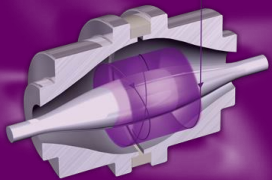


HRMS strategies for the target analysis and suspect screening of persistent and mobile substances in drinking, ground and surface water of Greece



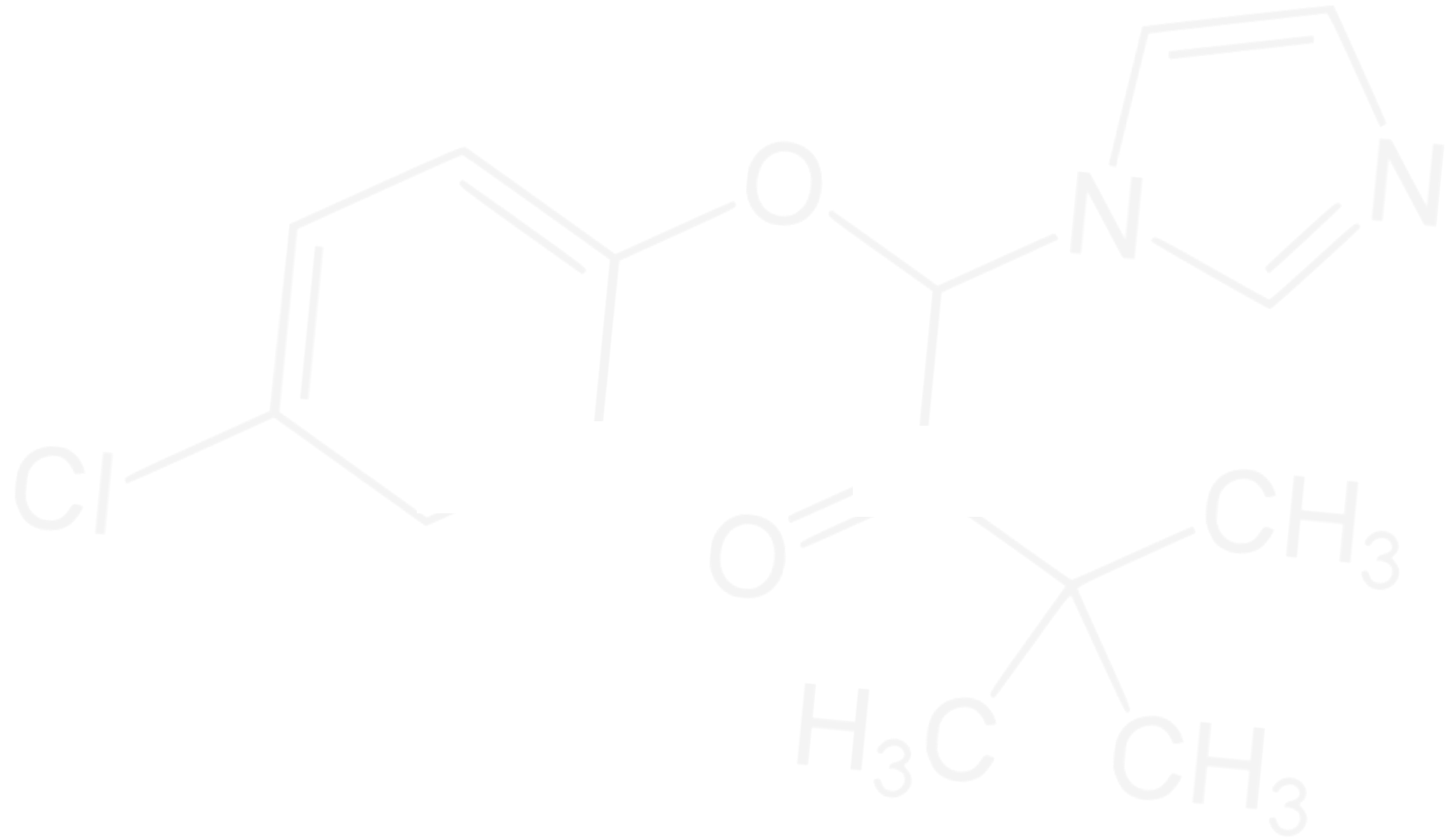
Kyriaki Anagnostopoulou, Eleni Evgenidou, Dimitra Lambropoulou

Laboratory of Environmental Pollution Control, Department of Chemistry, Aristotle University of Thessaloniki, Greece

Center for Interdisciplinary Research and Innovation (CIRI-AUTH), Balkan Center, Thessaloniki, 57001, Greece



R
E
A
C
H



PM Compounds



Identifiable organic chemical >0.1%

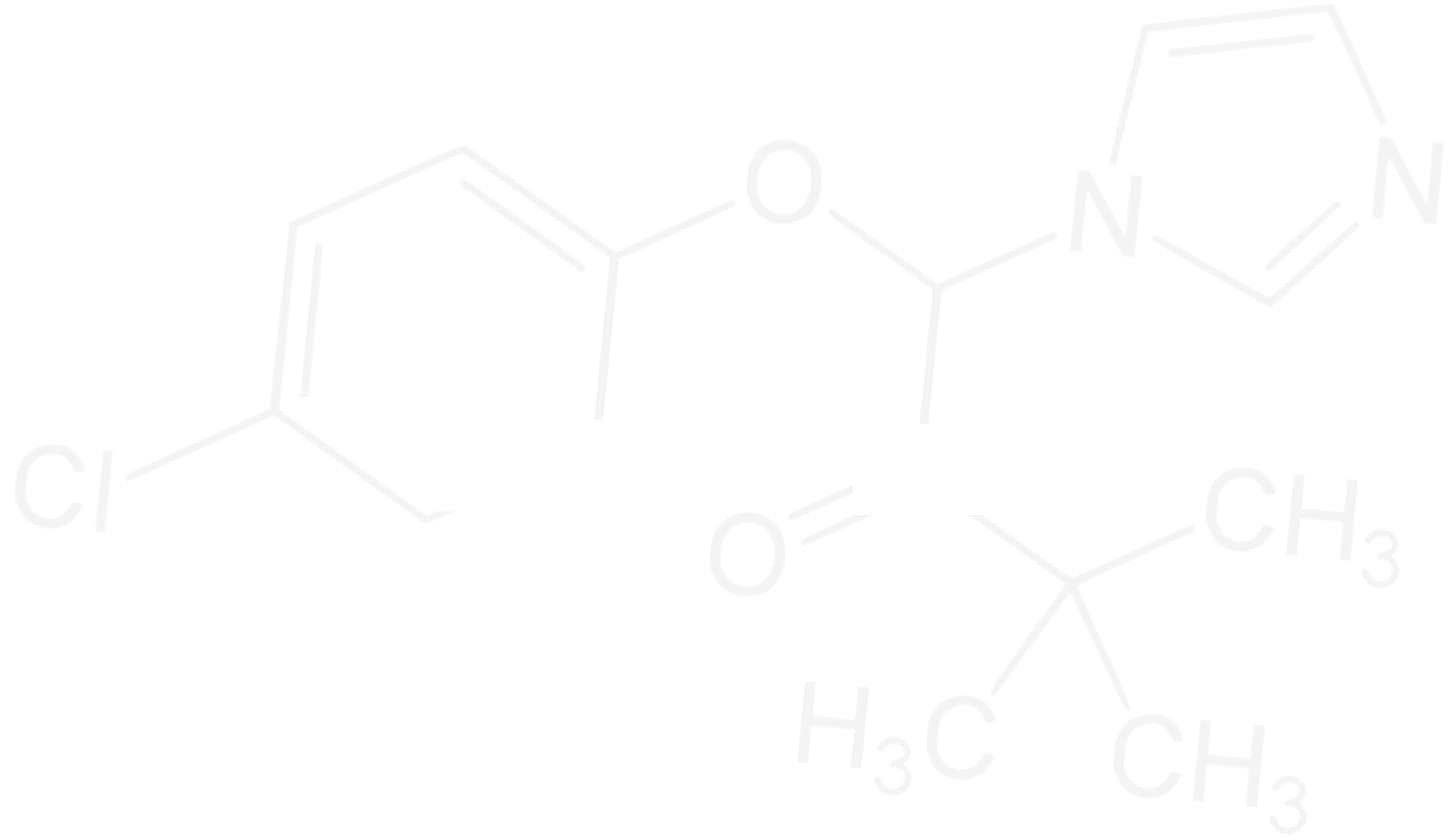
R

E

A

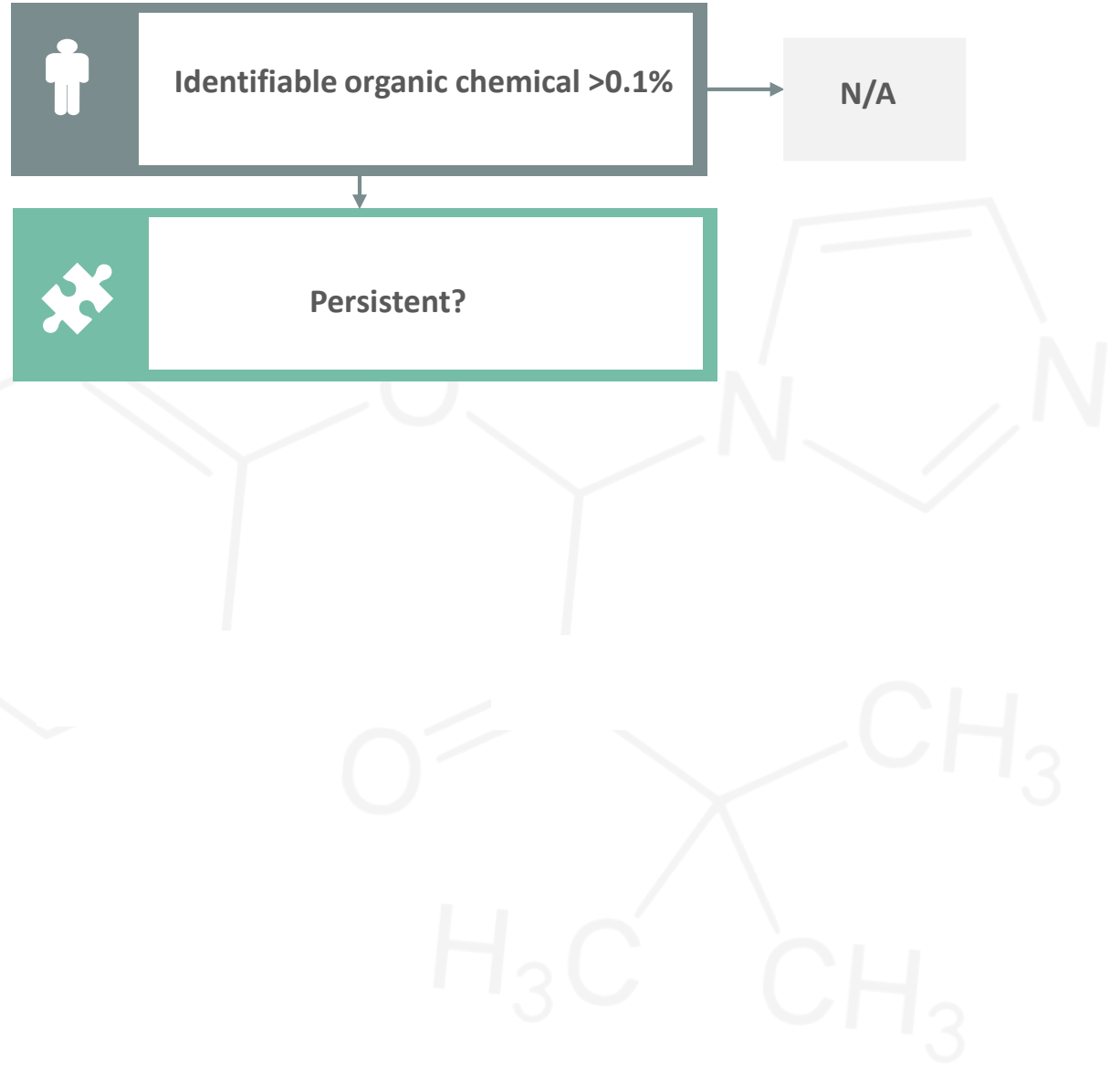
C

H



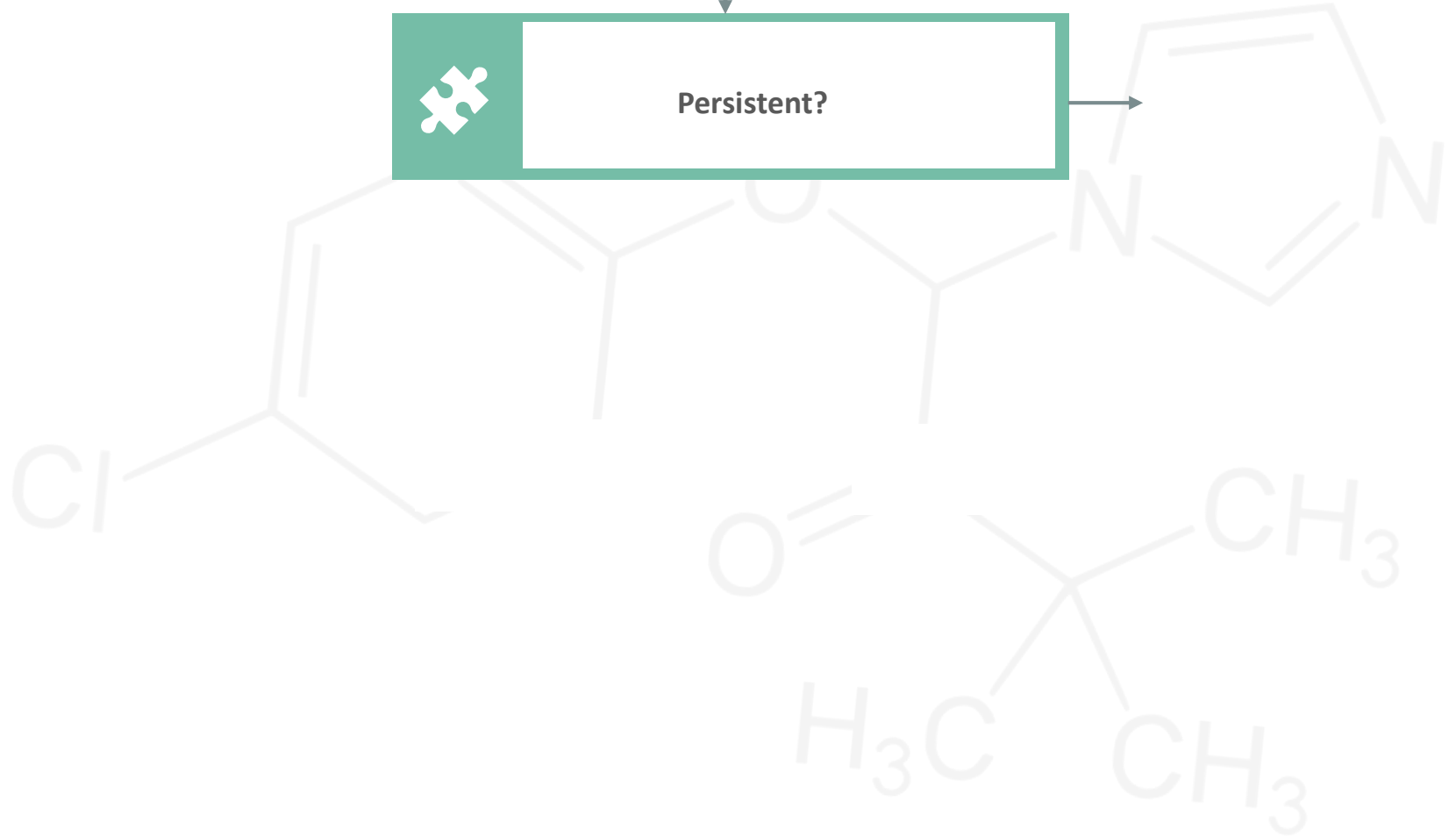
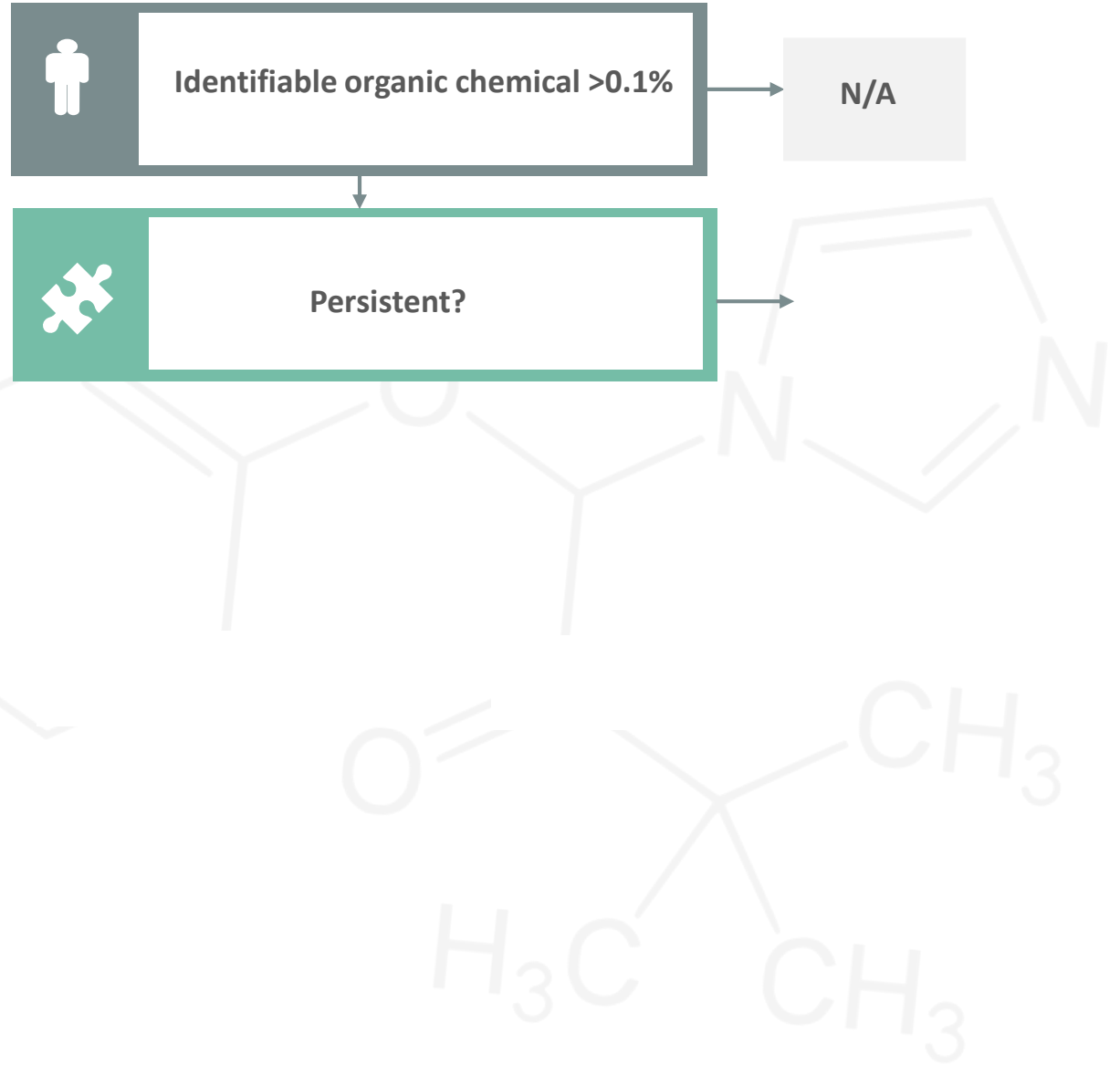
PM Compounds

