

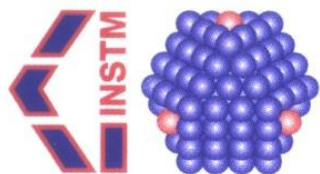
# Nanocarbon-metal oxide junctions in electrocatalytic energy processes

*Michele Melchionna*

*Email: [melchionnam@units.it](mailto:melchionnam@units.it)*



UNIVERSITÀ  
DEGLI STUDI DI TRIESTE



Ljubljana  
5<sup>th</sup> February  
2020

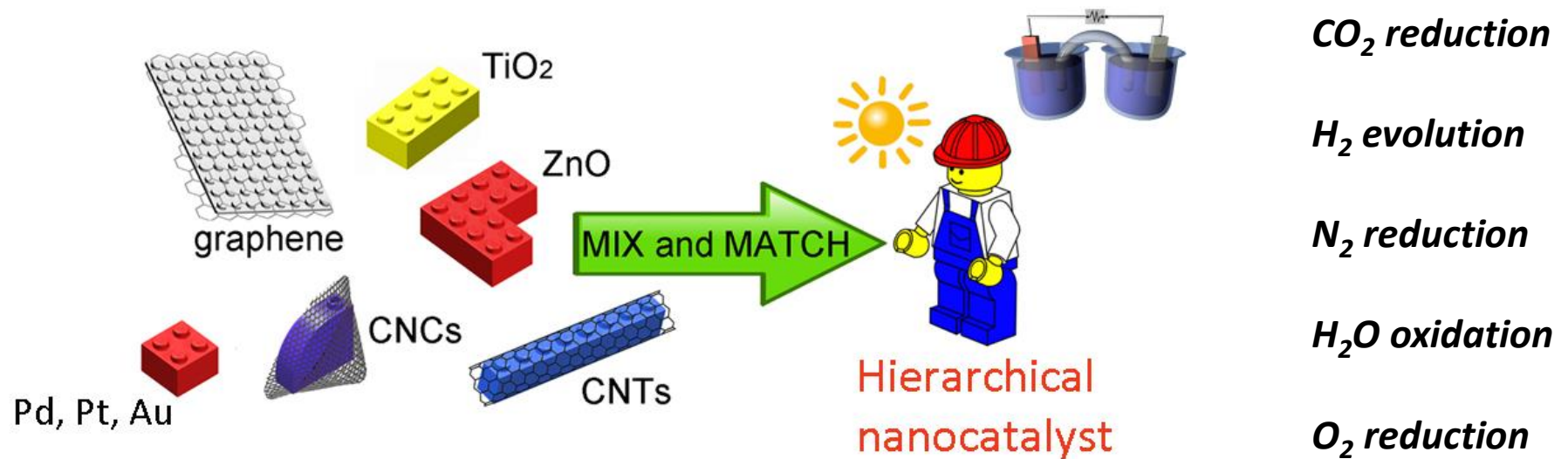
# Nanostructure-based (electro)catalysis

*“God made solids, but surfaces  
were the work of the Devil”*

Wolfgang Pauli



# Carbon nanostructures/metal oxide interfaces



## Photocatalysis:

**Semiconducting metal oxide:**  
Excited charges generation

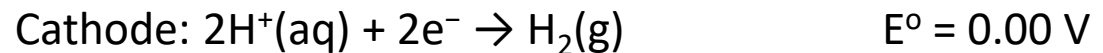
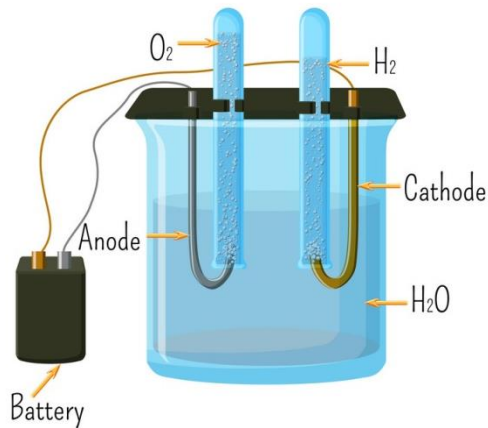
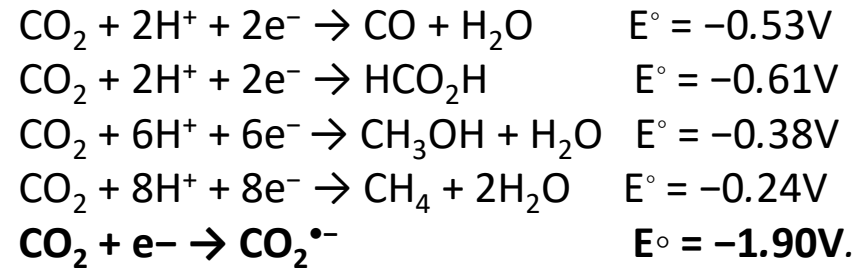
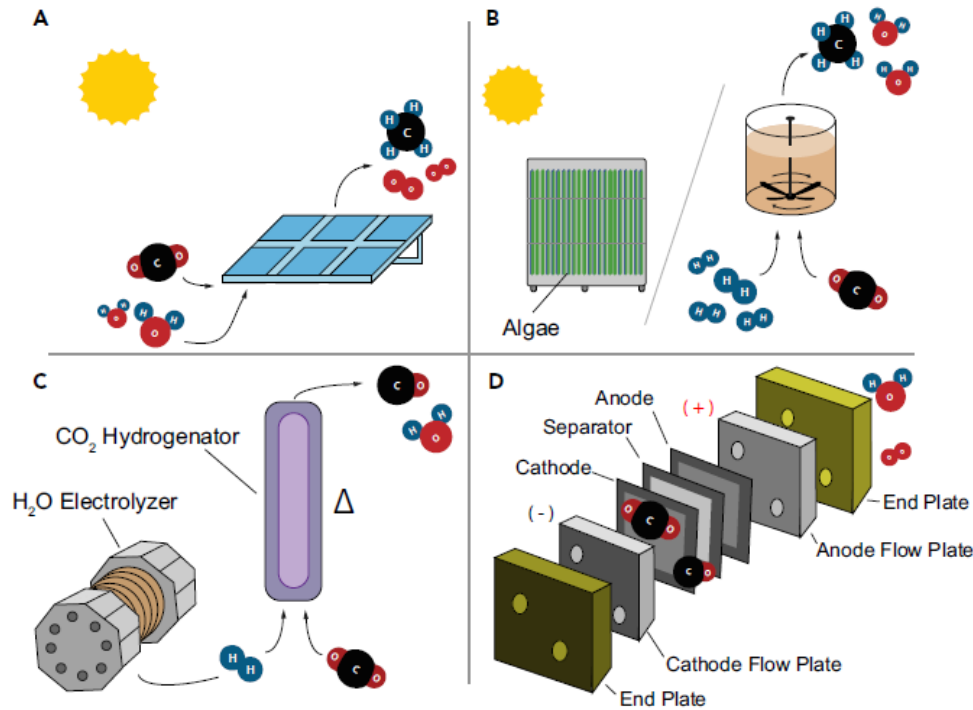
**Nanocarbon:** electron or hole sink.  
Retard of the recombination rate

