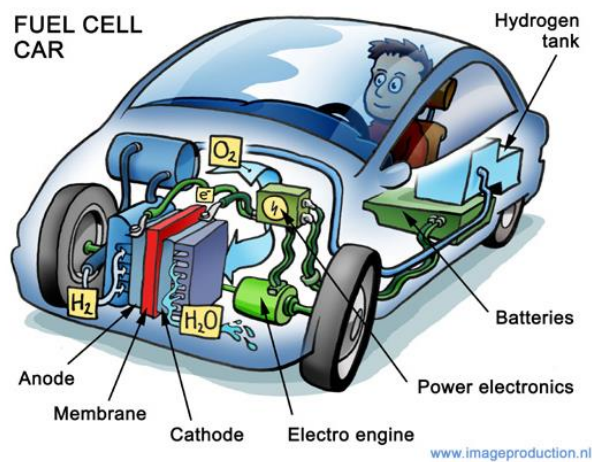


Electrocatalysts are Pt-based nanoparticles on high surface carbon.

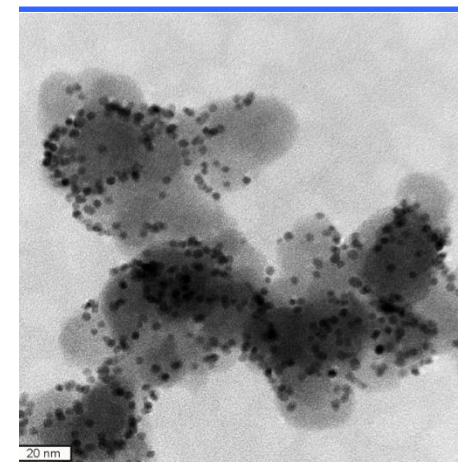


Proton exchange membrane fuel cell

No emissions – no CO₂! Only 2H₂ + O₂ -> 2H₂O
Higher efficiency!

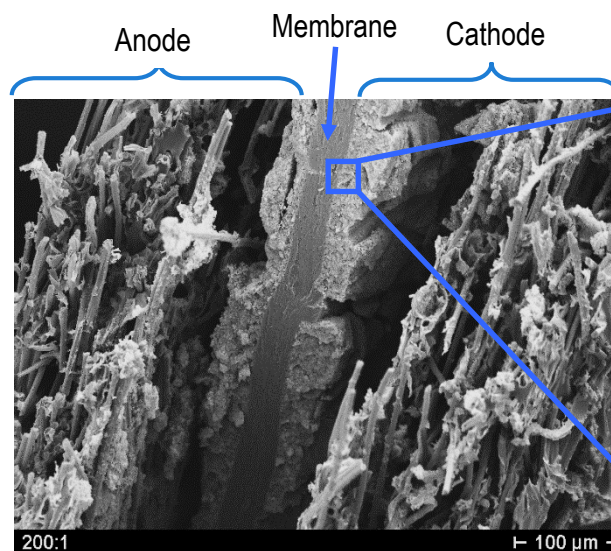
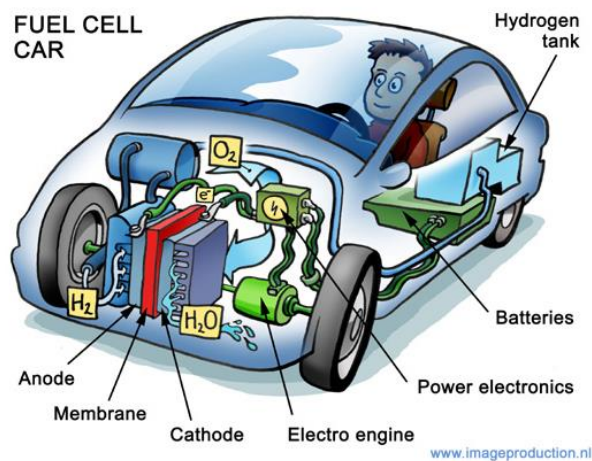


Electrocatalysts are Pt-based nanoparticles on high surface carbon.

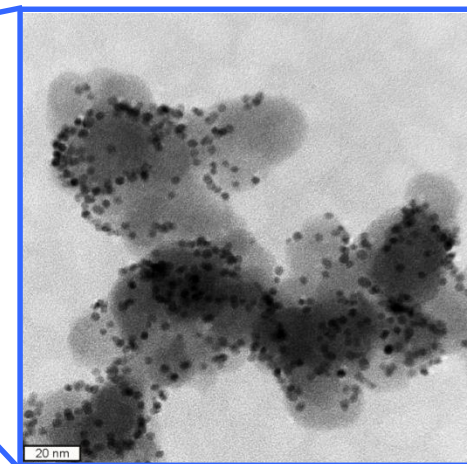


Proton exchange membrane fuel cell

No emissions – no CO₂! Only 2H₂ + O₂ -> 2H₂O
Higher efficiency!



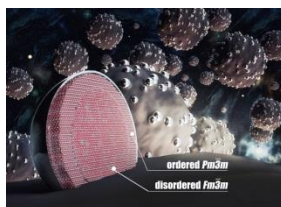
Electrocatalysts are Pt-based nanoparticles on high surface carbon.



PEM-FC electrocatalyst still requires improvement of performance & stability & cost !

Most promising replacement of Pt/C are Pt-alloys, at least activity wise – therefore it is important to study their electrochemical behaviour – their stability!

PtCu₃ nanoparticles...
- US 9147885 Patent



Proton exchange membrane fuel cell

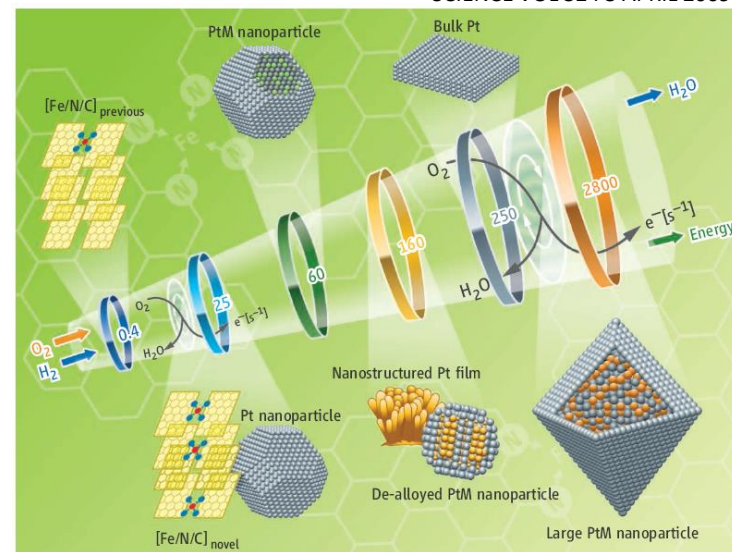
- What defines metal **electrocatalyst performance**:
 - double layer structure (what defines that...)

- nature of the metal
- surface facets (morphology)
- presence of defects, steps, kinks, ad-atoms (structure)
- presence of second metal (composition&structure)
- size
- support
- electrolyte (double layer)

Practically everything we can think of!

ligand and/or strain, morphology, confinement effect, surface segregation, proximity, ensemble, surface patterning, size, etc., effect

Gasteiger & Markovic
SCIENCE VOL 324 3 APRIL 2009



Proton exchange membrane fuel cell

- What defines metal **electrocatalyst performance**:
 - double layer structure (what defines that...)

- nature of the metal
- surface facets (morphology)
- presence of defects, steps, kinks, ad-atoms (structure)
- presence of second metal (composition&structure)
- size
- support
- electrolyte (double layer)

Practically everything we can think of!

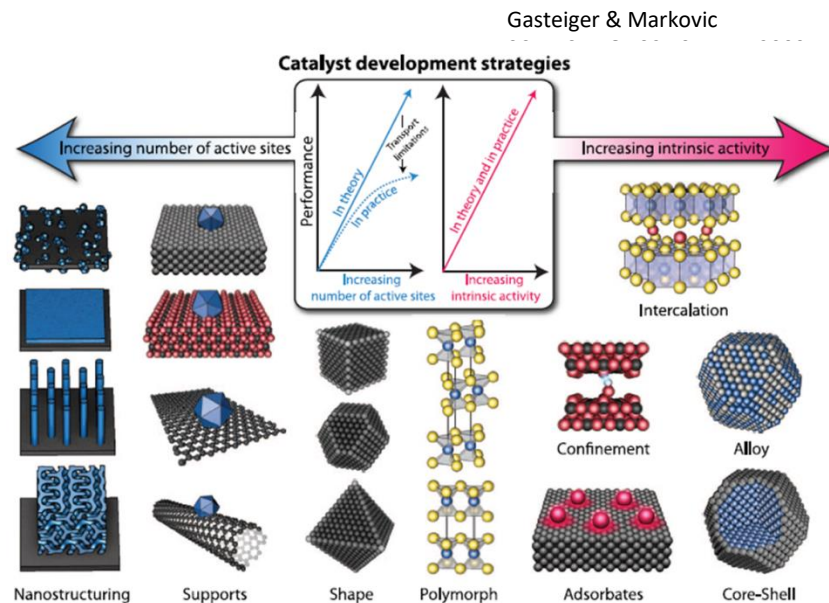
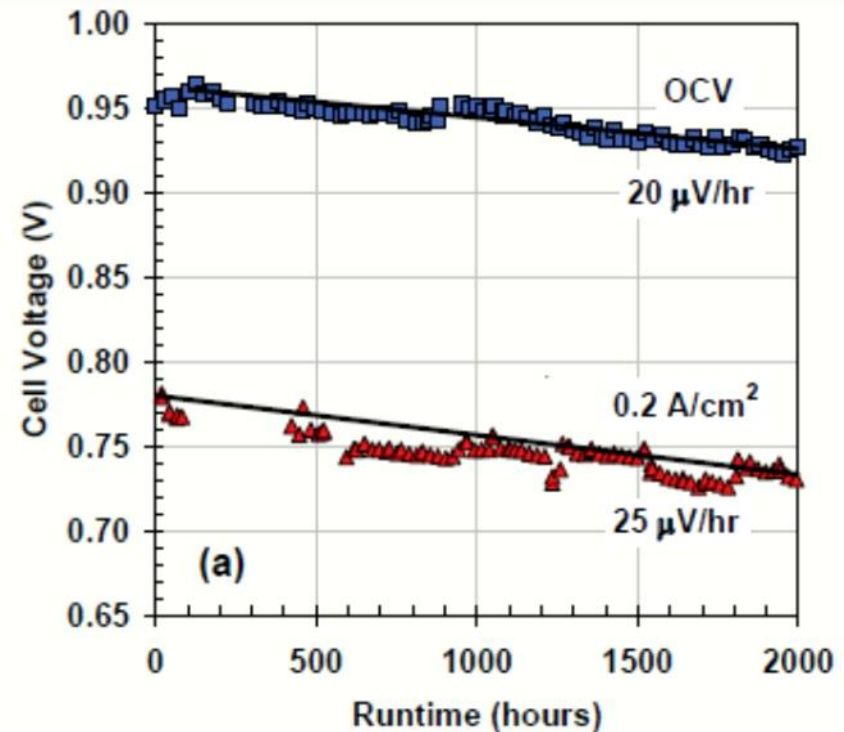
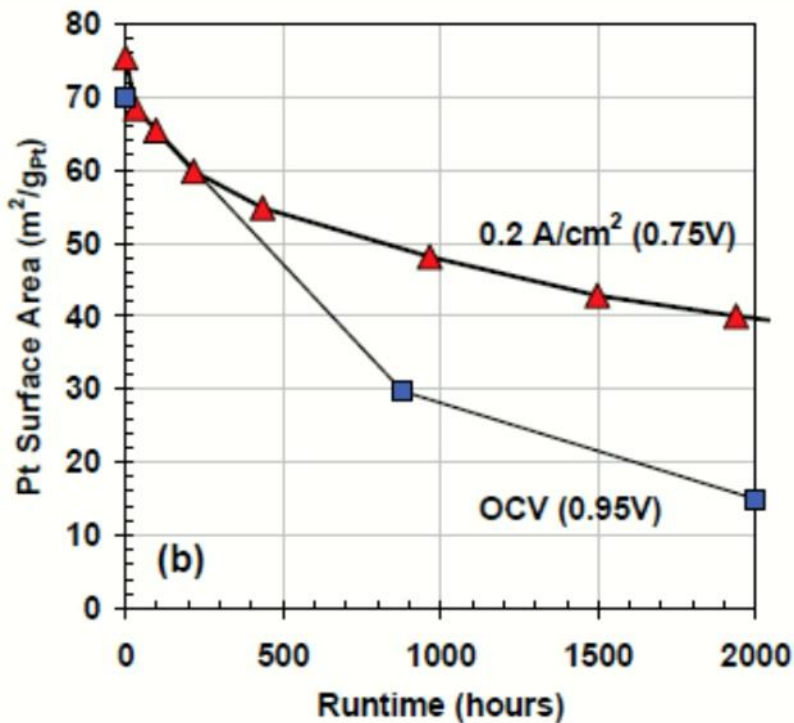


Fig. 2. Catalyst development strategies. Schematic of various catalyst development strategies, which aim to increase the number of active sites and/or increase the intrinsic activity of each active site.

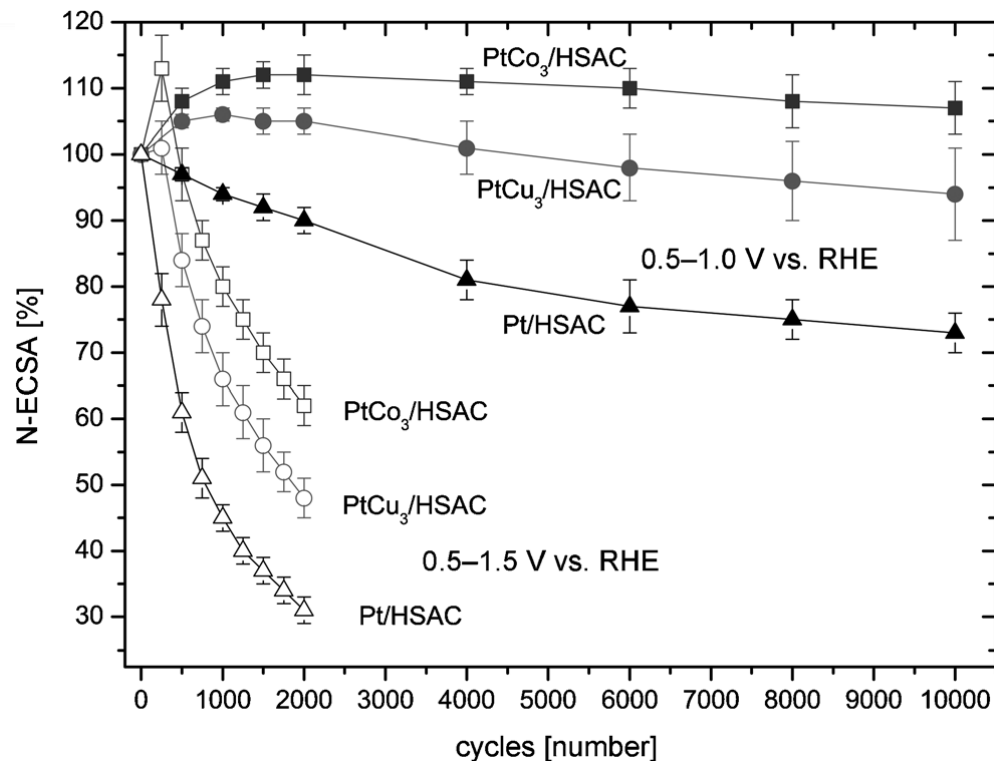
ligand and/or strain, morphology, confinement effect, surface segregation, proximity, ensemble, surface patterning, size, etc., effect

Proton exchange membrane fuel cell

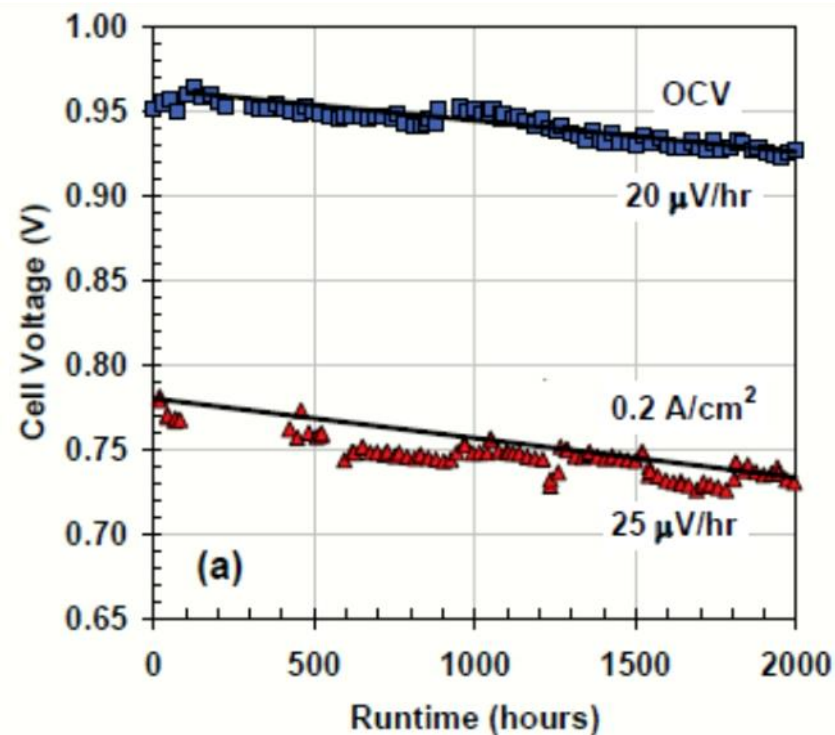


Is it possible to avoid this – we must truly understand fundamentals first → ex-situ tests

Proton exchange membrane fuel cell



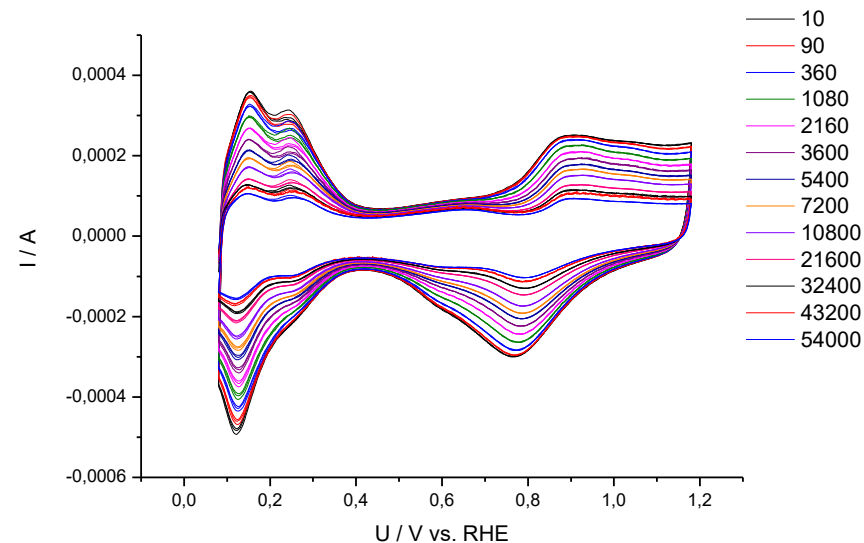
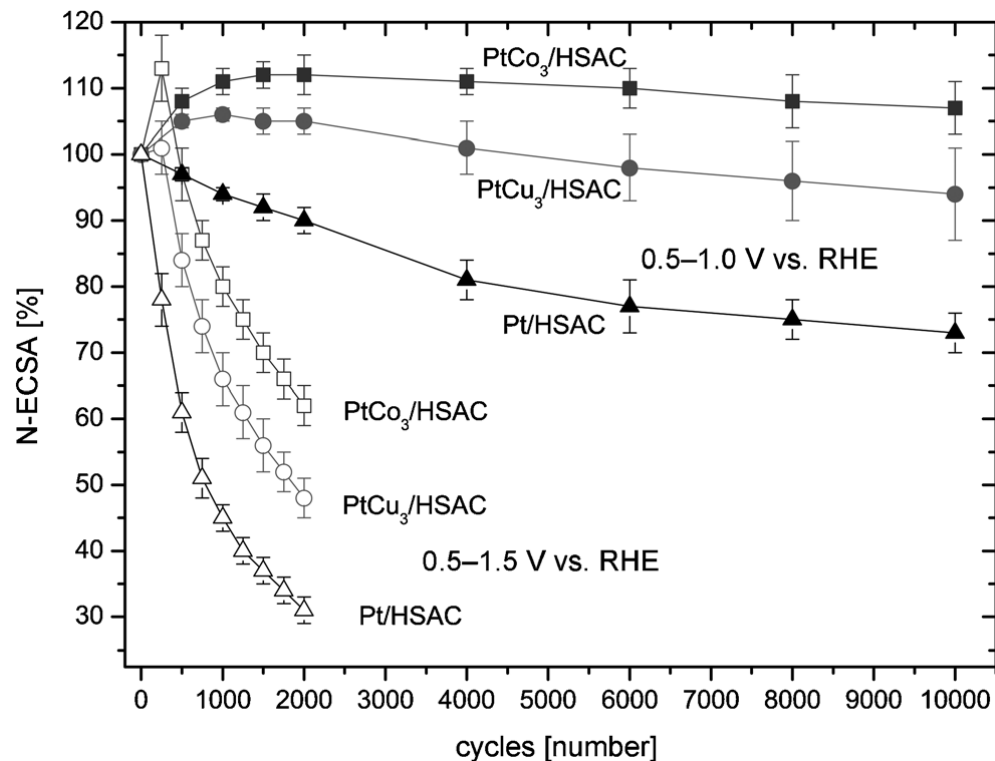
Frédéric Hasché, Activity, *ChemCatChem*, 3, (2011) 1805



Is it possible to avoid this – we must truly understand fundamentals first → ex-situ tests

Ferreira, *JES*, 2005, 152

Proton exchange membrane fuel cell



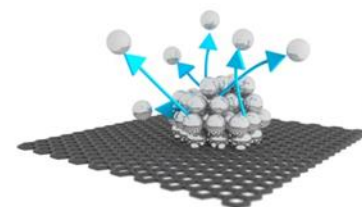
Frédéric Hasché, Activity, *ChemCatChem*, 3, (2011) 1805

Is it possible to avoid this – we must truly understand fundamentals first → ex-situ tests

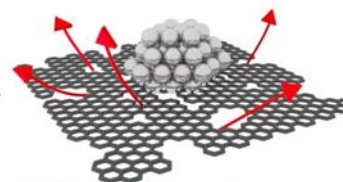
Ferreira, *JES*, 2005, 152

• Topic 2: Degradation studies

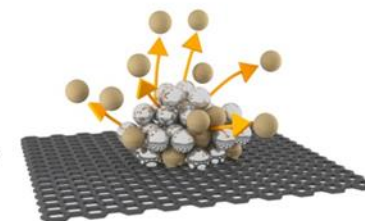
Mechanisms for the loss of active surface area:



Platinum Dissolution

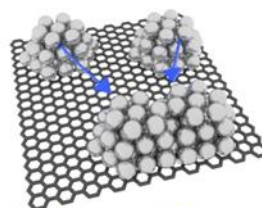


Carbon Corrosion

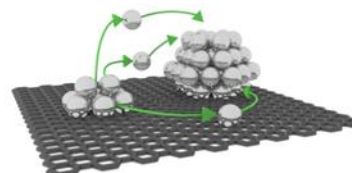


Dealloying

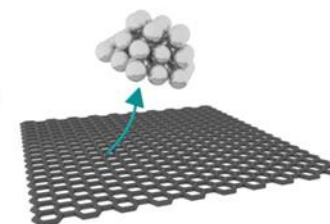
Most studies performed on Pt or PGM at low pH – Fuel Cell.



Agglomeration

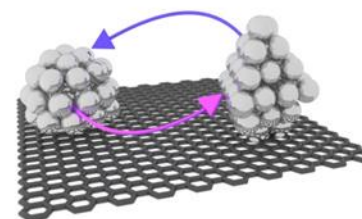


Ostwald Ripening



Particle Detachment

Degradation at oxidative potentials!

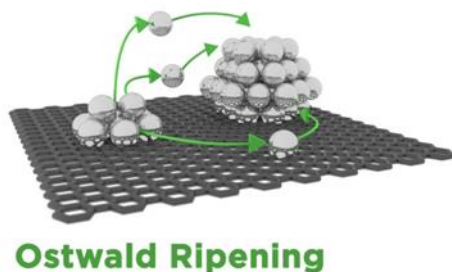


Reshaping

Hodnik, et al; *Acc. Chem. Res.*, **2016**, 49 (9), pp 2015–2022

How to distinguish between them?

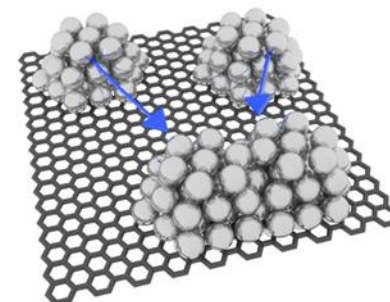
Postmortem analysis?
No!



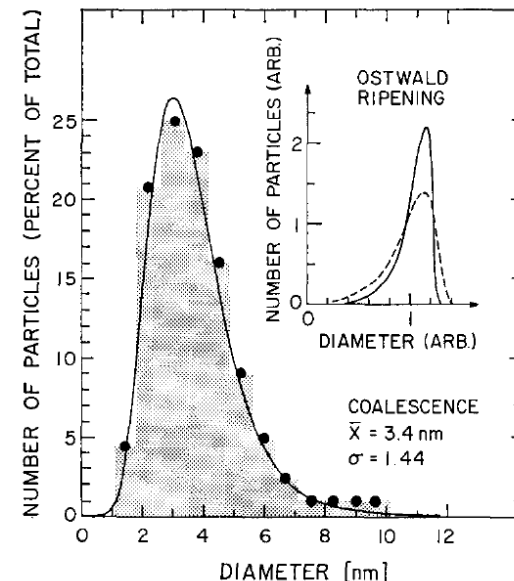
Ostwald Ripening

Bulk methods like particle size distribution (PSD) obtained by analyzed TEM images and XRD offer only averaged information: It was shown that PSDs with **tails to larger particle sizes** are associated with particle growth via **migration and coalescence**. In contrast, a PSD indicative of growth through electrochemical **Ostwald ripening** involves a peak toward large particle sizes with **tailing to smaller sizes**.

Problem: the simple models assume that only one growth mode occurs! – not true!



Agglomeration

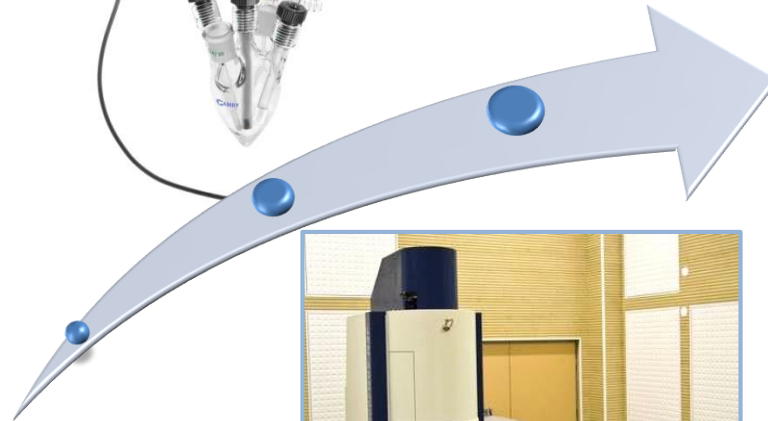
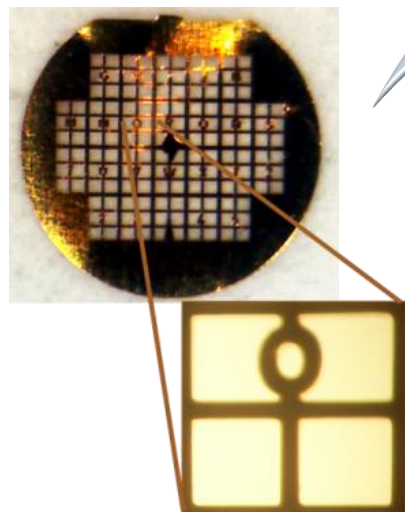


Simple principle: *Look at the grid in the TEM, take it out and perform the electrochemistry, and again look at the grid in the TEM!*

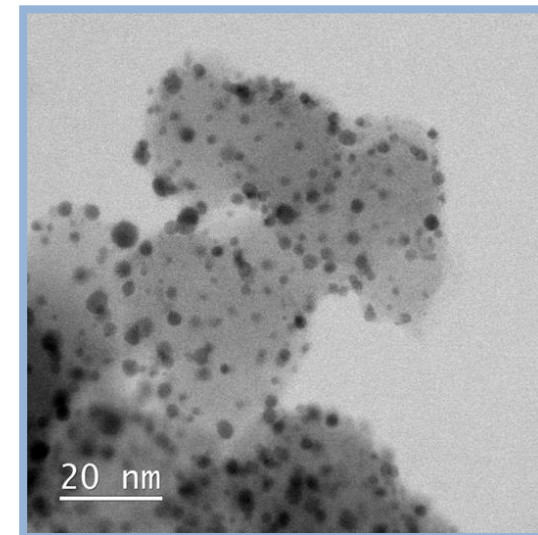
Electrochemical cell



The Gold finder TEM grid allows to retrieve the same spot throughout the electrochemical treatments.