R
E
A
C
H



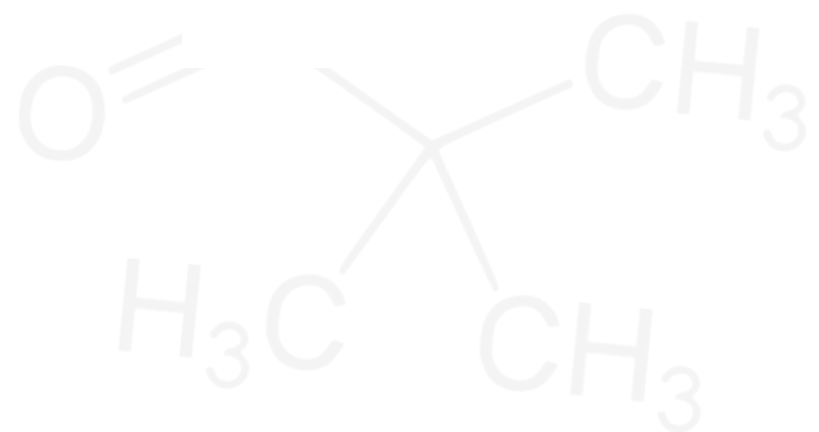
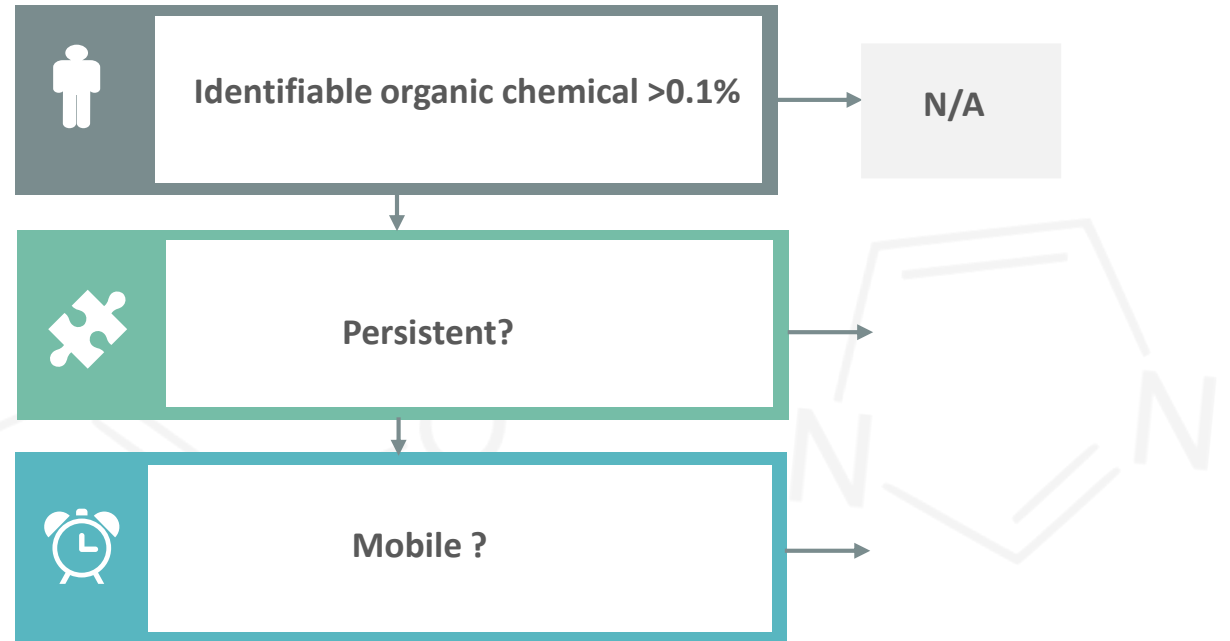
PM Compounds

R
E
A
C
H



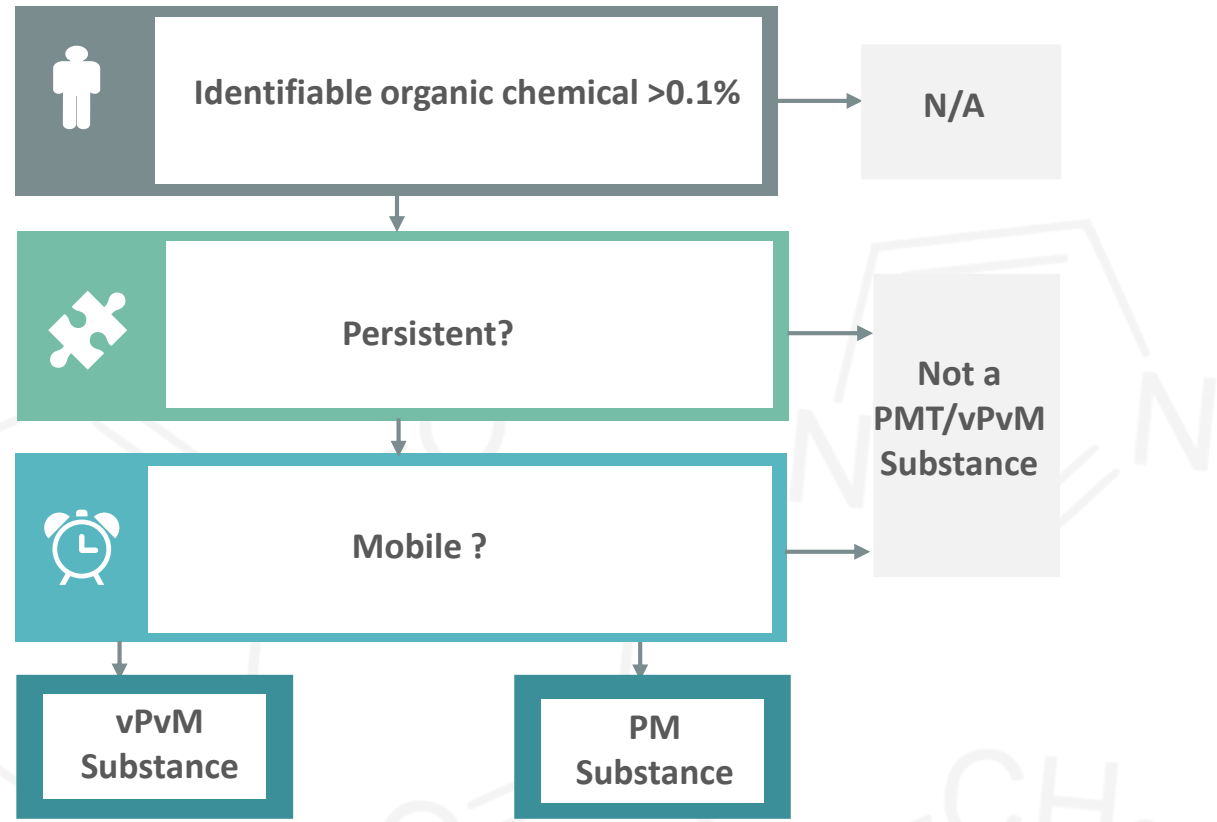
PM Compounds

R
E
A
C
H



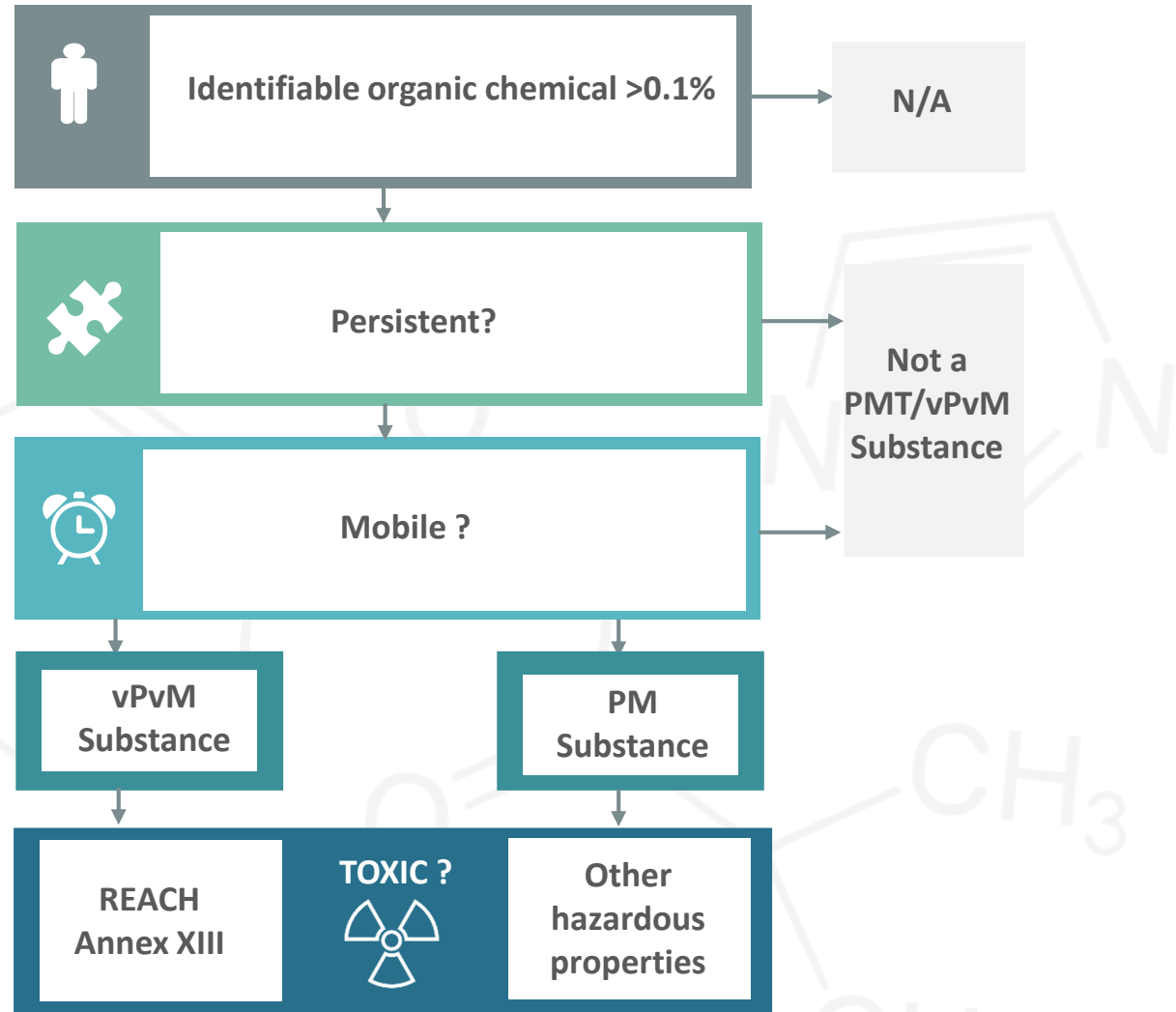
PM Compounds

R
E
A
C
H



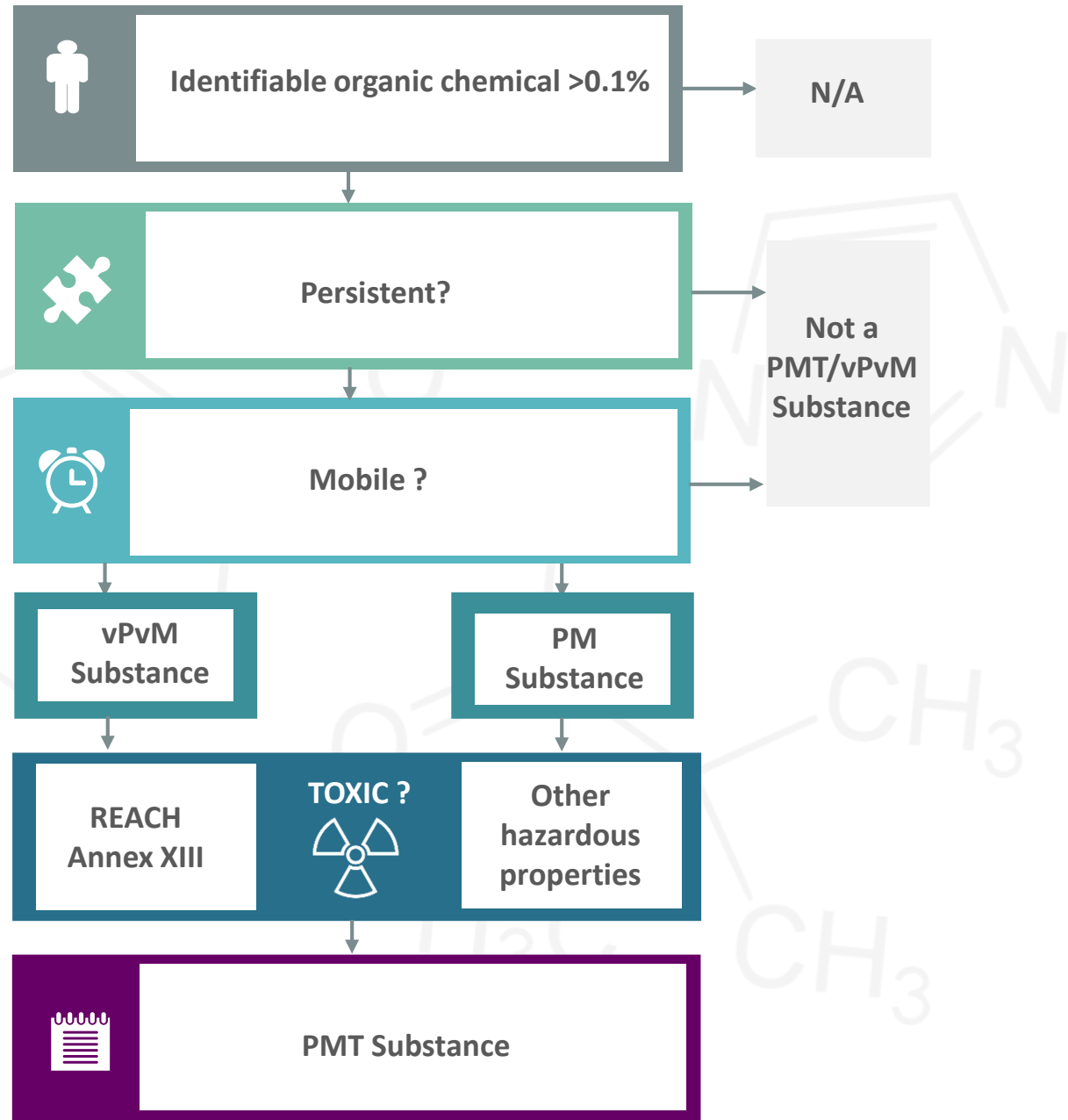
PM Compounds

R
E
A
C
H



PM Compounds

R
E
A
C
H



PM Compounds

P/vP Criteria

A substance fulfils the **persistent** / "**very persistent**" criterion (**P**) /(**vP**)

The degradation half-life:

- in marine water at 9 °C is higher than **60 days**
- in fresh or estuarine water at 12 °C and pH 4-9 is higher than **40 days /60 days**
- in marine sediment at 9 °C is higher than **180 days**
- in fresh or estuarine water sediment at 12 °C and pH 4-9 is higher than **120 days /180 days**
- in soil at 12 °C and pH 4-9 is higher than **120 days /180 days**