## Electrocatalysis:

**Semiconducting metal oxide:**  
Different roles (catalyst, capturing agent, co-catalyst...)

**Nanocarbon:** improve electron transfer

# Carbon nanostructures/metal oxide interfaces

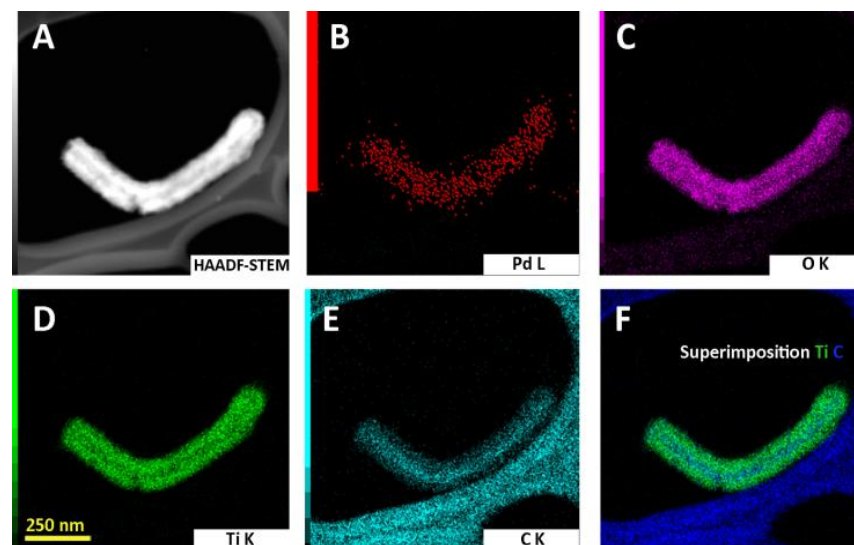
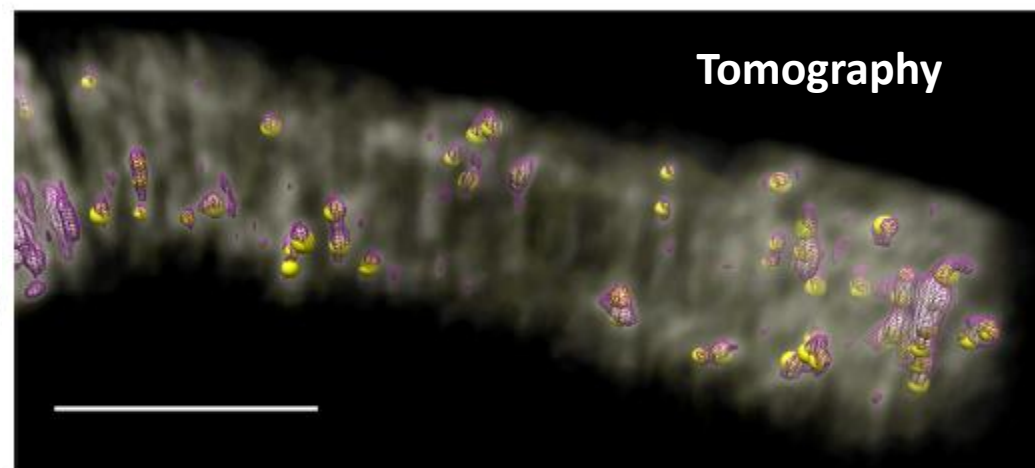
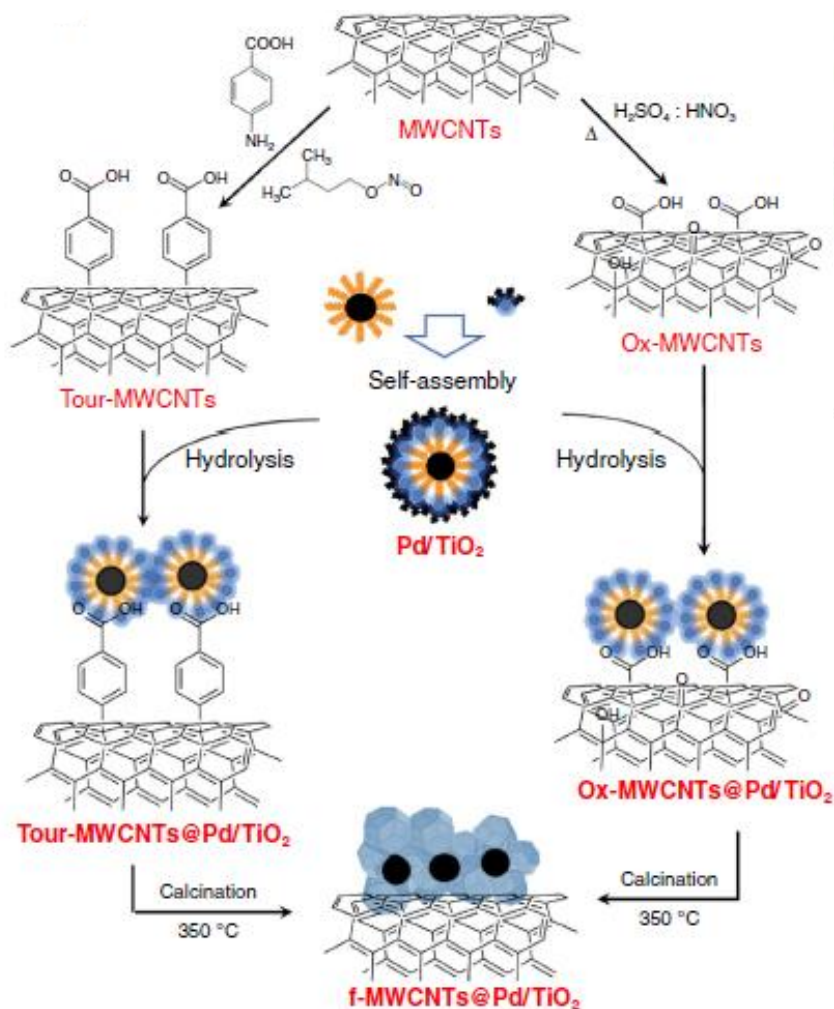