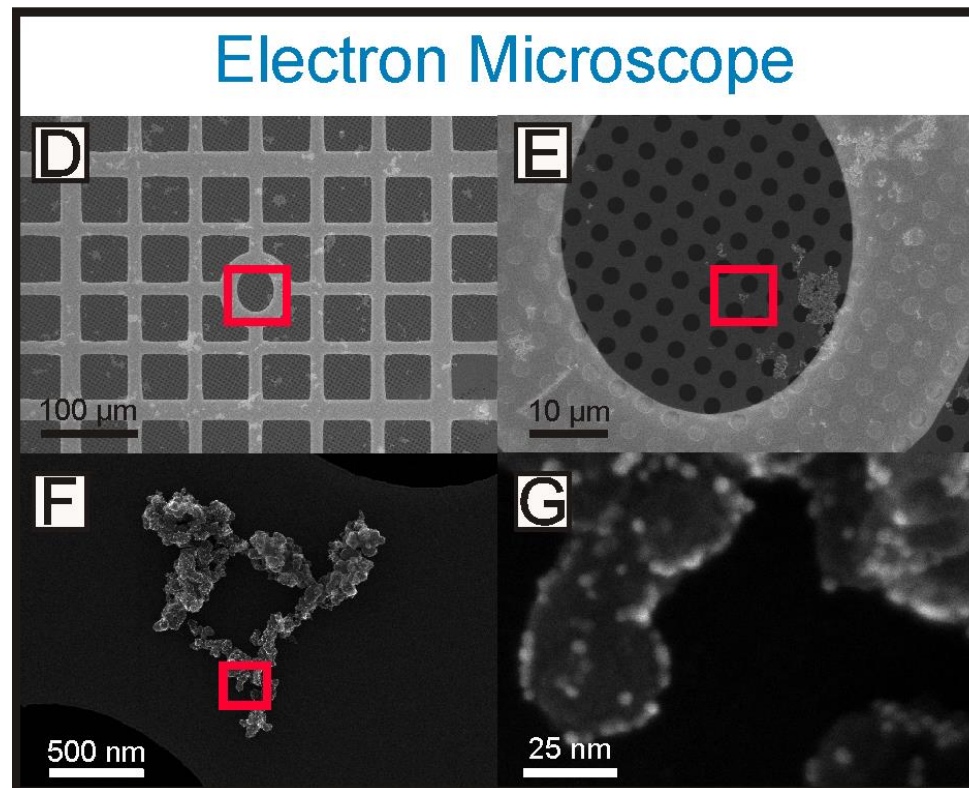
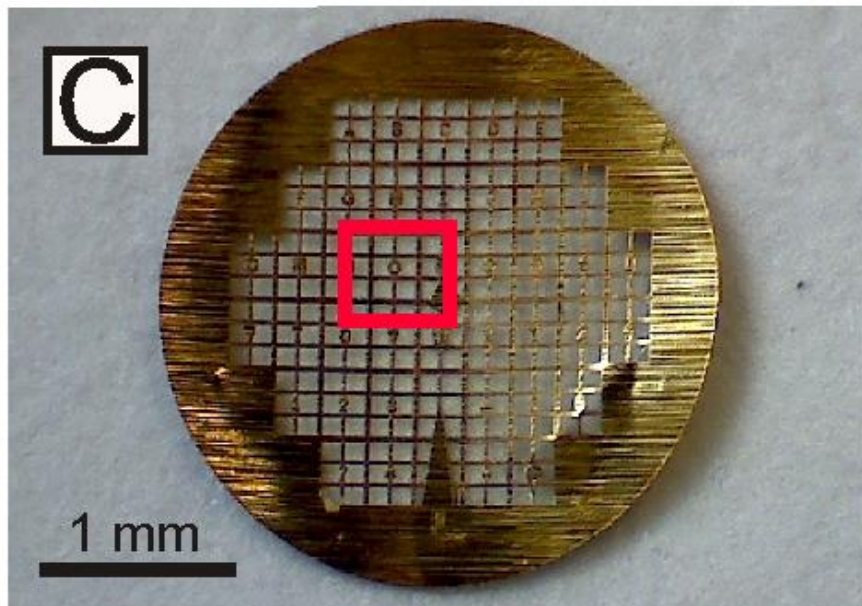


Atomic resolution STEM -
JEOL ARM 200 CF



Such approach has been proven extremely useful in providing a direct link between macroscopic and microscopic behavior

Mayrhofer, K., et al; *Electrochem. Commun.* **2008**, 10, 1144-1147.



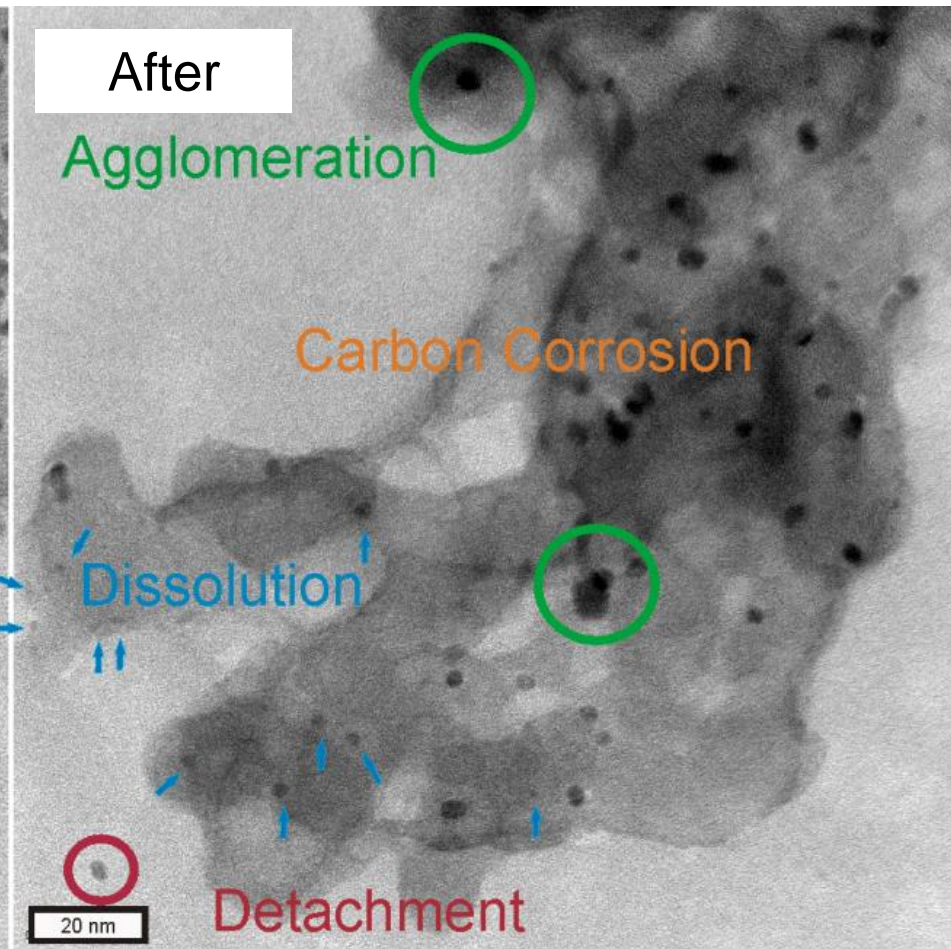
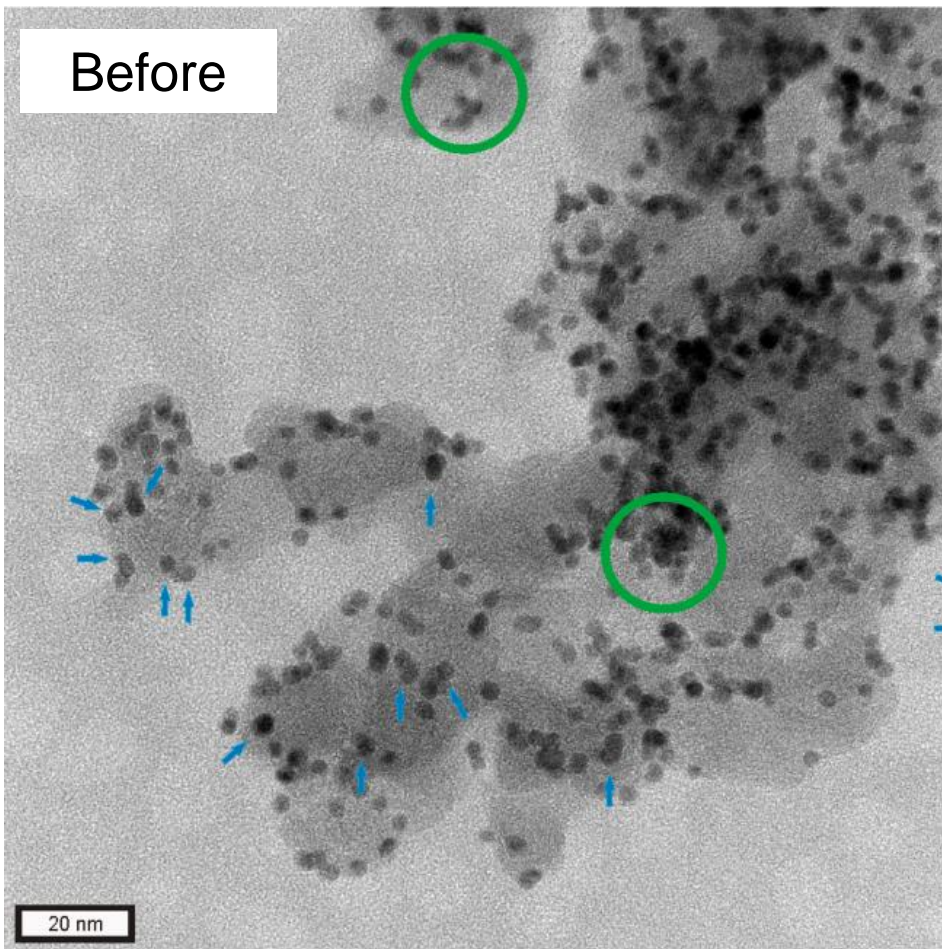
- Utilizing a TEM finder grid as a WE
 - Conducting, but inert Au

- Tracking of the catalyst particles
 - Alphanumerical signs for orientation

Pt/Vulcan
(20 wt.%, 3.6 nm)

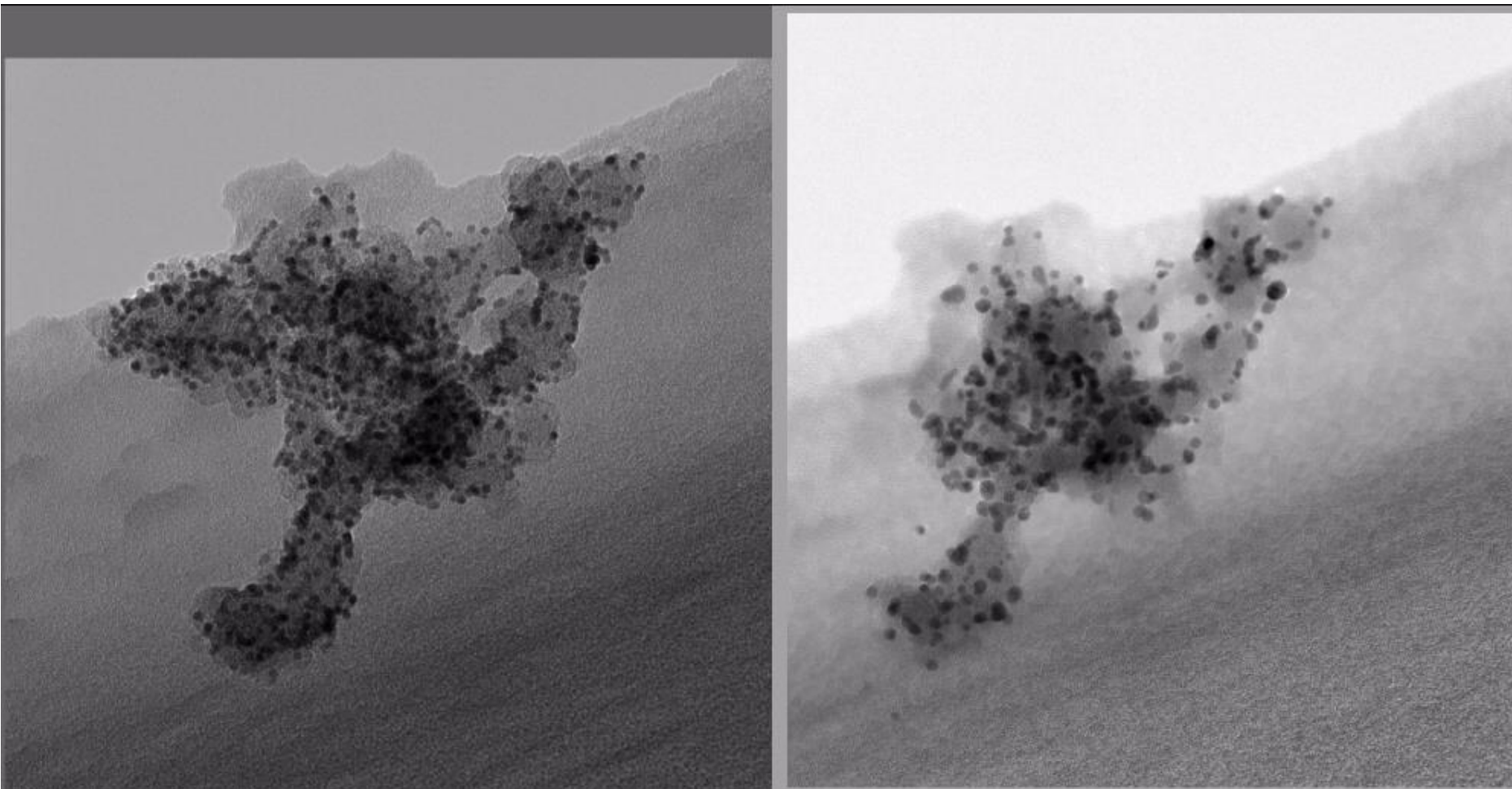


Overlapping degradation mechanisms – material dependant



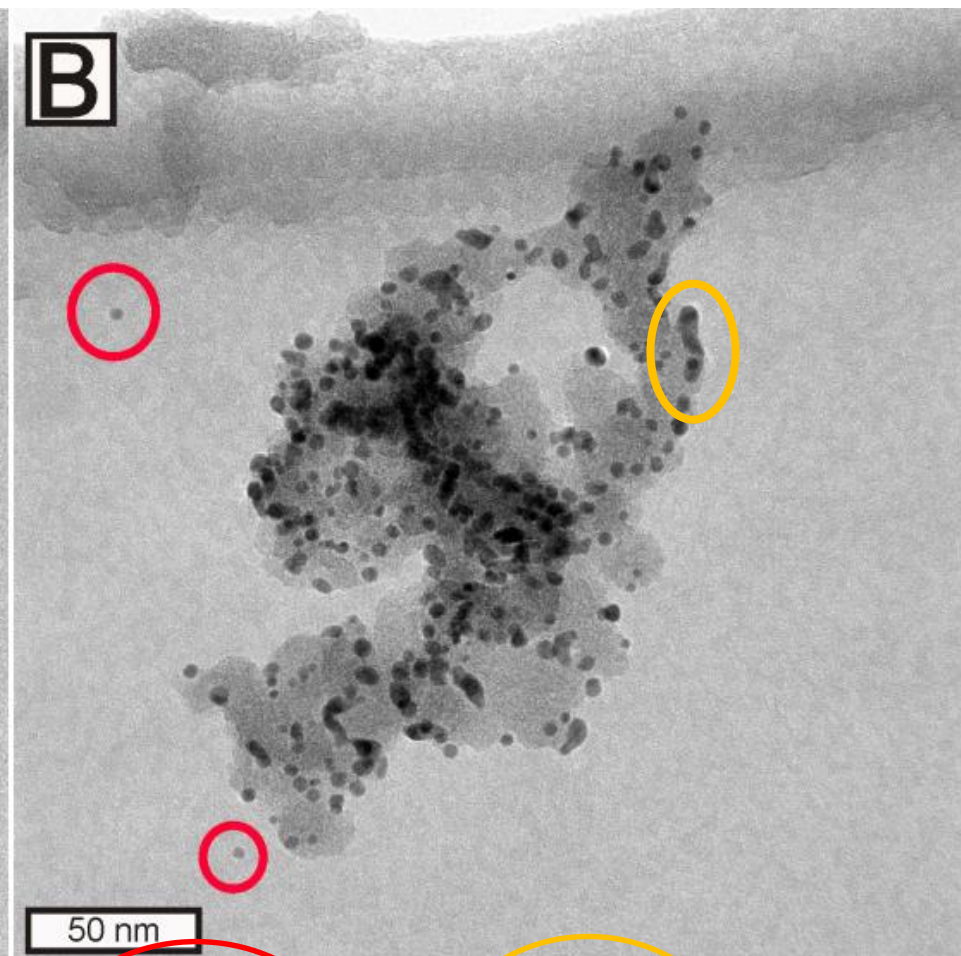
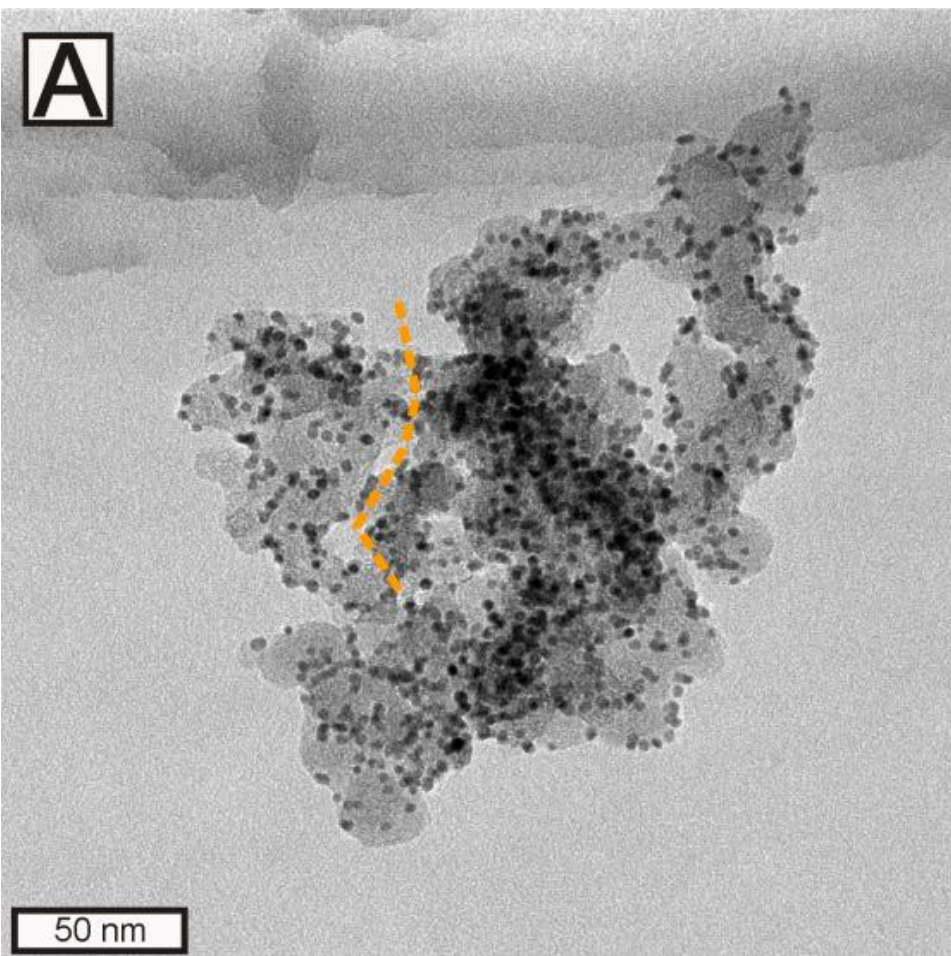
3600 potential cycles, 0.4 – 1.4 V_{RHE}, 0.1 M HClO₄, 1 Vs⁻¹

Meier, *ACS Catalysis*, **2012**, 2, 832-843.



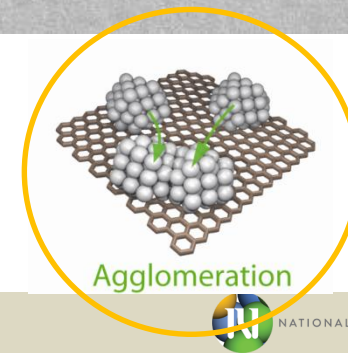
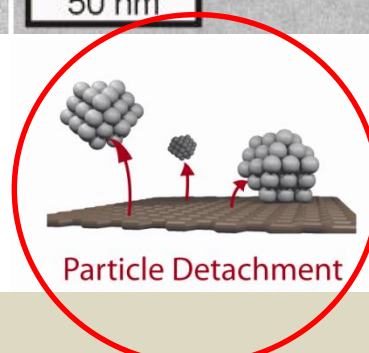
5000 degradation cycles between 0.4 and 1.4 VRHE in 0.1 M HClO₄ at a scan rate of 1 V s⁻¹ at room temperature.

Meier, *ACS Catalysis*, **2012**, 2, 832-843.



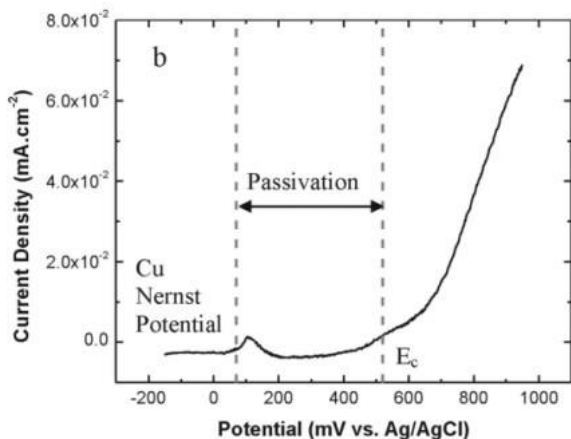
5000 degradation cycles between 0.4 and 1.4 VRHE in 0.1 M HClO₄ at a scan rate of 1 V s⁻¹ at room temperature.

Meier, *ACS Catalysis*, **2012**, 2, 832-843.

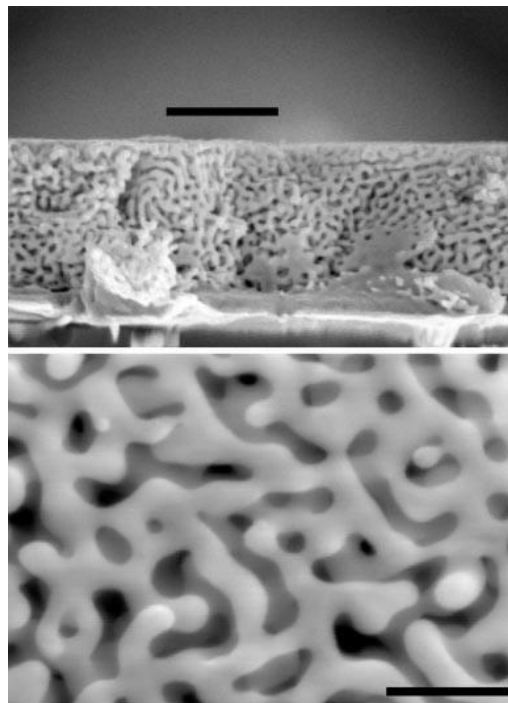


Dealloying is selective removal of less noble metal.

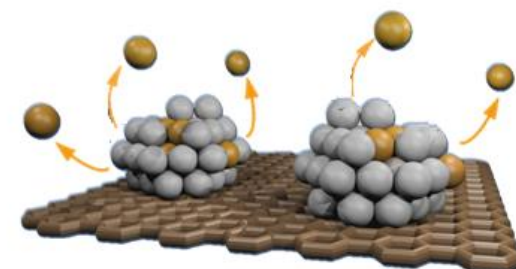
”It is a competition between surface diffusion of noble metal (Pt) and tendency of less noble metal (Cu) to dissolve.”



R. Yang et al, J. Phys. Chem. C
115 (2011) 9074–9080.



→ pore formation



Dealloying

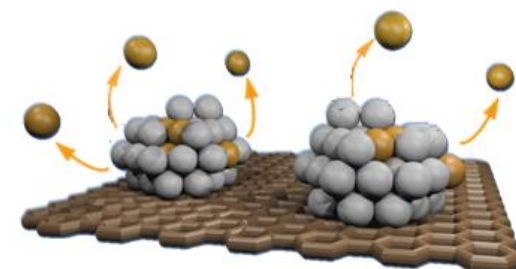
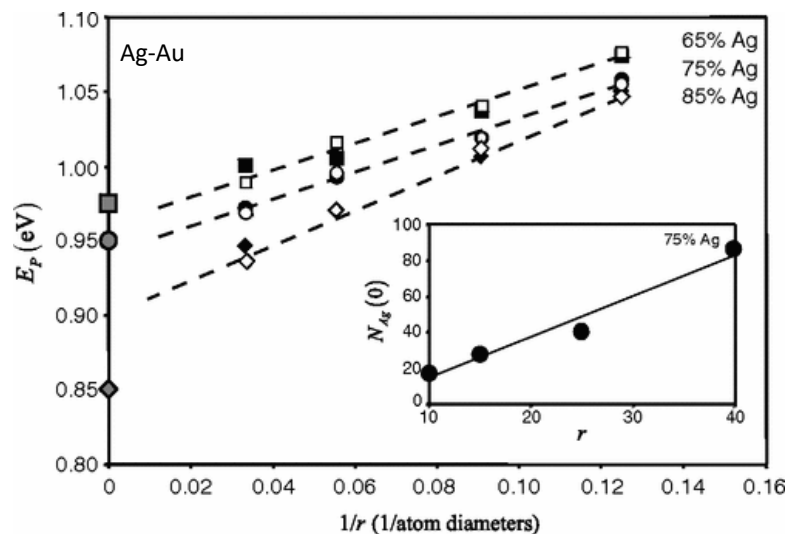


J. Erlebacher, M. J. Aziz, A. Karma, N. Dimitrov, and K. Sieradzki, Nature (London) 410, 450 (2001).

Dealloying is particle size dependant!

The Gibbs-Thomson effect: with $1/r$ there is increased mobility of surface atoms, the surface of the particle quickly passivates

Model:

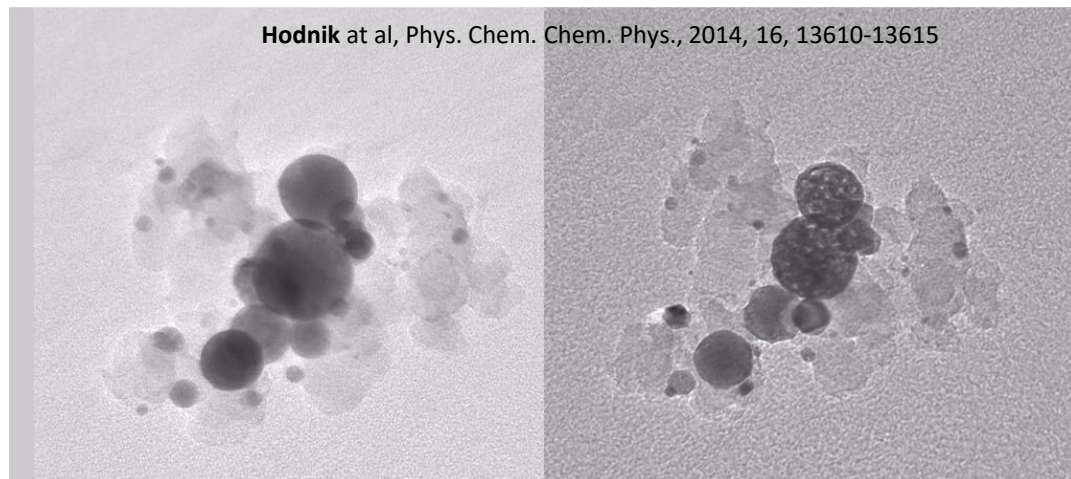


Dealloying

Simulation and model data for the potential for porosity evolution E_P versus of $1/r$.