A substance fulfils the **mobile** criterion (**M**):

- The lowest organic carbon-water coefficient $\log K_{OC}$ over the pH range of 4-9 is less than **4.0**

A substance fulfils the "**very mobile**" criterion (**vM**):

- the lowest organic carbon-water coefficient $\log K_{OC}$ over the pH range of 4-9 is less than **3.0**

If no KOC data is available

- *octanol-water partition coefficient (KOW) or*
- *pH-dependant octanol-water coefficient for ionisable substances (DOW)*

M/vM Criteria

PM Compounds

P/vP
Criteria

A substance

- The limit value is **4.0**

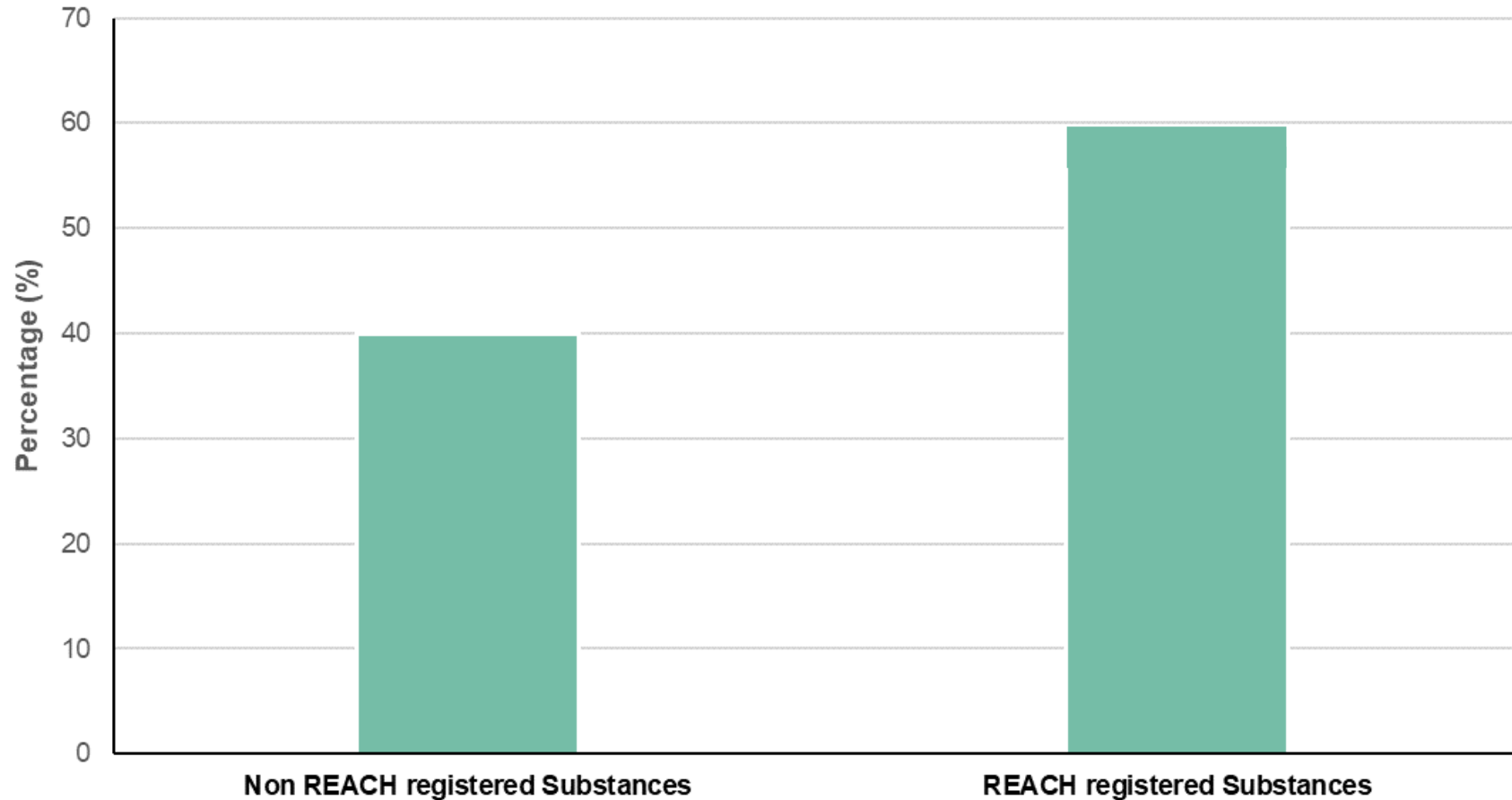
A substance

- the limit value is **10**

If no KOC is available

- octanol-water partition coefficient (KOW) or
- pH-dependant octanol-water coefficient for ionisable substances (DOW)

Drinking Water and Groundwater the frequency of detected substances reported over 0.1ug/L



/180 days

M/vM
Criteria

Science of the Total Environment 865 (2023) 161228



Contents lists available at ScienceDirect

Science of the Total Environment

journal homepage: www.elsevier.com/locate/scitotenv



Persistence and Mobility (defined as organic-carbon partitioning) do not correlate to the detection of substances found in surface and groundwater: Criticism of the regulatory concept of Persistent and mobile substances



Marie Collard ^a, Louise Camenzuli ^b, Delina Lyon ^{c,*}, David Saunders ^d, Nathalie Vallotton ^e, Pippa Curtis-Jackson ^f

^a Firmenich SA, Louvain-la-neuve, Belgium

^b ExxonMobil Petroleum and Chemical BV, Machelen, Belgium

^c Concawe, Boulevard du Souverain 165, B-1160 Brussels, Belgium

^d Shell International, The Hague, Netherlands

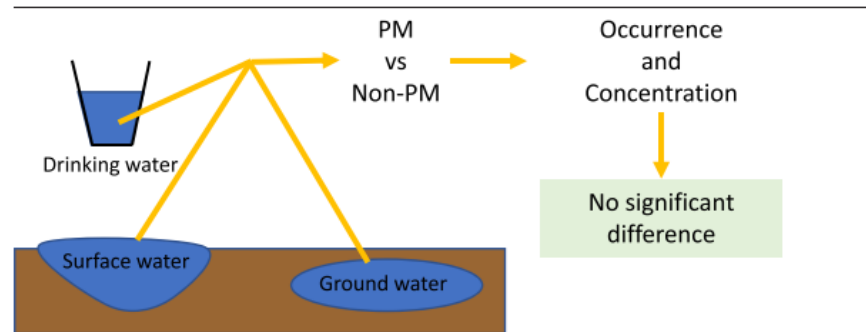
^e Dow Europe GmbH, Horgen, Switzerland

^f Environment Agency, Bristol, United Kingdom

HIGHLIGHTS

- PM substances do not have a higher likelihood of detection than non-PM substances.
- Likelihood of detecting a P substance in water is independent of the M criterion.
- There is no indication that PM substance accumulate in water bodies.
- No evidence of log Dow or log Koc as driver of contamination of water.

GRAPHICAL ABSTRACT



Science of the Total Environment 865 (2023) 161228



Contents lists available at ScienceDirect

Science of the Total Environment

journal homepage: www.elsevier.com/locate/scitotenv



Persistence and Mobility (defined as organic-carbon partitioning) do not correlate to the detection of substances found in surface and groundwater: Criticism of the regulatory concept of Persistent and mobile substances



Marie Collard ^a, Louise Camenzuli ^b, Delina Lyon ^{c,*}, David Saunders ^d, Nathalie Vallotton ^e, Pippa Curtis-Jackson ^f

^a Firmenich SA, Louvain-la-neuve, Belgium

^b ExxonMobil Petroleum and Chemical BV, Machelen, Belgium

^c Concawe, Boulevard du Souverain 165, B-1160 Brussels, Belgium

^d Shell International, The Hague, Netherlands

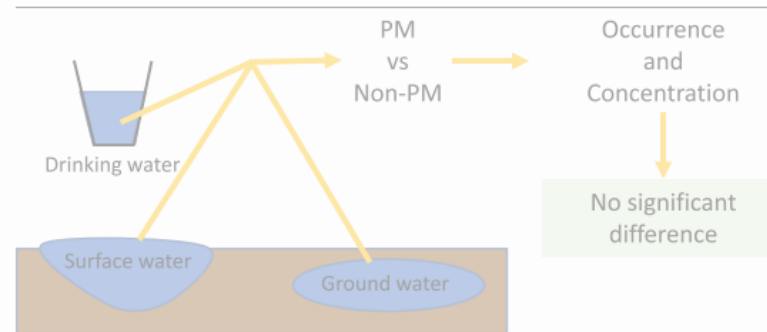
^e Dow Europe GmbH, Horgen, Switzerland

^f Environment Agency, Bristol, United Kingdom

HIGHLIGHTS

- PM substances do not have a higher likelihood of detection than non-PM substances.
- Likelihood of detecting a P substance in water is independent of the M criterion.
- There is no indication that PM substance accumulate in water bodies.
- No evidence of log Dow or log Koc as driver of contamination of water.

GRAPHICAL ABSTRACT






Science of the Total Environment 865 (2023) 161228

Contents lists available at ScienceDirect

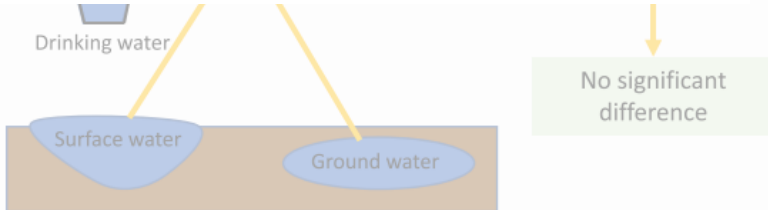
Science of the Total Environment

journal homepage: www.elsevier.com/locate/scitotenv



is-Jackson ^f

- PM substances do not have a higher likelihood of detection than non-PM substances.
- Likelihood of detecting a P substance in water is independent of the M criterion.
- There is no indication that PM substance accumulate in water bodies.
- No evidence of log Dow or log Koc as driver of contamination of water.



Drinking water

Surface water

Ground water

No significant difference

accumulate in water bodies.



No evidence of log Dow or log Koc as driver of contamination of water.

Science of the Total Environment 865 (2023) 161228

Contents lists available at ScienceDirect

Science of the Total Environment

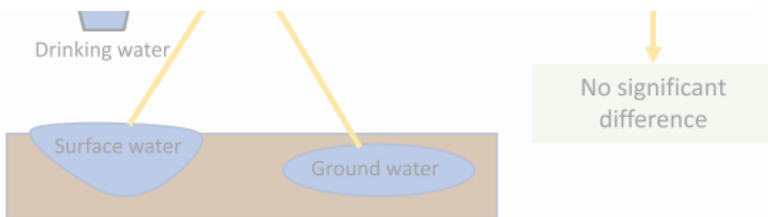
journal homepage: www.elsevier.com/locate/scitotenv



Check for updates

is-Jackson^f

- PM substances do not have a higher likelihood of detection than non-PM substances.
- Likelihood of detecting a P substance in water is independent of the M criterion.
- There is no indication that PM substance accumulate in water bodies.
- No evidence of log Dow or log Koc as driver of contamination of water.



Drinking water

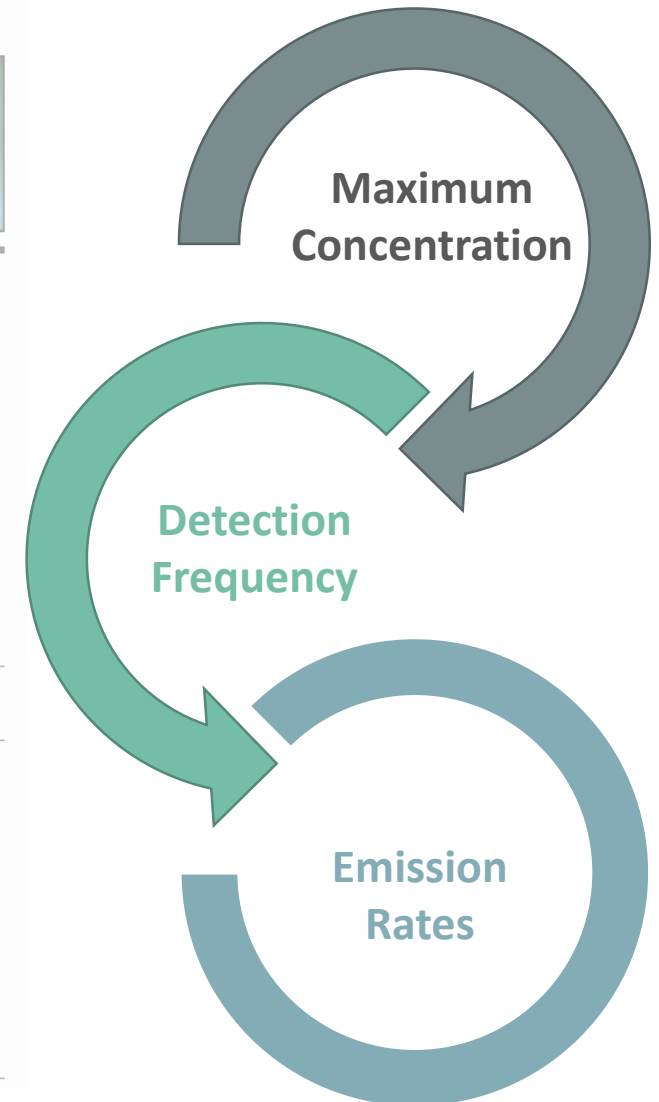
Surface water

Ground water

No significant difference

accumulate in water bodies.

No evidence of log Dow or log Koc as driver of contamination of water.



Water Research 204 (2021) 117645



Contents lists available at ScienceDirect

Water Research

journal homepage: www.elsevier.com/locate/watres



Water Research 150 (2019) 86–96



Contents lists available at ScienceDirect

Water Research

journal homepage: www.elsevier.com/locate/watres



Filling the knowledge gap: A suspect screening study for 1310 potentially persistent and mobile organic compounds in river systems

Water Research 153 (2019) 80–90

Isabelle Neuwald^{a,1}, Ma Till Meier^b, Jochen Kuc Thorsten Reemtsma^{b,d,*}

^a Hochschule Fresenius gmbh, Lissib
^b Department of Analytical Chemistry, Hel
^c Umweltbundesamt, Section Toxicology of
^d University of Leipzig, Institute for Analy



Contents lists available at ScienceDirect

Water Research

journal homepage: www.elsevier.com/locate/watres

ARTICLE INFO

Keywords:
PMT-substances
Wastewater
Groundwater
Drinking water
Water cycle
LC-MS

Occurrence of emerging persistent and mobile organic compounds in European water samples

Stefanie Schulze^a, Daniel Zahn^b, Rosa Montes^c, Rosario Rodil^c, José Thomas P. Knepper^b, Thorsten Reemtsma^{a,d}, Urs Berger^{a,*}

^a Helmholtz Centre for Environmental Research – UFZ, Department of Analytical Chemistry, Permoserstrasse 15, 04318, Leipzig, Germany
^b Hochschule Fresenius University of Applied Sciences, Institute for Analytical Research, Limburger Strasse 2, 65510, Idstein, Germany
^c Universidade de Santiago de Compostela, Department of Analytical Chemistry, Nutrition and Food Sciences, IIAA – Instituto de Investigación en Alimentos, 15782, Santiago de Compostela, Spain
^d University of Leipzig, Institute of Analytical Chemistry, Linnéstrasse 3, 04103 Leipzig, Germany

ARTICLE INFO

Article history:
Received 31 August 2018
Received in revised form 19 December 2018
Accepted 12 January 2019
Available online 18 January 2019

Keywords:
Persistent and mobile organic chemicals
PMOC
Water
Occurrence
Chromatography

ABSTRACT

The release of persistent and mobile organic chemical compounds into water resources at risk. PMOCs are challenging due to their high polarity. The aim of this study was to develop novel methods for the detection and identification of PMOCs. A prioritized list of industrial chemicals that were mobile and persistent in the environment. Analytical screening methods based on solid phase extraction (SPE), reversed phase LC, or supercritical fluid chromatography (SFC) were successfully developed and applied to surface and groundwater samples from three European countries. A total of 43 PMOCs were identified, of which 23 PMOCs have not been reported before. The chemical structures of these novel PMOCs were methyl sulfate, 2-trimethylammonium, benzyldimethylamine, trifluoromethylamine, and 1,3-di-*o*-tolylguanidine occurring in ≥ 51 ng L⁻¹ up to $\mu\text{g L}^{-1}$ range. The approach of focusing on mobile and persistent organic chemical analysis proved to be highly efficient in revealing a large suite of novel as well as scarcely investigated PMOCs in surface and groundwater.