$$\Delta G = 237\text{ kJ mol}^{-1}$$

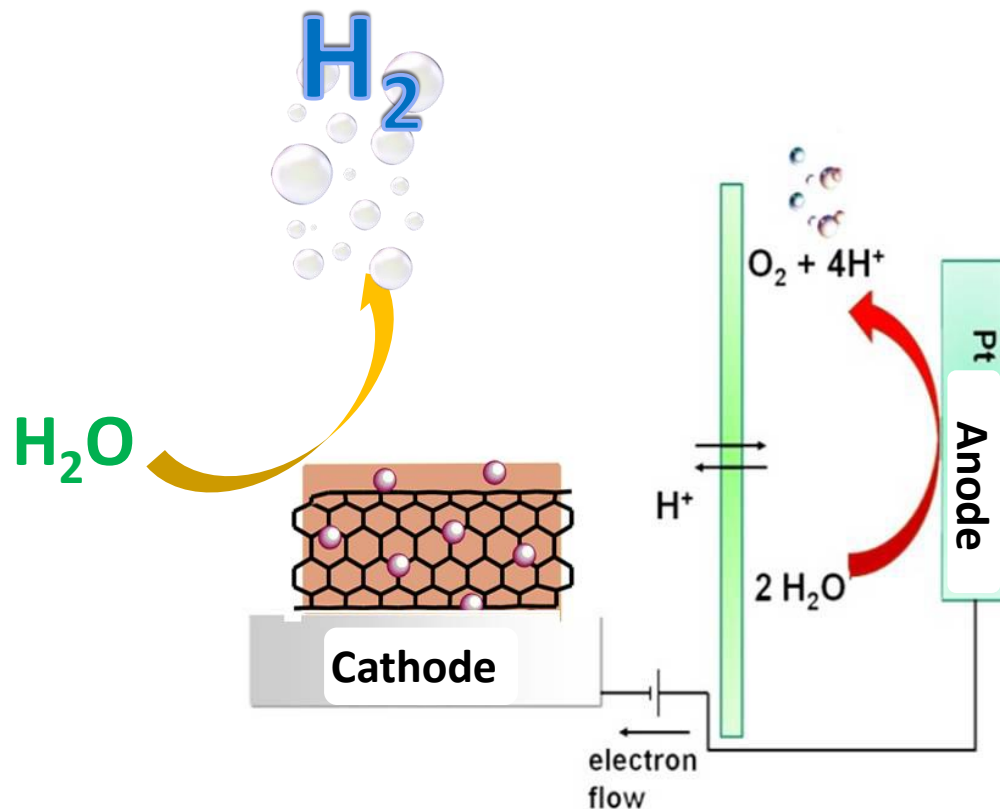
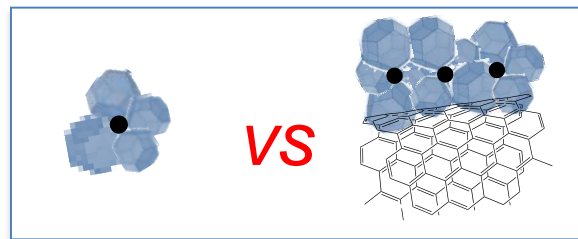
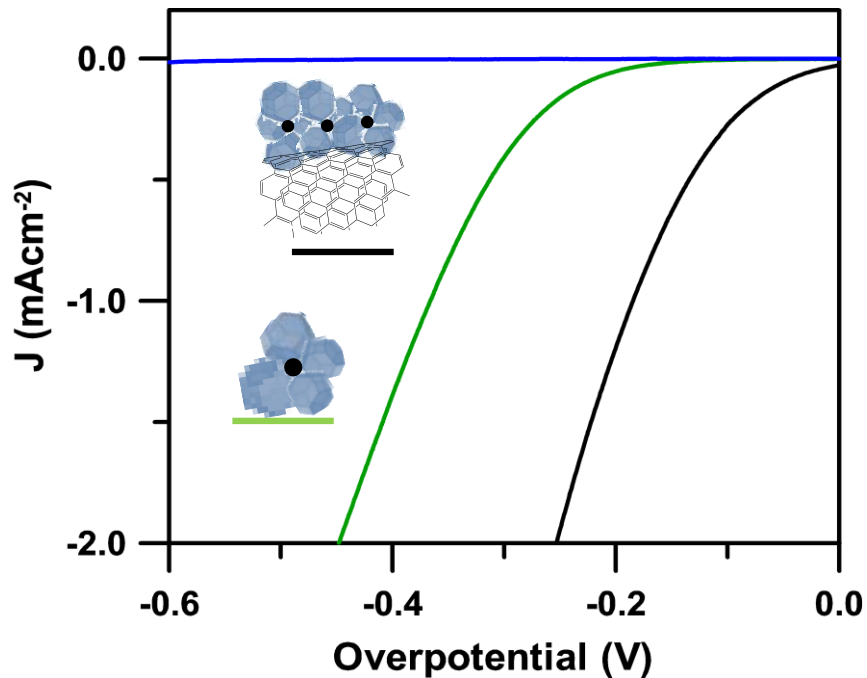