I. McCue, J. Snyder, X. Li, Q. Chen, K. Sieradzki, and J. Erlebacher Phys. Rev. Lett. 108, 225503 – Published 31 May 2012

Pot. Hold @ 1.2 V PtCu₃



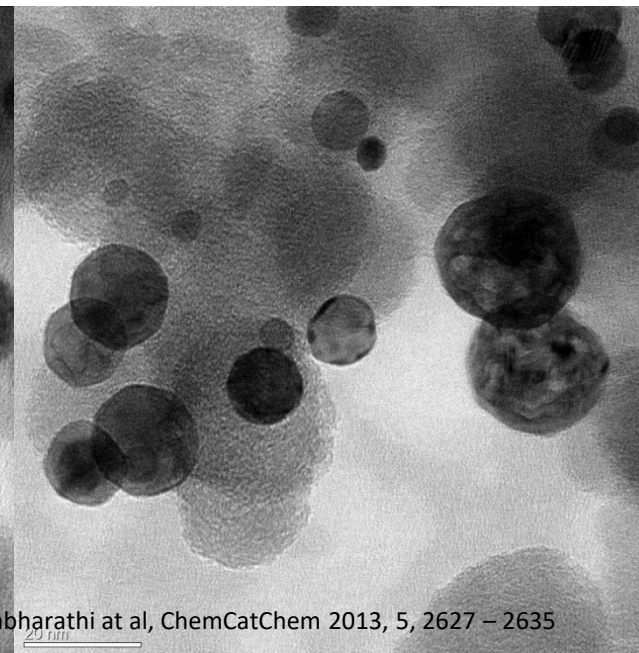
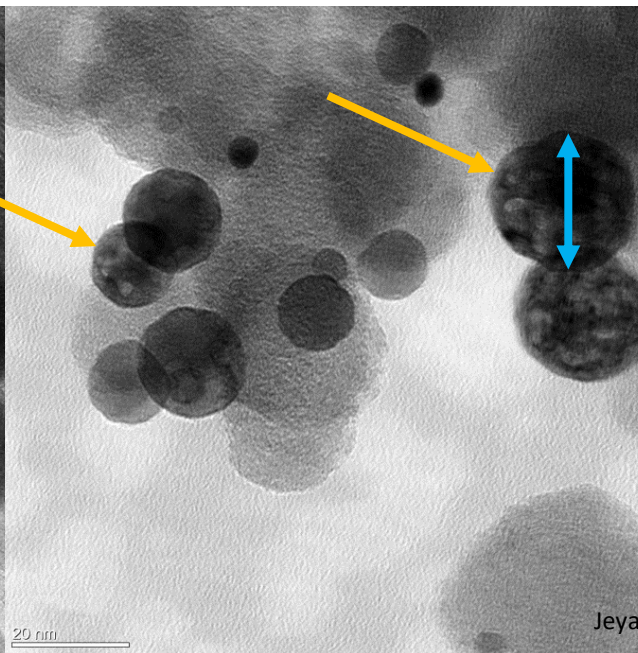
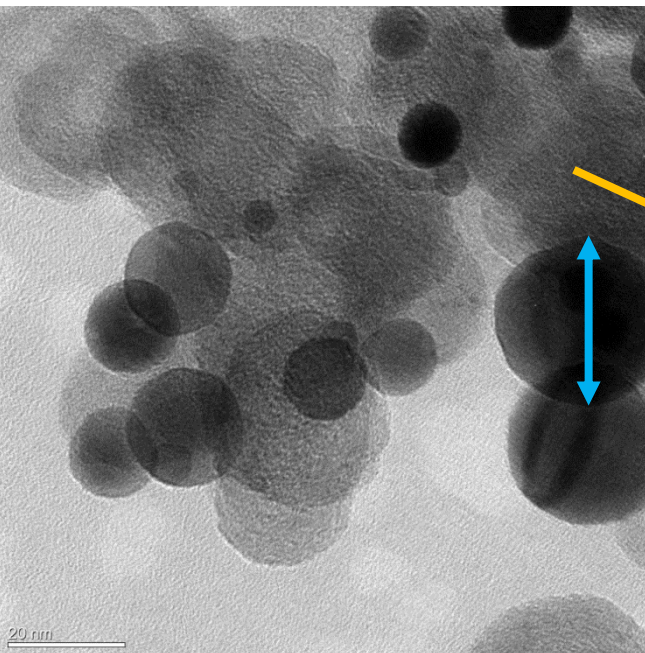
Hodnik* et al. *PCCP* (2014) 16(27), pp. 13610-13615

Before dealloying

~60-75 % of Cu removed

Hold at 1.2V for 2 Hrs

Cycling: 0.6 to 1.2 V,
scan rate=1 V/s; 7000 cycles

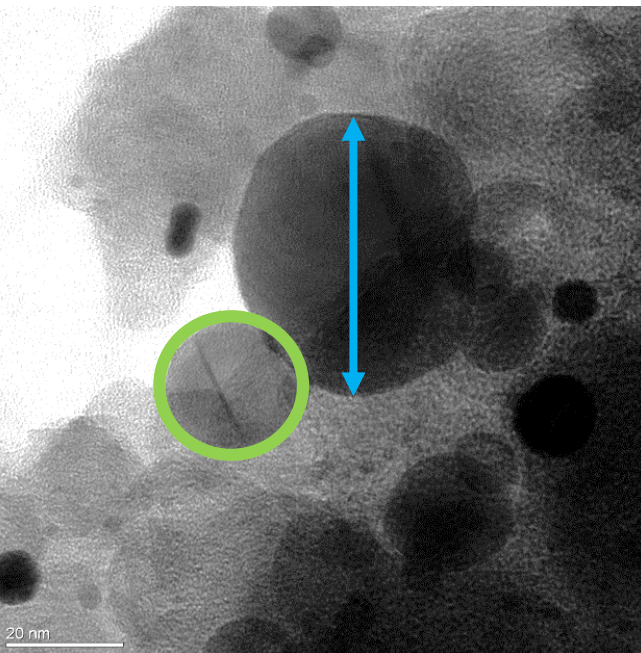


Jeyabharathi et al, *ChemCatChem* 2013, 5, 2627 – 2635

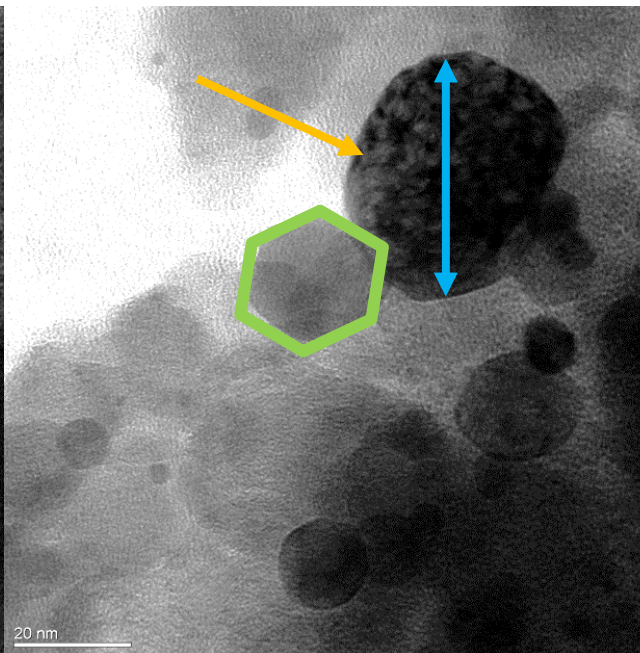
Cyc. Up to 1.2 V PtCu₃

Hodnik* et al. *PCCP* (2014) 16(27), pp. 13610-13615

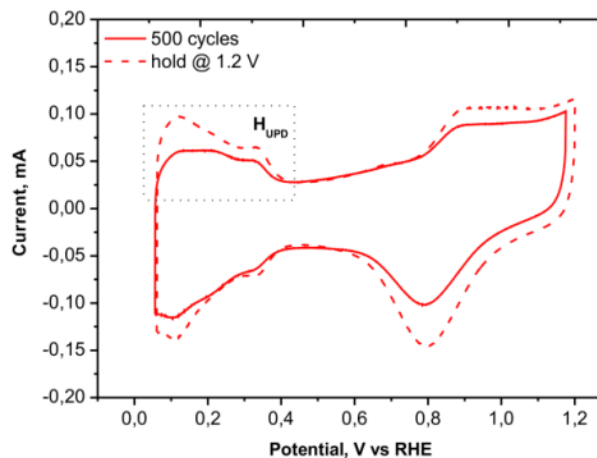
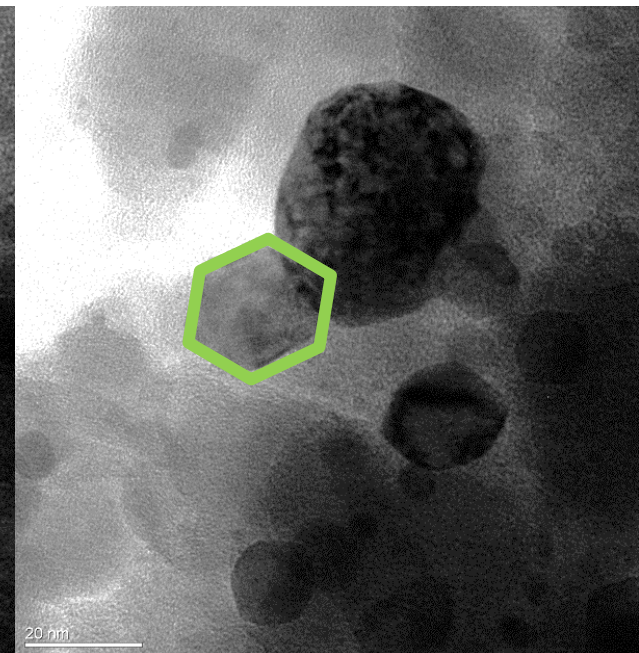
Before dealloying



Cycling: 0.05 to 1.2 V, scan rate=0.5 V/s; 500 cycles (2 Hrs)



Cycling: 0.6 to 1.2 V, scan rate=1 V/s; 7000 cycles

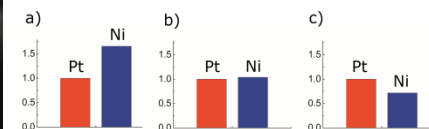
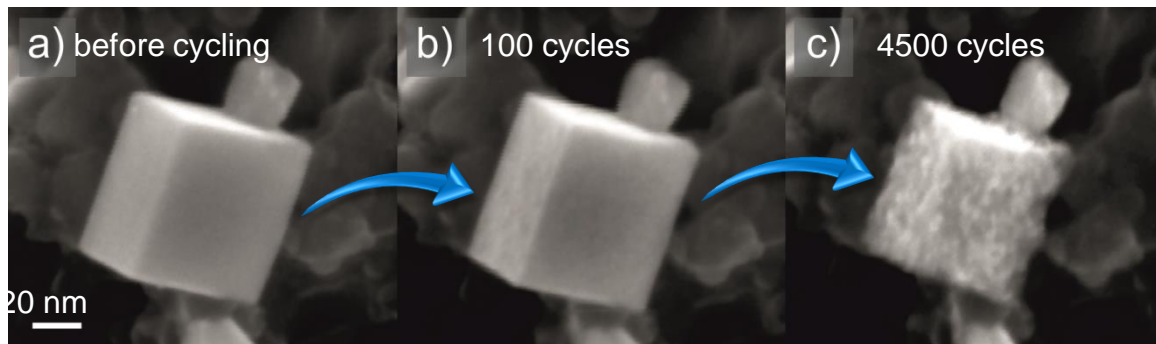


- Cubo Octahedron
- Truncated Octahedron
- Cube
- Tetrahexahedral

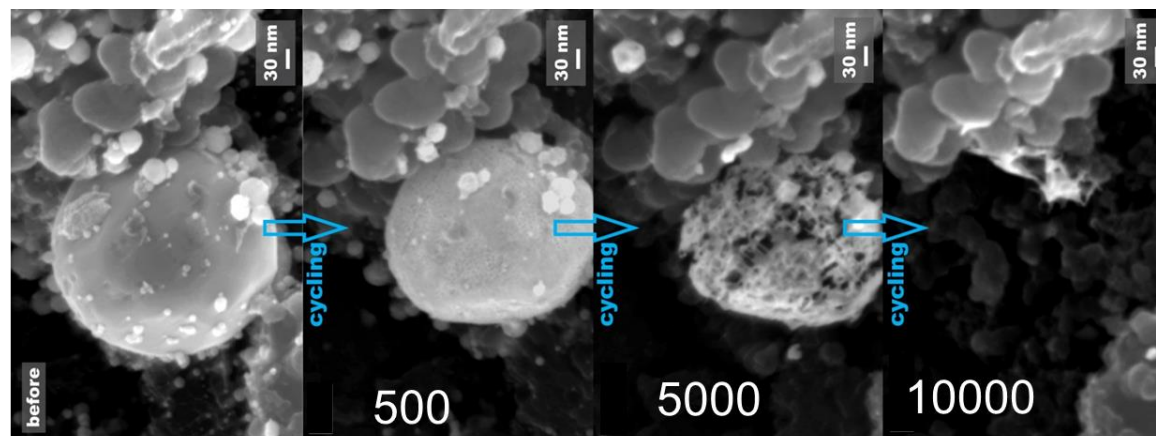
Na Tian, *et al Science* 316, 732 (2007)

-0.10 and -0.20 V to 1.20 V vs. SCE

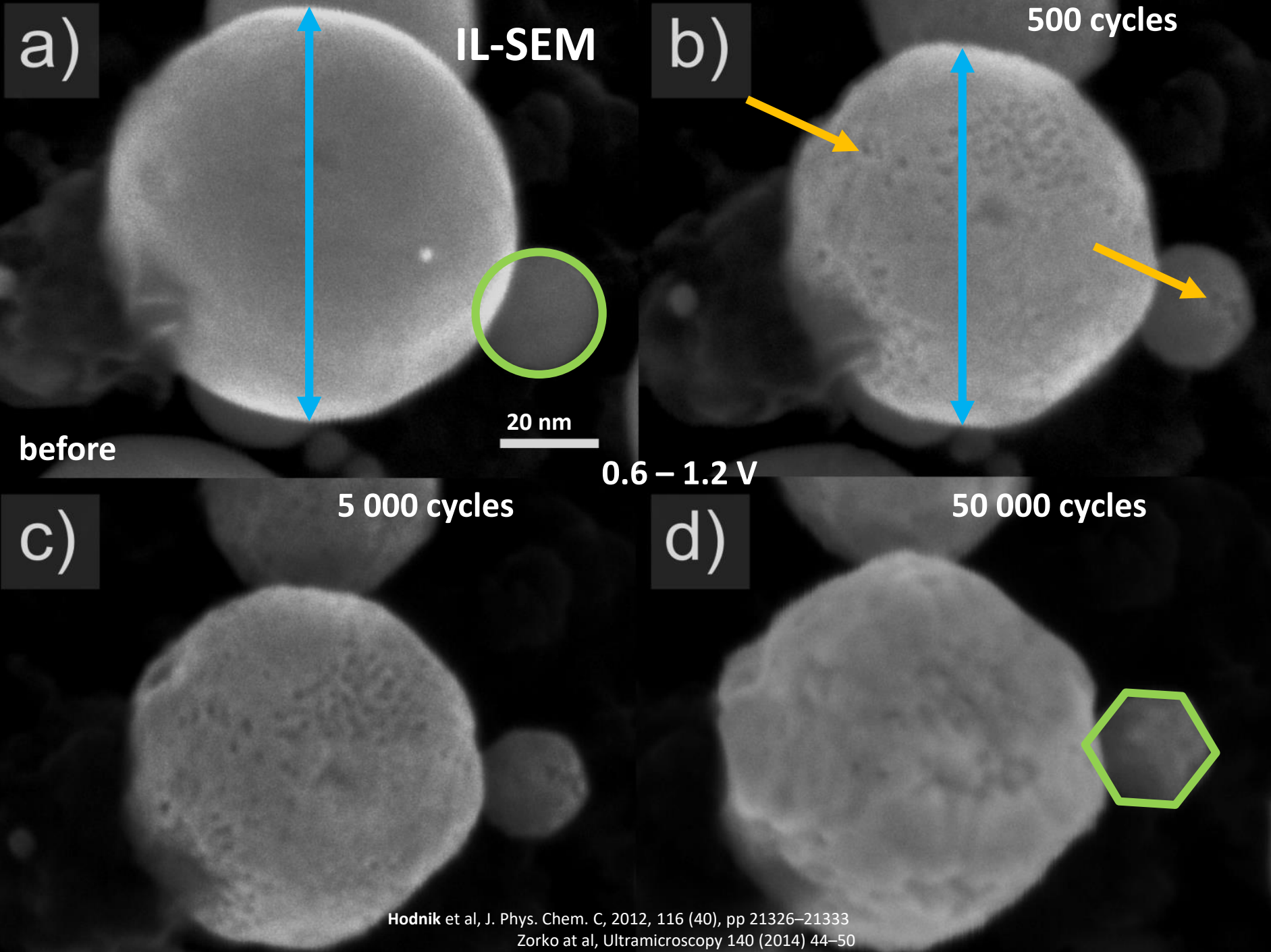
NiPt

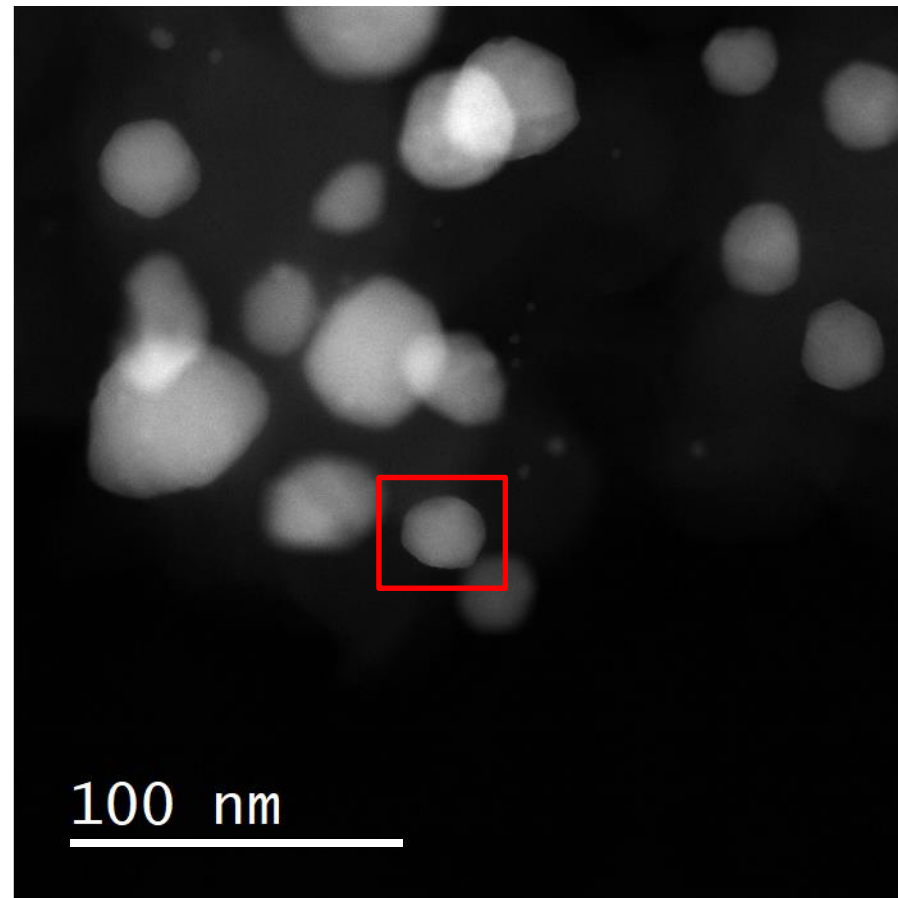
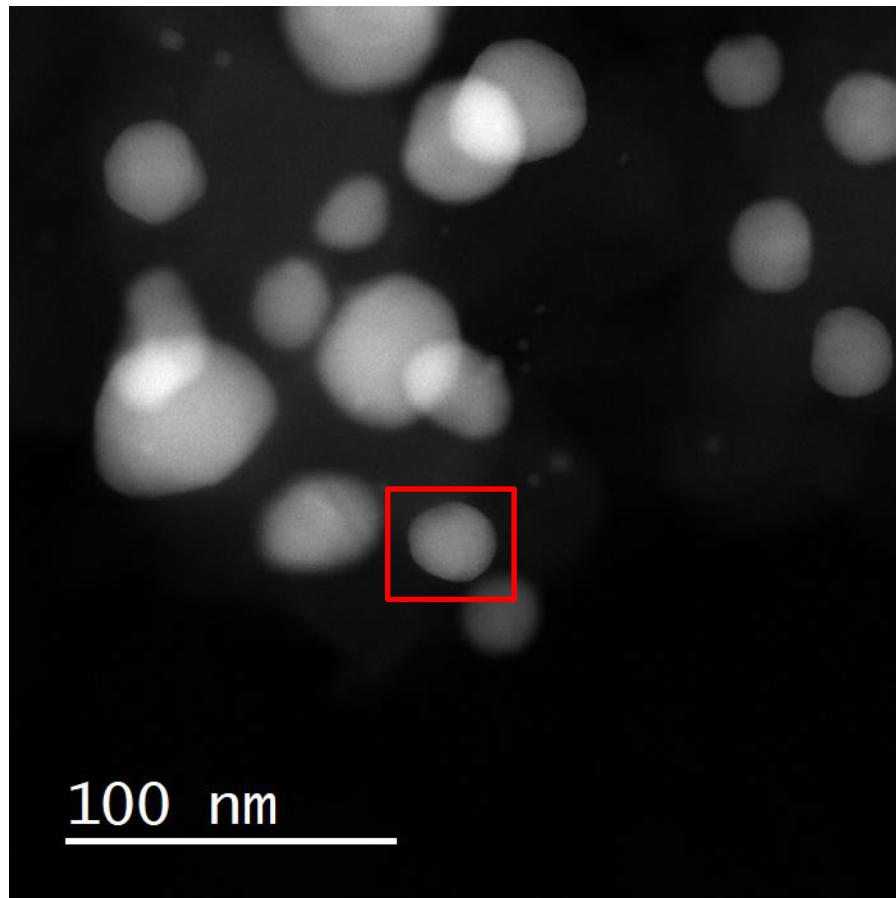


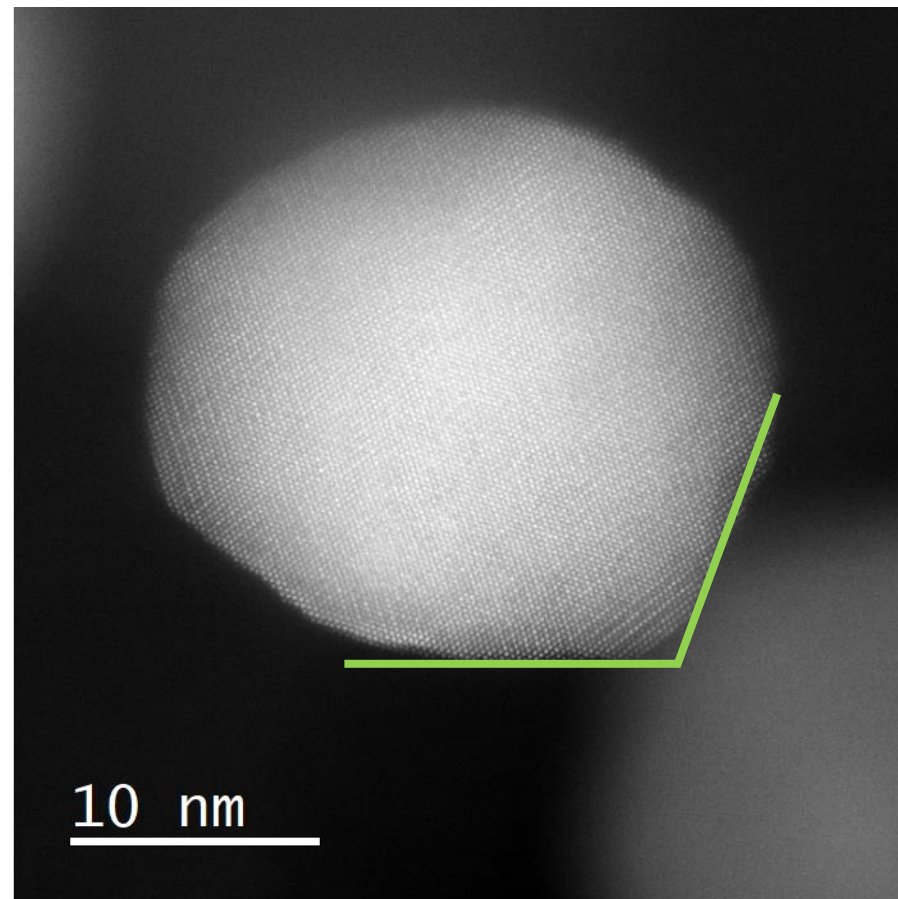
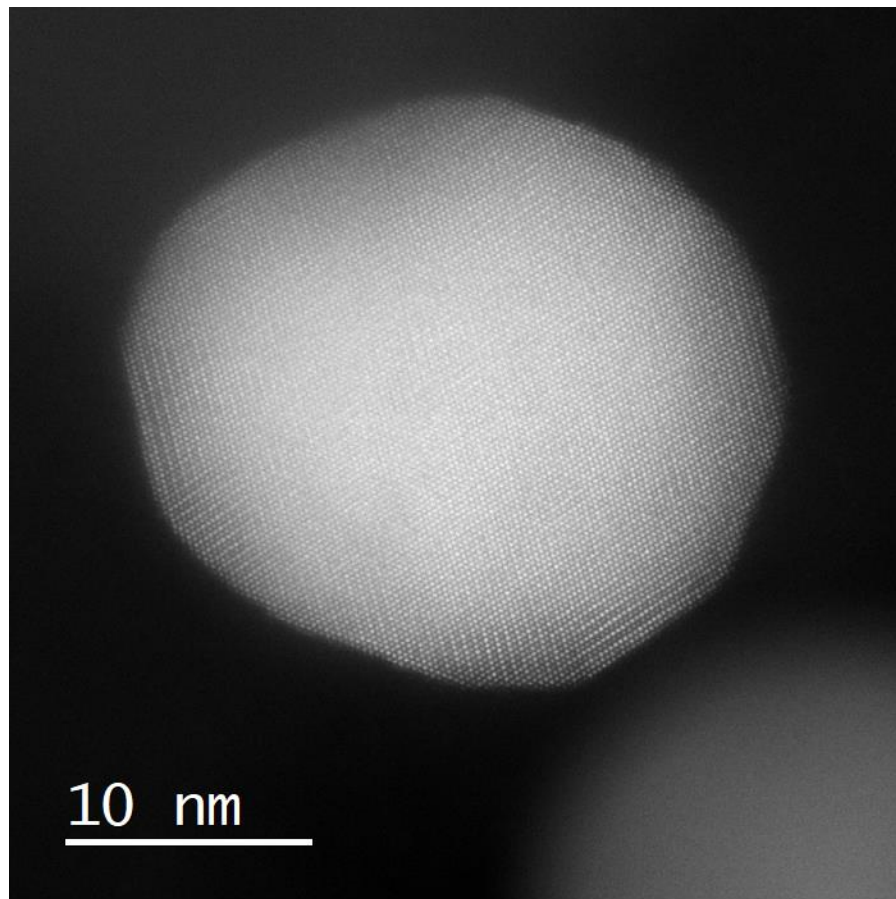
IL-SEM-EDX

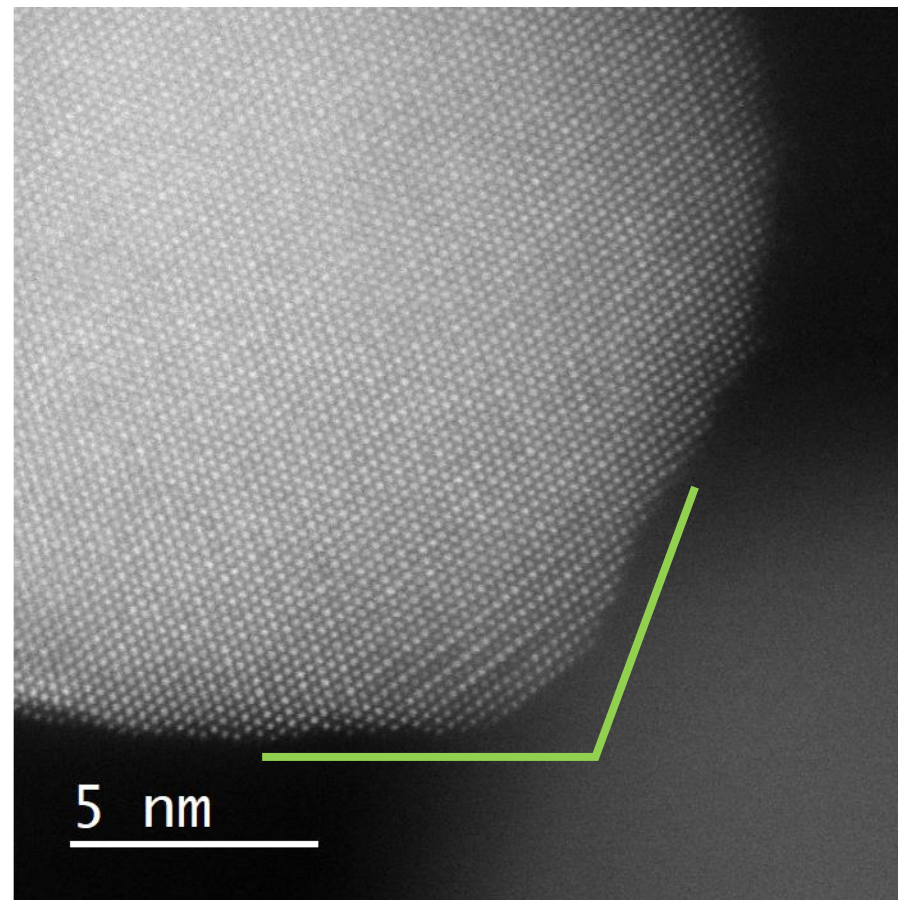
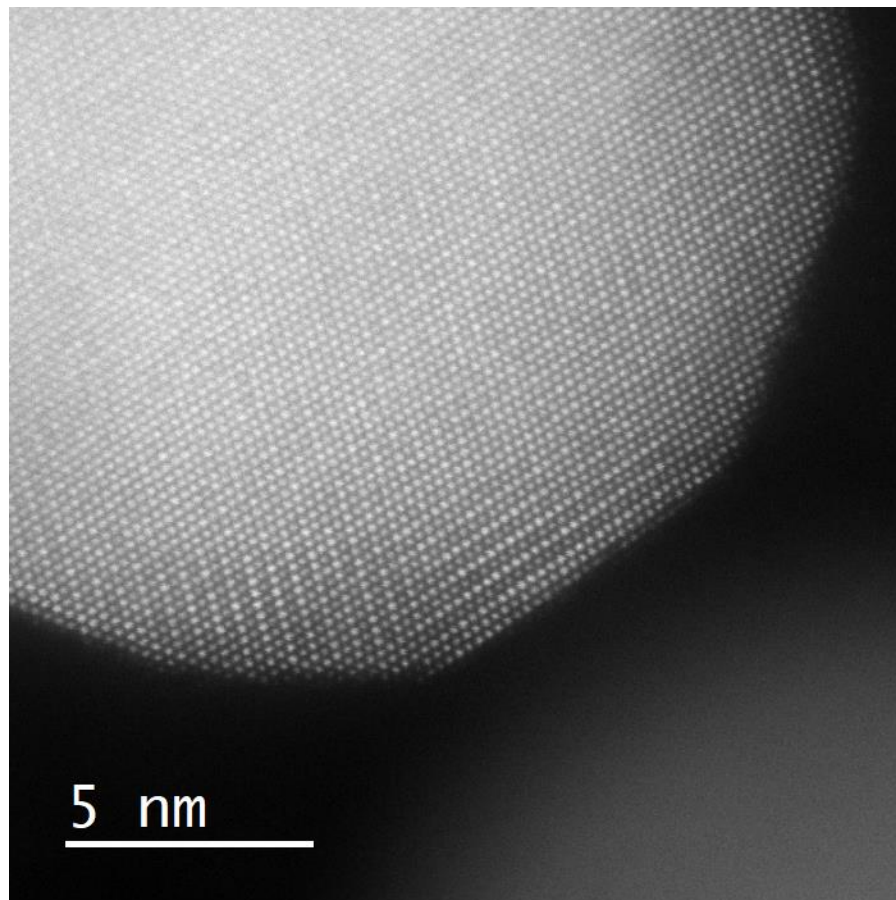
 Cu₃Pt

 SEM Microscope (HI-resolution):
 FE-SEM Zeiss SUPRA 35VP


- Utilizing versatile SEM graphite holdes or any other conductive sample
- Enabels visual tracking of nanoscale morphological changes upon any treatment