Identification of potentially mobile and persistent transformation products of REACH-registered chemicals and their occurrence in surface waters

Journal of Hazardous Materials Letters 2 (2021) 100026



Contents lists available at ScienceDirect

Journal of Hazardous Materials Letters

journal homepage: www.elsevier.com/locate/hazl



The distribution of persistent, mobile and toxic (PMT) pharmaceuticals and personal care products monitored across Chinese water resources

Chen Huang^{a,b,c}, Biao Jin^{a,b,c,*}, Min Han^{a,b,c}, Yang Yu^d, Gan Zhang^{a,b}, Hans Peter H. Arp^{e,f}

^a State Key Laboratory of Organic Geochemistry, Guangzhou Institute of Geochemistry, Chinese Academy of Sciences, Guangzhou, 510640, China
^b CAS Center for Excellence in Deep Earth Science, Guangzhou, 510640, China
^c University of Chinese Academy of Sciences, Beijing, 100069, China
^d Solid Waste and Chemicals Management Center, Ministry of Ecology and Environment (MEE), Beijing, 100029, China
^e Norwegian Geotechnical Institute (NGI), P.O. Box 3930, Ullevål Stadion, N-0806, Oslo, Norway
^f Norwegian University of Science and Technology (NTNU), NO-7491, Trondheim, Norway

ARTICLE INFO

Keywords:
Drinking water
Groundwater
Surface water
vPvM
PPCP

ABSTRACT

Hazard classifications have recently been introduced for persistent, mobile and toxic (PMT) and very persistent and very mobile (vPvM) substances, which are those that negatively impact water resources if substantially emitted into the environment. Many pharmaceuticals and personal care products (PPCPs) may meet this classification. Our study focused on 169 detected PPCPs in surface water, groundwater and drinking water in China among a total of 432 PPCPs that were monitored for across 75 studies. We assessed if these could be classified as PMT/vPvM substances based on the recent European criteria for industrial substances. For most PPCPs, persistence half-life data were lacking; therefore, definitive classifications could not be made. In surface water, 52 (37.7%) of the detected PPCPs met, or were strongly suspected to meet, the PMT/vPvM criteria. Over half of these were antibiotic compounds. The industrialized Yangtze, Haihe and Pearl river basins were the most impacted. We hypothesized the proportion of PPCPs monitored meeting the PMT/vPvM criteria in drinking water and groundwater would be larger than surface water. This was supported as 44.3% of PPCPs in these media met, or based on weight-of-evidence likely meet, the PMT/vPvM criteria; though, a definitive comparison was hindered by a lack of persistence data.

known persistent and mobile organic compounds (PMOCs) in the environment. Herein we investigated the occurrence of PMOC precursors among various water resources, and the transformation products (TPs) formed during transformation. The transformation products (TPs) were identified as well as a target list of PMOCs. The results revealed the occurrence of PMOCs in drinking water, surface water, and groundwater. The transformation products (TPs) were identified as well as a target list of PMOCs. The results revealed the occurrence of PMOCs in drinking water, surface water, and groundwater. The transformation products (TPs) were identified as well as a target list of PMOCs. The results revealed the occurrence of PMOCs in drinking water, surface water, and groundwater.

1. All rights reserved.

Target List

120 Compounds



Suspect List

2500 Compounds





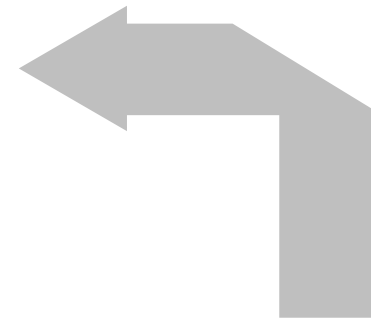
Analytical Methods

● Working Plan

2nd ISO-FOOD Symposium Portorož, Slovenia, April 24 – 26, 2023 ●

Literature – Inclusion List

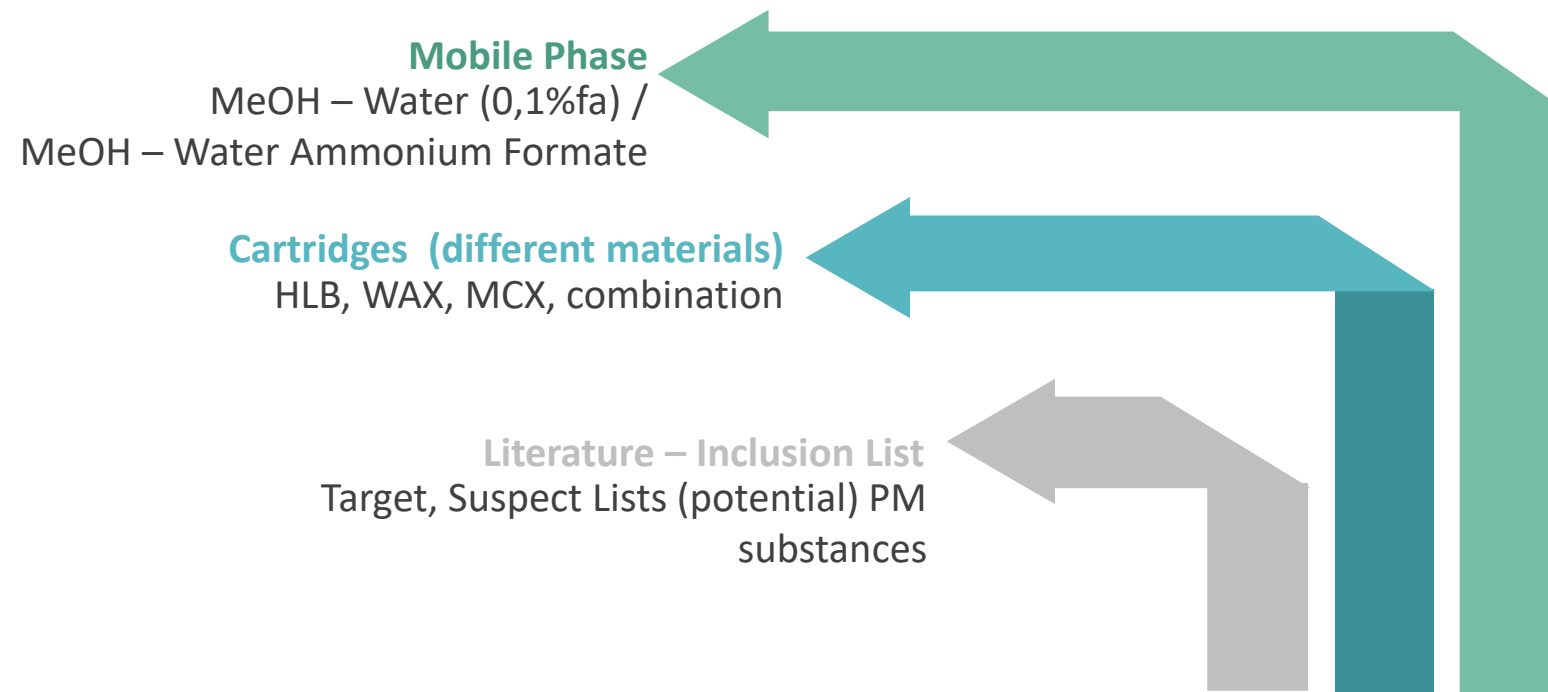
Literature – Inclusion List
Target, Suspect Lists (potential) PM
substances

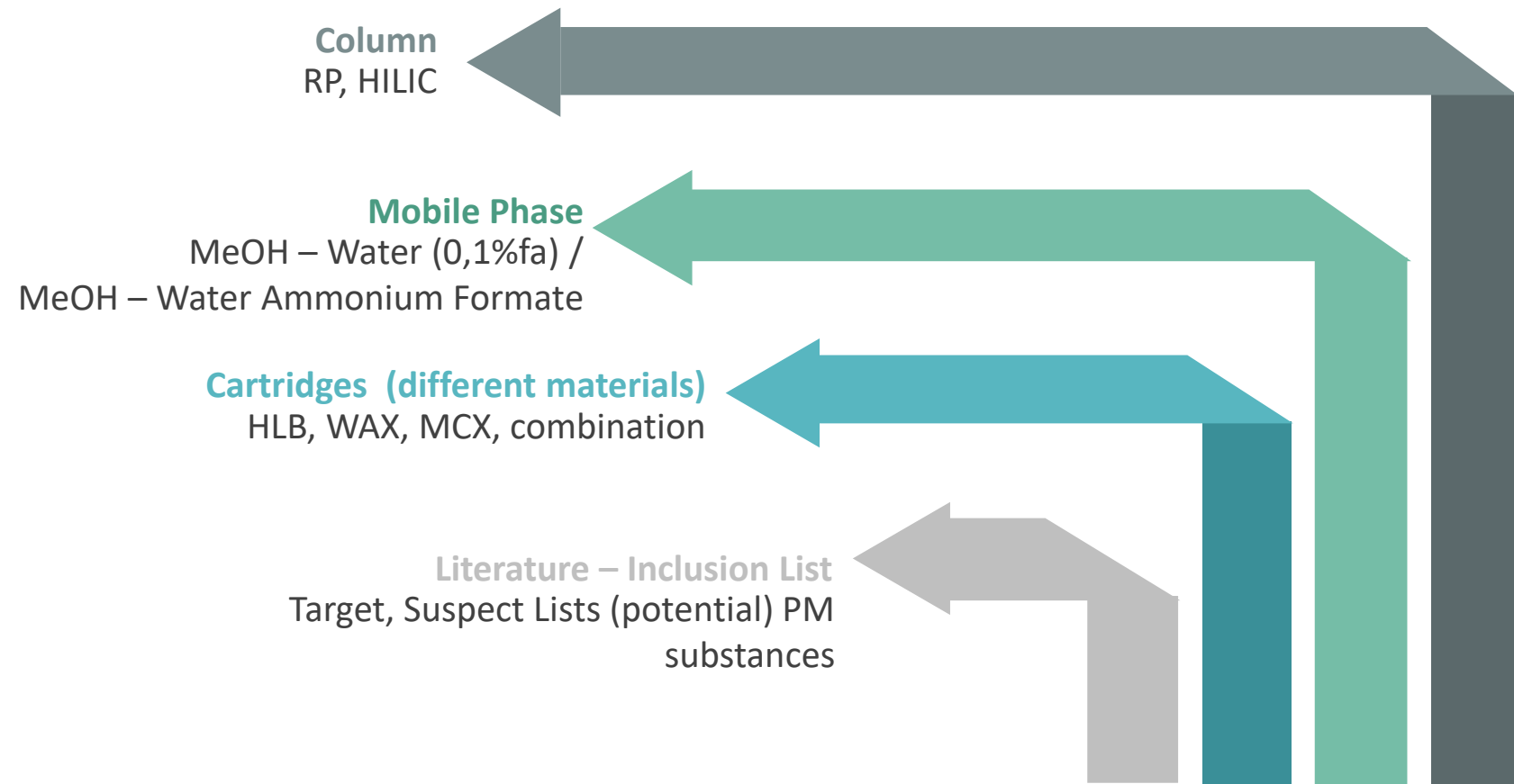


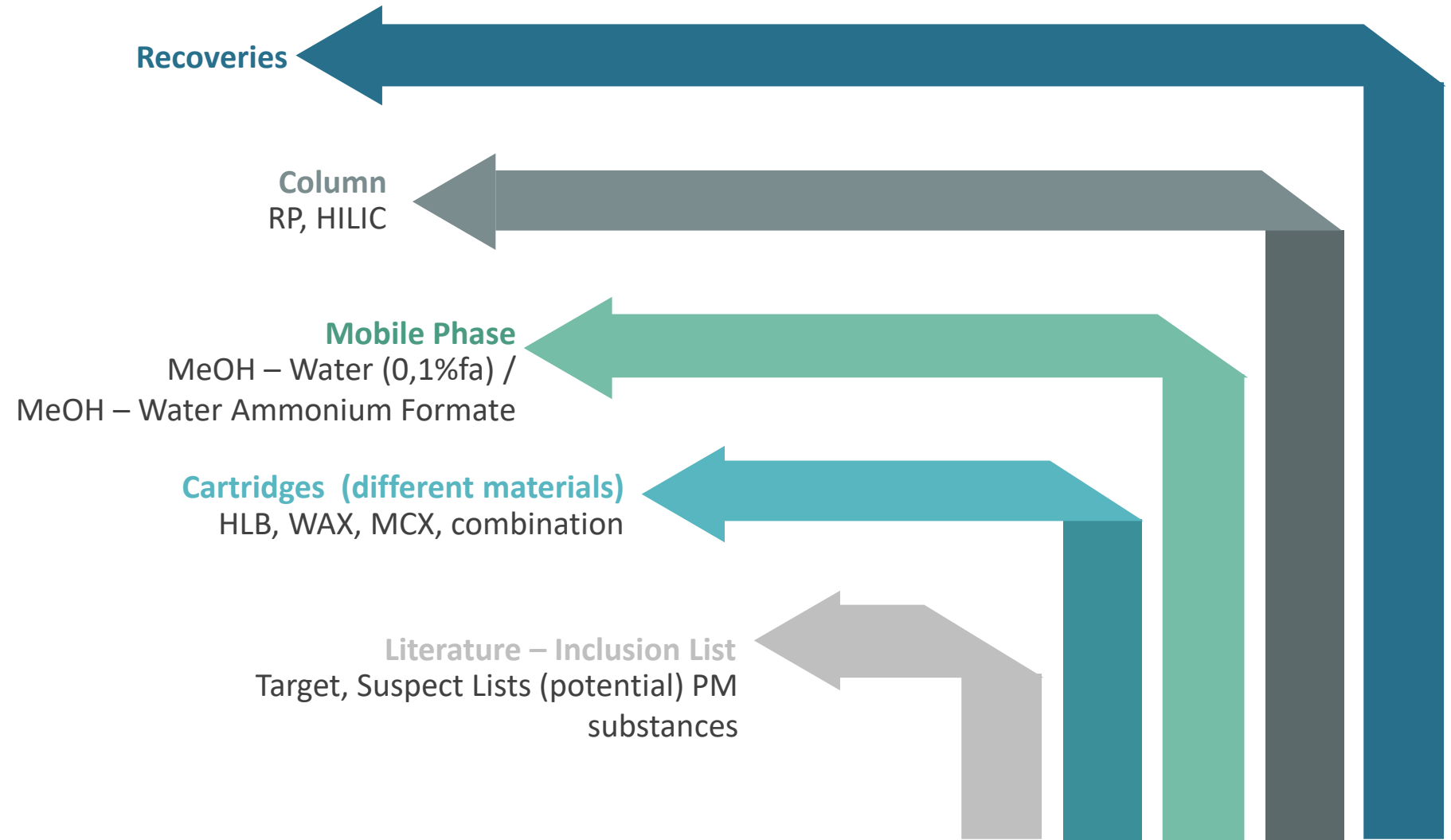
Cartridges (different materials)
HLB, WAX, MCX, combination

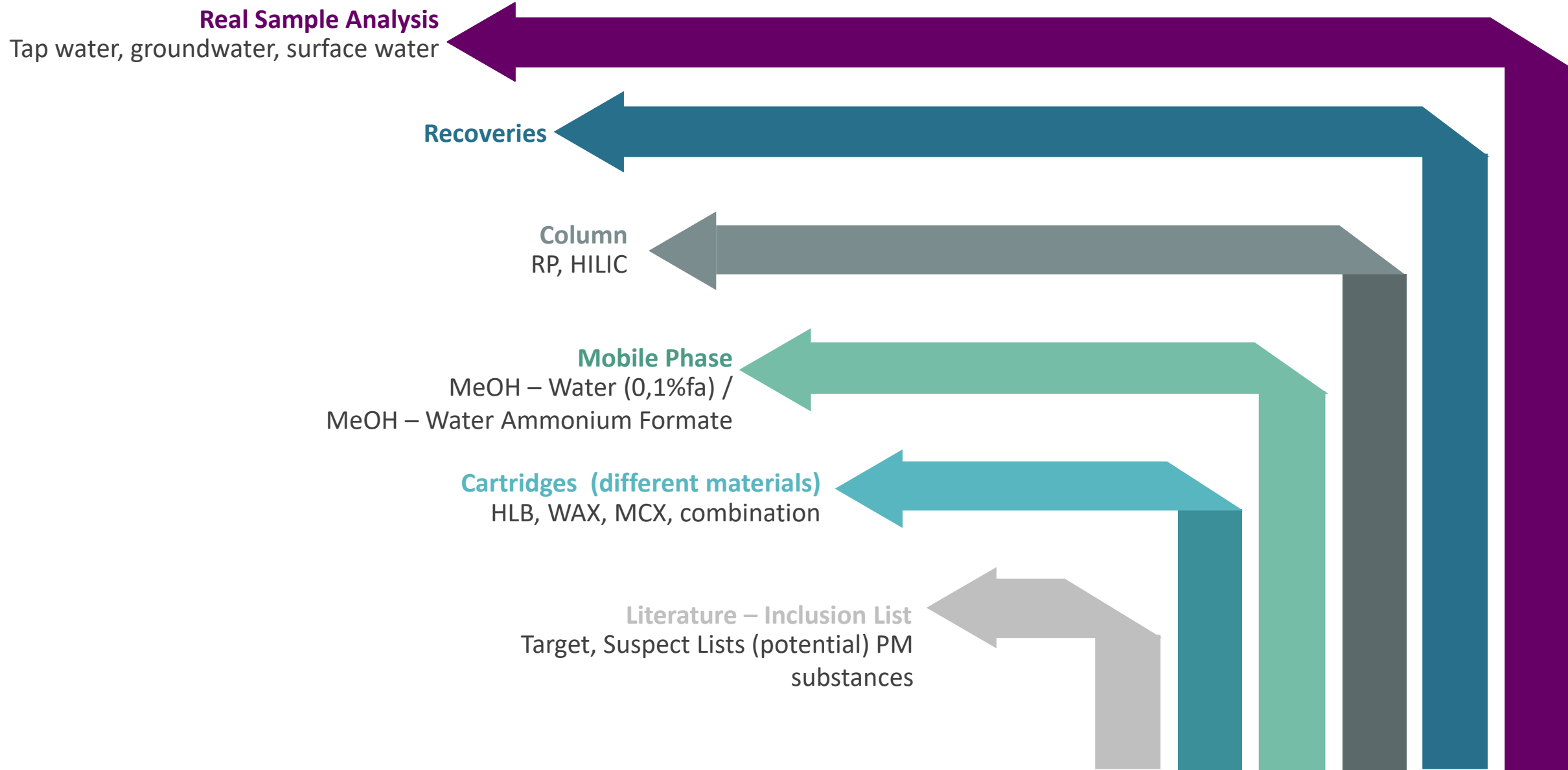
Literature – Inclusion List
Target, Suspect Lists (potential) PM
substances