# A versatile hierarchical material: CNT@TiO<sub>2</sub>@Pd

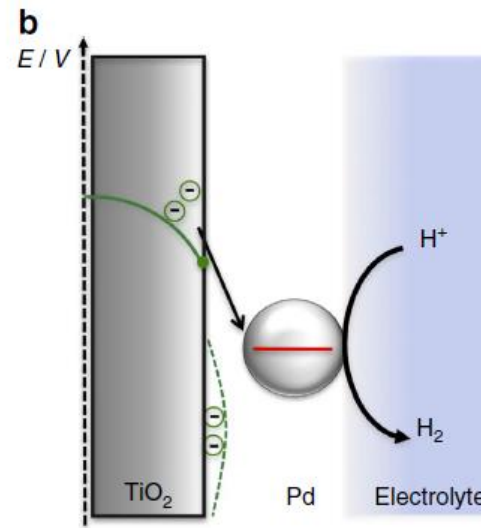
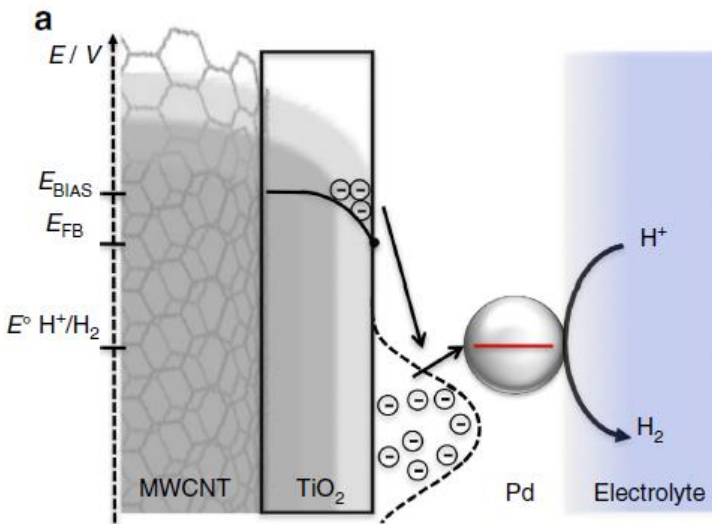
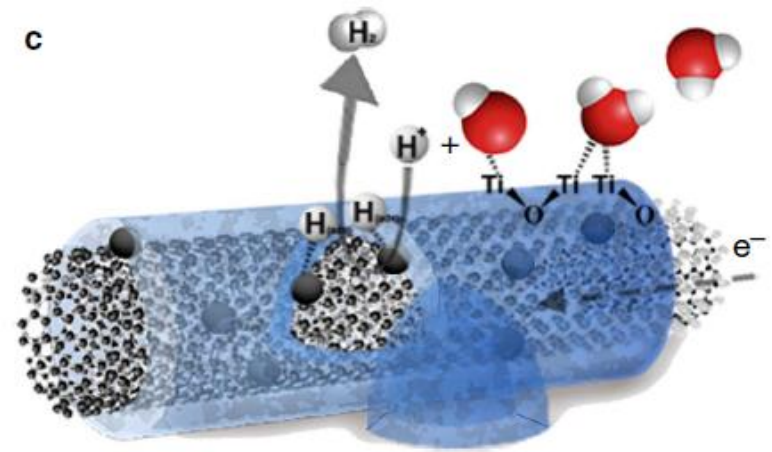
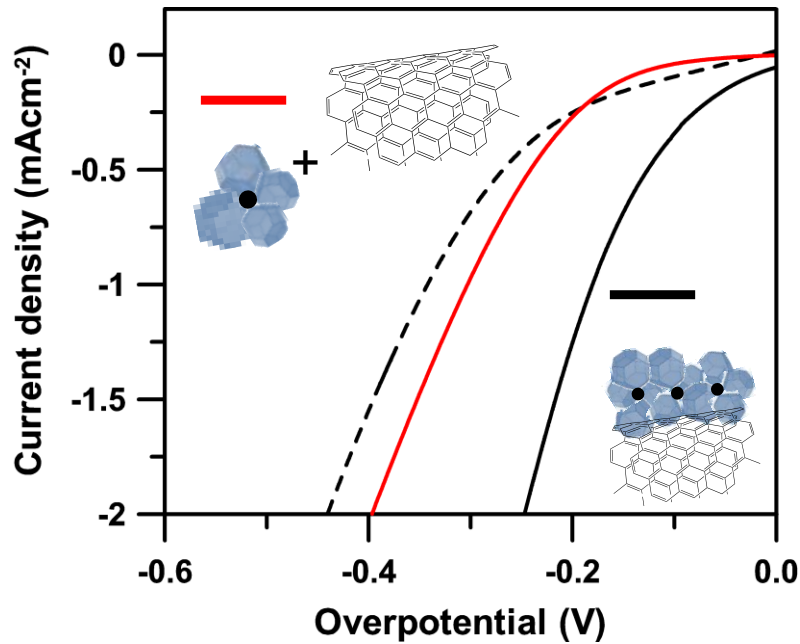