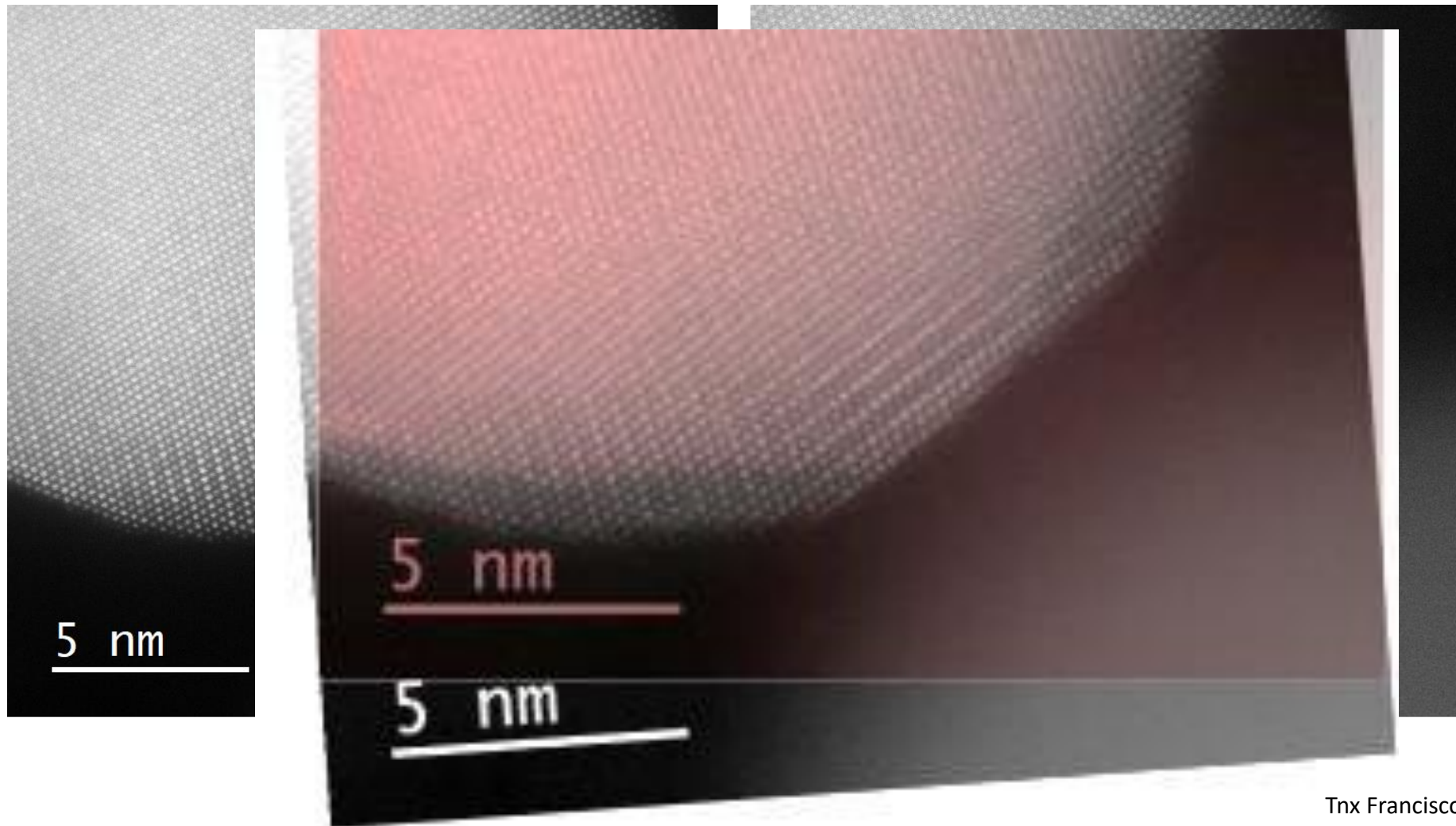






Tnx Francisco

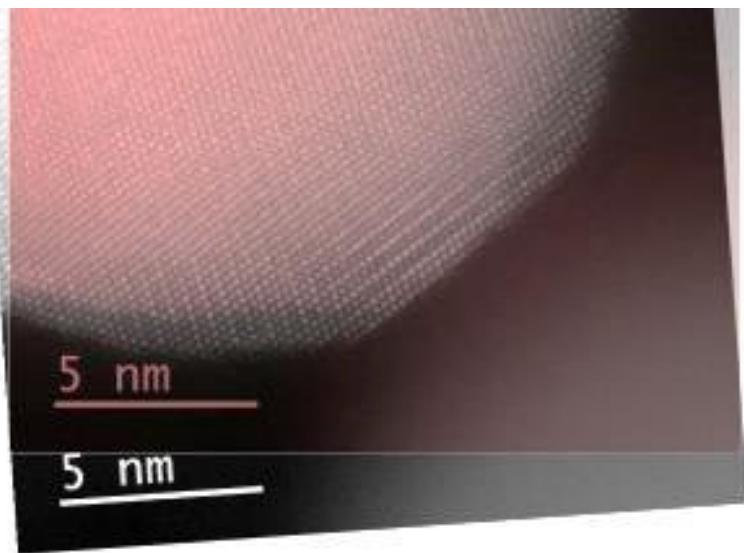
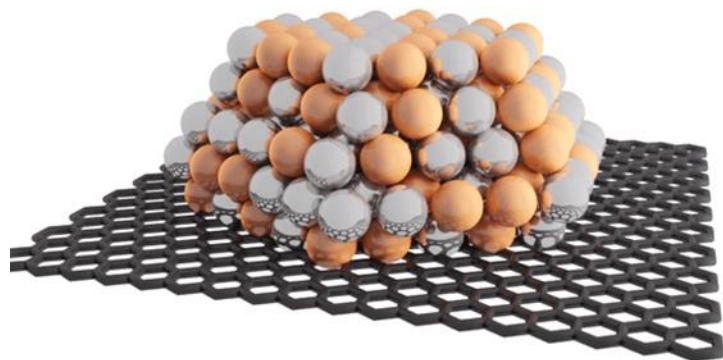
unpublished



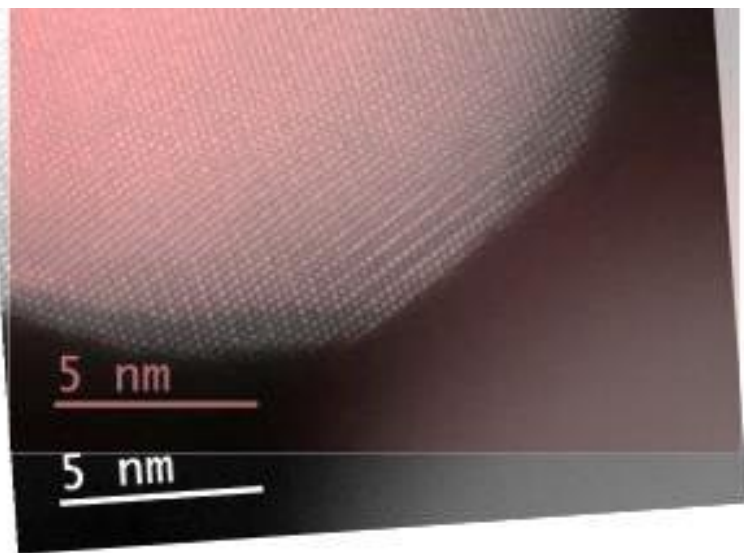
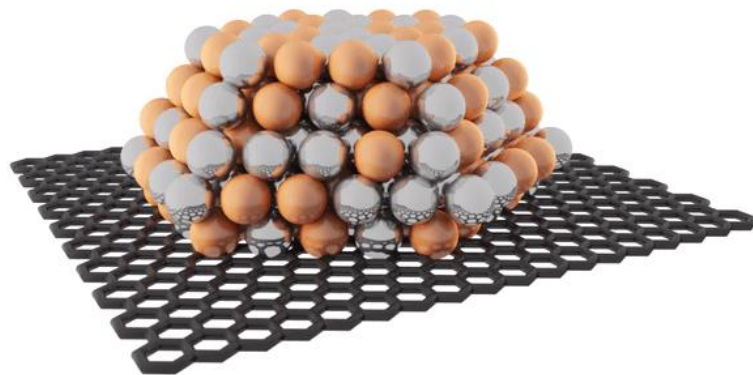
Tnx Francisco

unpublished

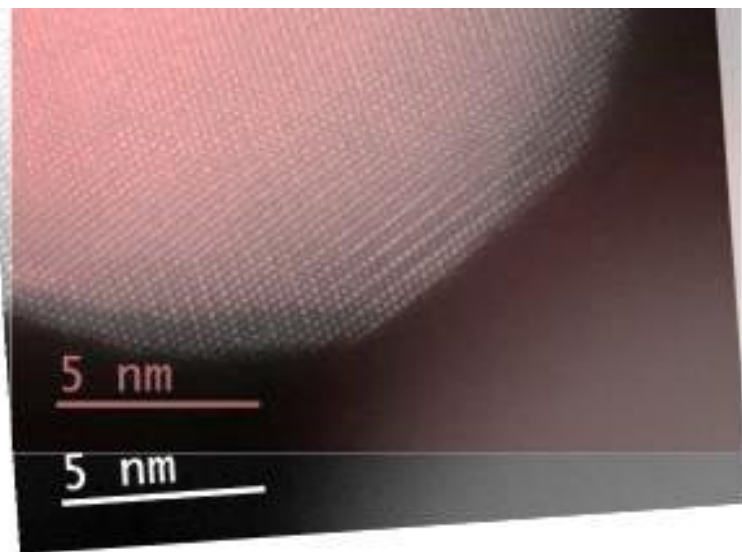
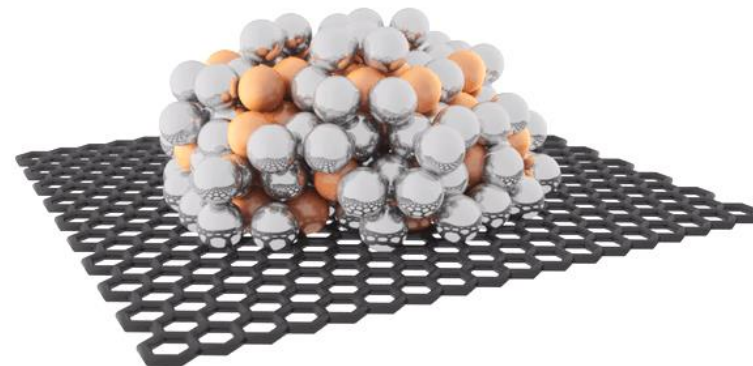
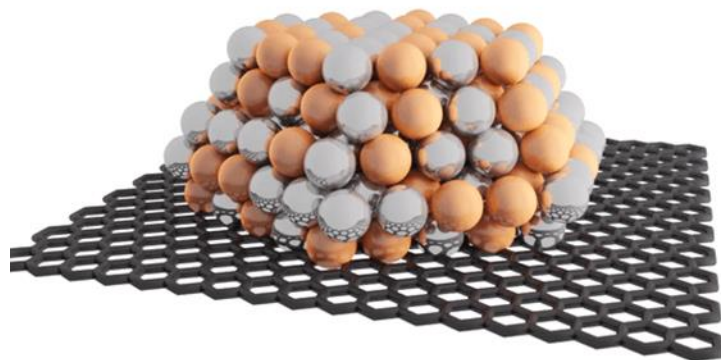
Atomic-scale atoms diffusion of NM nanoparticles changes in liquid environments



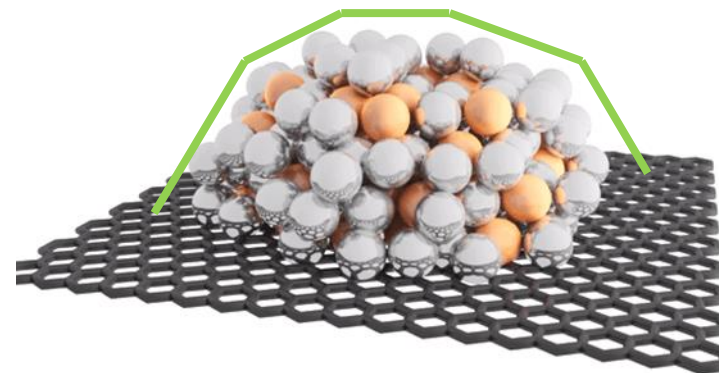
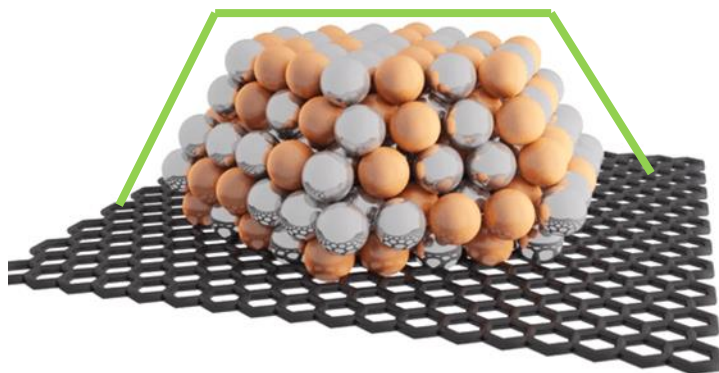
Atomic-scale atoms diffusion of NM nanoparticles changes in liquid environments



Atomic-scale atoms diffusion of NM nanoparticles changes in liquid environments

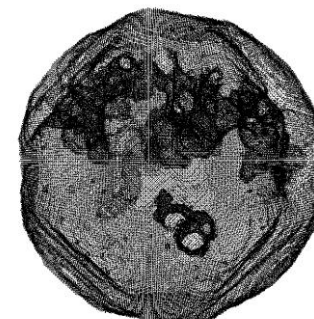
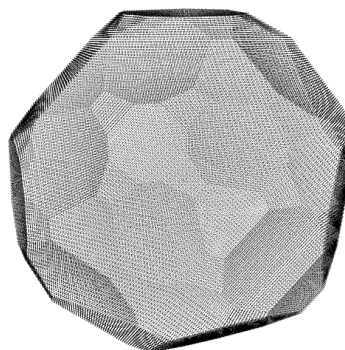
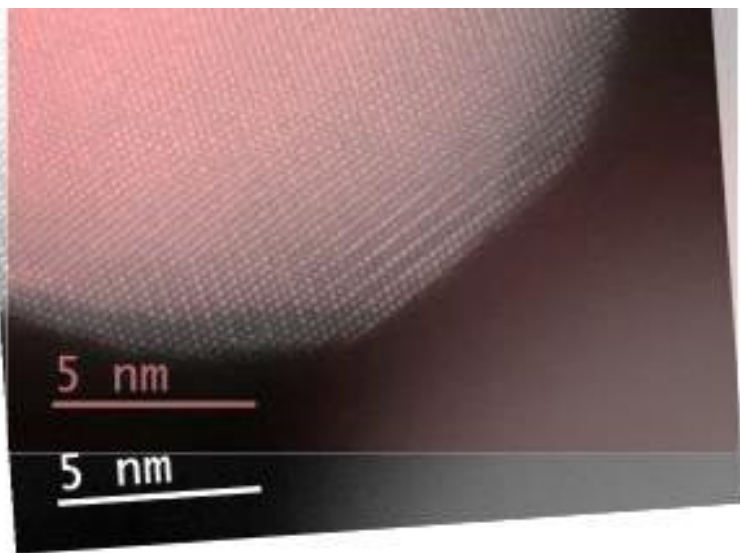


Atomic-scale atoms diffusion of NM nanoparticles changes in liquid environments



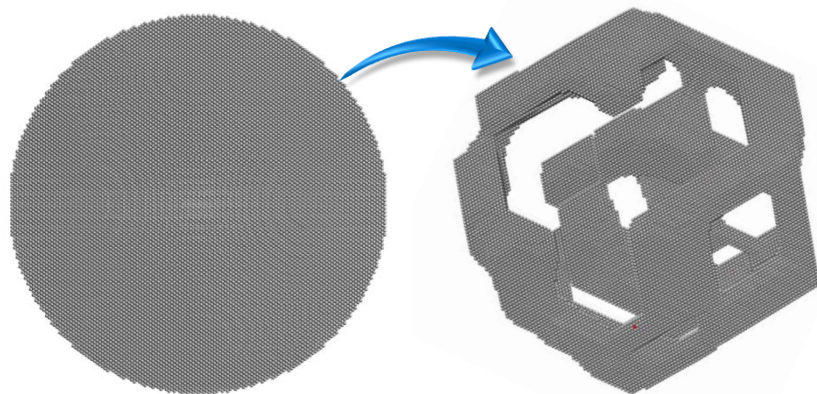
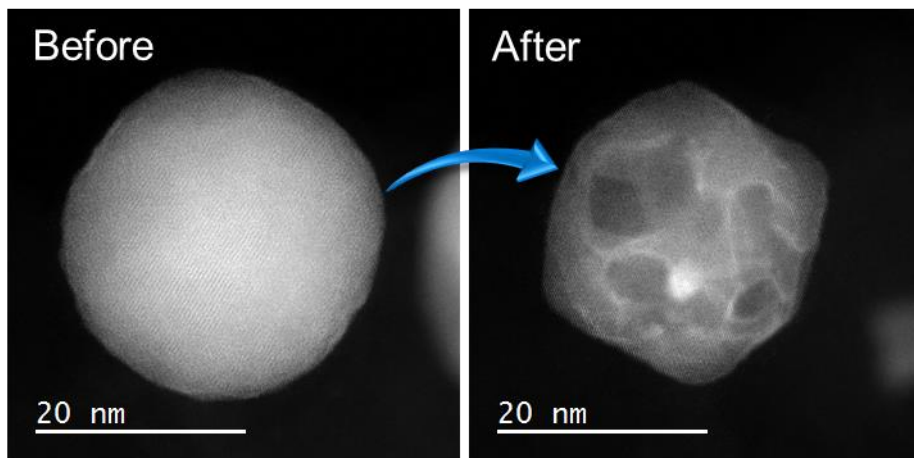
We need 3D insights!

- TEM Tomography
- 3D atomic simulations



Tnx Andraž

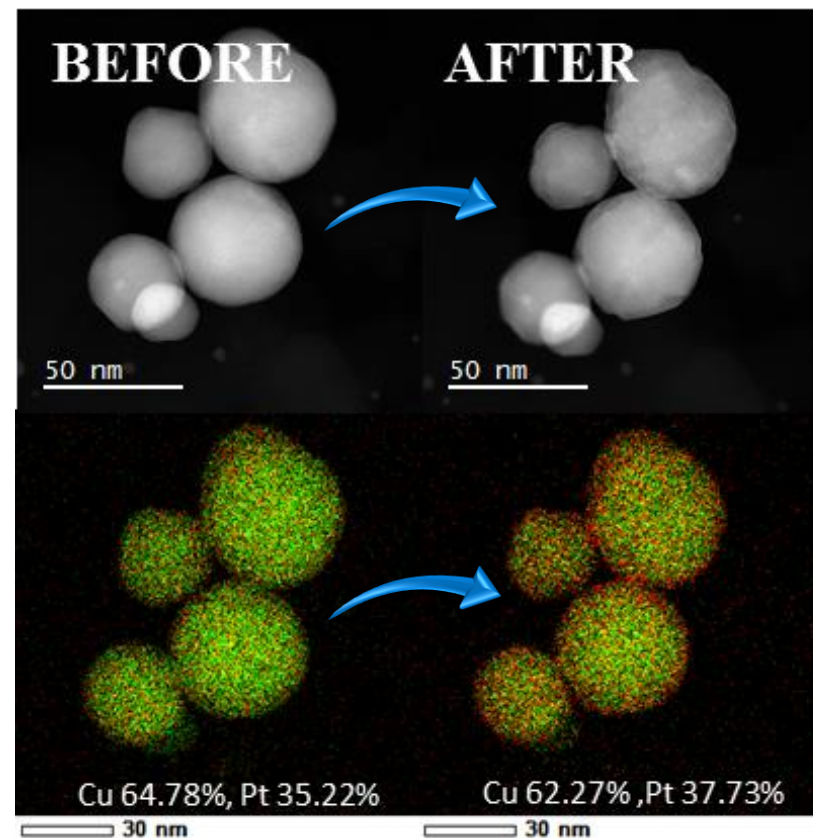
unpublished

PtCu₃ dealloying – selective Cu dissolution


Multi-scale process Monte Carlo & micro-kinetics modeling of dealloying process and mass transport

Pavlišič, Hodnik, ACS Catal., 2016, 6 (8), pp 5530–5534

Zepeda, Hodnik, accepted in ChemCatChem, 2017

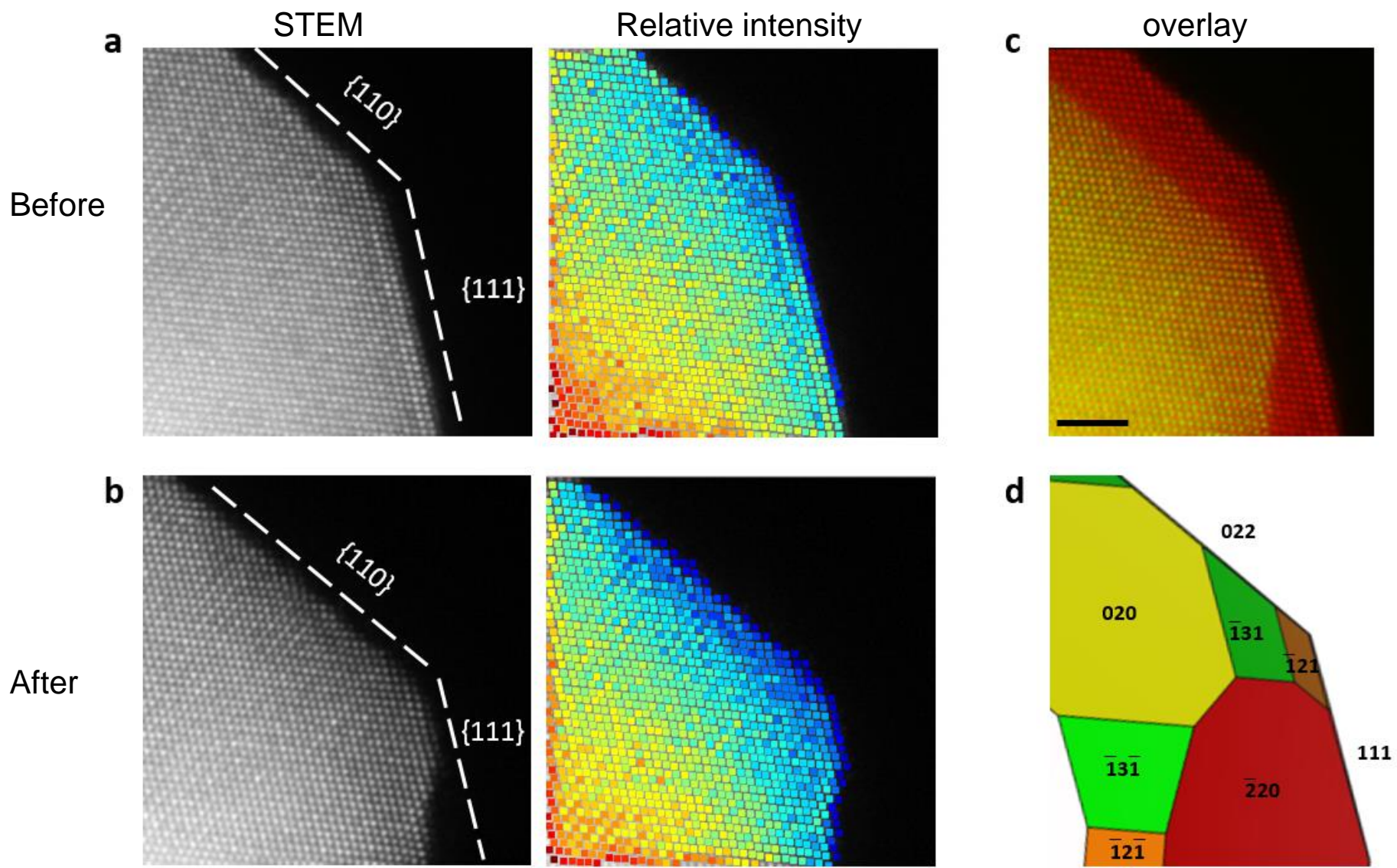
 PtCu₃Au dealloying

IL-STEM-EDX

-we can use any other detector available in the TEM

Zepeda, Hodnik, accepted in ChemCatChem, 2017

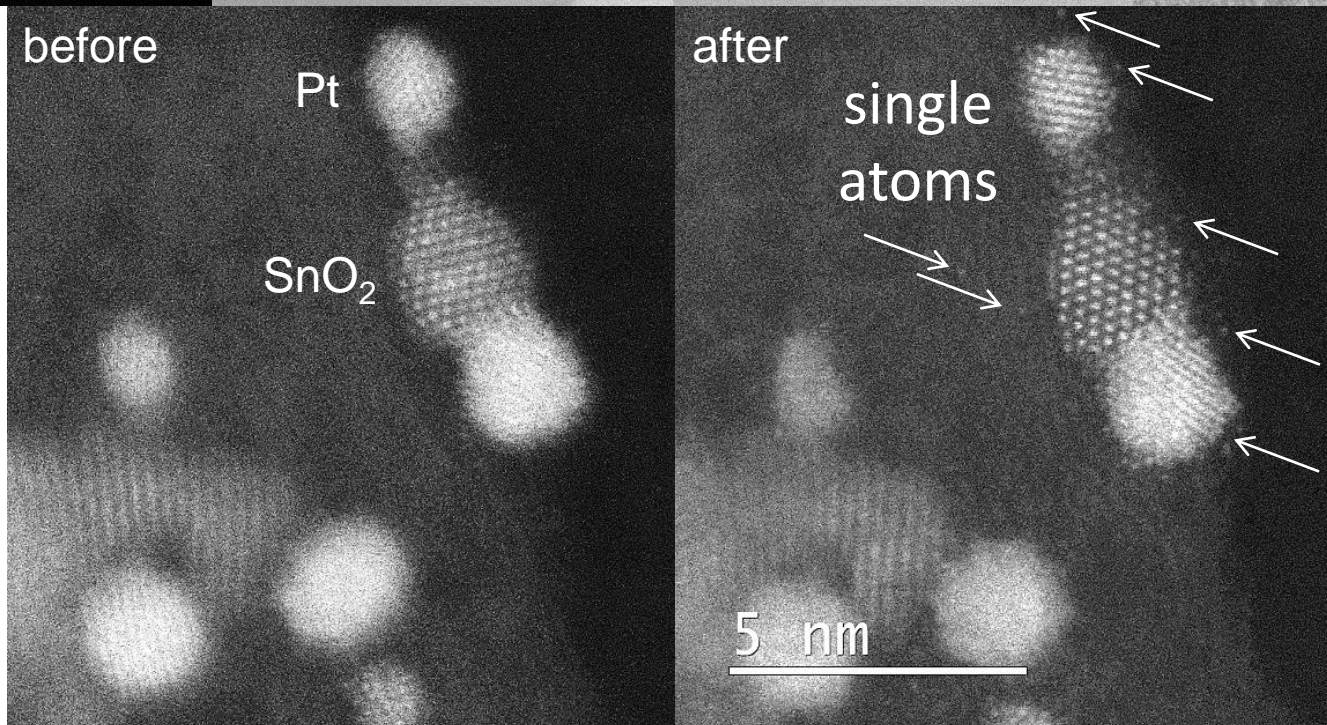
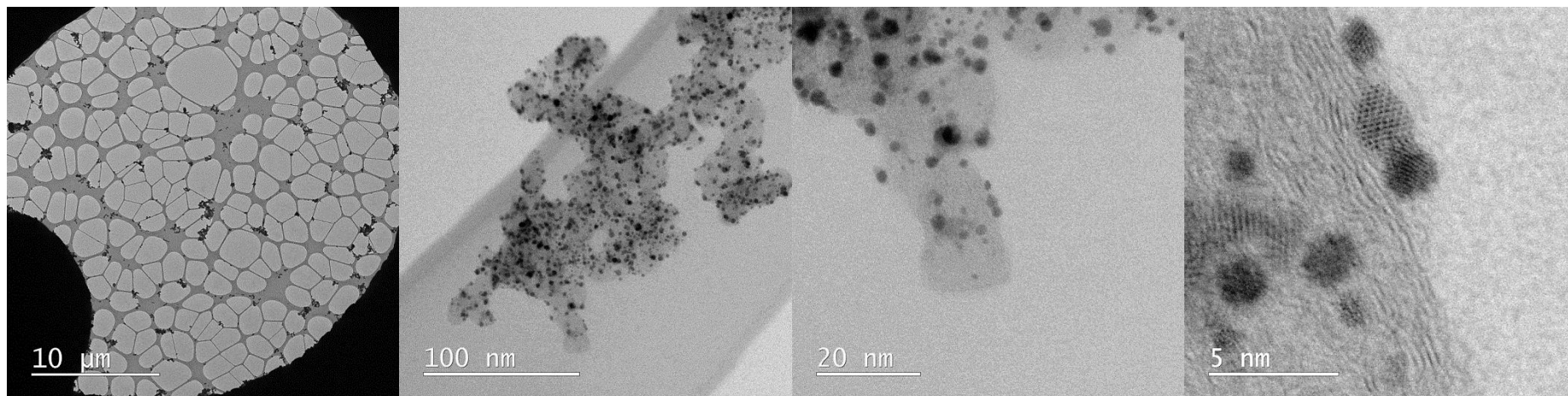
Gatalo, Hodnik, J. Electrochem. Sci. Eng., 2018, 8 (1) pp. 87-100

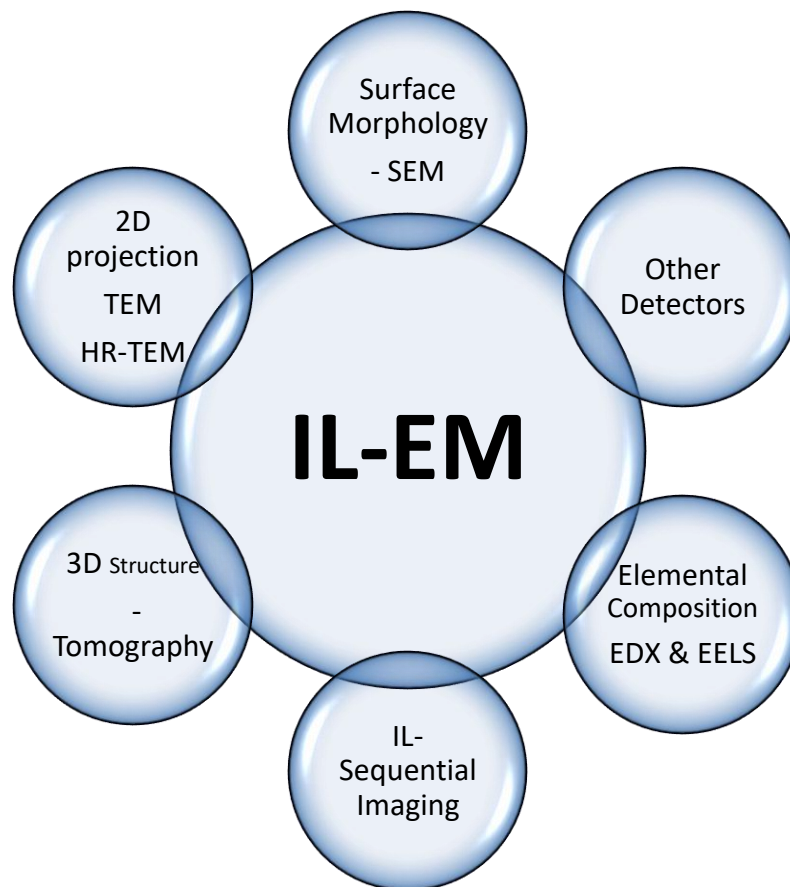
Example 1



Tnx Francisco

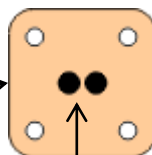
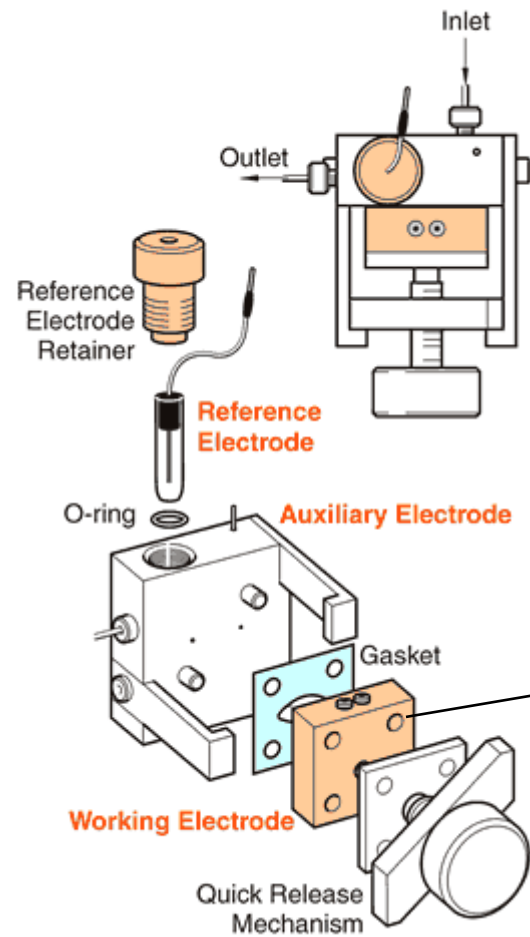
unpublished

Example 2




- other materials
- synthesis
- heterogeneous catalysis
- any kind of treatment

We have structure! Now how to measure dissolution?