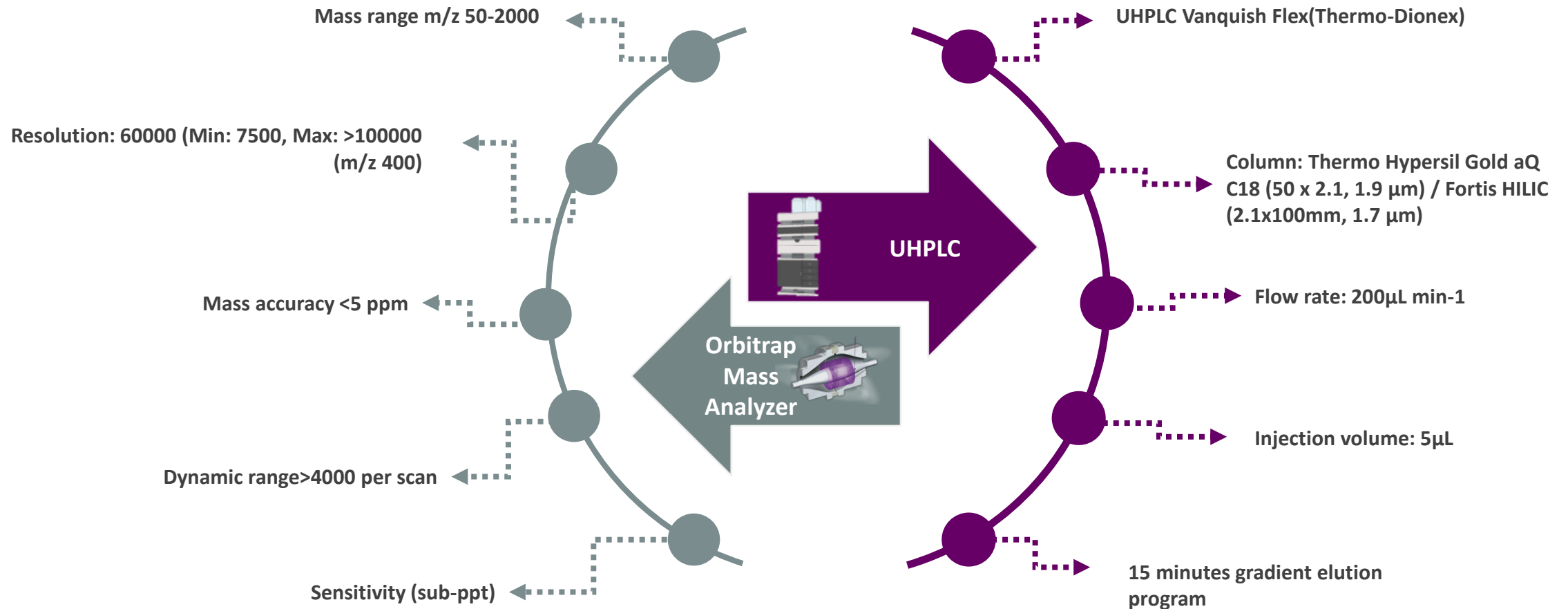








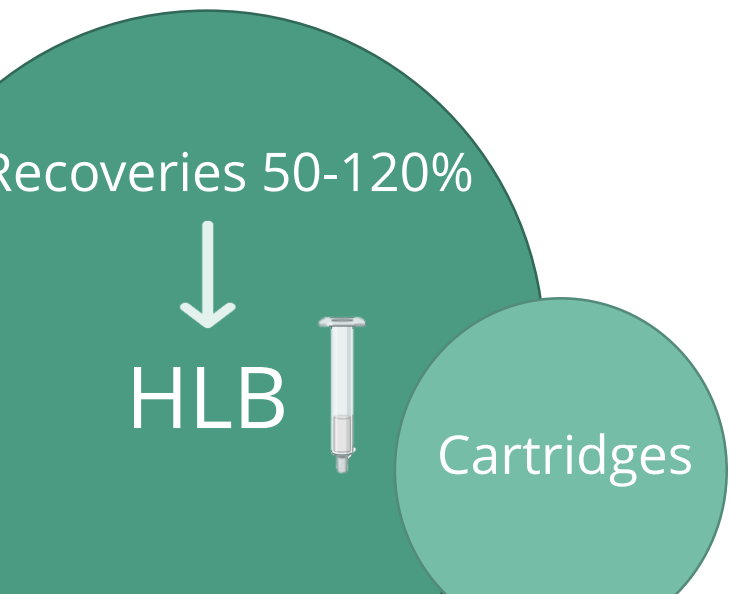
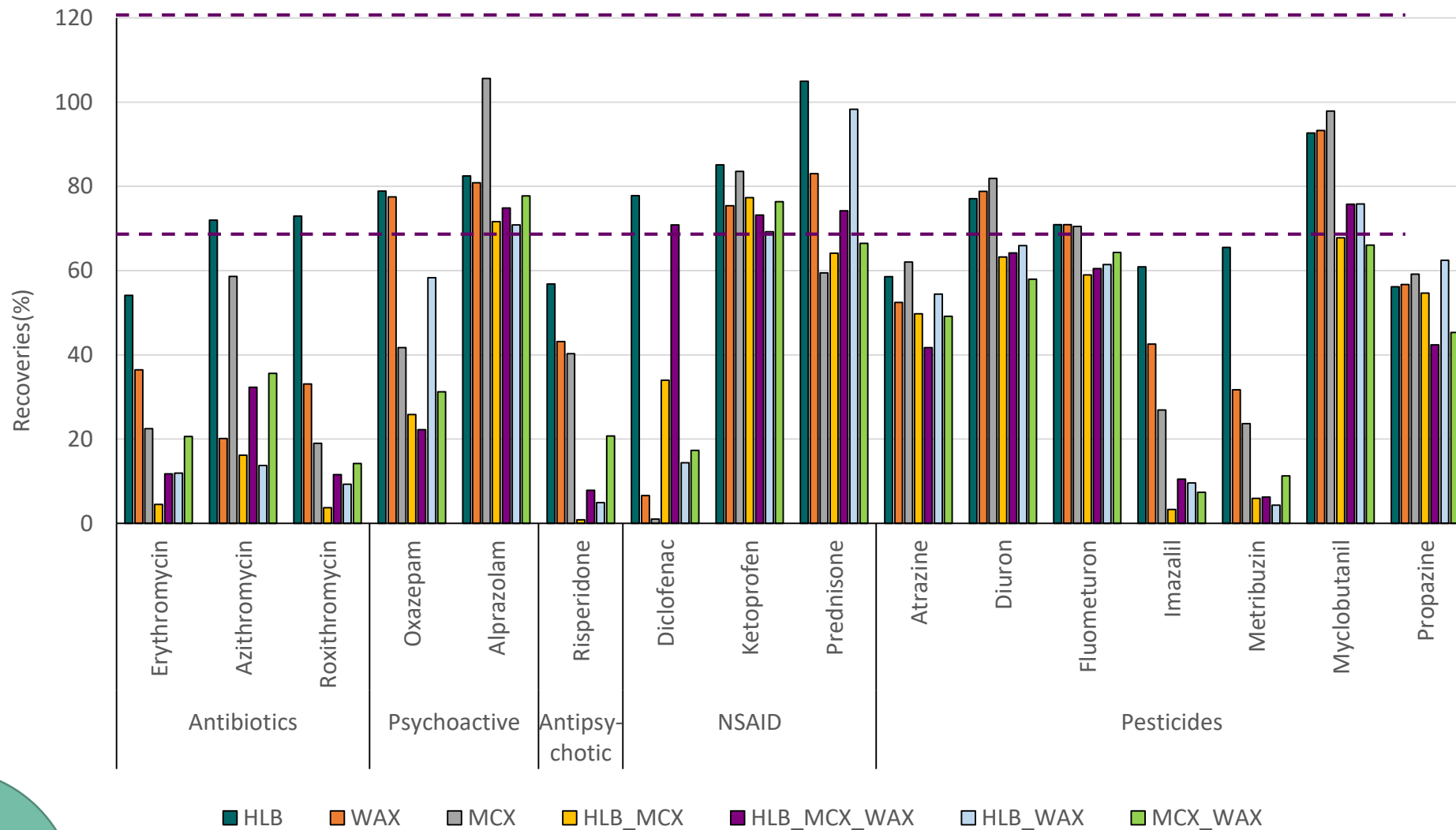
Analysis



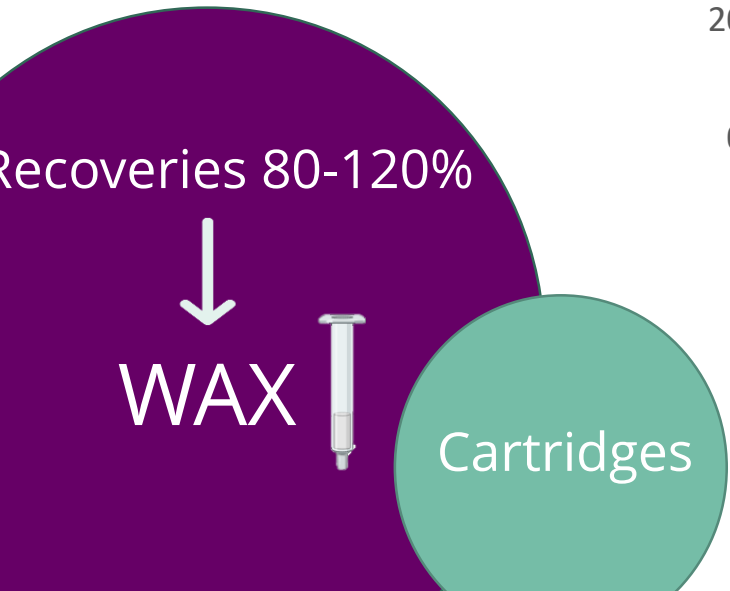
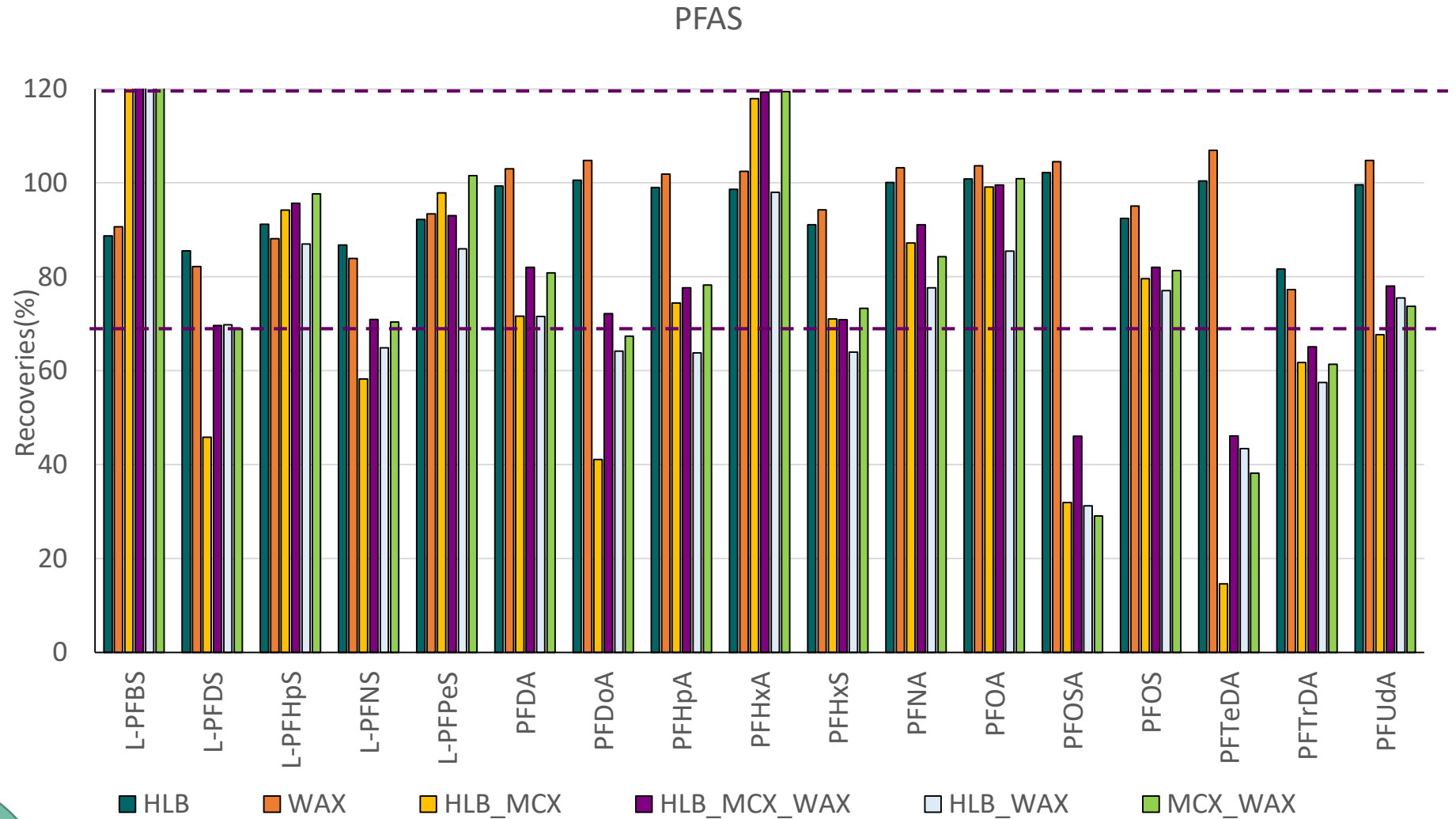


RESULTS

Recoveries

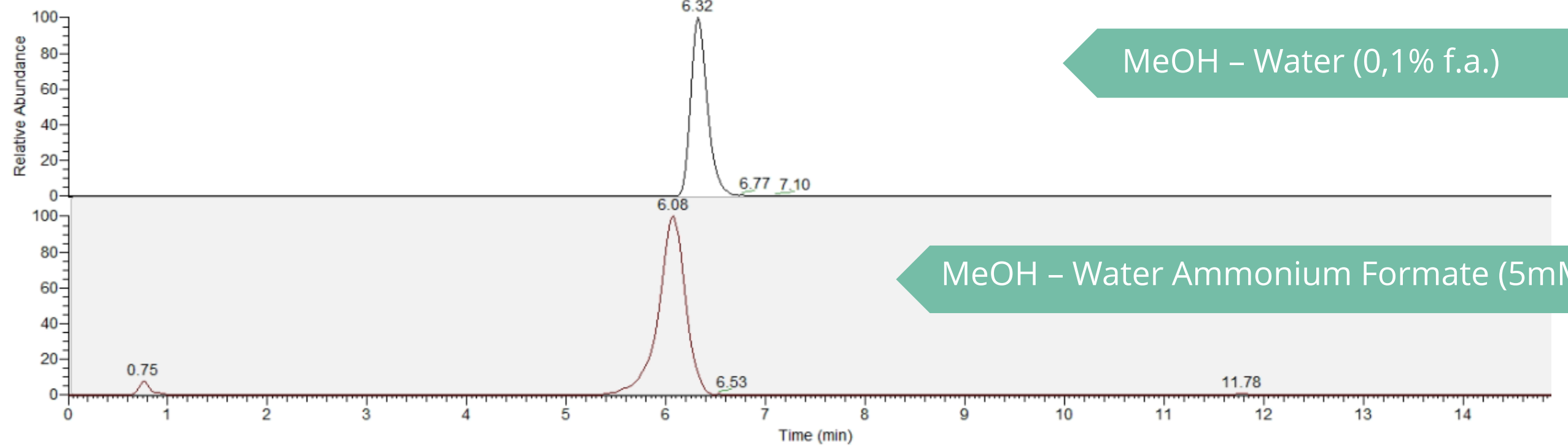


Recoveries



Peak Shape

RT: 0.00 - 15.01 SM: 11G



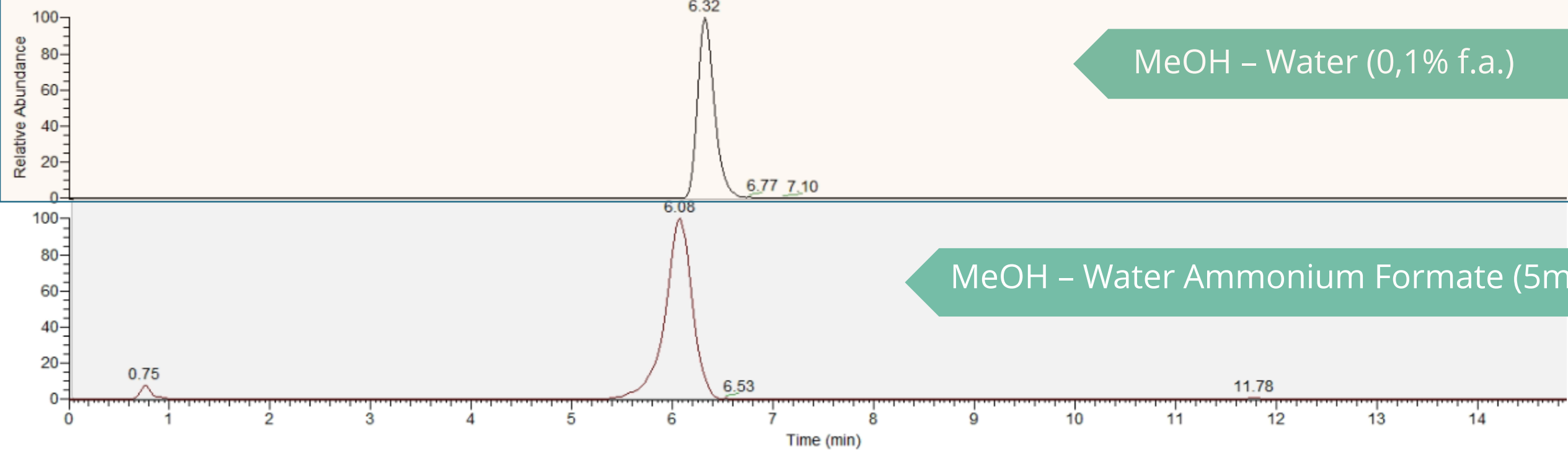
MeOH - Water (0,1% f.a.)