# Electrocatalytic H<sub>2</sub> production: Effect of the CNTs



**200 mV reduction in the overpotential**

# Effect of the interfacial and nanostructuring



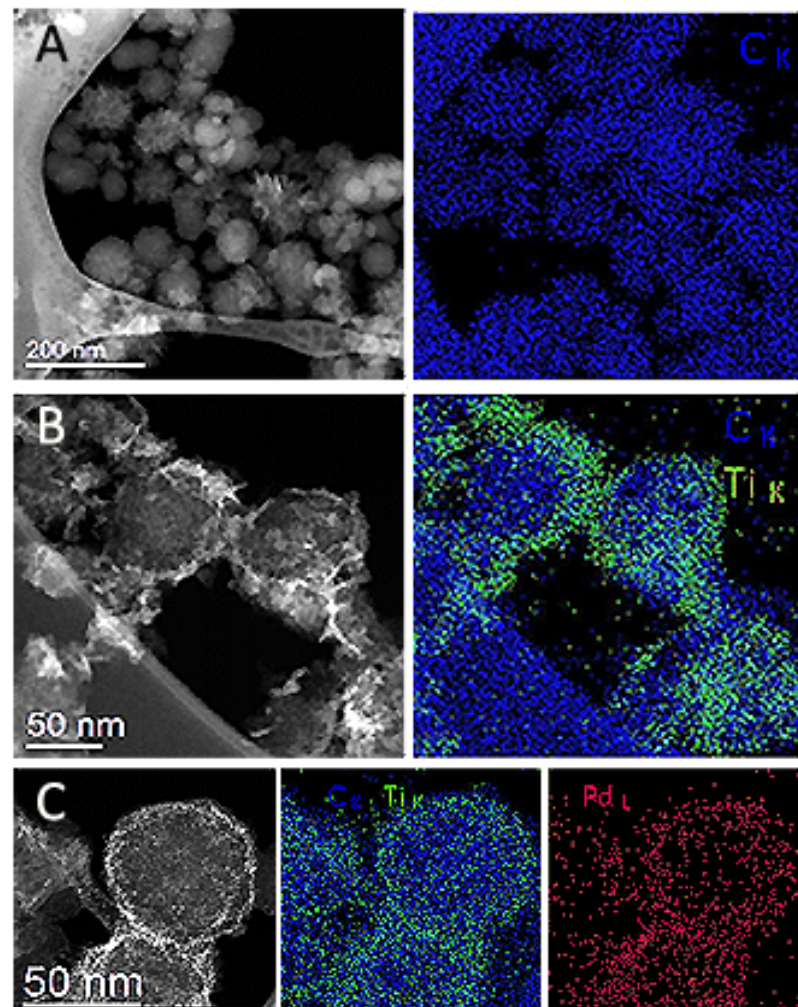
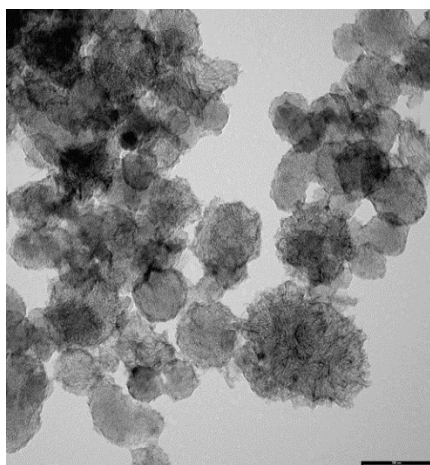
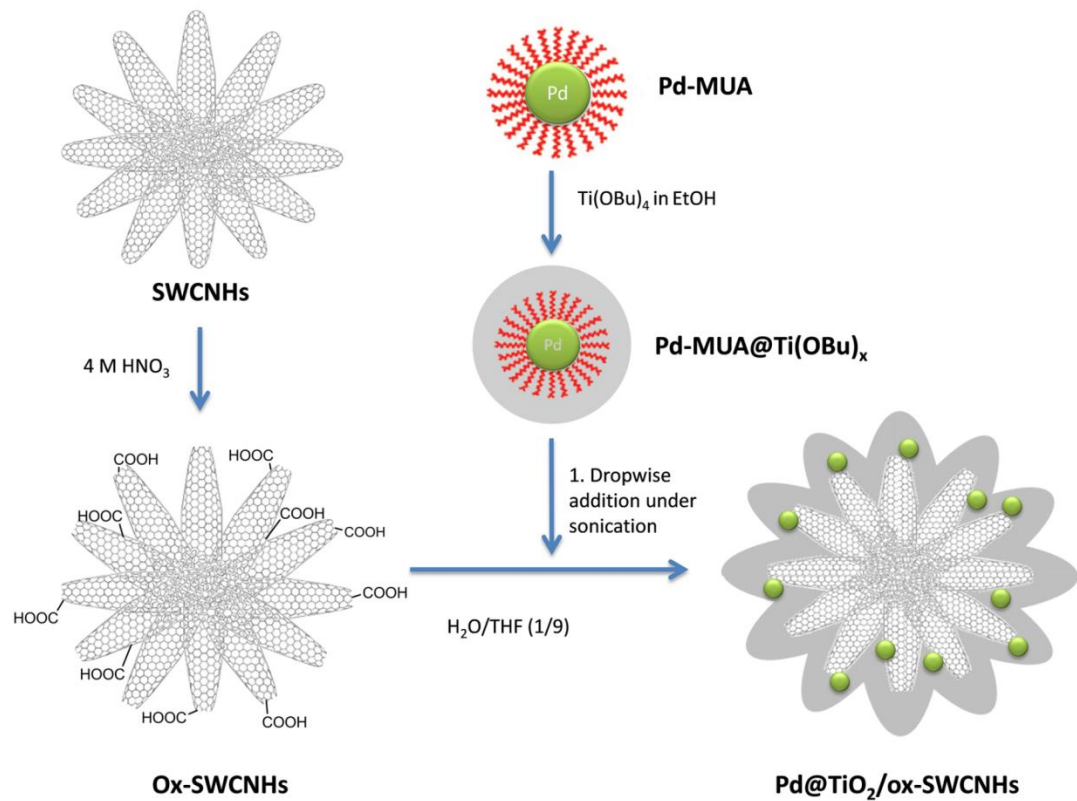
Presence of surface state  
at the interface  
CNT/TiO<sub>2</sub> : free electrons