← Can be replaced by steel or any other metal plate

Glassy carbon

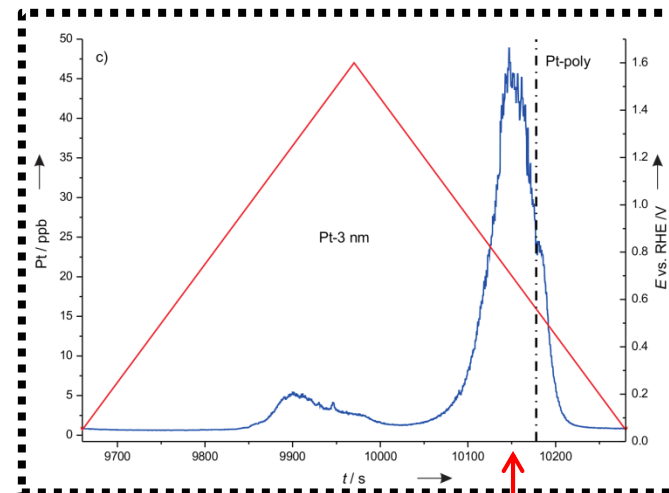
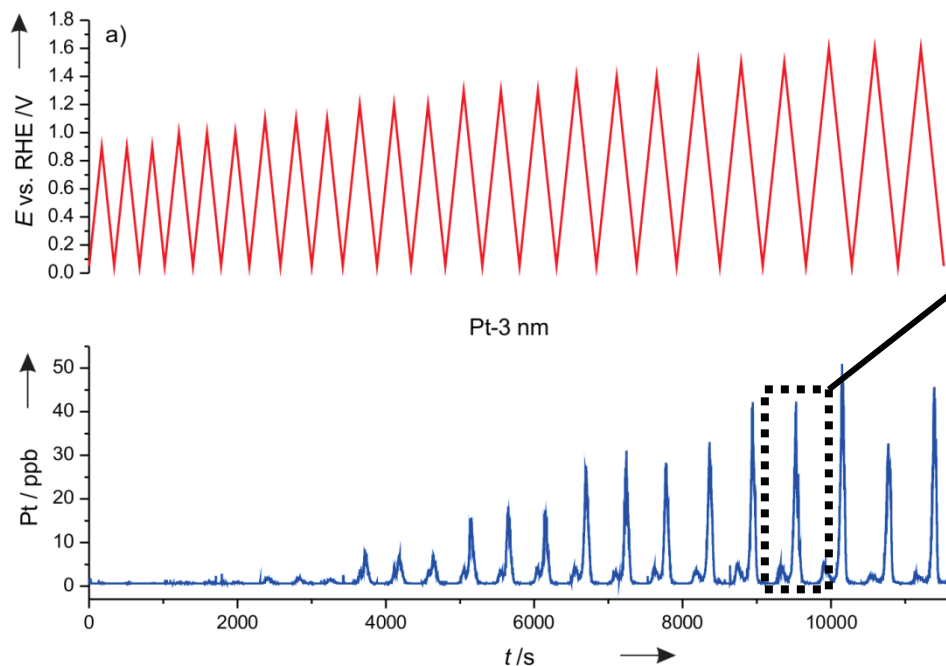
- many different electrolytes
- can couple to ICP-MS and ICP-OES
- either peristaltic pump or syringe
- can analyze corrosion of any metallic samples



Agilent 7500ce ICP-MS instrument

Extremely sensitive potential- and time- resolved dissolution profiles of metals

PEM Fuel Cell catalyst: Pt 3 nm



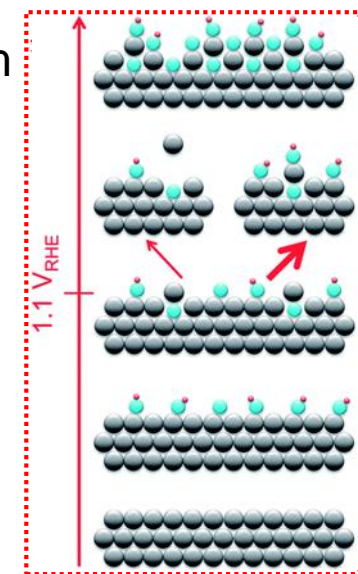
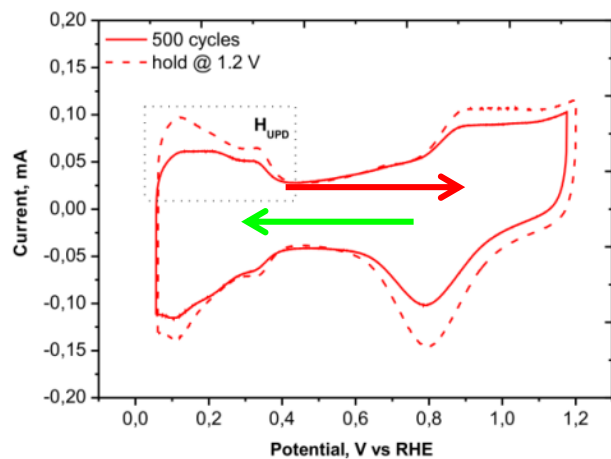
„transient dissolution“

1. Similar behavior as Pt-poly
2. In one cycle with vertex limit 1.6 V:

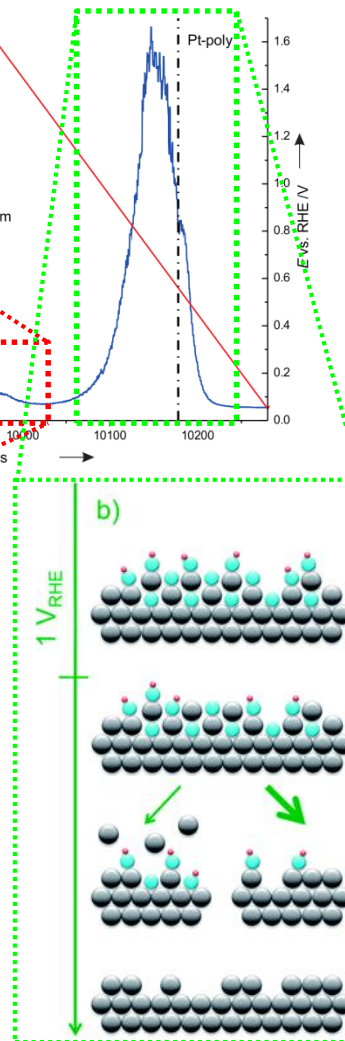
Pt/C	0.43%	or	1.4% of a monolayer
Pt/black	0.04%	or	0.7% of a monolayer
PtCu ₃ /C	0.22%	or	1.1% of monolayer

Extremely sensitive potential- and time- resolved dissolution profiles of metals

PEM Fuel Cell catalyst: Pt 3 nm



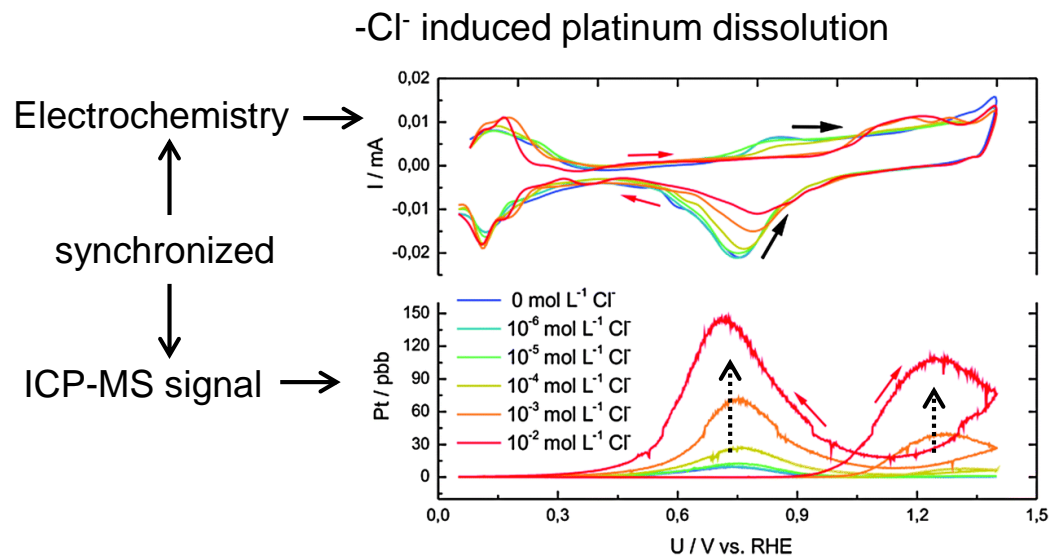
Anodic dissolution



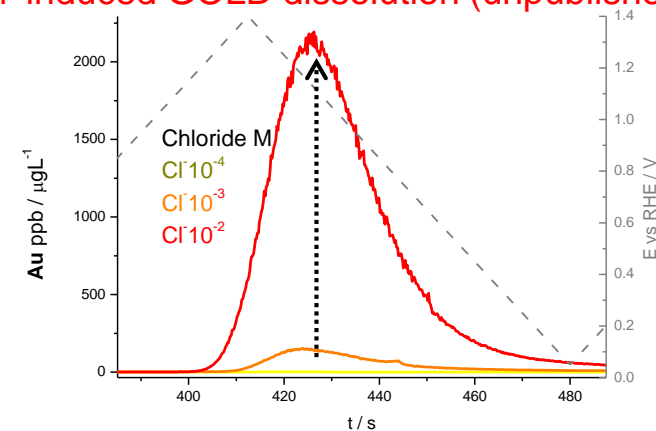
Cathodic dissolution

Extremely sensitive potential- and time- resolved dissolution profiles of metals

PEM Fuel Cell catalyst: Pt/C – 3 nm



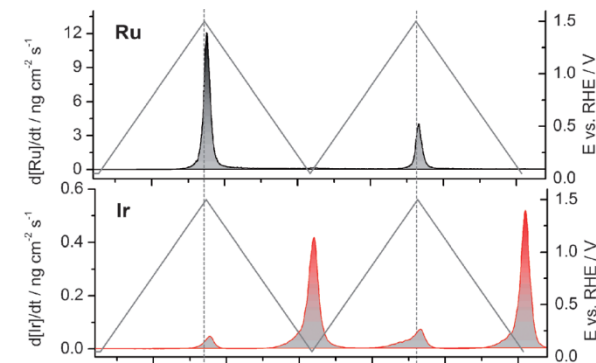
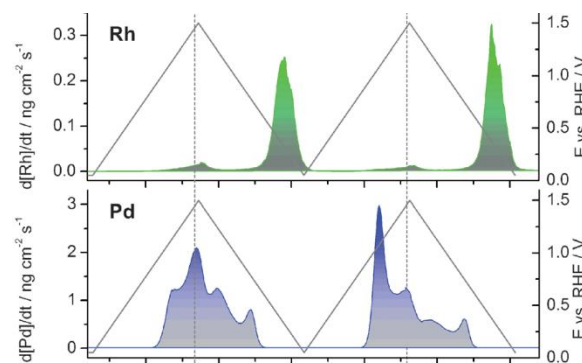
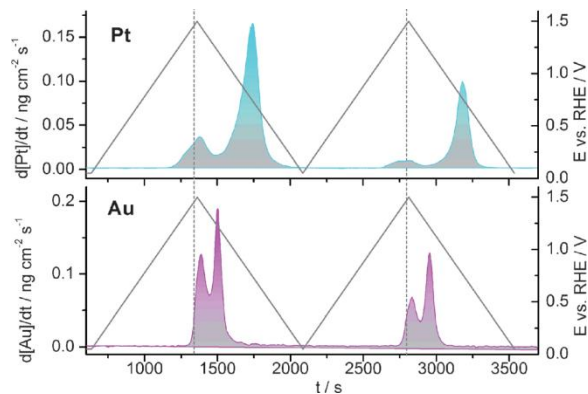
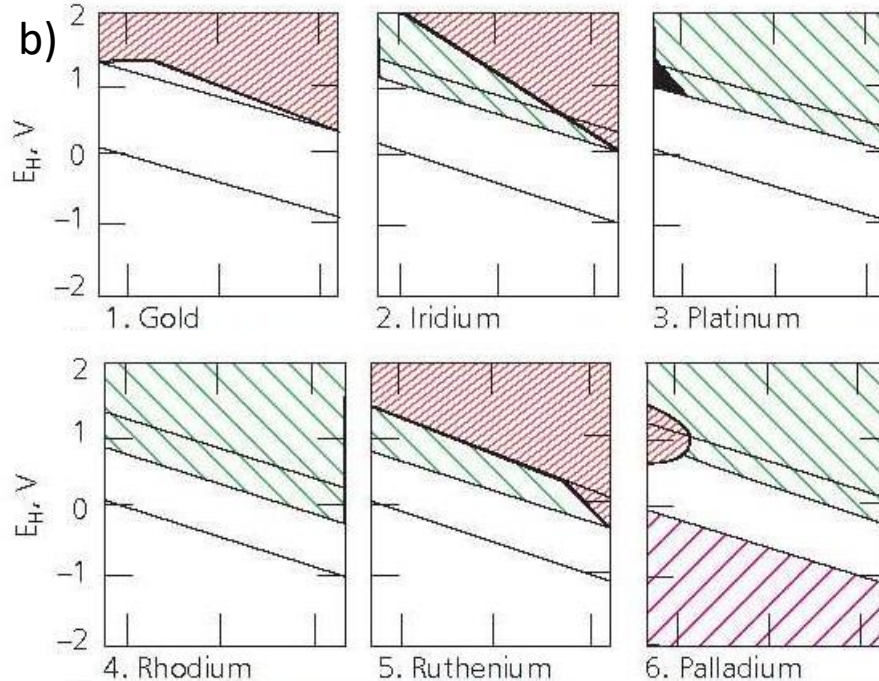
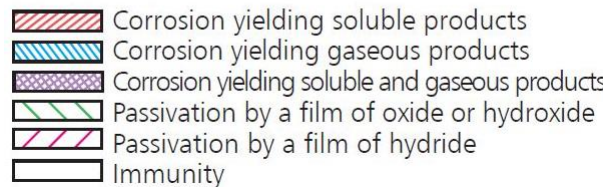
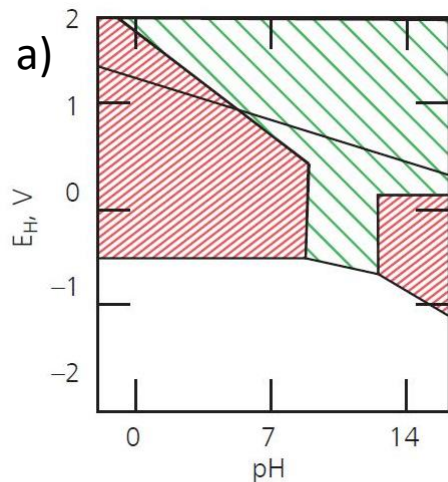
-Cl⁻ induced GOLD dissolution (unpublished)



Tnx Primož

Dissolution of PGM

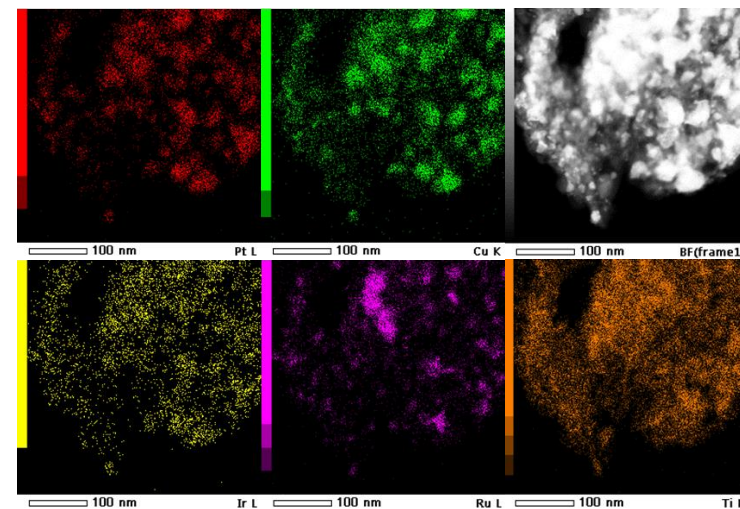
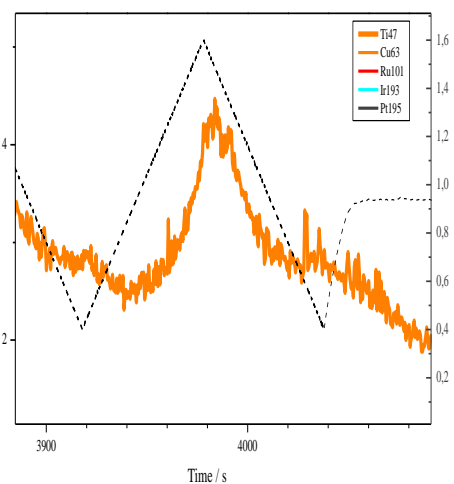
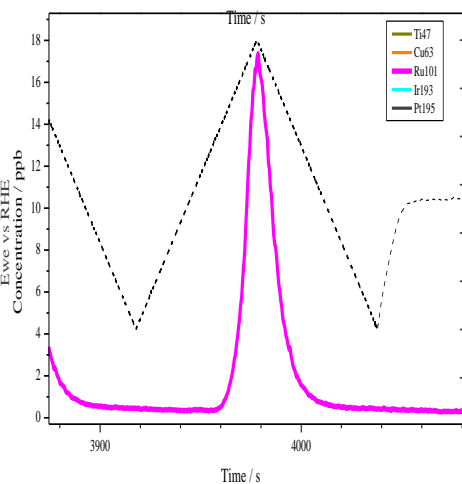
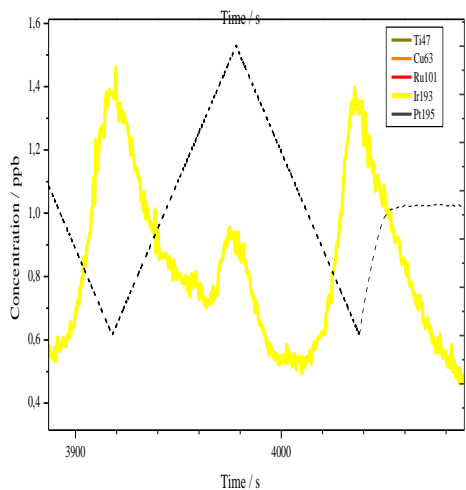
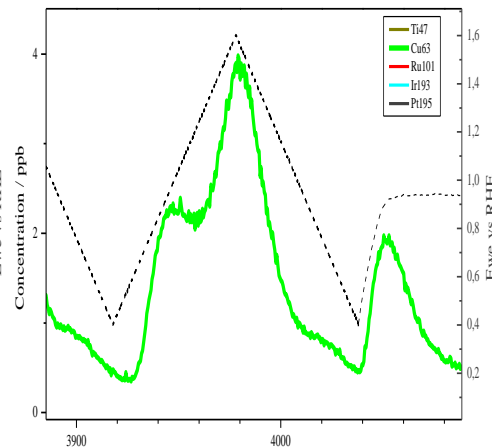
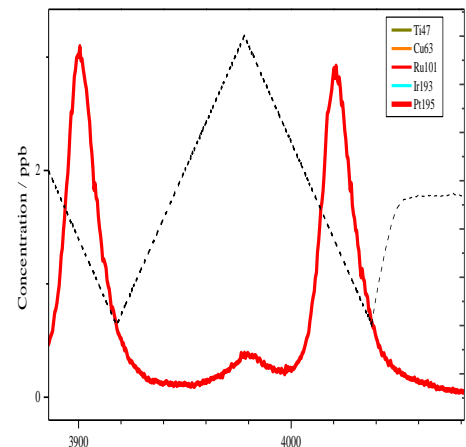
Pourbaix diagrams



Dissolution of PGM

“all-in-one“ or a good electrocatalyst for everything

Pt, Cu, Ir, Ru, Ti, C nanocomposite

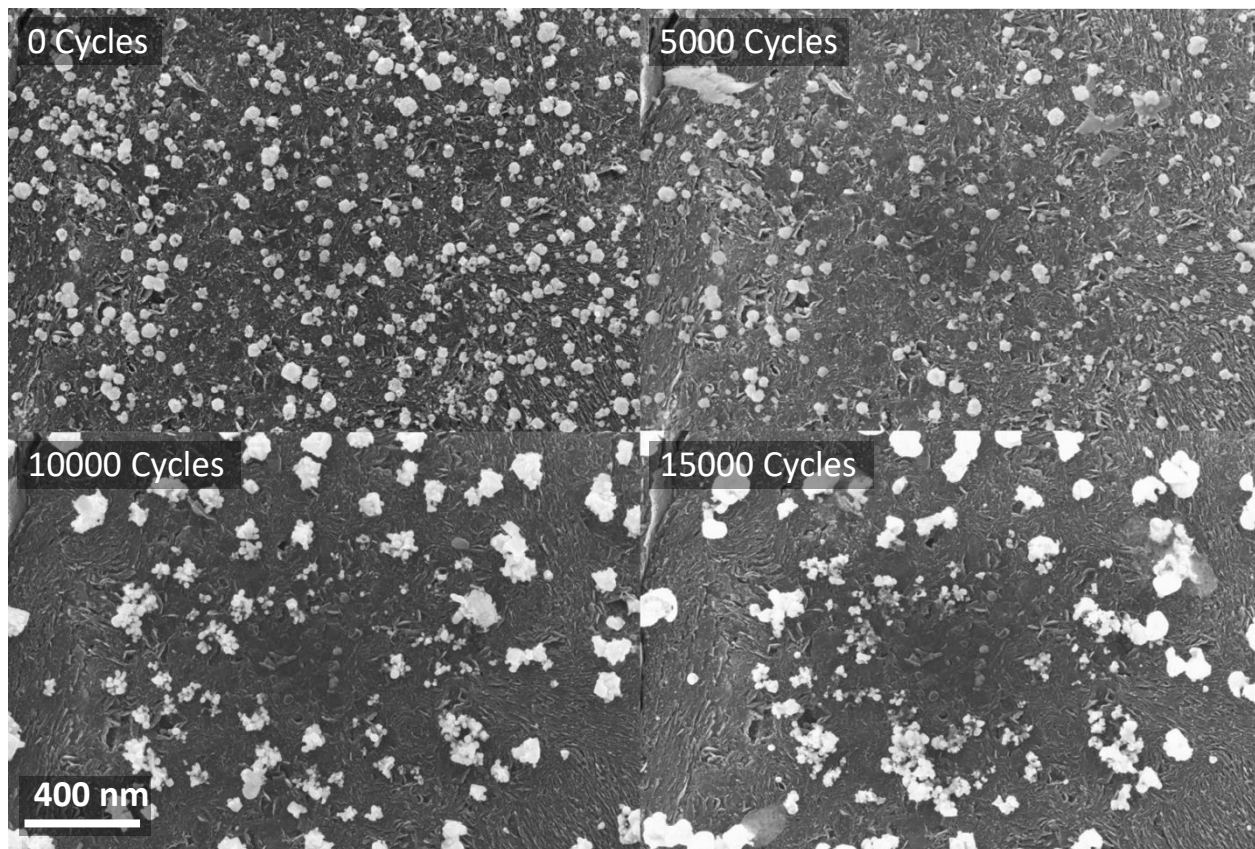


Tnx Leonard

unpublished

What about degradation at reductive potentials?

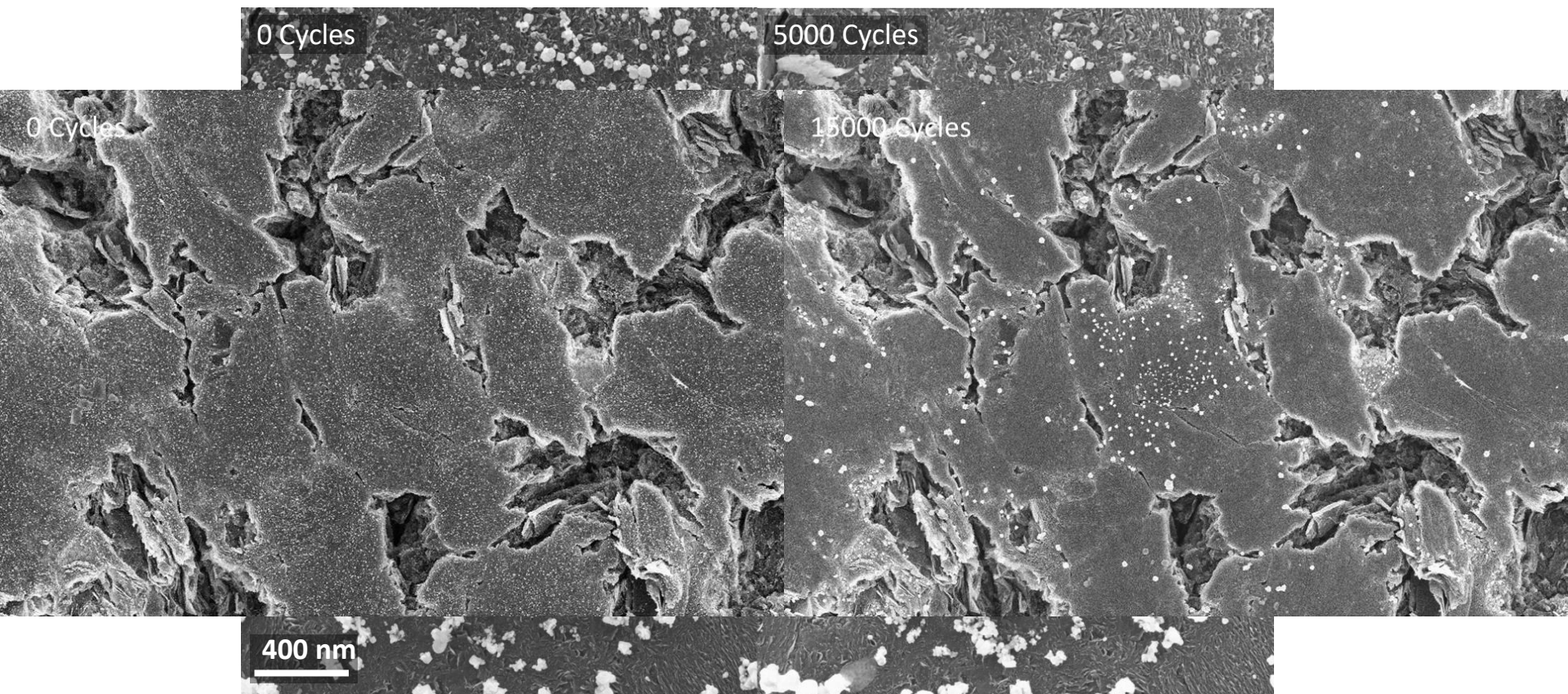
Degradation Ag via coalescence: 0, 5000, 10000, 15000 cycles



Vanrenterghm, Hodnik, Applied Catalysis B: Environmental, 2018, 226, pp 396–402.

