



Calculation of results for Po-210 in water by alpha spectrometry

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Bi-210 in-growth from Pb-210 at different sample thickness







Calculation of results

$$A_{\rm Pb-210} = \frac{\left(R_{\rm beta} - R_{\rm b,beta}\right)}{\eta_{\rm Pb-210} \,\varepsilon_{\rm Pb-210} \,V_{\rm s}}$$
(1)
$$R_{X} = \frac{N_{\rm X}}{t_{\rm m}}$$
(2)
$$\eta_{\rm Pb-210} = \frac{m_{\rm PbSO4} \,M_{\rm Pb}}{m_{\rm Pb} \,M_{\rm PbSO4}}$$
(3)

$$A_{Pb-210} \rightarrow activity concentration of Pb - 210 [Bq/L]$$

 $R_{\text{beta}} \rightarrow \text{beta count rate } [1/s]$

$$R_{b,beta} \rightarrow beta background count rate [1/s]$$

$$\eta_{Pb-210} \rightarrow Pb - 210$$
 chemical recovery

$$\epsilon_{Pb-210} \rightarrow Pb - 210$$
 detection efficiency

 $V_{\rm s} \rightarrow \text{sample volume [L]}$

$$R_{\rm X} \rightarrow \text{count rate of radionuclide X or background } [1/s]$$

 $N_{\rm X} \rightarrow$ number of counts of radionuclide X or background

 $t_{\rm m} \rightarrow$ measurement time [s]

 $m_{PbSO4} \rightarrow mass of PbSO_4$ on planchet after separation [g] $M_{PbSO4} \rightarrow molar mass of PbSO_4$ [g/mol] $m_{Pb} \rightarrow mass of added Pb carrier before separation [g]$ $M_{Pb} \rightarrow molar mass of Pb [g/mol]$





Calculation of measurement uncertainty

$$u_{c,Pb-210} = A_{Pb-210} \sqrt{\left(\frac{u_{R_{beta}-R_{b,beta}}}{R_{beta}-R_{b,beta}}\right)^{2} + \left(\frac{u_{\eta_{Pb-210}}}{\eta_{Pb-210}}\right)^{2} + \left(\frac{u_{\varepsilon_{Pb-210}}}{\varepsilon_{Pb-210}}\right)^{2} + \left(\frac{u_{V_{s}}}{V_{s}}\right)^{2}}$$
(4)
$$u_{R_{beta}-R_{b,beta}} = \sqrt{\left(u_{R_{beta}}\right)^{2} + \left(u_{R_{b,beta}}\right)^{2}}$$
(5)
$$u_{R_{X}} = \frac{1}{\sqrt{N_{X}}}$$
(6)

 $u_{c,Pb-210} \rightarrow \text{combined standard uncertainty for Pb} - 210[Bq/L]$ $u_X \rightarrow \text{standard uncertainty of X}$





Reporting of the results

 $U_{\rm Pb-210} = k \, u_{\rm c,Pb-210} \quad (7)$

 $U_{Pb-210} \rightarrow expanded uncertainty for Pb - 210 activity concentration [Bq/L]$ $k \rightarrow coverage factor (k = 2 for 95\% coverage)$

 $A_{\rm Pb-210} = A_{\rm Pb-210} \pm U_{\rm Pb-210}$







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