Better coupling with Pd  
nanoparticles

Faster kinetics

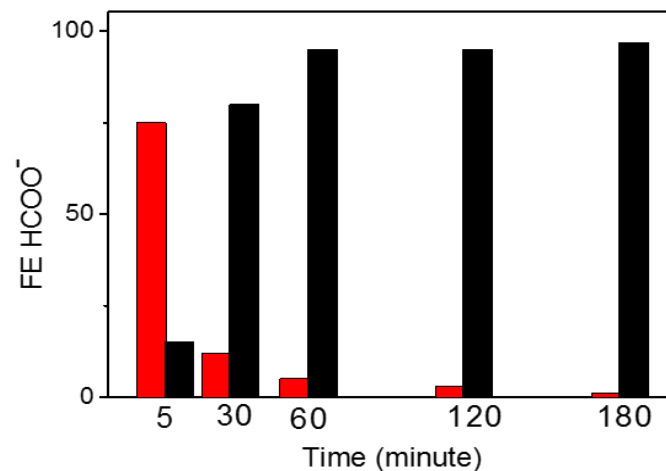
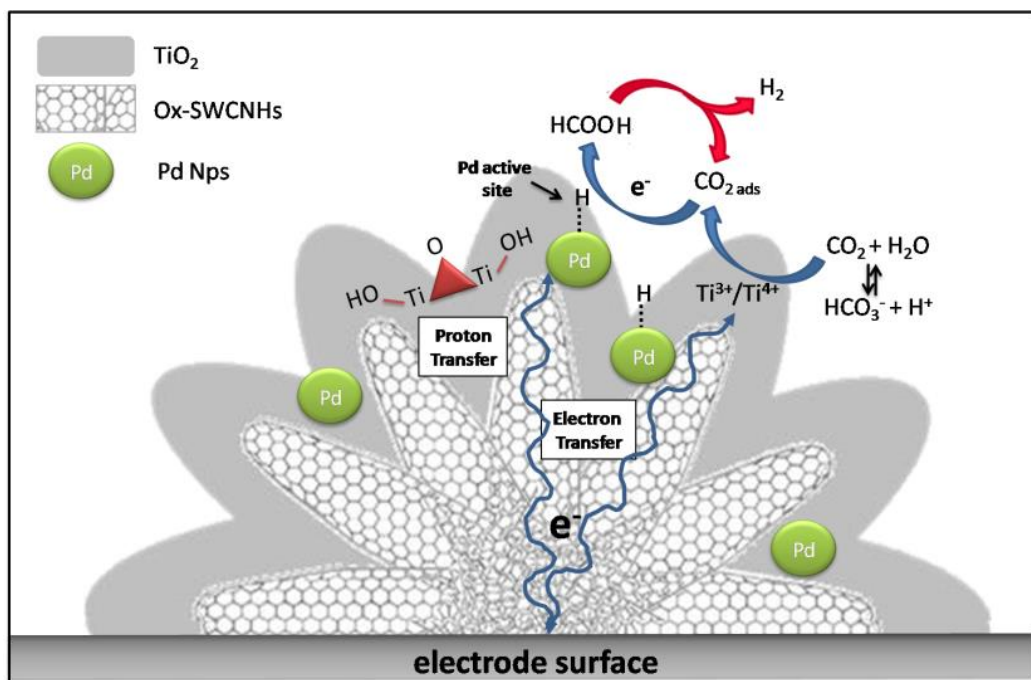
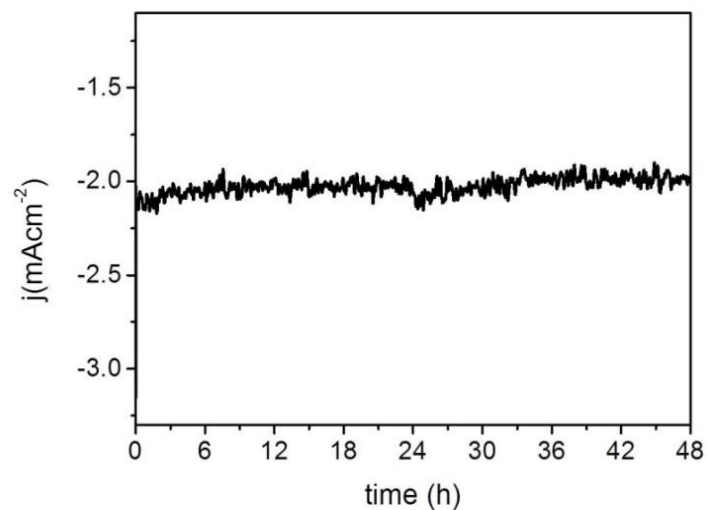
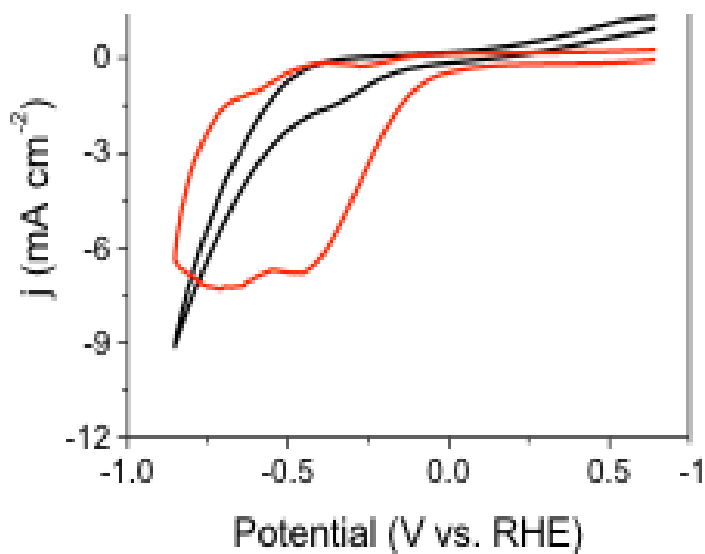


# CO<sub>2</sub> reduction: Carbon nanohorns@TiO<sub>2</sub>@Pd

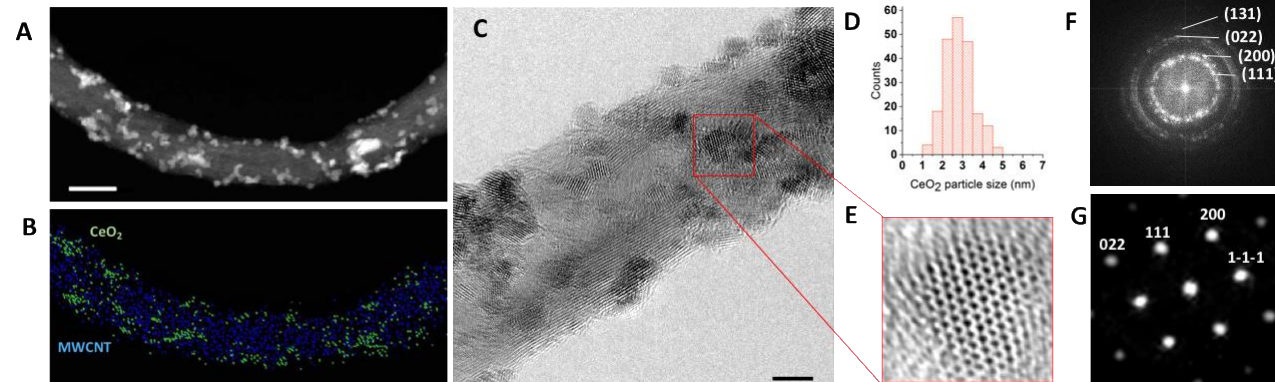
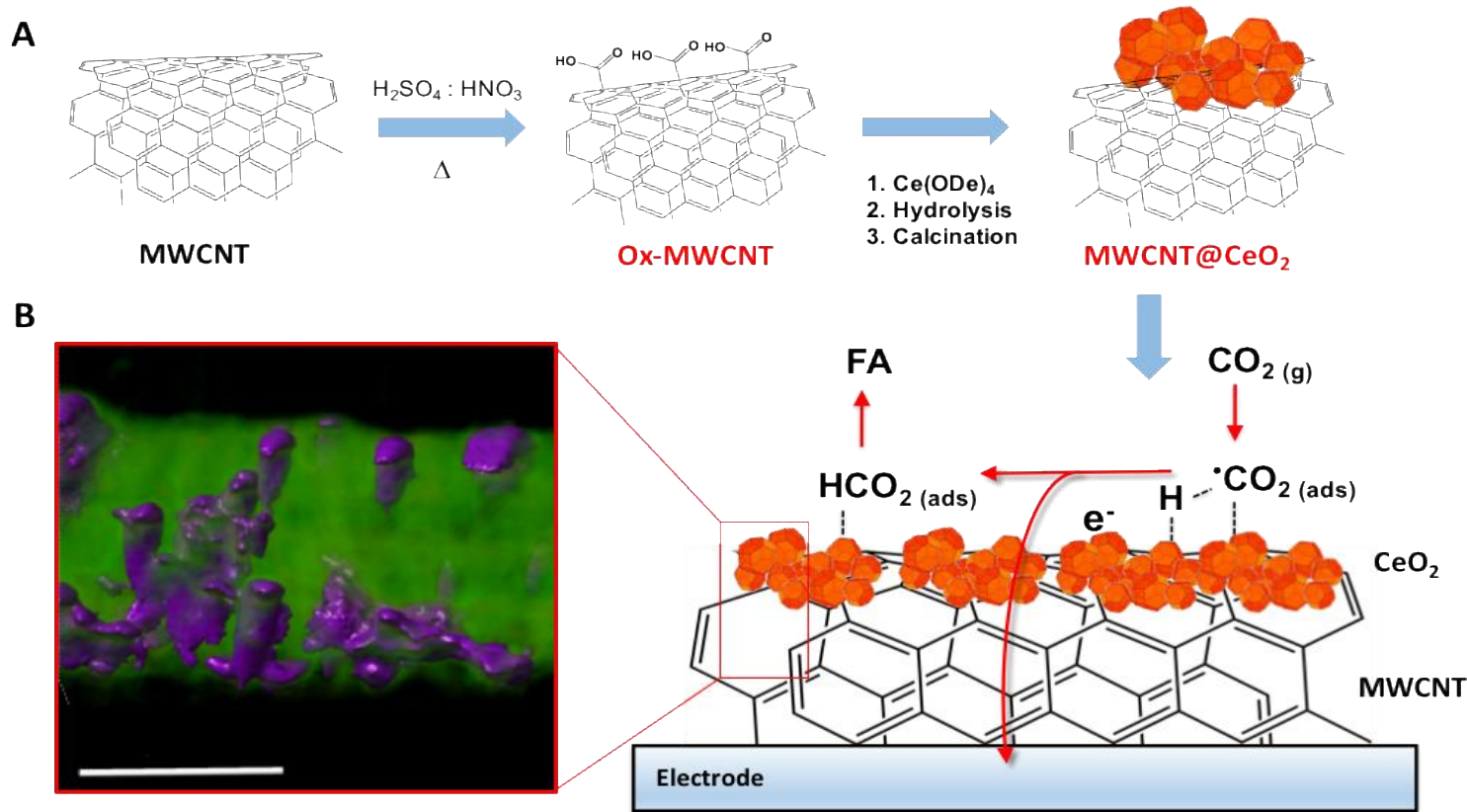