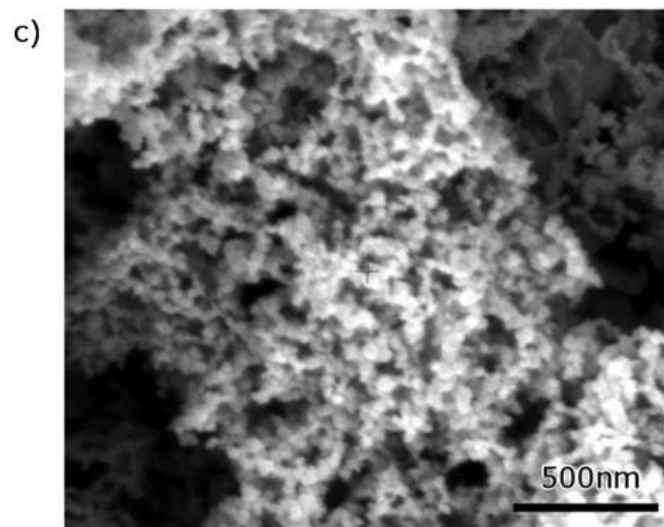
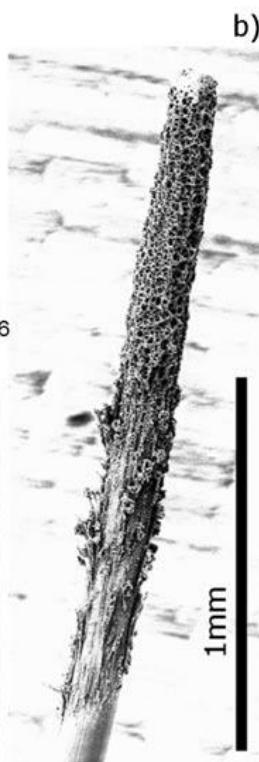
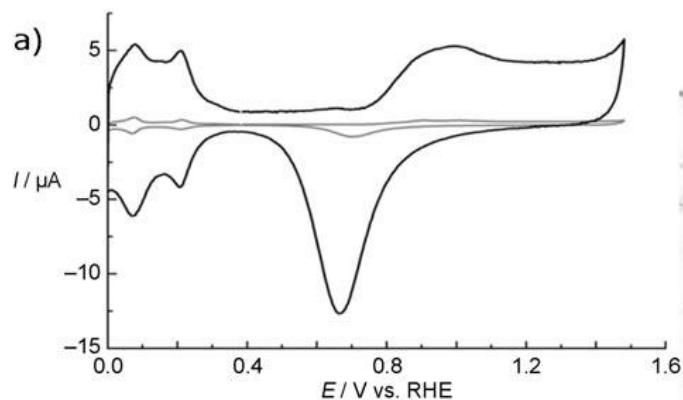
What about degradation at reductive potentials?

Degradation Ag via coalescence: 0, 5000, 10000, 15000 cycles



There is something else occurring at negative potentials!

Cathodic Corrosion: A Quick, Clean, and Versatile Method for the Synthesis of Metallic Nanoparticles

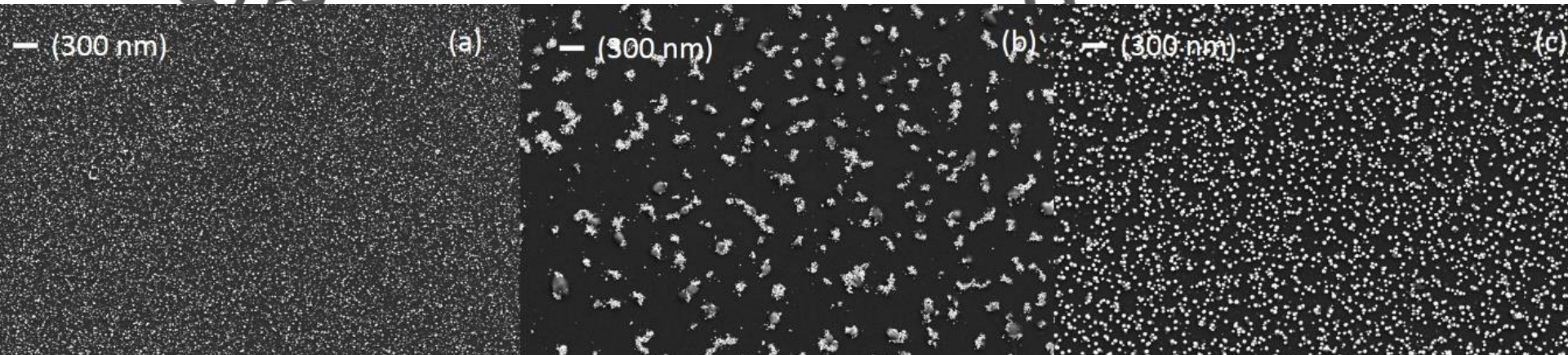
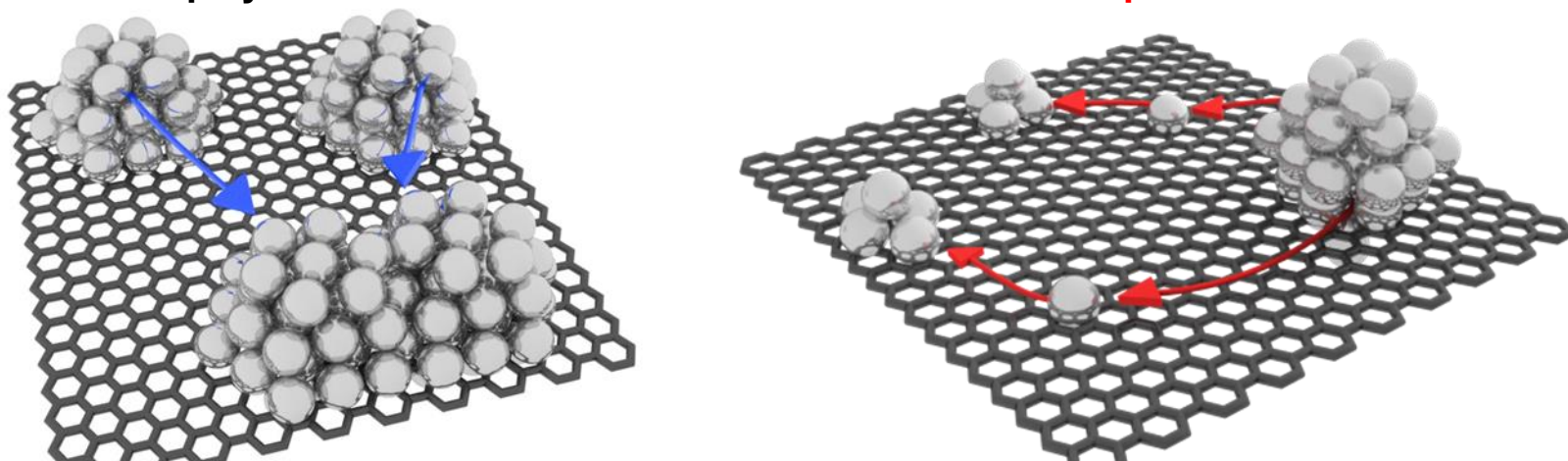


Cyclic voltammogram of a Pt wire 135 μm in diameter, submerged by 1 mm in 0.5 M H_2SO_4 before (gray) and after (black) the wire was held for 1000 s at a dc of 10 V (7.2 V vs. HgO) in 10 M NaOH. Graphite is used as anode to rule out the formation of interfering species by anodic dissolution. Sweep rate: 50 mVs⁻¹. b,c) Typical scanning electron microscopic images of a well-rinsed Pt electrode after cathodic treatment.

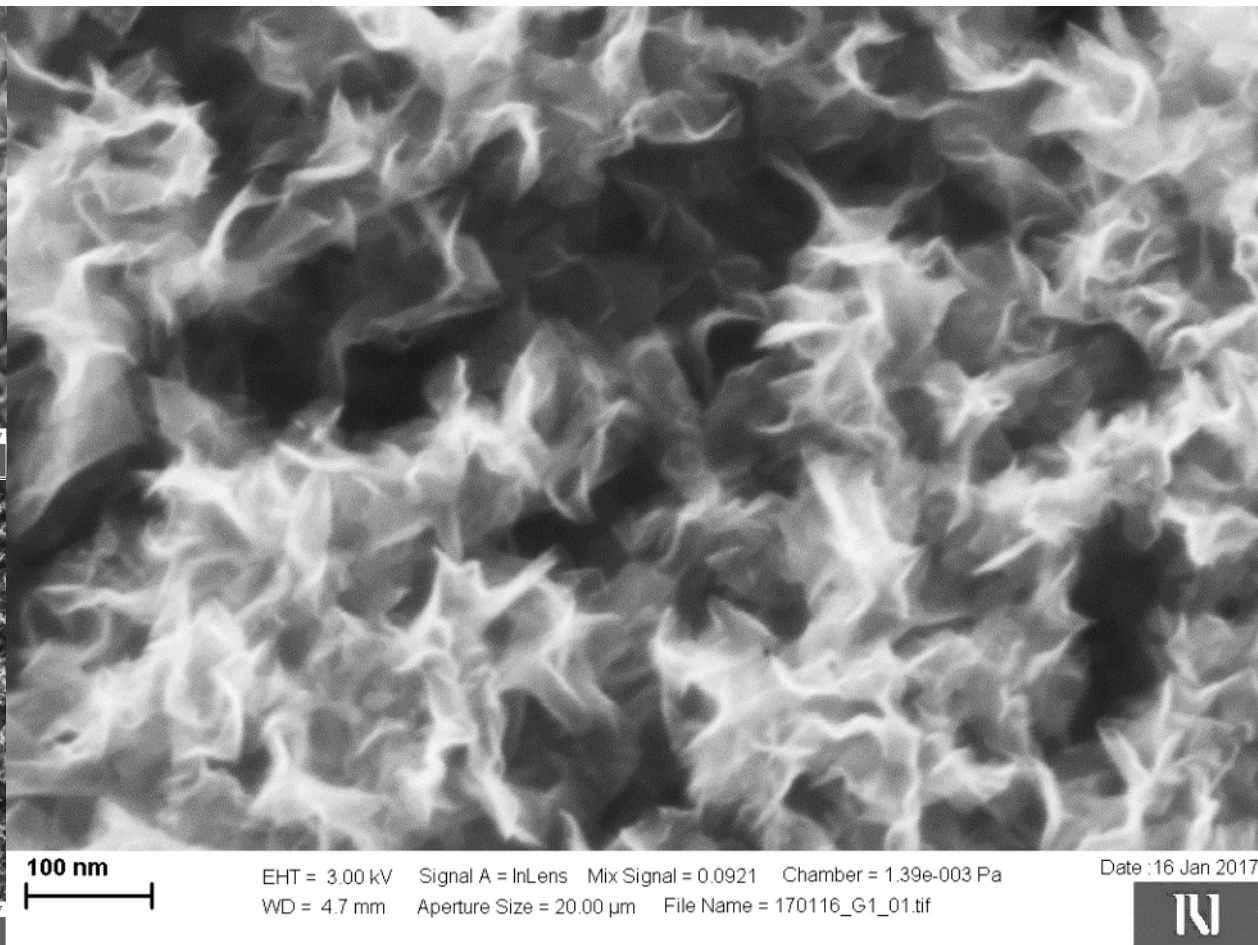
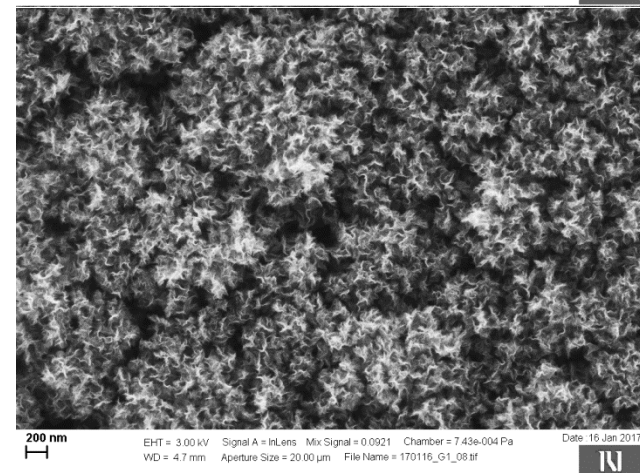
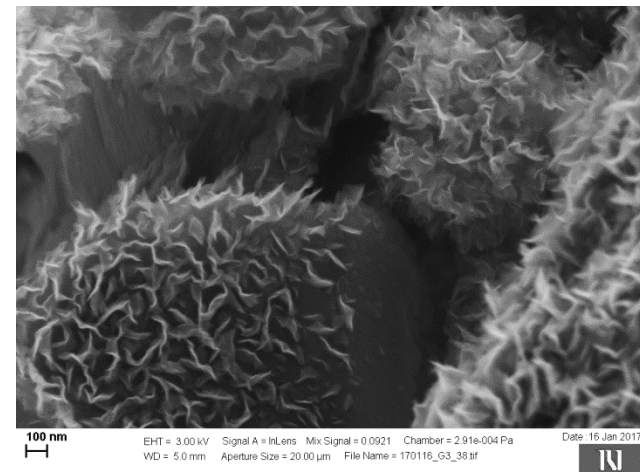
What about degradation at reductive potentials?

Electrochemical induced metal nanoparticle Coalescence / Dispersion

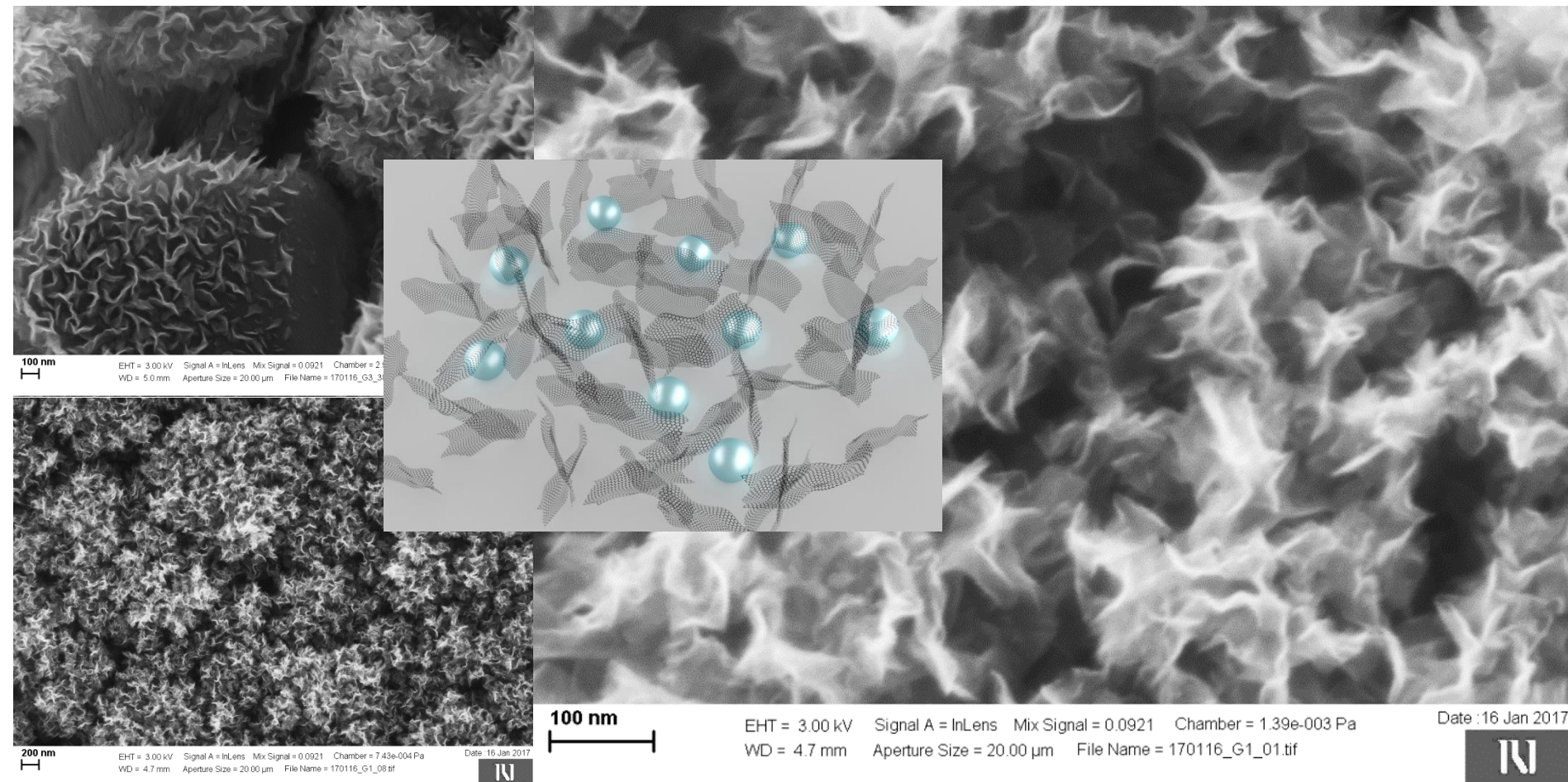
Interplay between Potential induced **Coalescence** vs. **Dispersion**



Plasma Grown Vertically Aligned Graphene



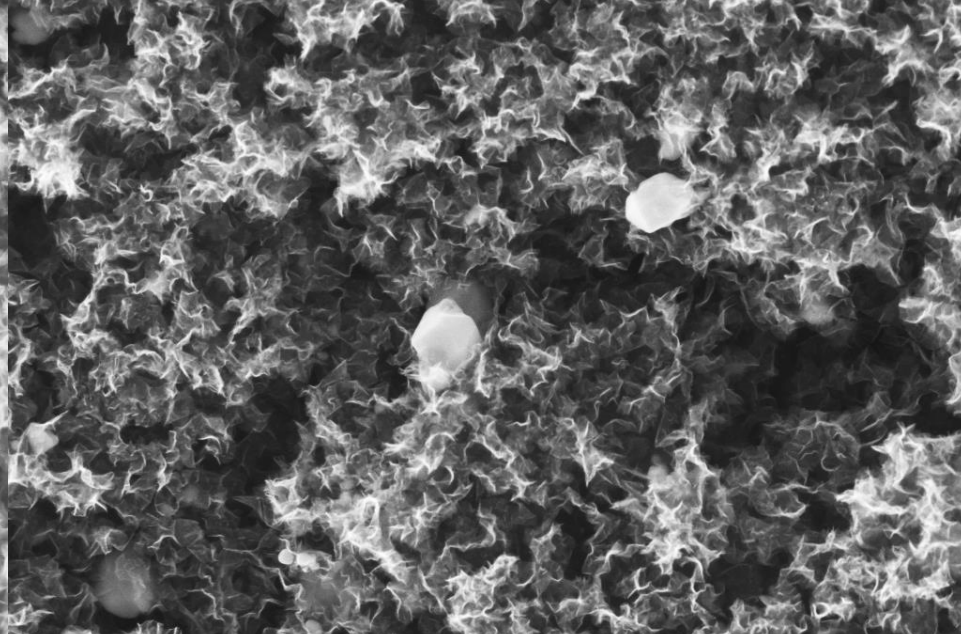
Plasma Grown Vertically Aligned Graphene





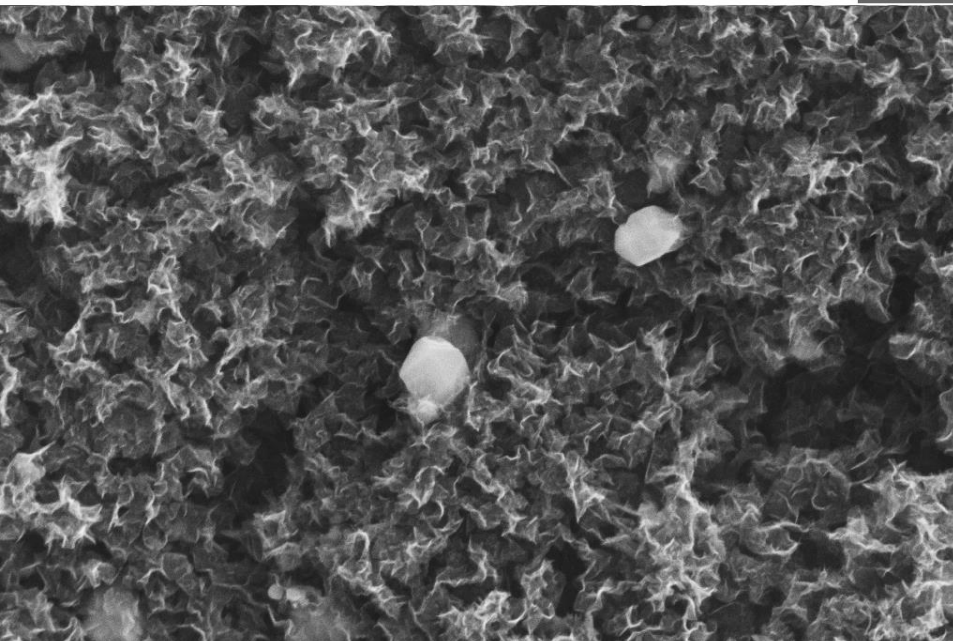
EHT = 3.00 kV Signal A = InLens Mix Signal = 0.0921 Chamber = 1.80e-004 Pa
WD = 4.8 mm Aperture Size = 20.00 μ m File Name = 170116_T3_11.tif

Date : 16 Jan 2017



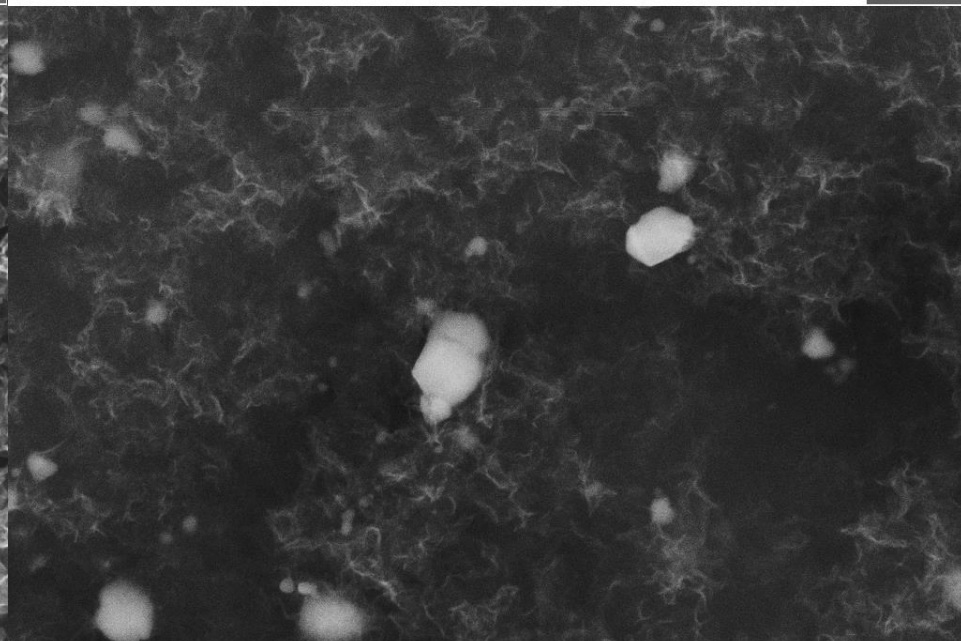
EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0000 Chamber = 1.38e-004 Pa
WD = 4.0 mm Aperture Size = 20.00 μ m File Name = 170117_T3_06.tif

Date : 17 Jan 2017



EHT = 15.00 kV Signal A = InLens Mix Signal = 0.0000 Chamber = 1.68e-004 Pa
WD = 8.4 mm Aperture Size = 20.00 μ m File Name = 170117_T3_s139.tif

Date : 17 Jan 2017



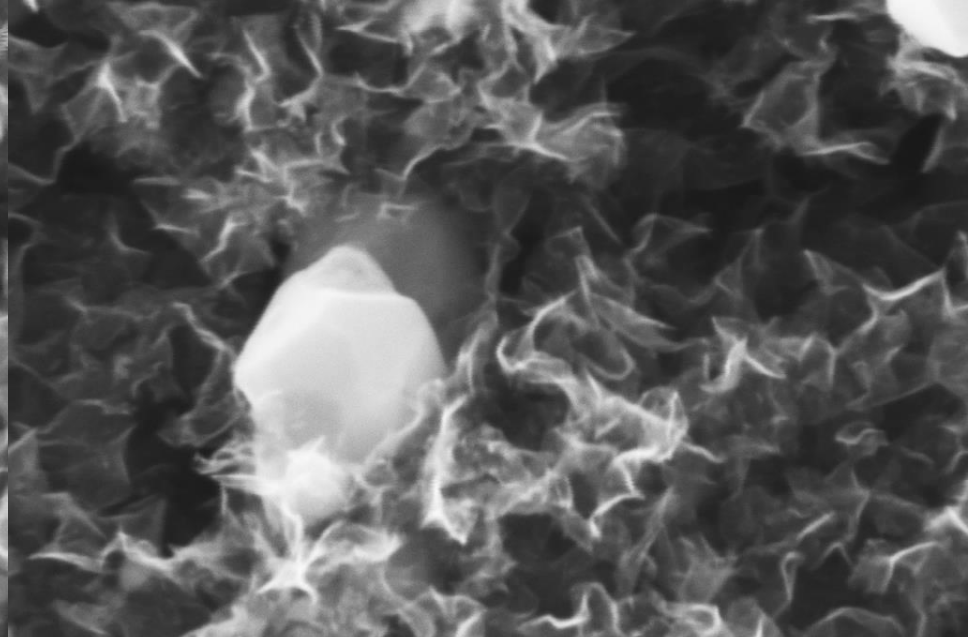
EHT = 15.00 kV Signal A = SE2 Mix Signal = 0.0000 Chamber = 1.68e-004 Pa
WD = 8.4 mm Aperture Size = 20.00 μ m File Name = 170117_T3_s140.tif

Date : 17 Jan 2017

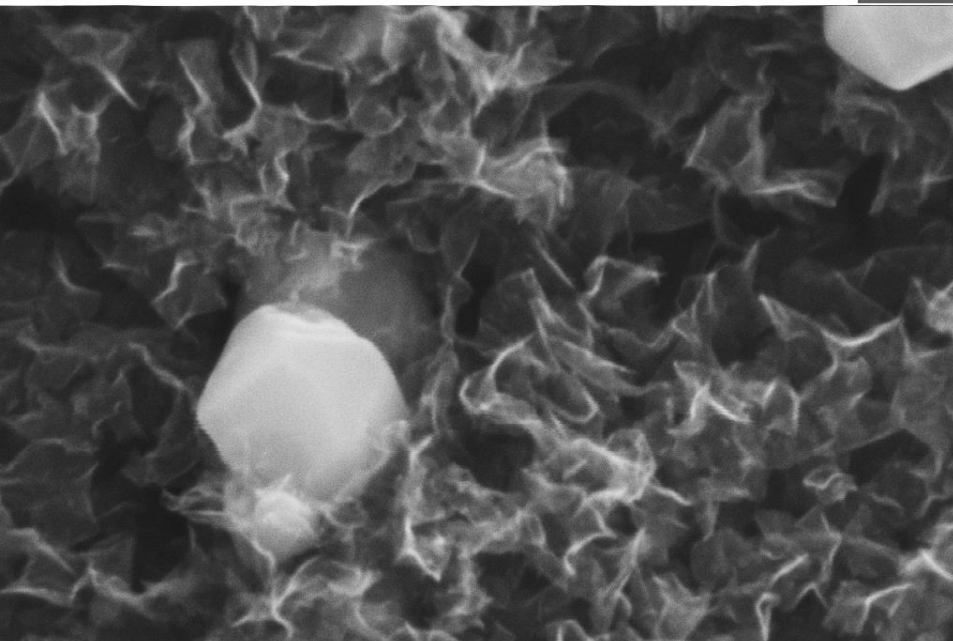




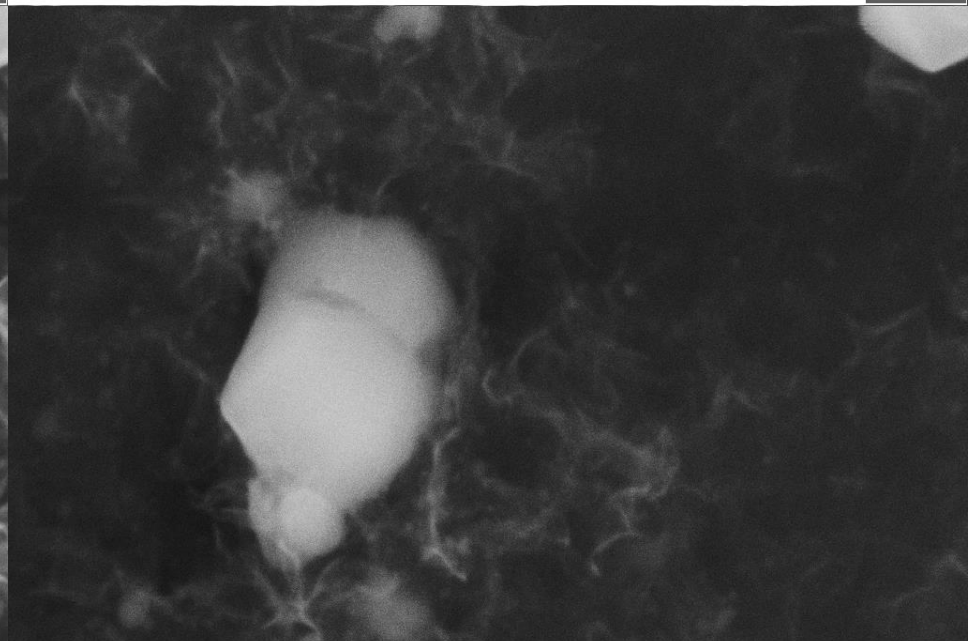
100 nm EHT = 3.00 kV Signal A = InLens Mix Signal = 0.0921 Chamber = 1.84e-004 Pa Date :16 Jan 2017
WD = 4.8 mm Aperture Size = 20.00 μ m File Name = 170116_T3_10.tif



100 nm EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0000 Chamber = 1.47e-004 Pa Date :17 Jan 2017
WD = 4.0 mm Aperture Size = 20.00 μ m File Name = 170117_T3_04.tif

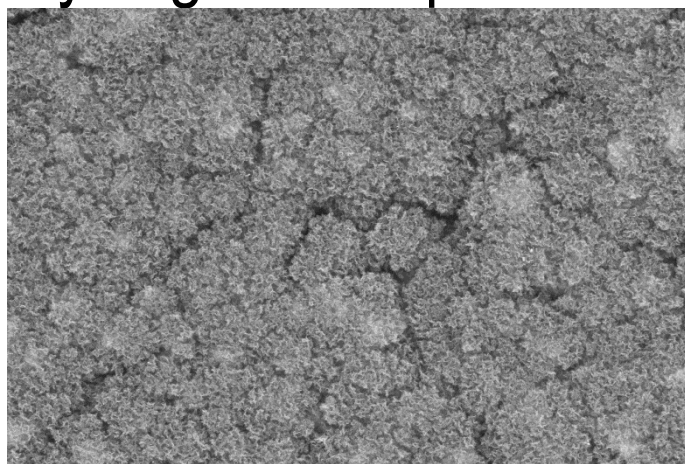


100 nm EHT = 15.00 kV Signal A = InLens Mix Signal = 0.0000 Chamber = 1.72e-004 Pa Date :17 Jan 2017
WD = 8.4 mm Aperture Size = 20.00 μ m File Name = 170117_T3_s136.tif

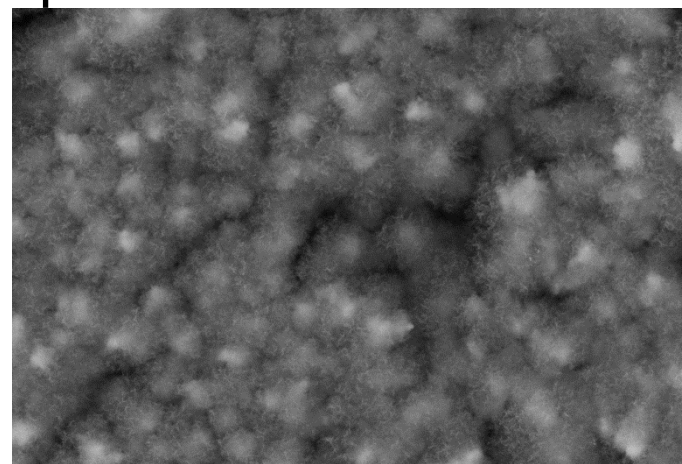


100 nm EHT = 15.00 kV Signal A = SE2 Mix Signal = 0.0000 Chamber = 1.72e-004 Pa Date :17 Jan 2017
WD = 8.4 mm Aperture Size = 20.00 μ m File Name = 170117_T3_s137.tif

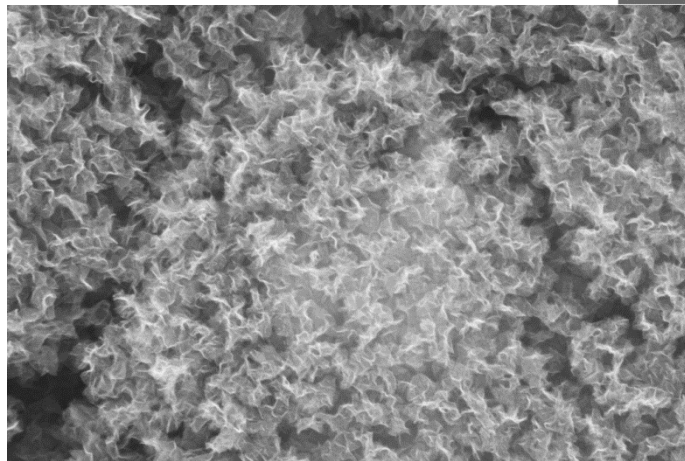
Increase of Electrodeposited Catalysts Stability via Plasma Grown Vertically Aligned Graphene Nanoparticles Movement Restriction!