MeOH - Water Ammonium Formate (5mM)

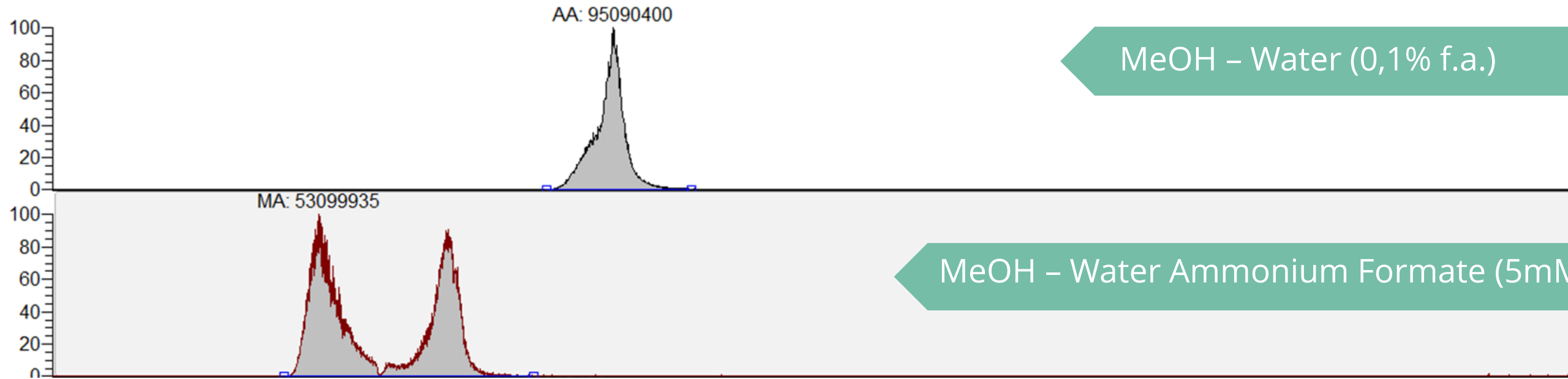


Peak Shape

RT: 0.00 - 15.01 SM: 11G



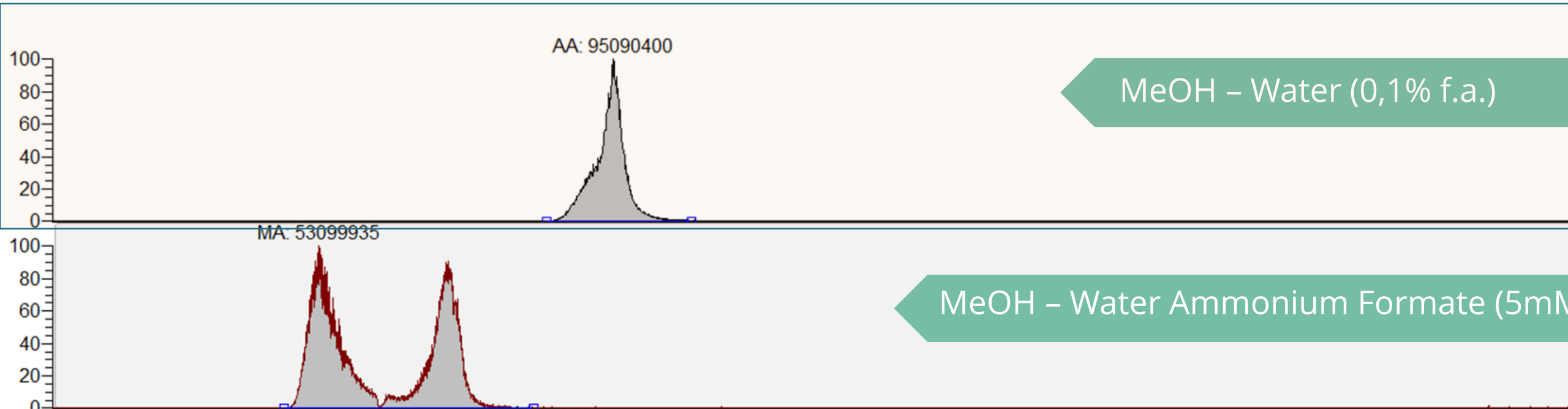
Peak Shape



Mobile
Phase

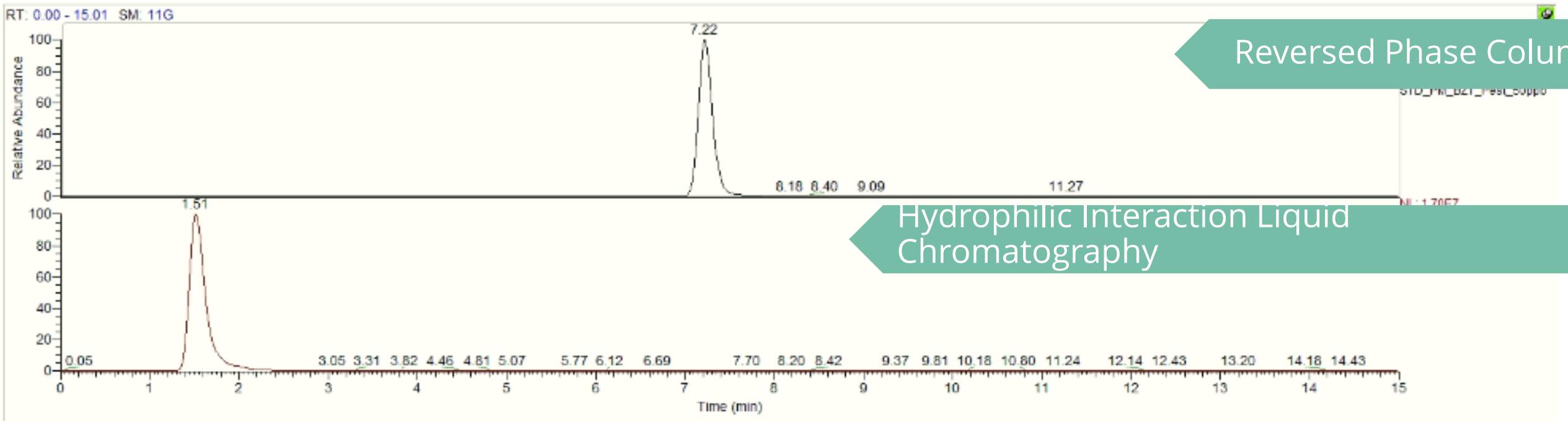


Peak Shape



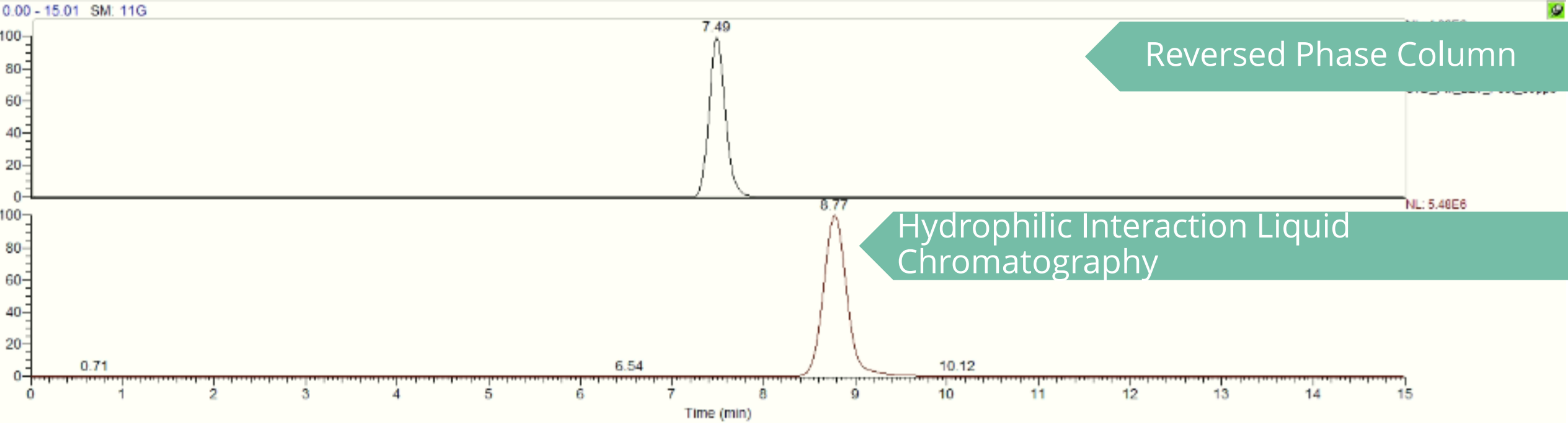
Peak Shape

Column



Peak Shape

Column

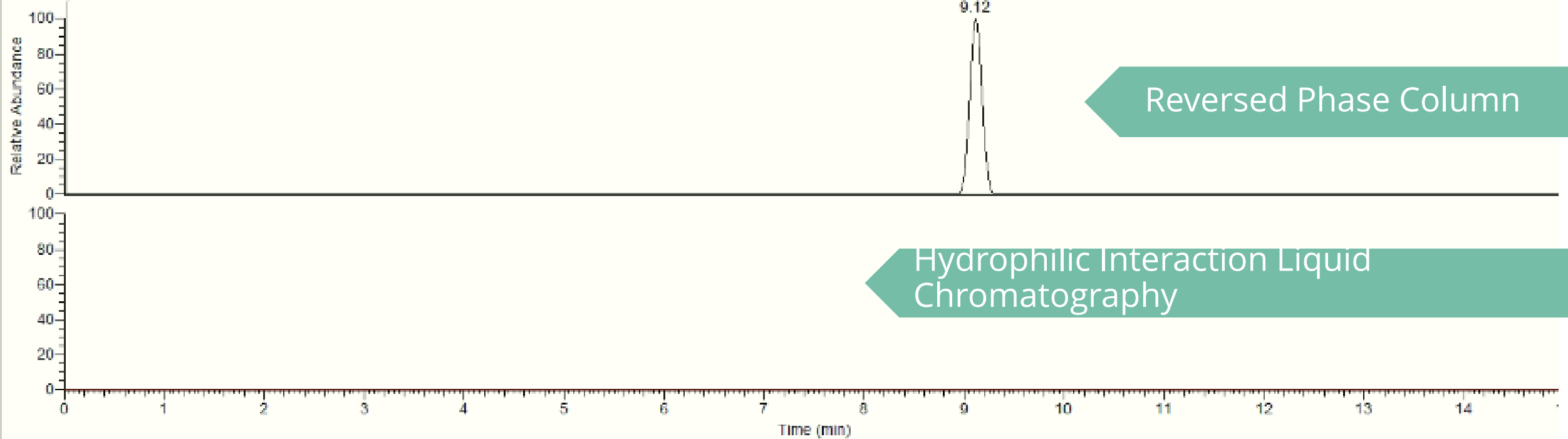


Peak Shape

Column



RT: 0.00 - 15.01 SM: 11G

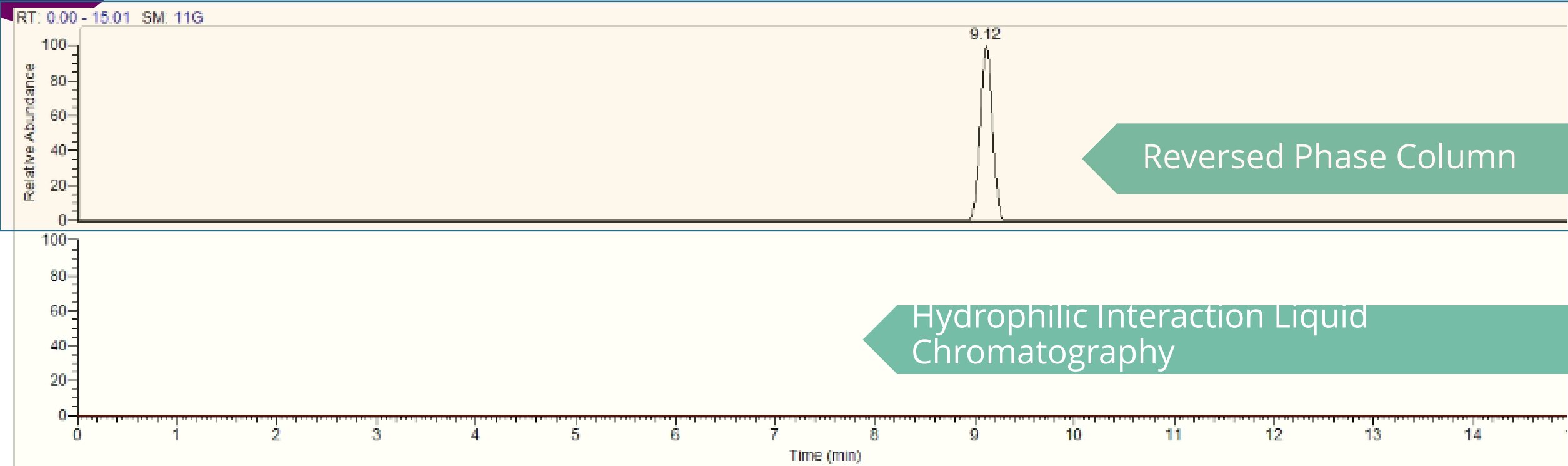


Reversed Phase Column

Hydrophilic Interaction Liquid Chromatography

Peak Shape

Column



Real Sample Analysis

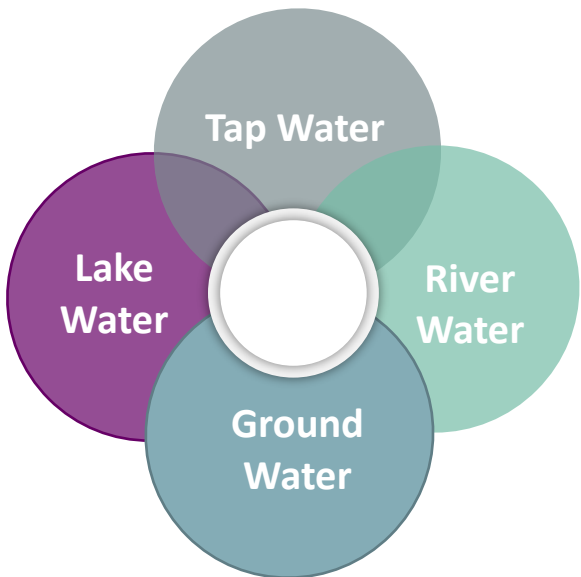
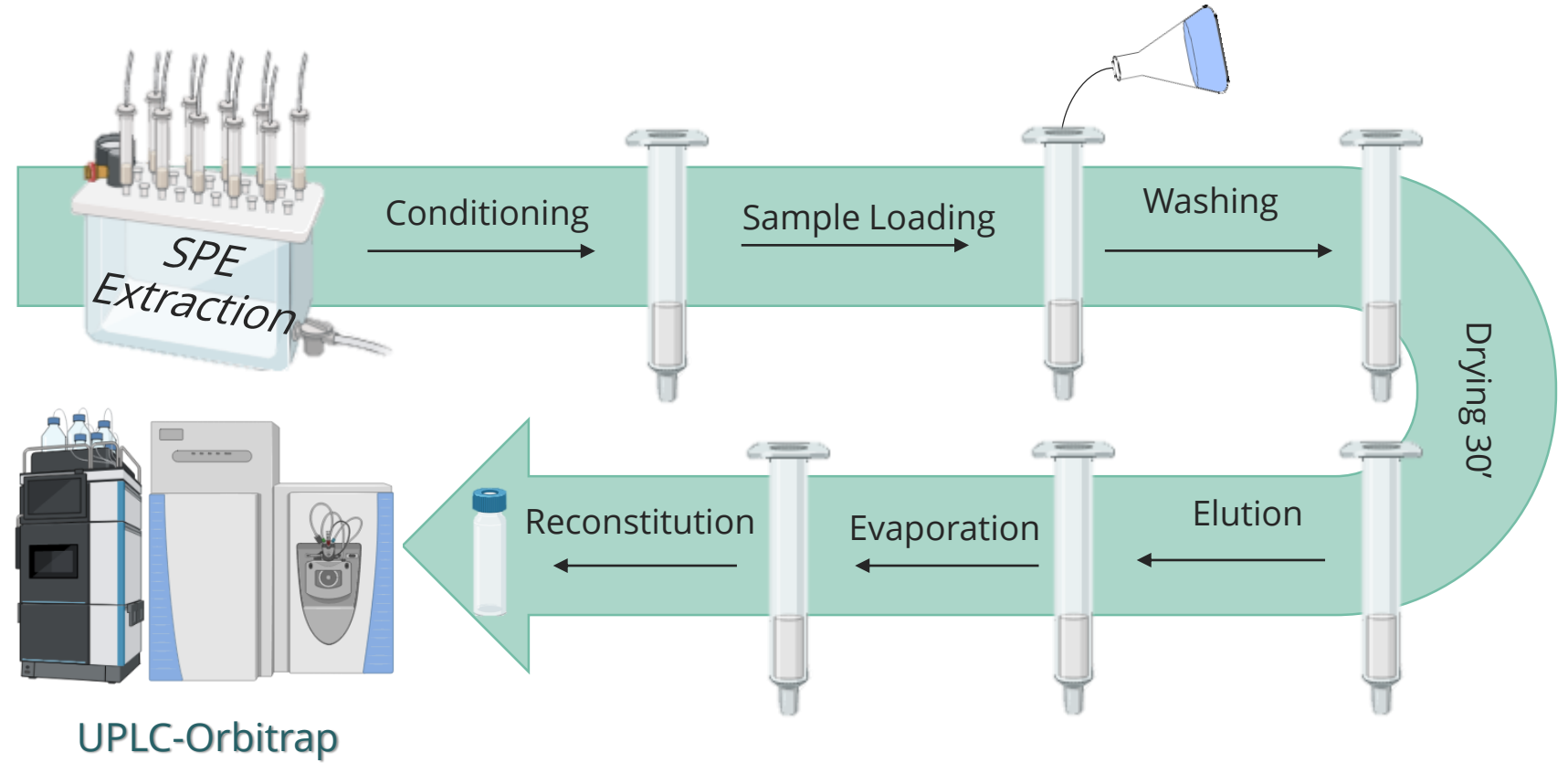