# CO<sub>2</sub> reduction: CO<sub>2</sub> electroreduction to formic acid

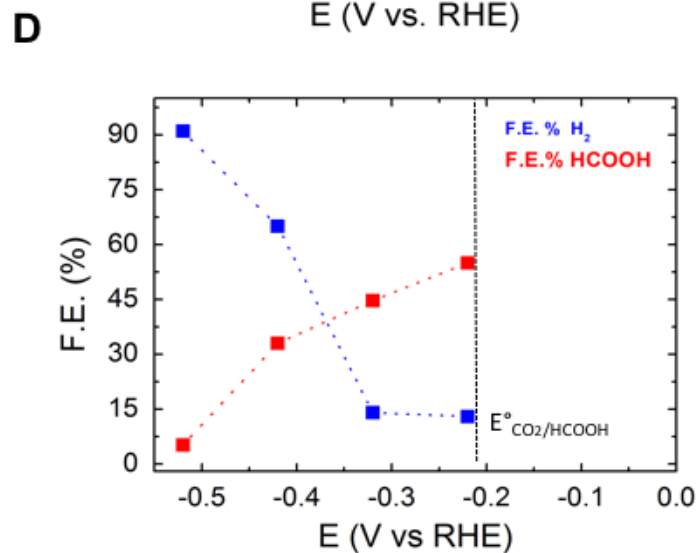
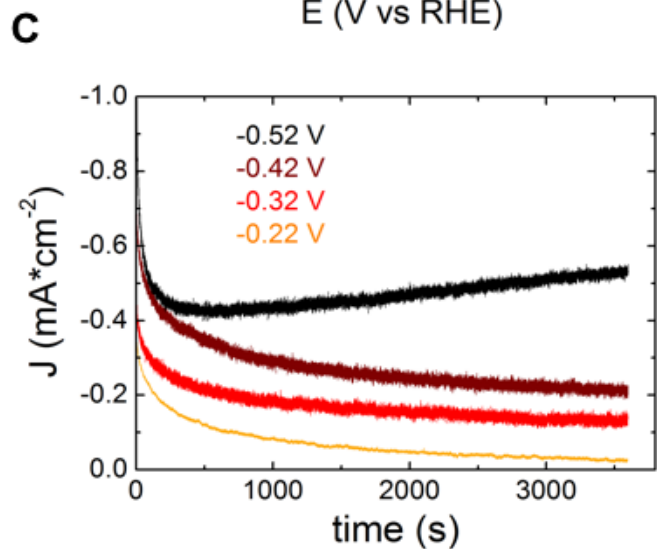
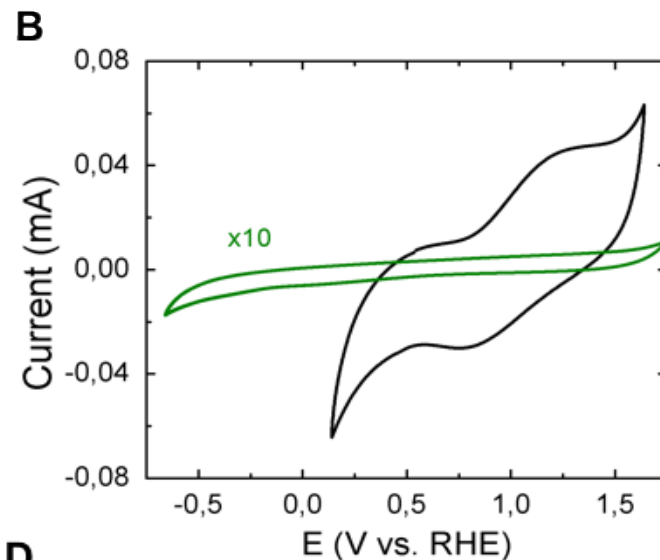
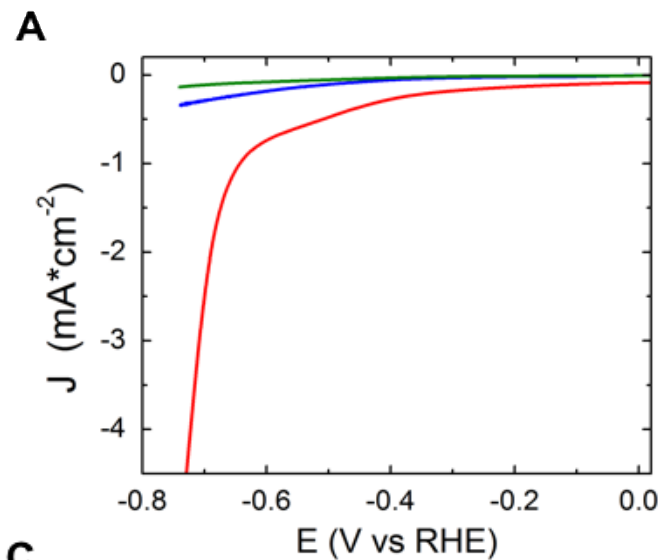


