



Introduction to sampling and sample pre-treatment techniques

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Sampling

- Sampling is the most important step errors made here can not be corrected afterwards
- We can take different kind of samples (water, soil, sediment, biological material, air, dust particles, concrete,...)
- Sampling must be representative
- Prevent contamination and cross-contamination during sampling









Preserving sample integrity

- To prevent adverse change in analyte concentration or speciation in time between sampling and analysis in the laboratory
- Example: water sample
 - filtration right after the sampling (0.45 μ m)
 - change pH to acidic by adding acid to prevent radionuclide loss from water







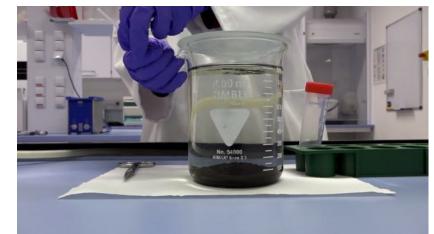


Sample pre-treatment: water

- Activity concentrations are usually very low => need to pre-concentrate radionuclides
- Co-precipitation:

 $FeOH_3$ MnO₂ PbSO₄ CaPO₃

- Evaporation
- Addition of suitable tracers for recovery before sample pre-treatment







Sample pre-treatment: biological material

- Separation of specific organs or parts of interes or species Fish: muscle, bones, liver, spleen,...
 Plants: leaves, shots, roots
- Drying (air, oven, freeze drying, depending on volatility of radionuclide)
- Homogenisation (grinding, milling, sieving, mixing, subsampling)
- Reducing sample mass and removing organic material by ashing (for non-volatile radionuclides)
- Addition of suitable tracers for recovery
- Digestion, decomposition, leaching with acids, either on hot plate, microvawe or alkaline fusion









Sample pre-treatment: soil, sediment

- Drying (air, oven, depending on volatility of radionuclide)
- Removal of large stones and roots
- Homogenisation (grinding, milling, sieving, mixing, subsampling)
- Removing organic material by ashing for high organic content soil or sediment (for non-volatile radionuclides)
- Addition of suitable tracers for recovery
- Digestion, decomposition, leaching with acids, either on hot plate, microvawe or alkaline fusion











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