Pesticides,
Pharmaceuticals,
PCPs,
Industrial Chemicals

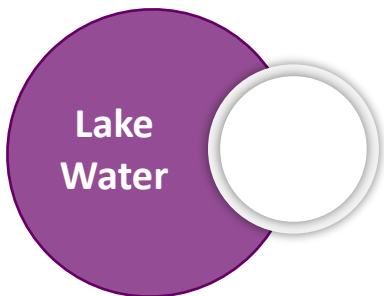
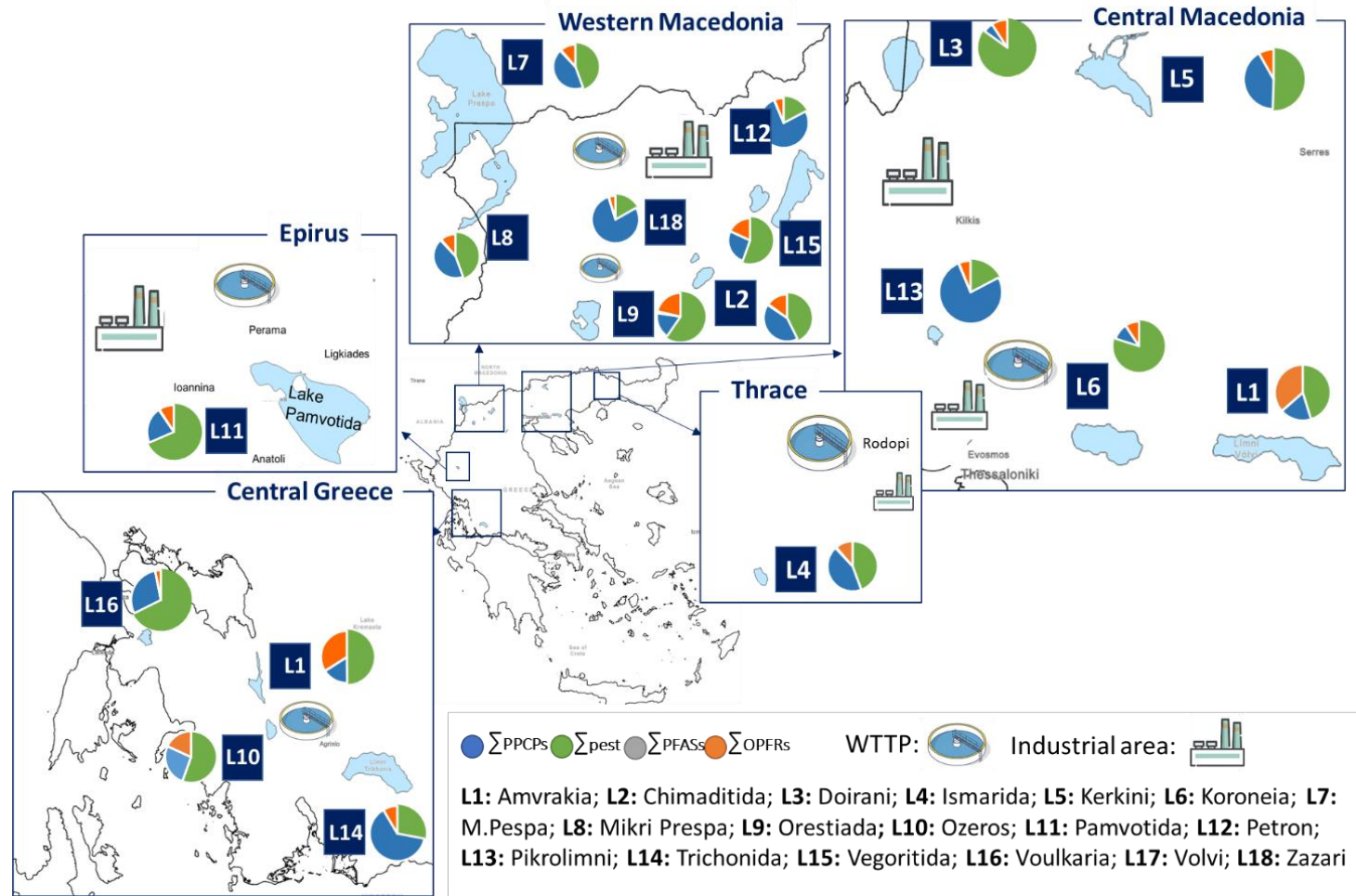


PFAS

Real Samples

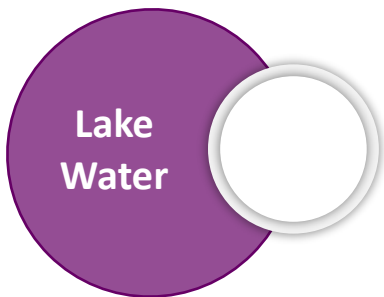
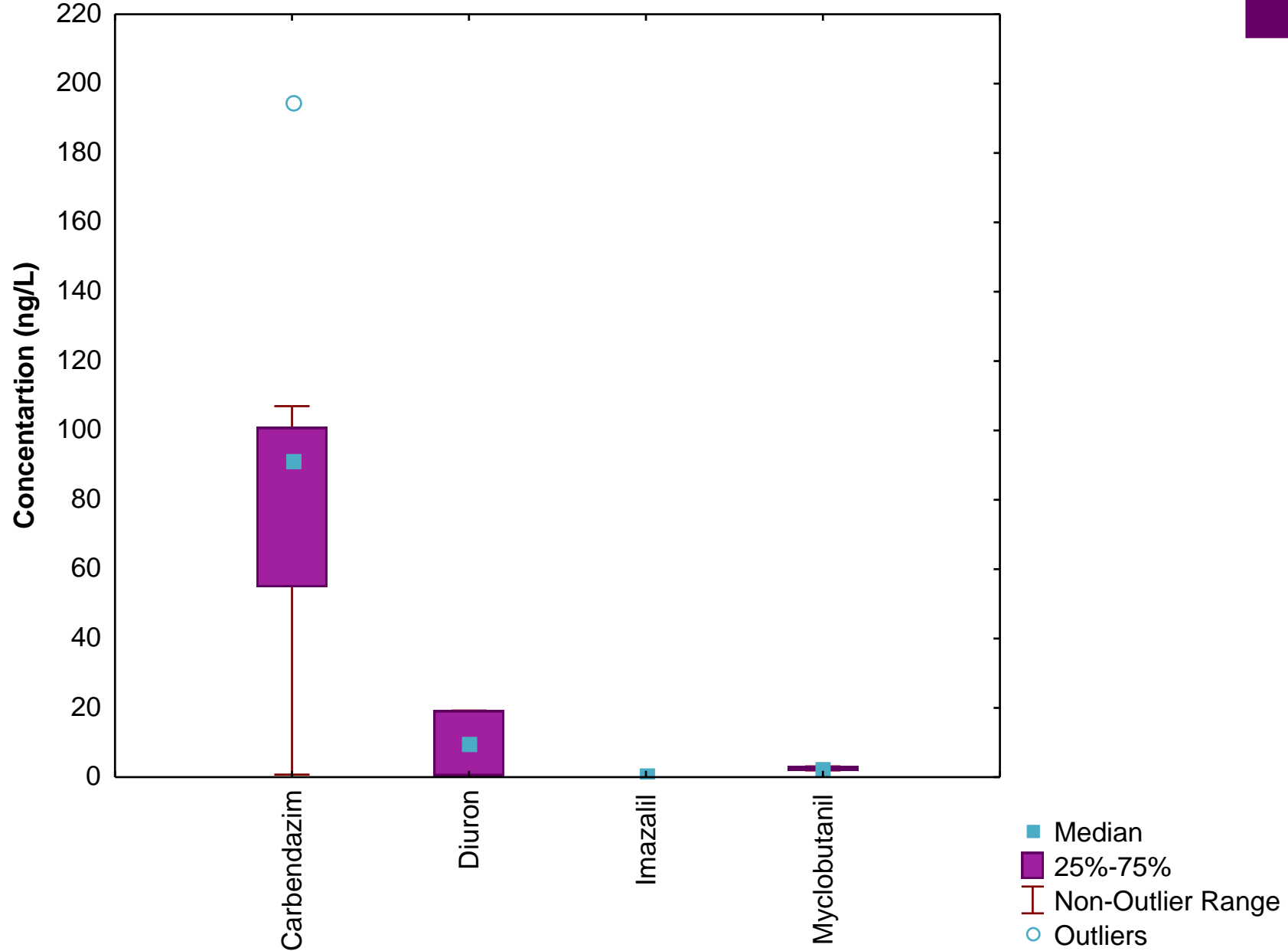


Real Samples



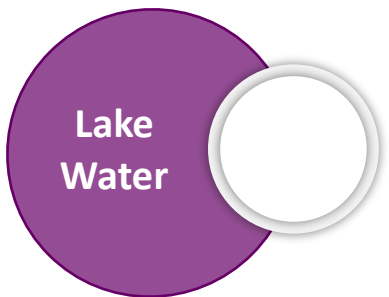
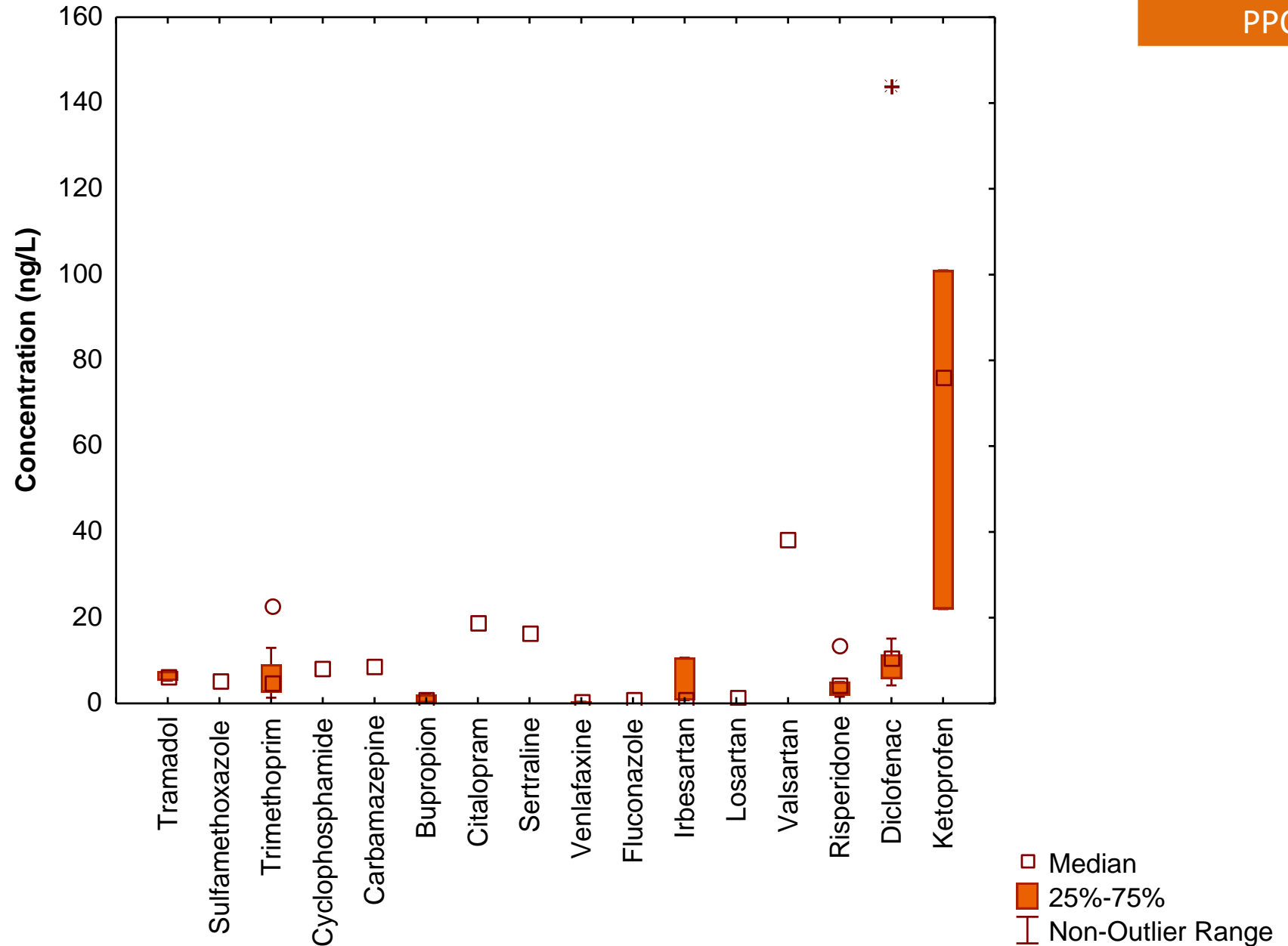
Real Samples

Pesticides



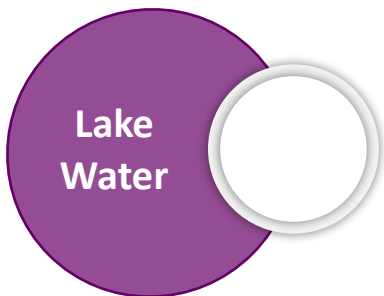
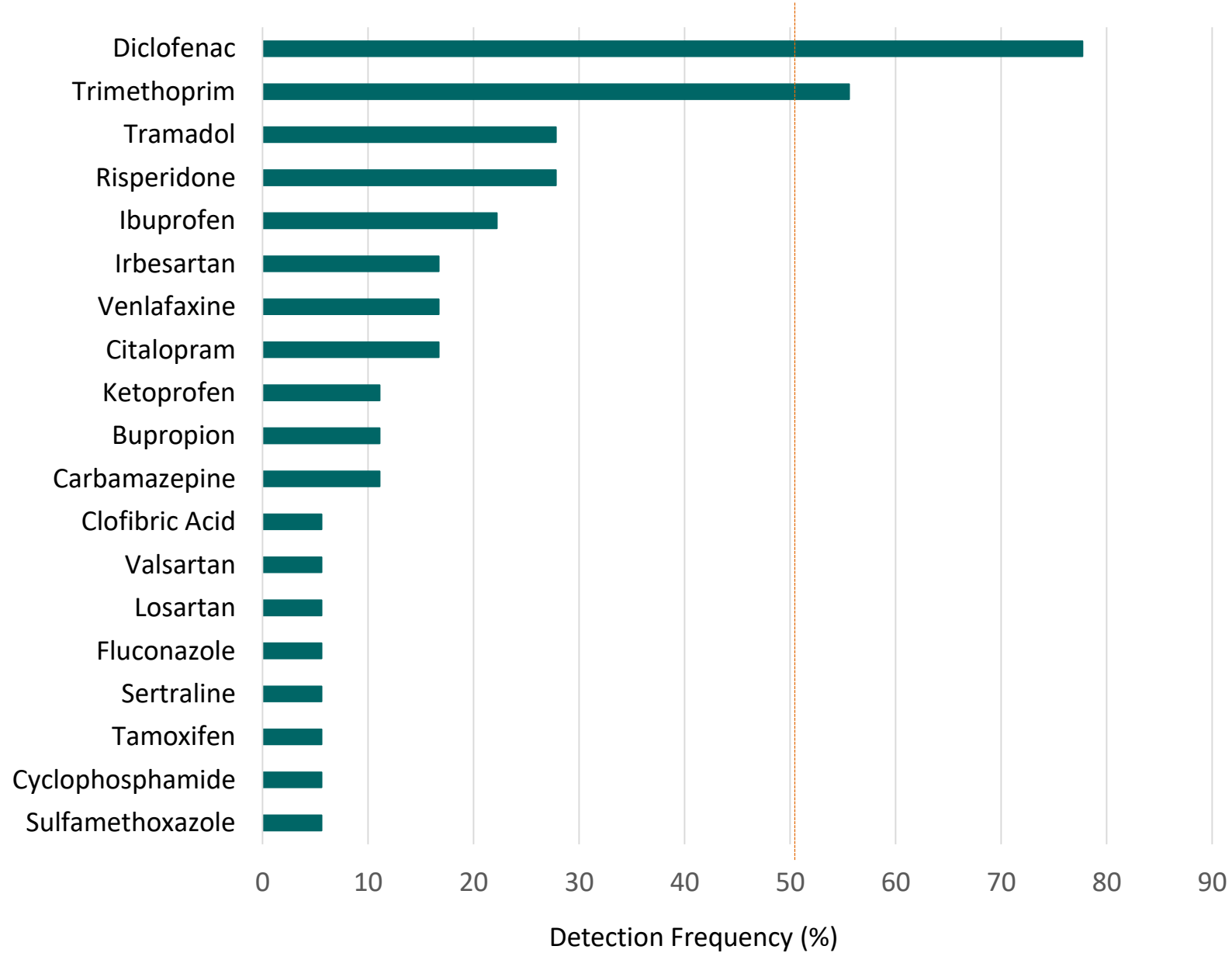
Real Samples