# Get rid of noble metals: Carbon nanotubes/CeO<sub>2</sub>



*Submitted (under review)*

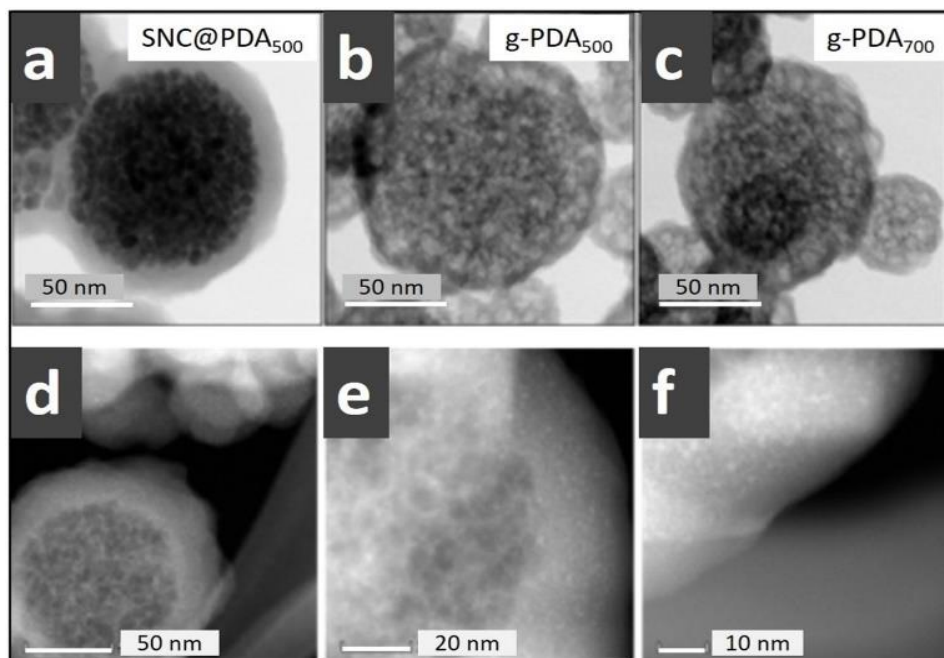
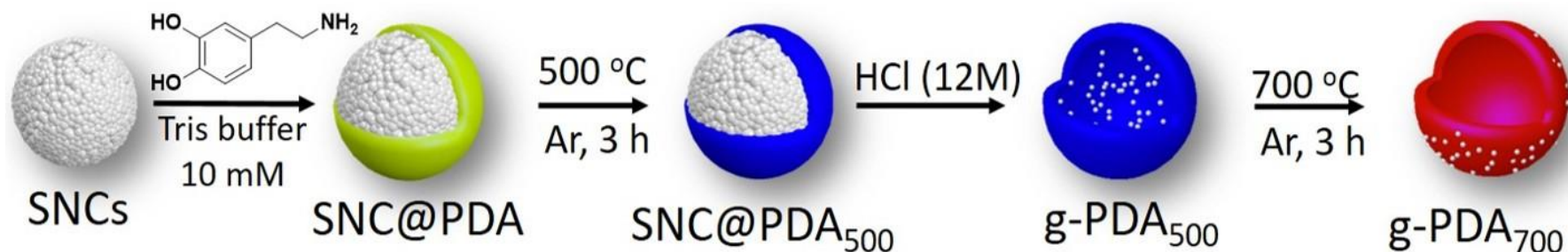
# Get rid of noble metals: Carbon nanotubes/CeO<sub>2</sub>





# *O<sub>2</sub> reduction: Ex-solution Surface modification*

SNC = superparamagnetic nanoparticle clusters of  $\gamma\text{-Fe}_2\text{O}_3$



Sample	$E_{\text{onset}}$ (V)	$E_{\text{applied}}$ (V)	$j^{[1]}$ (mA cm <sup>-2</sup> )	$x_{\text{H}_2\text{O}_2}$ (%)
SNC@PDA <sub>500</sub>	0.67	0.6	-0.098	72
		0.5	-0.22	62
g-PDA <sub>500</sub>	0.64	0.6	-0.11	63
		0.5	-0.14	28
g-PDA <sub>700</sub>	0.60	0.6	-0.30	9
		0.5	-0.42	8

**Collaboration with Prof. Darko Makovec  
and Dr. Slavko Kralj**

# ***Conclusions and perspectives***

**Nanocarbons can be useful components in the design of hierarchical catalysts for Electrocatalytic applications**

**Opportune interface with other phases and make it functional (rational synthetic design)**

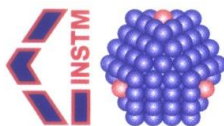
**Translation of fundamental knowledge to large scale devices**

**For catalysis: Advanced techniques (*in situ* or *in operando*) to aid understanding of mechanism**

**Computational chemistry to help design**

# ACKNOWLEDGEMENTS

University of Trieste  
Materials, Environment and Energy  
Research Group



Thank You!



Prof. Marcella Bonchio (Universita' di Padova)  
Prof. Francesco Paolucci (Universita' di Bologna)  
Prof. Claudio Tavagnacco (Universita' di Trieste)  
Dr. Victoria Bracamonte (National University of Cordoba)  
Dr. Giovanni Valenti (Universita' di Bologna)  
Dr. Lucia Nasi (IMEM-CNR, Italy)  
Dr. Manuela Bevilacqua (ICCOM-CNR)



# MultiComp CA15107

is currently composed of 34 COST countries,  
represented by 63 Management Committee  
members.

**FREE REGISTRATION !**  
**12-13 Mar 2020**  
**TRIESTE**

**ORAL SLOTS  
AVAILABLE!**

## Confirmed Speakers

Prof. Paola Ayala (University of Vienna, Austria)

Prof. Jan Peters MBE (Katalytic, UK)

Prof. Maurizio Prato (University of Trieste & CIC BiomaGUNE, Spain)

Dr. Lok Kumar Shrestha (NIMS, Japan)

Dr. Jonathan P. Hill (IPC, Japan)

Felicite Ruddock (LJMU, UK)