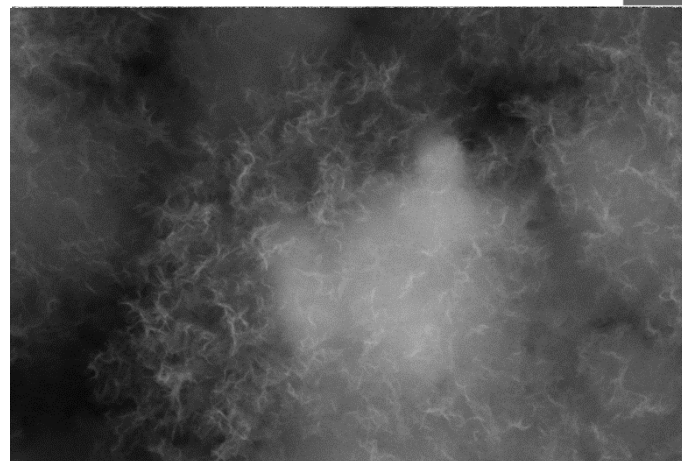
1 µm EHT = 15.00 kV Signal A = InLens Mix Signal = 0.0000 Chamber = 6.35e-004 Pa Date: 17 Jan 2017
WD = 8.2 mm Aperture Size = 20.00 µm File Name = 170117_G_1_24.tif



1 µm EHT = 15.00 kV Signal A = SE2 Mix Signal = 0.0000 Chamber = 6.07e-004 Pa Date: 17 Jan 2017
WD = 8.2 mm Aperture Size = 20.00 µm File Name = 170117_G_1_25.tif



300 nm EHT = 15.00 kV Signal A = InLens Mix Signal = 0.0000 Chamber = 5.42e-004 Pa Date: 17 Jan 2017
WD = 8.2 mm Aperture Size = 20.00 µm File Name = 170117_G_1_26.tif



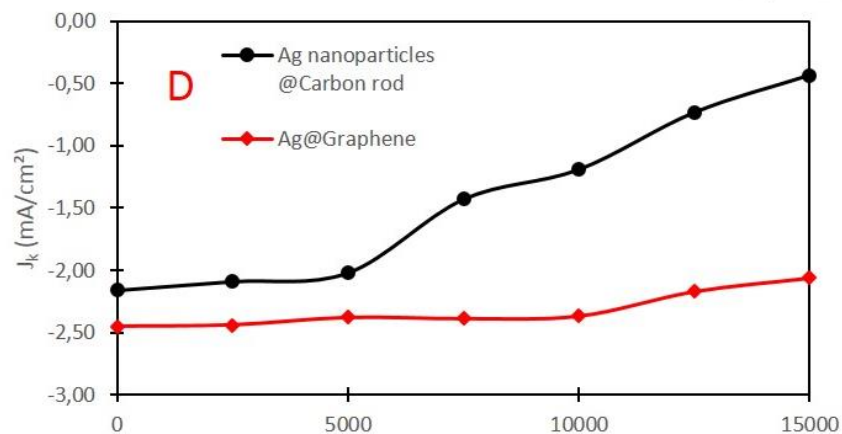
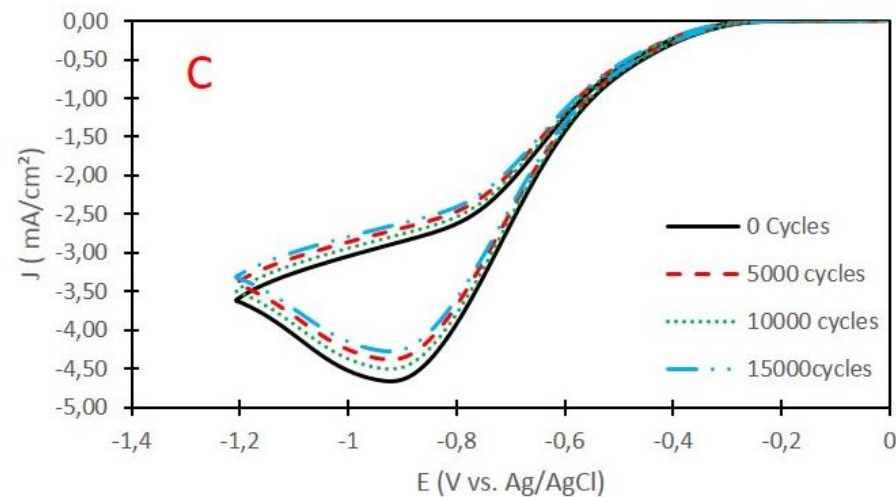
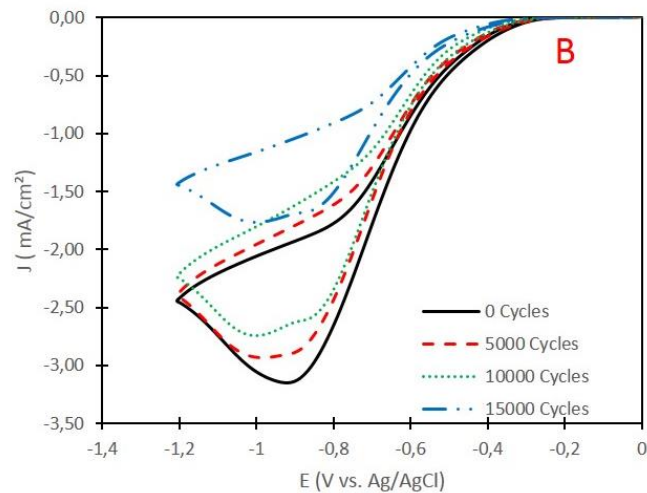
300 nm EHT = 15.00 kV Signal A = SE2 Mix Signal = 0.0000 Chamber = 5.42e-004 Pa Date: 17 Jan 2017
WD = 8.2 mm Aperture Size = 20.00 µm File Name = 170117_G_1_28.tif

in-lens
detector

secondary
electron
detector

gh . 53, pp.9340-9343.

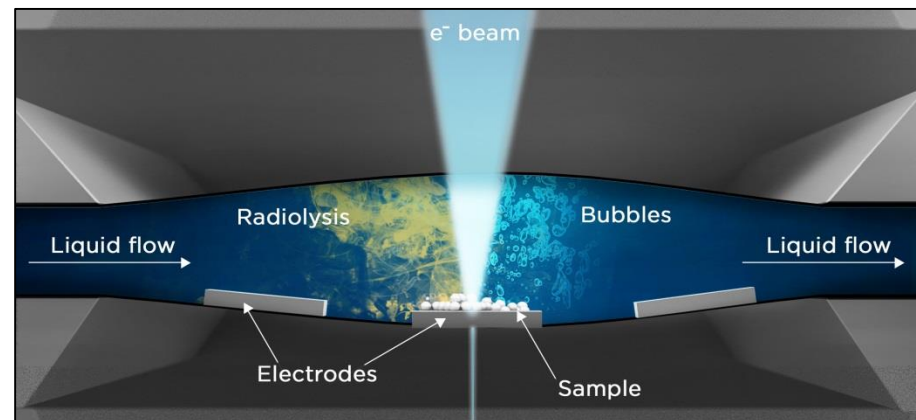
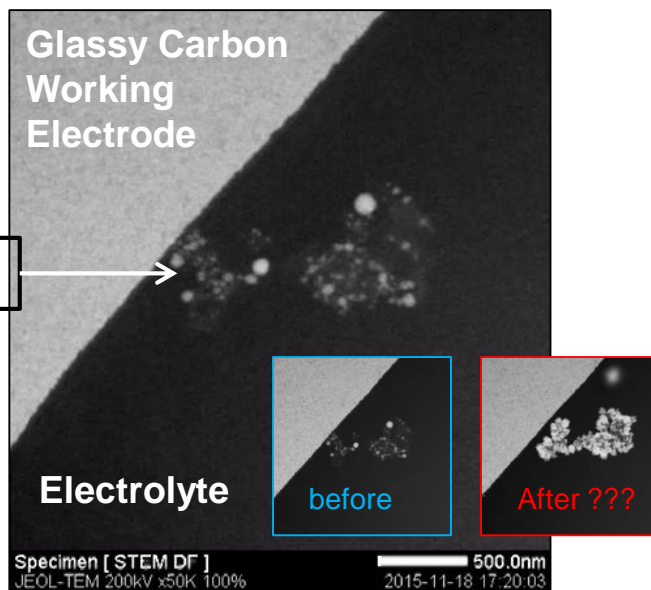
After electrochemical deposition



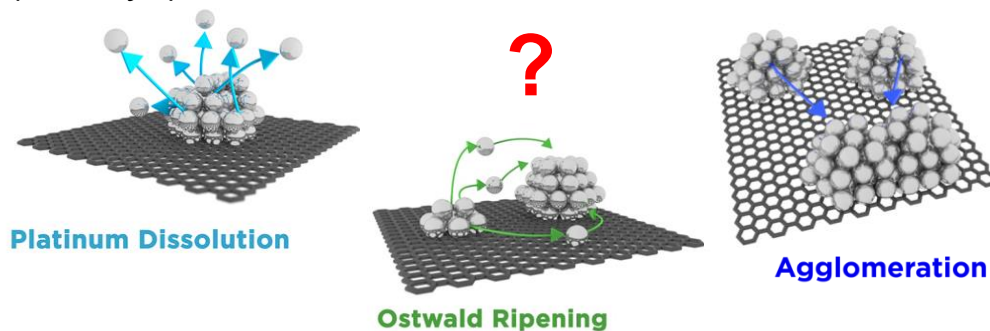
time- and
potential-
resolved
structural
changes:
Video!

Electrochemical in-situ liquid TEM

Agglomeration
or deposition
due to potential
cycling
(150 cyc) ?



Hodnik* et al. *Acc. Chem. Res.*, **2016**, 49 (9), pp 2015–2022



Most likely an artifact of electron beam!
To be sure it must be compared to IL-TEM that serves
as a benchmark. In-situ liquid TEM needs IL-TEM!





Noble or Precious metals

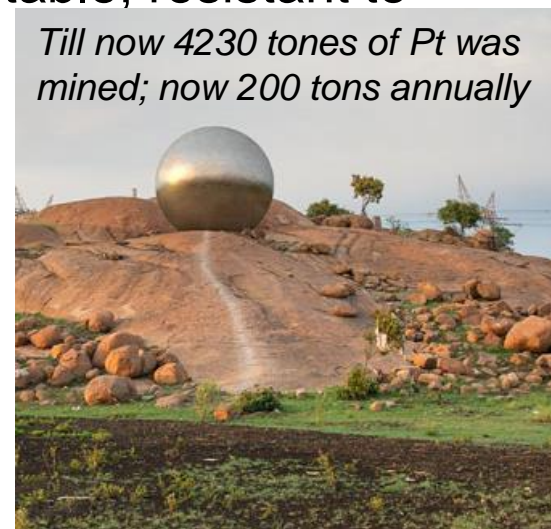
- Known from “4000 BC” and with a reach history since
- Useful due to unique functional properties: they are stable, resistant to oxidation, biocompatible and catalytically active

Mostly used as:

- Jewelry
- Electrical contacts
- (Electro)Catalysts

Ru 44 Ruthenium Electric Switches	Rh 45 Rhodium Searchlight Reflectors	Pd 46 Palladium Pollution Control	Ag 47 Silver Jewelry
Os 76 Osmium Pen Points	Ir 77 Iridium Spark Plugs	Pt 78 Platinum Labware	Au 79 Gold Jewelry

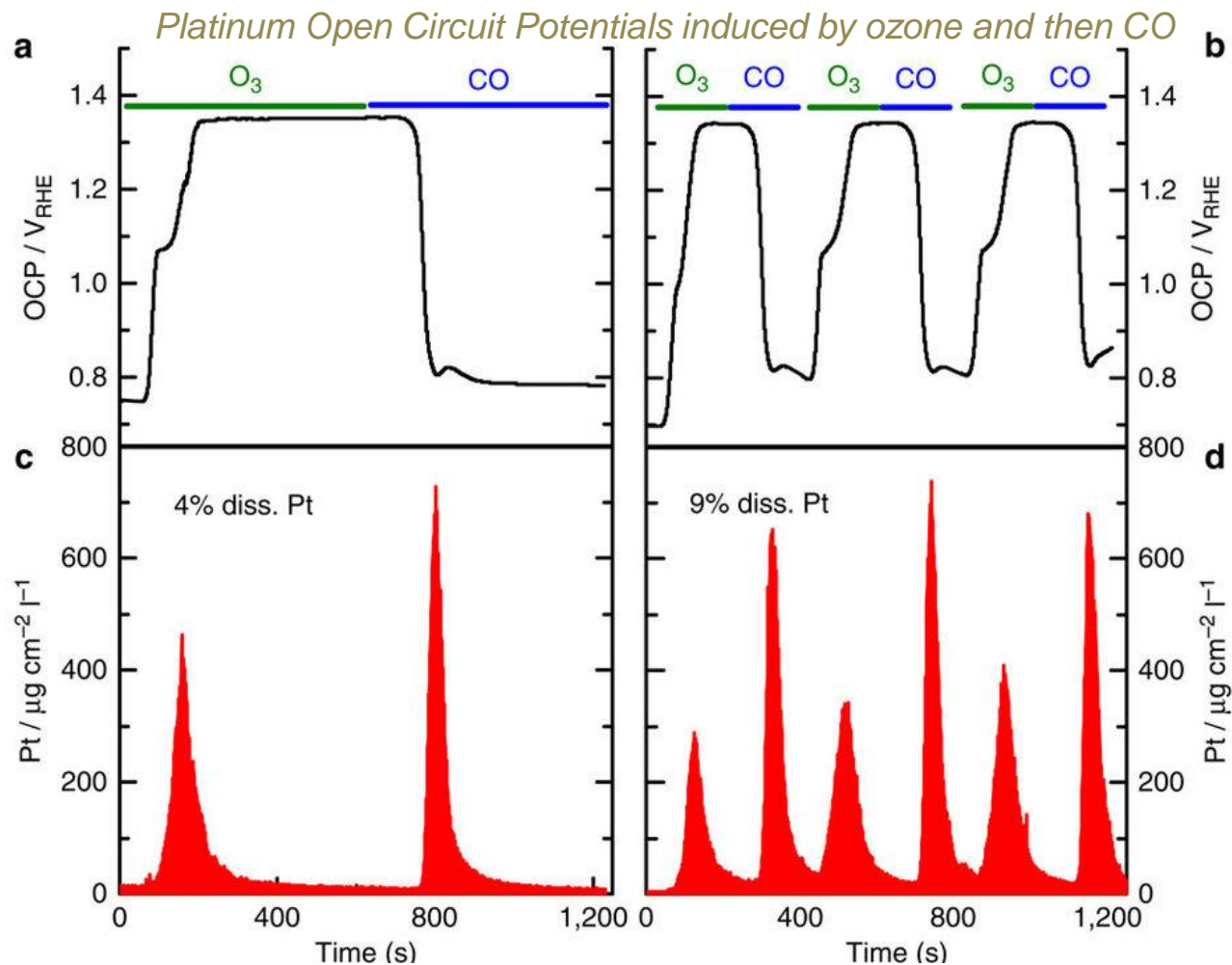
Till now 4230 tones of Pt was mined; now 200 tons annually



<http://dillonmarsh.com/pgm01.html>

- **In the future everything will be digitalized and free energy will come from sun**
- Due to their price and scarceness we must break them down to nanoparticles – properties change; stability issues
- Perfect for el. microscopy and ICP-MS (high-z methodology approach)

recycling of PGM



„transient
dissolution“

• recycling of PGM

Platinum Open Circuit Potentials induced by ozone and then CO

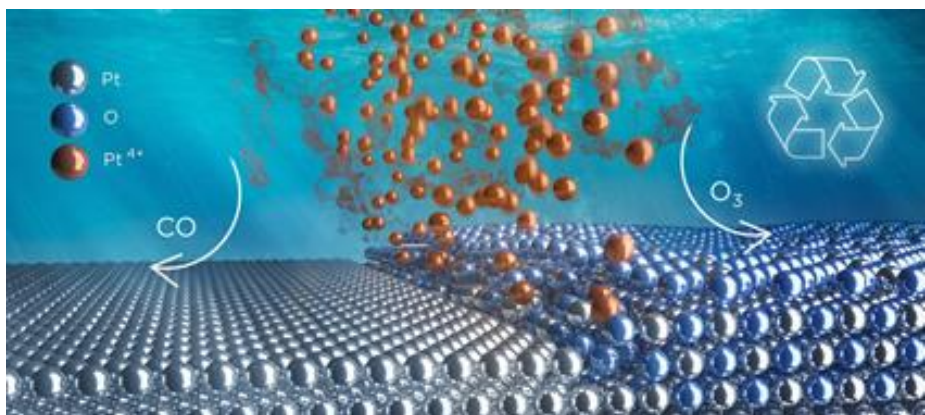
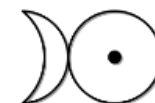
- We make use of **Transient Dissolution** phenomena

Compared to the State of the art: **Aqua Regia (alchemy)**

- 3:1 molar ratio of concentrated boiling nitric and hydrochloric acid

My new patent pending process is:

- Two orders of magnitude lower concentrated acids ($c = 0.1 \text{ M}$, $\text{pH} \sim 1$)
- Room temperatures ($20 \text{ }^\circ\text{C}$)
- „No toxicity” (toxic gasses can be avoided)
- Environmentally and user Friendly (simple)
- Based on completely new chemistry (unpredicted by Pourbaix)



Platinum!

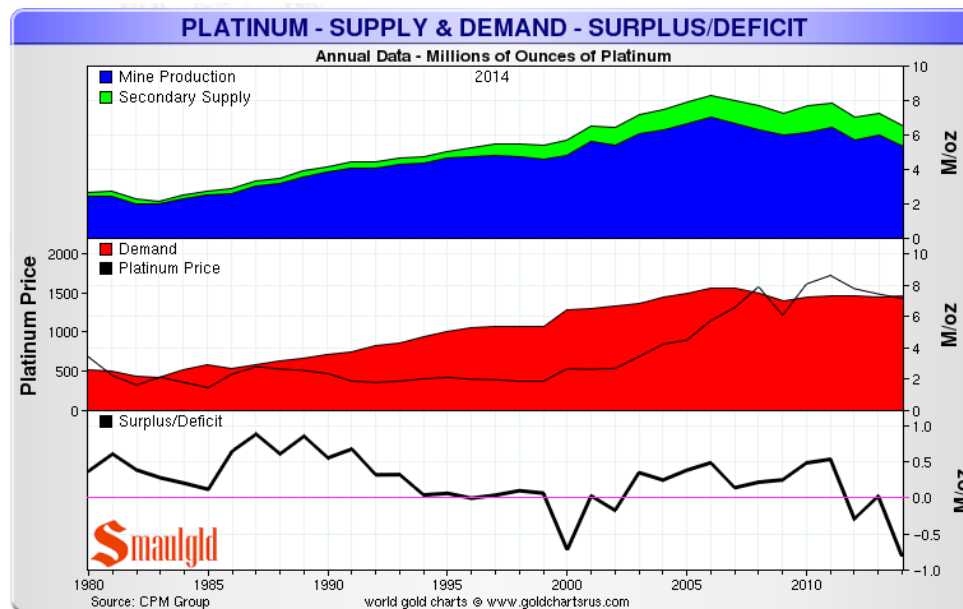


Important application areas for PGMs					
Application area	Platinum group metal				
	Platinum	Palladium	Rhodium	Iridium	Ruthenium
Catalysts	✓	✓	✓	✓	✓
Electronics	✓	✓		✓	✓
Fuel cells	✓	✓	✓		✓
Glass, ceramics and pigments	✓		✓		
Medical/dental	✓	✓		✓	
Pharmaceuticals	✓	✓			✓
Photovoltaic					✓
Super-alloys					✓

Source: *Recycling the Platinum Group Metals: A European Perspective*

Source: <http://en.wikipedia.org/wiki/Platinum>; <https://smaulgld.com/platinum-supply-and-demand>,
http://www.canplats.ca/html/Investor_Information/Platinum_Market/index.cfm, www.platinuminvestment.com

Platinum!

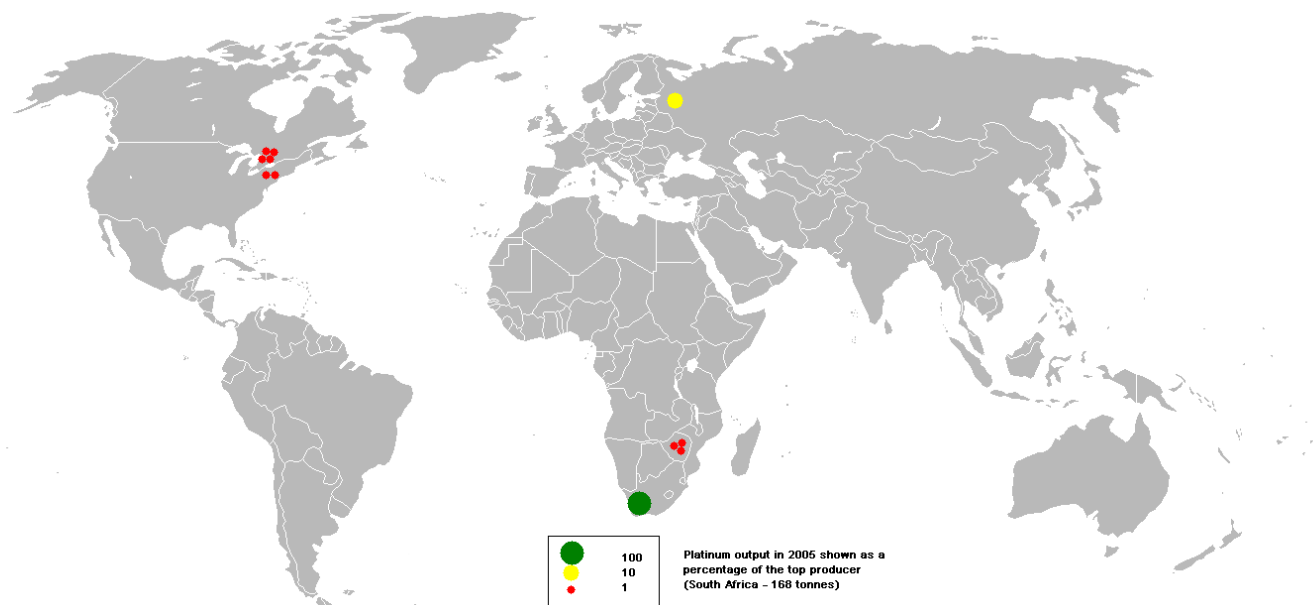


Source: <http://en.wikipedia.org/wiki/Platinum>; <https://smaulglld.com/platinum-supply-and-demand>,
http://www.canplats.ca/html/Investor_Information/Platinum_Market/index.cfm, www.platinuminvestment.com

Platinum!



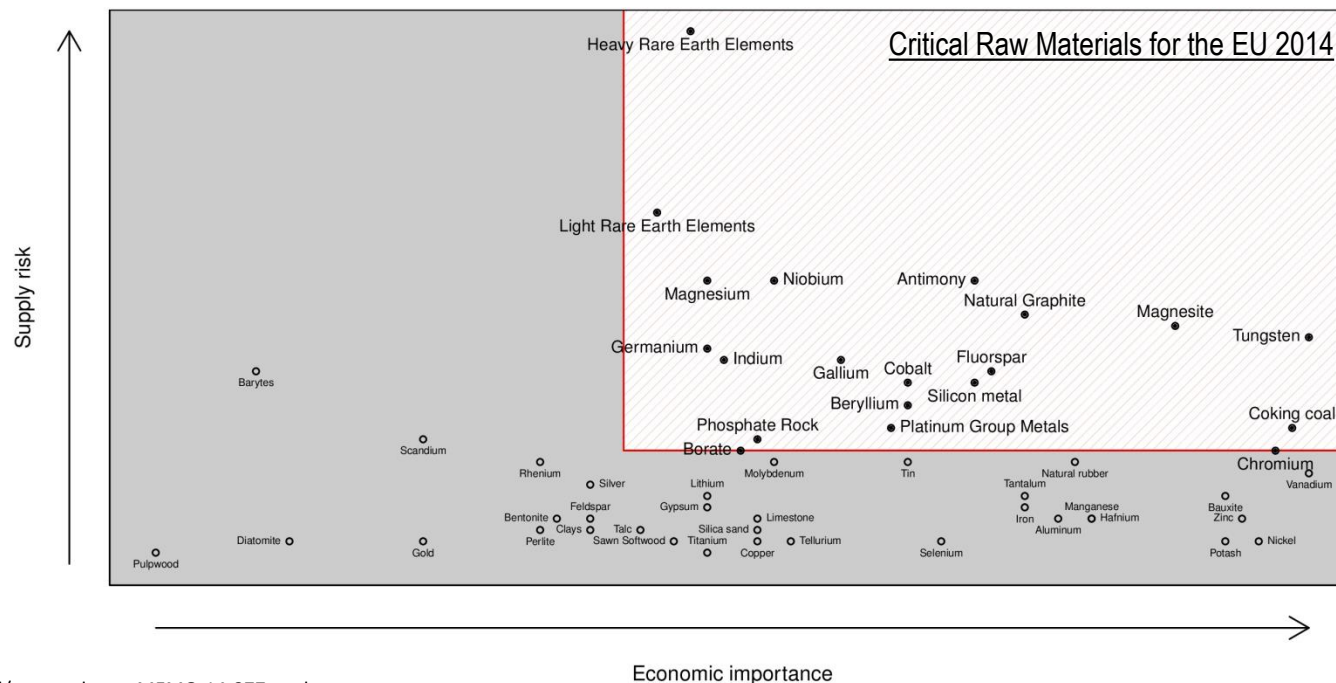
For EU urban mining is the obvious way to minimize import dependence from foreign countries.



Source: <http://en.wikipedia.org/wiki/Platinum>; <https://smaulgld.com/platinum-supply-and-demand>, http://www.canplats.ca/html/Investor_Information/Platinum_Market/index.cfm, www.platinuminvestment.com

Critical Raw Materials - CRMs

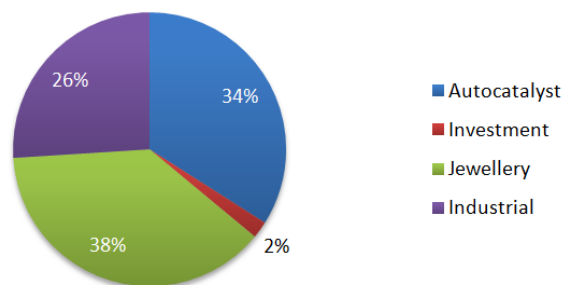
- EU makes a list of materials that are of supply risk to our well-being
- CRMs demand is increasing – High tech equipment like cell phones contain CRMs (also PGM)
- **Solution – Recycle and not export to third world countries (environment and health problems)**
- **Circular & Hydrogen Economy**



Platinum!

“Till now 4230 tons of Pt was mined.” – 6x6x6m cube

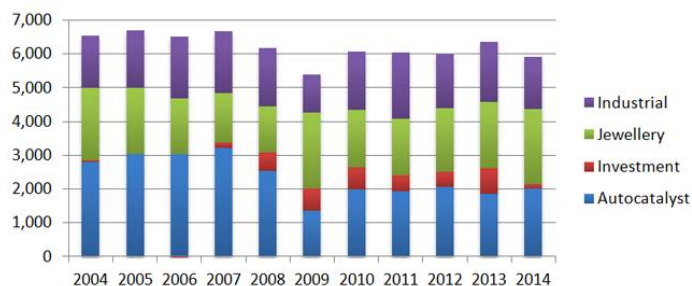
Platinum Net Demand By Application 2014 Total 184 tons net



Automotive Catalytic Converters -Now



Platinum Demand By Application

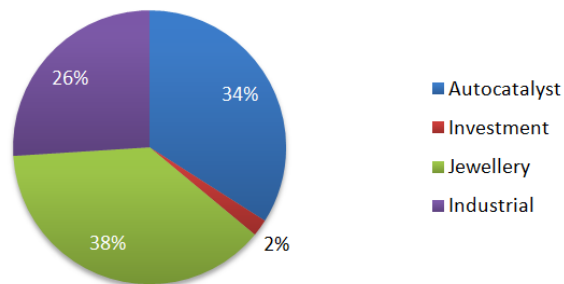


Source: <http://en.wikipedia.org/wiki/Platinum>; <https://smaulgl.com/platinum-supply-and-demand>, http://www.canplats.ca/html/Investor_Information/Platinum_Market/index.cfm, www.platinuminvestment.com

Platinum!

"Till now 4230 tons of Pt was mined." – 6x6x6m cube

Platinum Net Demand By Application 2014 Total 184 tons net

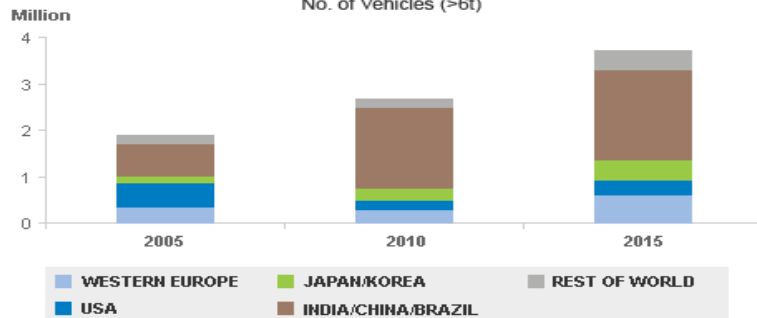


Automotive Catalytic Converters
-Now



Platinum Demand By Application

HEAVY DUTY VOLUMES WILL GROWTH
No. of Vehicles (>6t)



Source: Johnson Matthey, JD Power and IHS Automotive¹, 2011

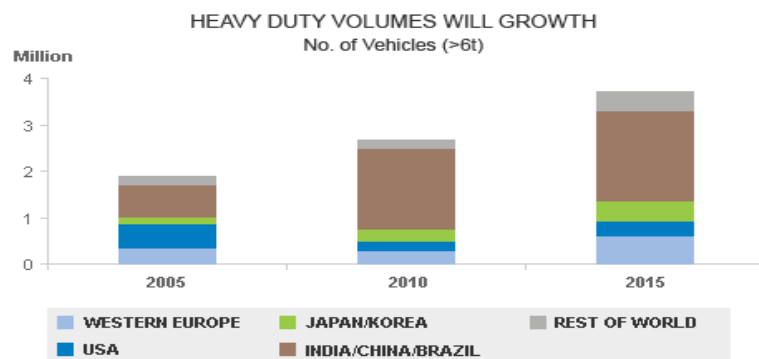
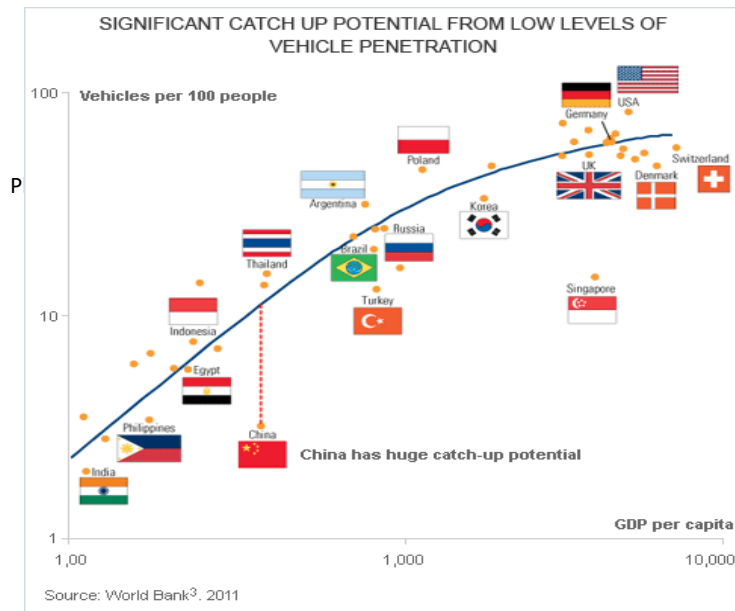
Source: <http://en.wikipedia.org/wiki/Platinum>; <https://smaulgld.com/platinum-supply-and-demand>,
http://www.canplats.ca/html/Investor_Information/Platinum_Market/index.cfm, www.platinuminvestment.com

“Till now 4230 tons of Pt was mined.” – 6x6x6m cube

Platinum!



Automotive Catalytic Converters
-Now



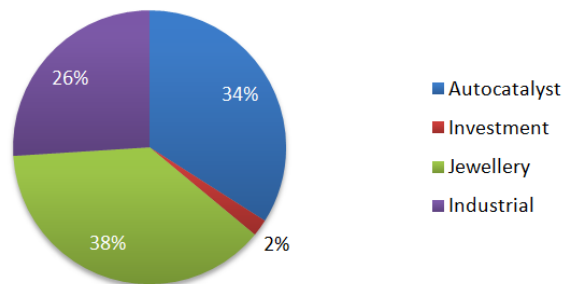
Source: Johnson Matthey, JD Power and IHS Automotive¹, 2011

Source: <http://en.wikipedia.org/wiki/Platinum>; <https://smaulgld.com/platinum-supply-and-demand>,
http://www.canplats.ca/html/Investor_Information/Platinum_Market/index.cfm, www.platinuminvestment.com

Platinum!

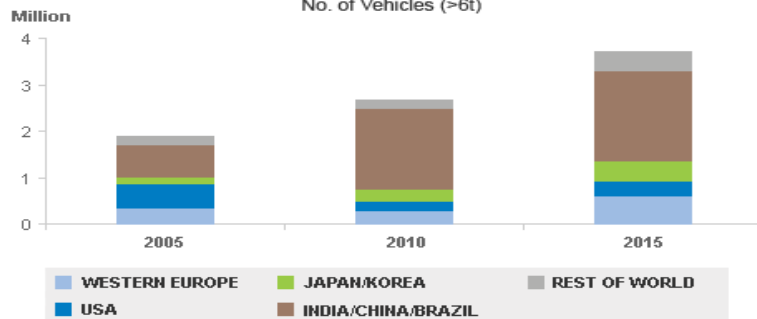
“Till now 4230 tons of Pt was mined.” – 6x6x6m cube

Platinum Net Demand By Application 2014 Total 184 tons net



Platinum Demand By Application

HEAVY DUTY VOLUMES WILL GROWTH
No. of Vehicles (>6t)



Source: Johnson Matthey, JD Power and IHS Automotive¹, 2011

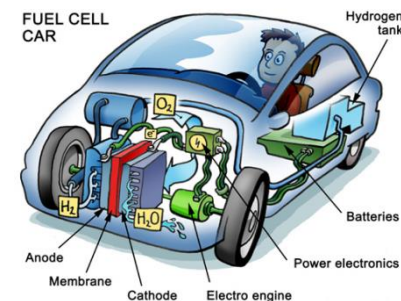


Automotive Catalytic Converters
-Now



3-15 g of Pt

Fuel Cell Cars
-Future



30-40 g of Pt

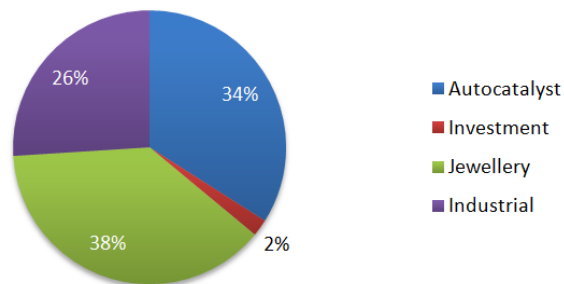
- Proven resources: 30,000 tons respectively.
- Annual production: 200 tons respectively (10% recycle).
- Reserves: in the range of 100 years.

Source: <http://en.wikipedia.org/wiki/Platinum>; <https://smaulgld.com/platinum-supply-and-demand>,
http://www.canplats.ca/html/Investor_Information/Platinum_Market/index.cfm, www.platinuminvestment.com

Platinum!

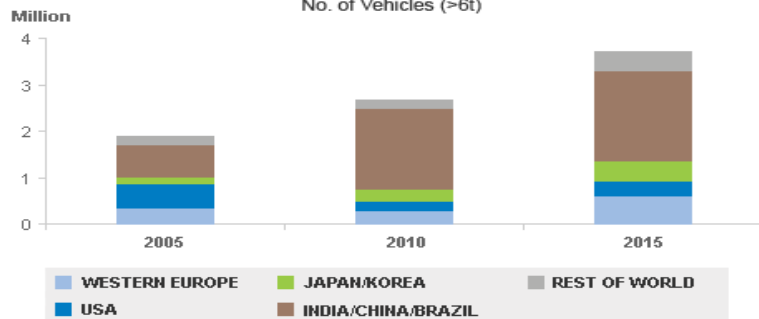
“Till now 4230 tons of Pt was mined.” – 6x6x6m cube

Platinum Net Demand By Application 2014 Total 184 tons net



Platinum Demand By Application

HEAVY DUTY VOLUMES WILL GROWTH
No. of Vehicles (>6t)



Source: Johnson Matthey, JD Power and IHS Automotive¹, 2011

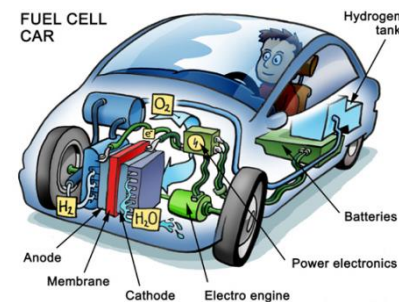


Automotive Catalytic Converters
-Now



3-15 g of Pt

Fuel Cell Cars
-Future



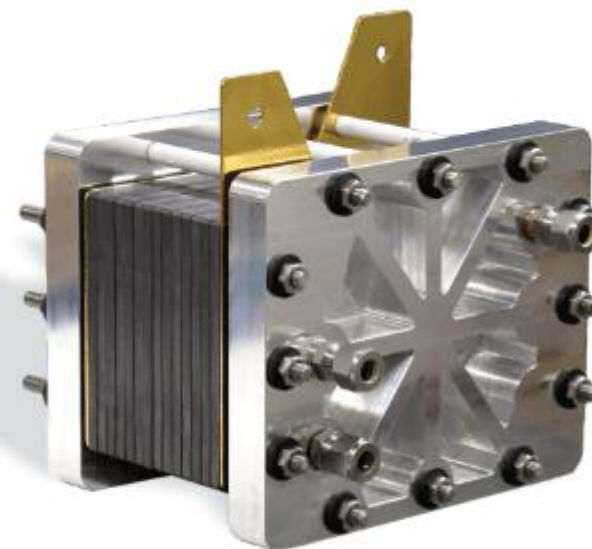
30-40 g of Pt

Mining Ores
-not in EU



Source: <http://en.wikipedia.org/wiki/Platinum>; <https://smaulgld.com/platinum-supply-and-demand>, http://www.canplats.ca/html/Investor_Information/Platinum_Market/index.cfm, www.platinuminvestment.com

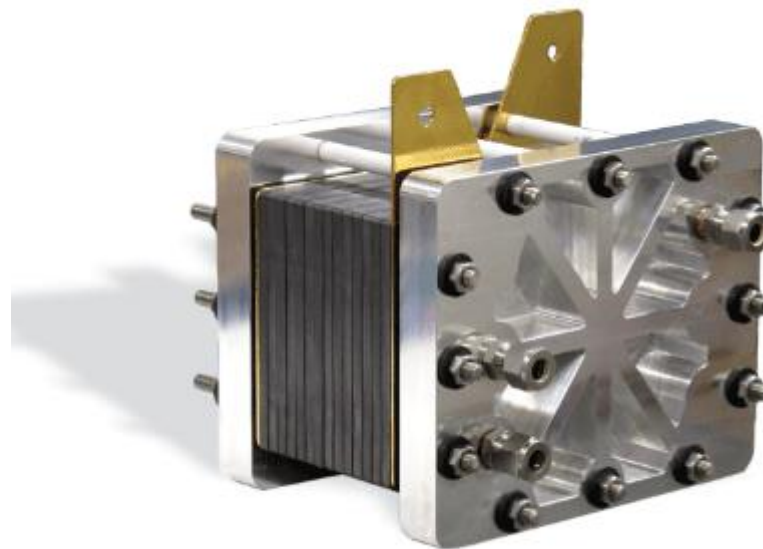
- recycling of PGM



- recycling of PGM



5 g



5 g ?

High Value Asteroid Materials

ASTEROID ELEMENTAL ABUNDANCE RELATIVE TO EARTH'S CRUST



Potable Water
Radiation Shielding
Fuel
Refrigerant
Agriculture Metallurgy

VOLATILES AND H₂O
to fuel the growth of
humanity into new frontiers



INDUSTRIAL METALS
to construct and
sustainably service space
platforms



Catalytic Converters
LCDs
Advanced materials
Cancer treatments

PLATINUM GROUP METALS
to support demand growth on
Earth



Despite desire to reduce dependency,
one-in-four manufactured goods require PGMs.

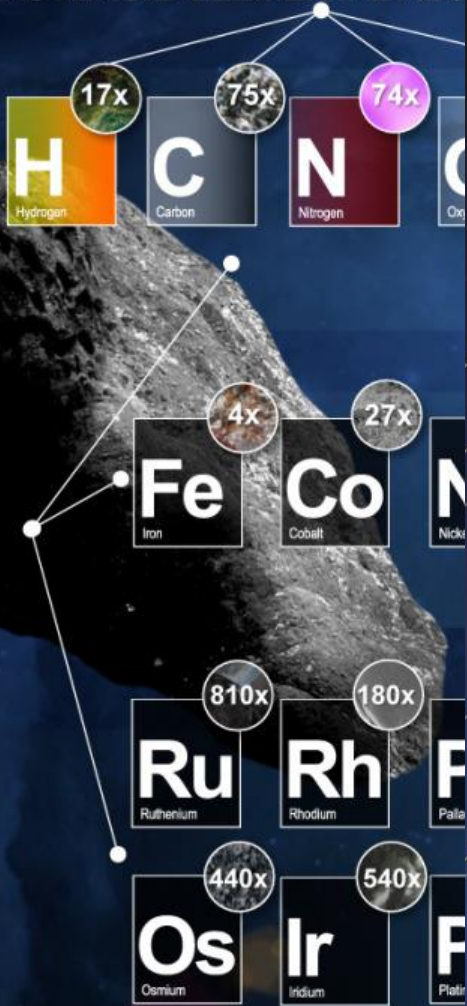
High Value

KNIGHT FRANK ASTEROID INDEX



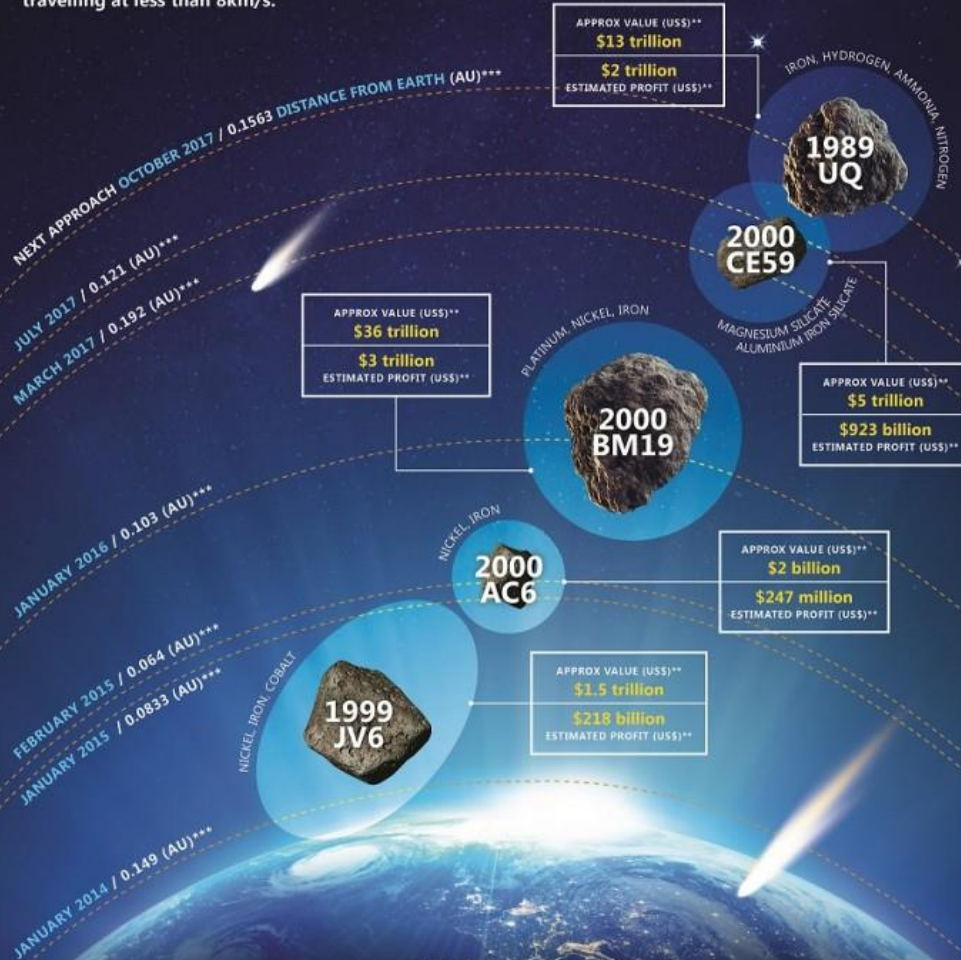
als

ASTEROID ELEMENTAL ABUNDANCE



ASTERIODS TO WATCH

Near-earth objects (including their main minerals) that will make an approach to Earth of less than 0.2AU*, travelling at less than 8km/s.



The Wealth Report 2014 highlights the rising trend for private wealth investment in space research — from asteroid mining to sub-orbital space travel. Knight Frank has identified more than 70 wealthy individuals, with a combined wealth of over US\$200 billion, who are targeting this sector and the potentially huge rewards that, as our graphic shows, could be on offer. To find out how this investment is even set to transform earth-based property markets go to KnightFrank.com/WealthReport

* 1AU or Astronomical unit is equivalent to 93 million miles

** Source: Asterank, <http://www.asterank.com/> 14 Jan 2014 Value estimates are based on the mass of a given asteroid and its spectral type.

Asteroid spectra is used to infer composition, which, in conjunction with current market prices, determine potential value.

*** Source: NASA <http://neo.jpl.nasa.gov/>

S AND H₂O
growth of
to new frontiers

AL METALS
t and
service space

GROUP METALS
demand growth on

dependency,
and goods require PGMs.

WEEE is a big environmental and health problem in developing countries!

- e-waste disposal is still not controlled; 75% ends up illegally in developing countries.
– Huge business and Huge problem!
- Only 12.5% of global e-waste is recycled where issues are that recycling is difficult, dangerous, expensive and toxic. Developing countries are dumping sites for e-waste.
- Noble metals demand will only increase in the future!

WEEE is a big environmental and health problem

COMMENT

ECONOMICS Joseph Stiglitz's guide to the failures of eurozone policymakers **p.26**



DEVELOPMENT Parenting tips from the lab of a 'Berkeley bubble' **p.27**

CHINA A bold and epic portrait of a civilization shaped by water **p.28**

CAPACITY-BUILDING Mentoring scheme supports female scientists in Cameroon **p.30**



An electronic-waste recycling factory in Hubei, China.

Take responsibility for electronic-waste disposal

International cooperation is needed to stop developed nations simply offloading defunct electronics on developing countries, argue Zhaohua Wang, Bin Zhang and Dabo Guan.

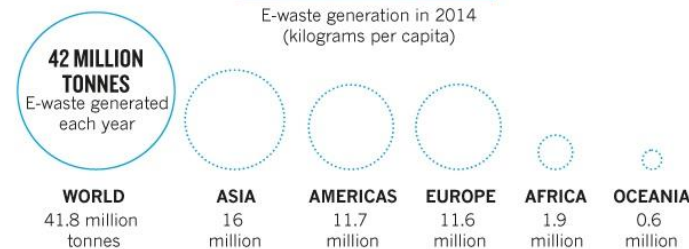
The world is producing ever more electrical and electronic waste. The quantity of dumped computers, telephones, televisions and appliances doubled between 2009 and 2014, to 42 million tonnes per year globally^{1,2}. Developed countries, especially in North America and Europe, produce

the most e-waste (see 'Unfair flow'). The United States generates the largest amount, and China the second most³. Much of this waste ends up in the developing world, where regulation is lax. China processed about 70% of the world's e-waste in 2012⁴; the rest goes to India and other countries in eastern Asia and Africa,

including Nigeria⁵. Non-toxic components — such as iron, steel, copper and gold — are valuable, so are more frequently recycled than toxic ones⁶. Disposal plants release toxic materials, volatile organic chemicals and heavy metals, which can harm the environment and human health. Lead levels sampled in the blood ▶

UNFAIR FLOW

Most electronic waste from developed countries ends up in poor nations that lack regulation. China processed around 70% of the world's e-waste in 2012; the rest goes to India and other countries in eastern Asia and Africa, including Nigeria.



- The United States produces the largest total amount of e-waste per year, at 7.1 million tonnes.
- Norway generates the most e-waste per person, at 28.3 kg per capita.
- African nations produce little e-waste, with Equatorial Guinea creating most (10.8 kg per capita).
- China ranks second for total e-waste generation (6 million tonnes), but low relative to its population size (4.4 kg per capita).

©nature

WEEE is a big environmental and health problem

COMMENT

e-

ECONOMICS Joseph Stiglitz's guide to the failures of...

DEVELOPMENT Parenting tips from the lab of a 'Backlash bubble'...

CHINA A bold and epic portrait of a civilization shaped by...

CAPACITY-BUILDING Mentoring scheme supports female scientists in Cameroon...


UNFAIR FLOW

Most electronic waste from developed countries ends up in poor nations that lack regulation. China processed around 70% of the world's e-waste in 2012; the rest goes to India and other countries in eastern Asia and Africa, including Nigeria.



- 1 The United States produces the largest total amount of e-waste per year, at 7.1 million tonnes.
- 2 Norway generates the most e-waste per person, at 28.3 kg per capita.
- 3 African nations produce little e-waste, with Equatorial Guinea creating most (10.8 kg per capita).
- 4 China ranks second for total e-waste generation (6 million tonnes), but low relative to its population size (4.4 kg per capita).

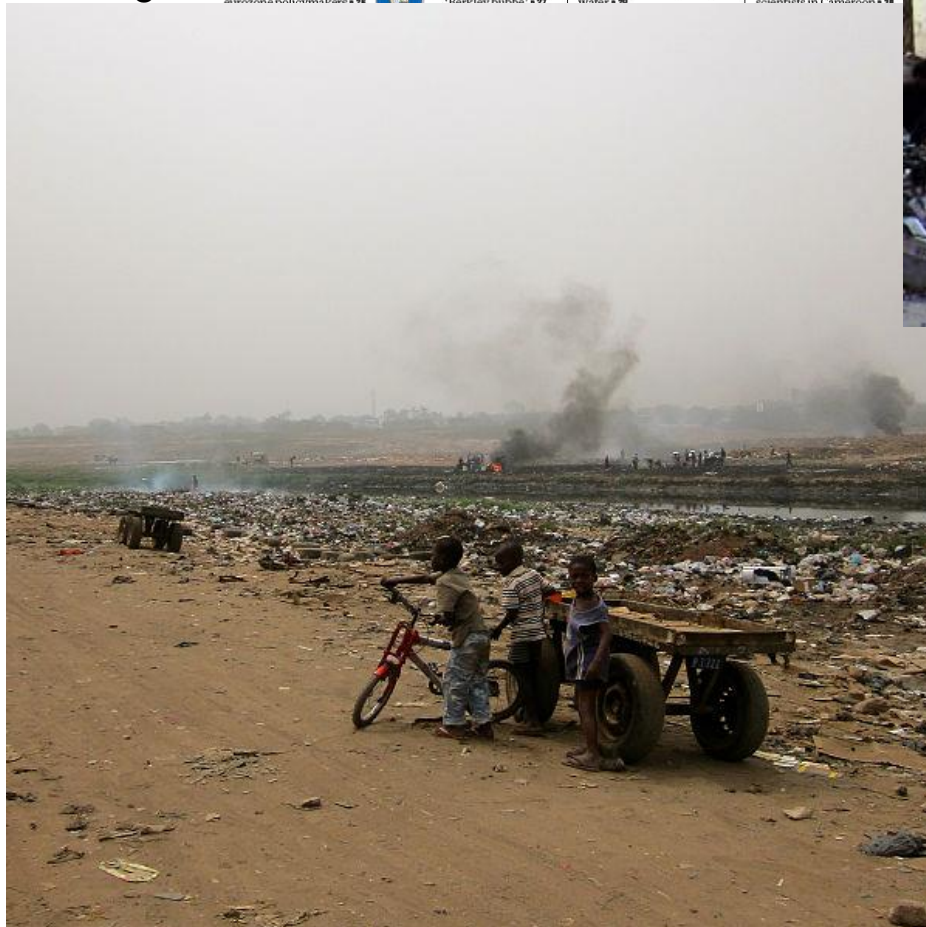
©nature

S.
t,
ste.

WEEE is a big environmental and health problem

COMMENT

E- **ECONOMICS** Joseph Stiglitz's guide to the failures of **EURO** **DEVELOPMENT** Parenting tips from the lab of a **CHINA** A bold and epic portrait of a civilization shaped by **CAPACITY-BUILDING** Mentoring scheme supports female scientists in Cameroon



- The United States produces the largest total amount of e-waste per year, at 7.1 million tonnes.
- Norway generates the most e-waste per person, at 28.3 kg per capita.
- African nations produce little e-waste, with Equatorial Guinea creating most (10.8 kg per capita).
- China ranks second for total e-waste generation (6 million tonnes), but low relative to its population size (4.4 kg per capita).

©nature

WEEE is a big environmental and health problem

COMMENT

• e-

ECONOMICS Joseph Stiglitz's guide to the failures of



DEVELOPMENT Parenting tips from the lab of a

CHINA A bold and epic portrait of a civilization shaped by

CAPACITY-BUILDING Mentoring scheme supports female scientists in Cameroon



WEEE is a big environmental and health problem



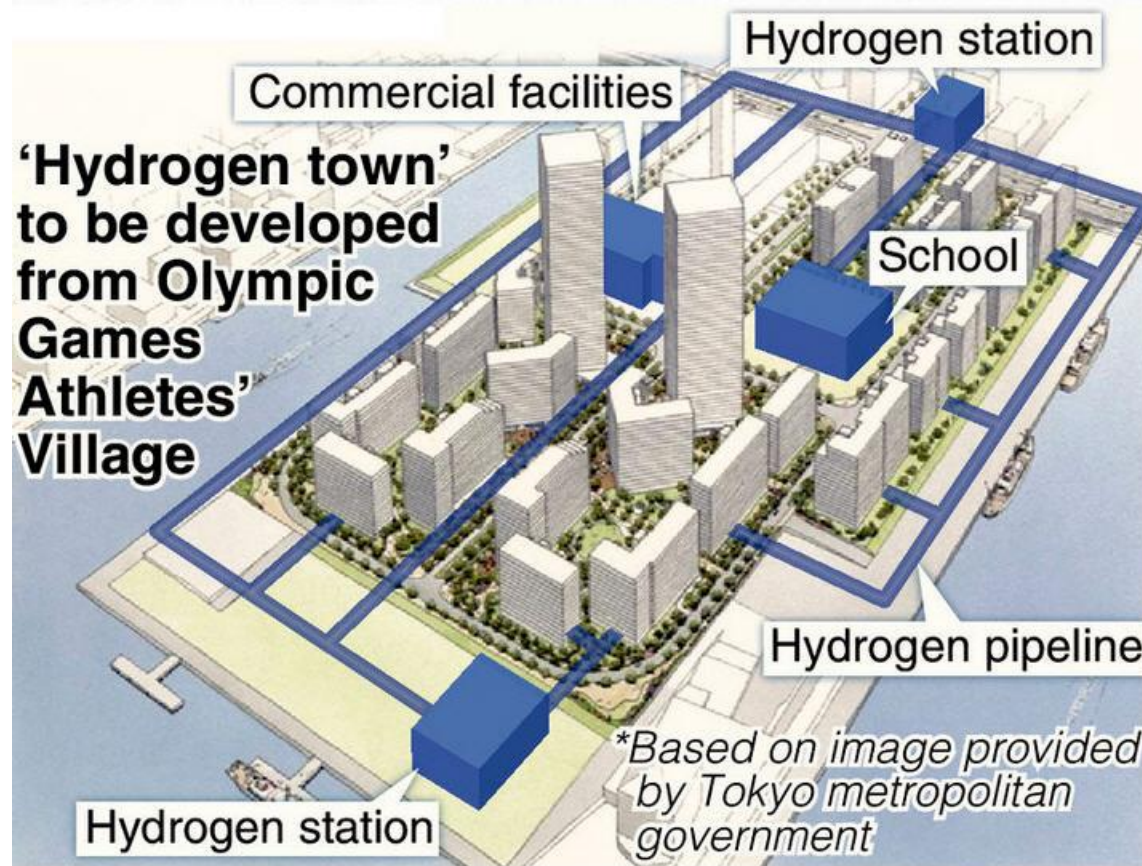
2020 Japan Olympics

“Tokyo 2020 medals could be made of electronic trash: Olympic organizers look to 'urban mine' of e-waste to find gold and silver.”



2020 Japan Olympics

“Japan is planning to turn the athletes' village for the 2020 Olympics in Tokyo into a "hydrogen town", where electricity and hot water are generated from hydrogen.”



The Yomiuri Shimbun

Thank you for your attention!

“Besides improving the technology we should also try to adapt human habits.”