PPCPs



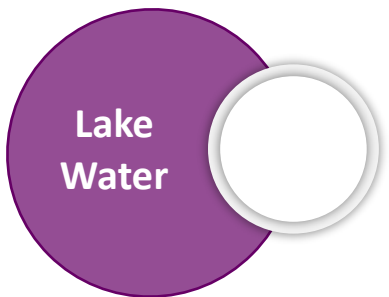
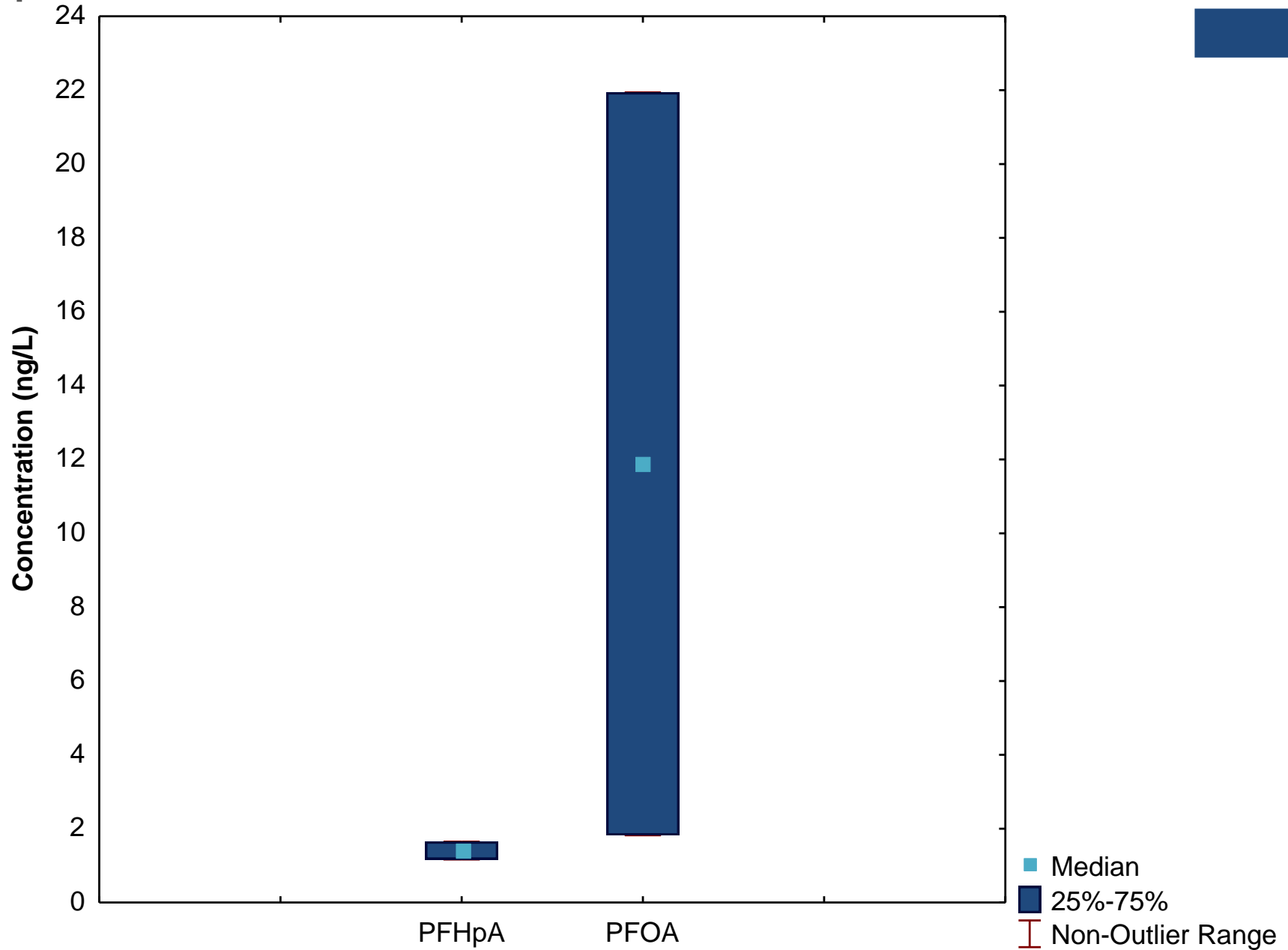
Real Samples

PPCPs



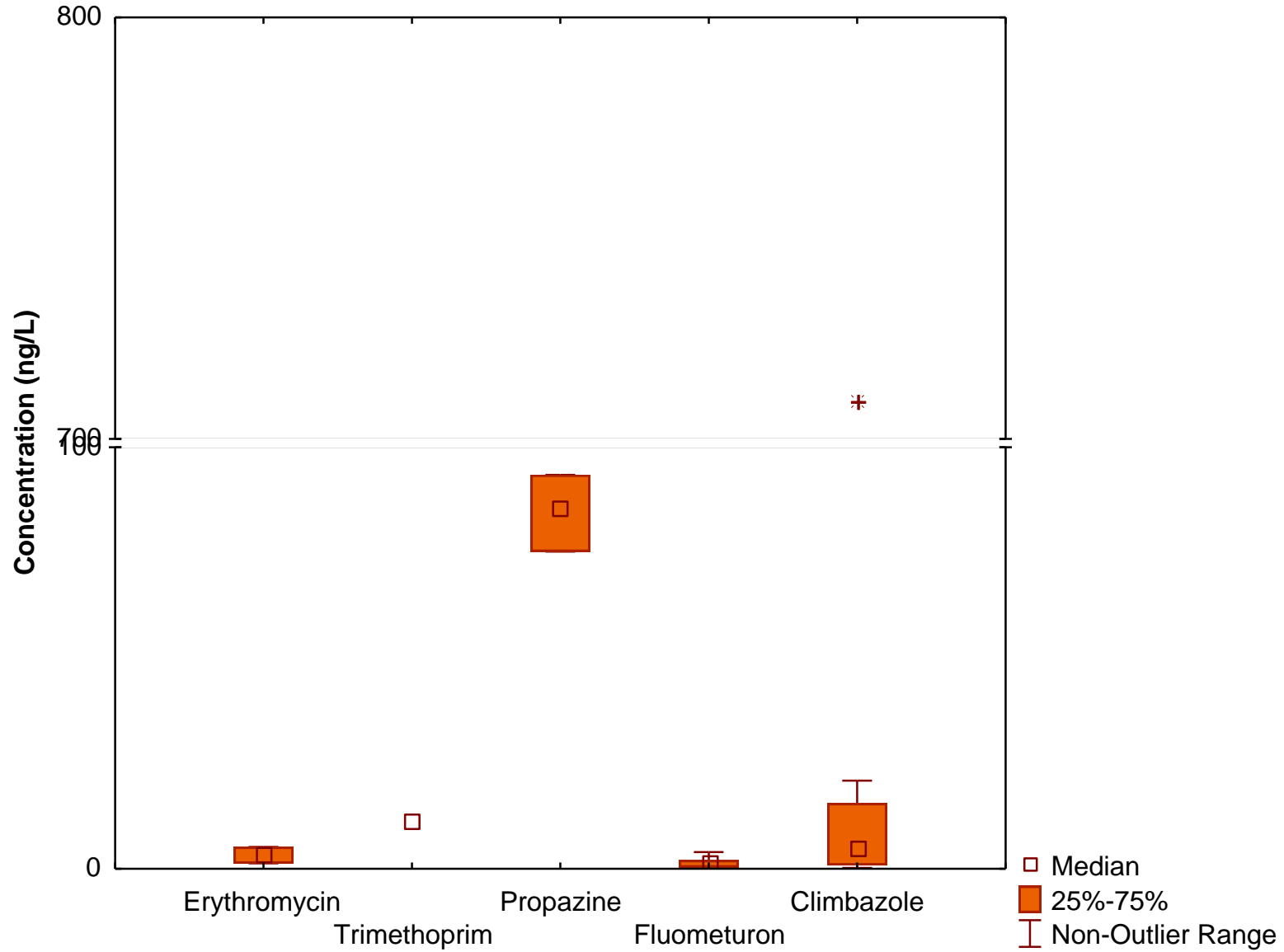
Real Samples

PFAS



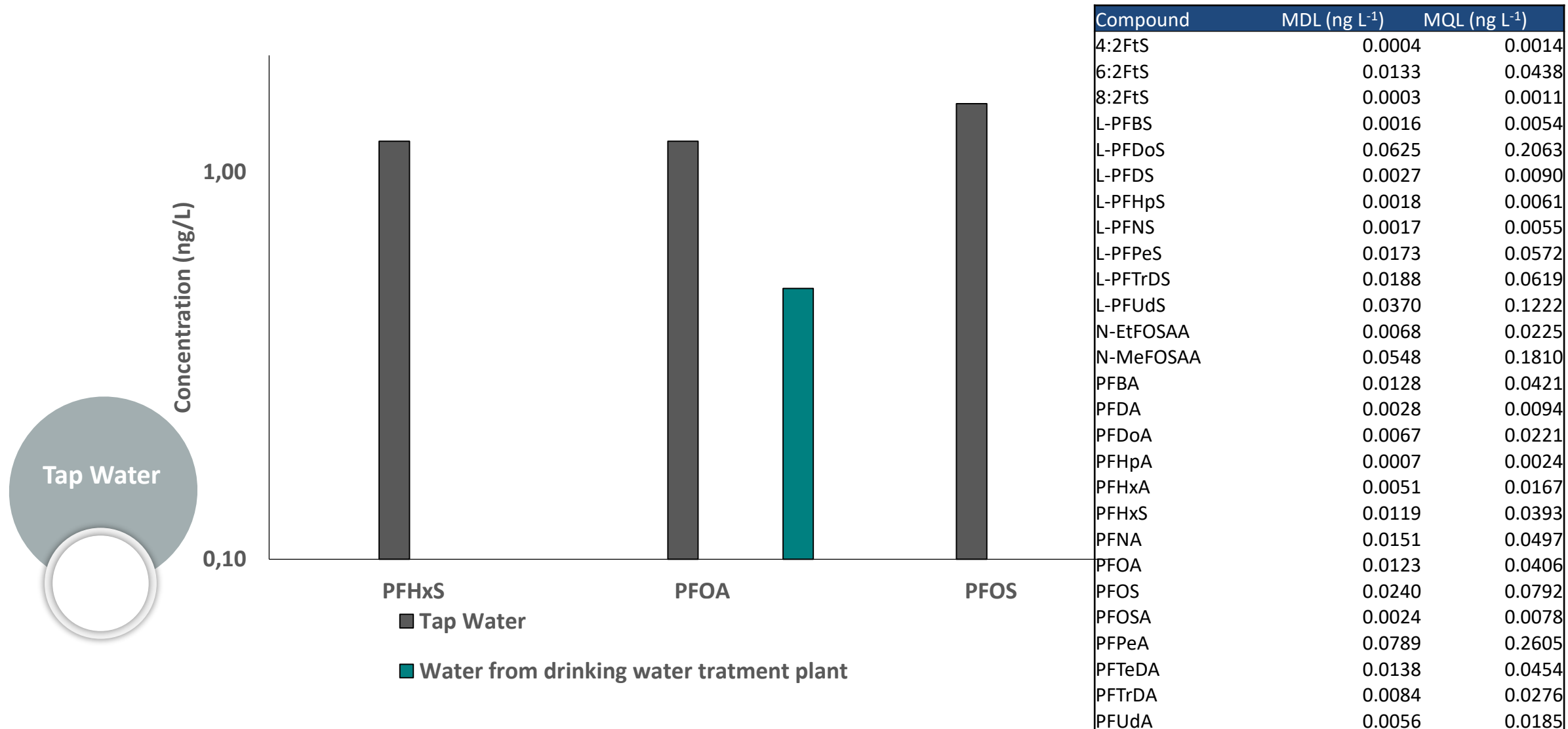
Real Samples

All ECs



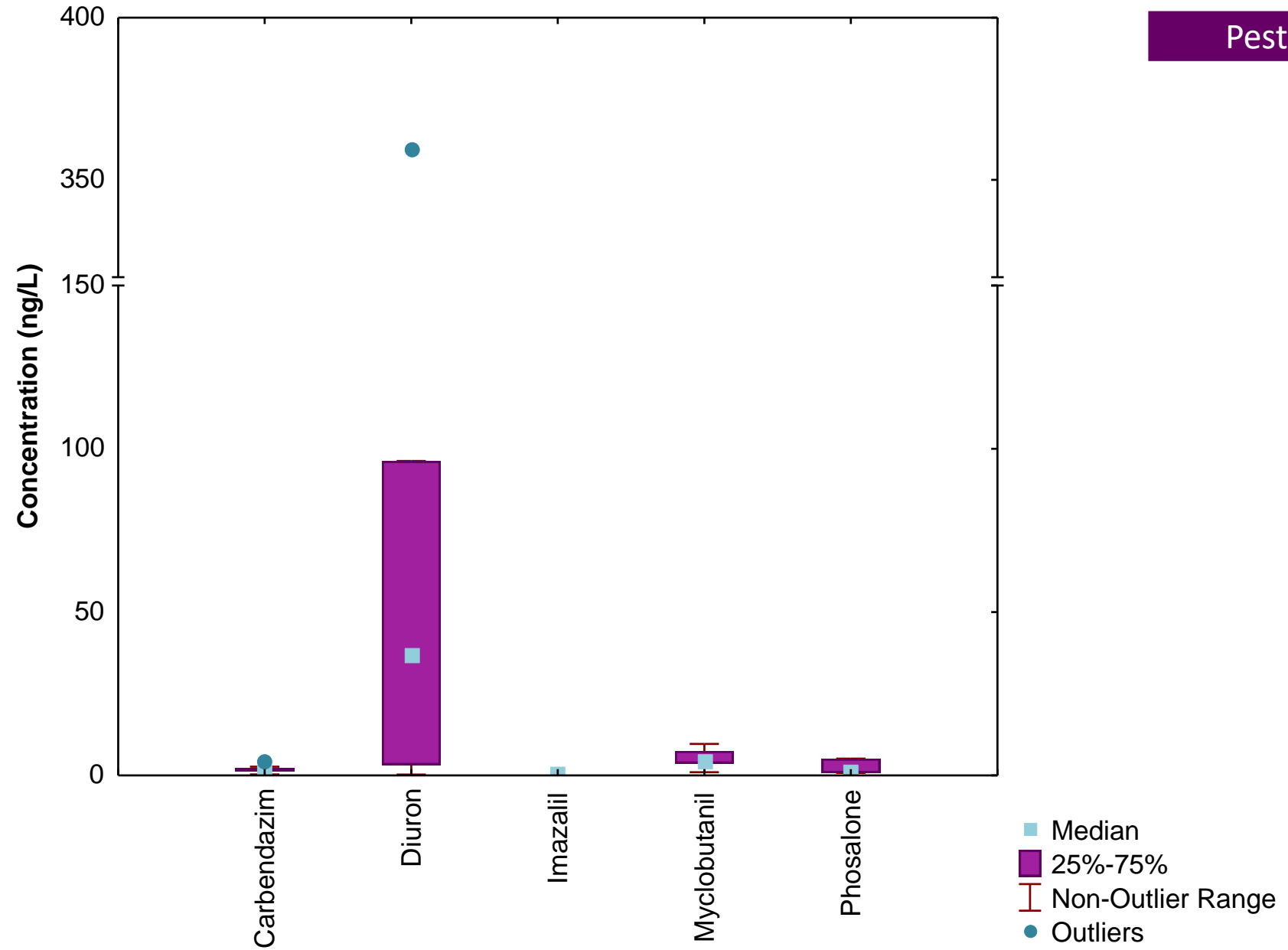
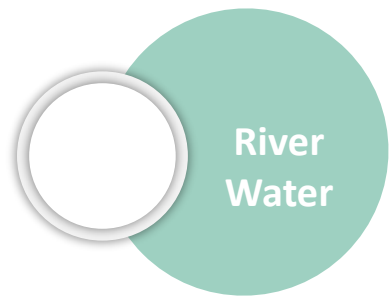
Real Samples

PFAS

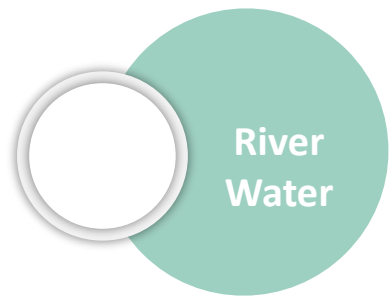


Real Samples

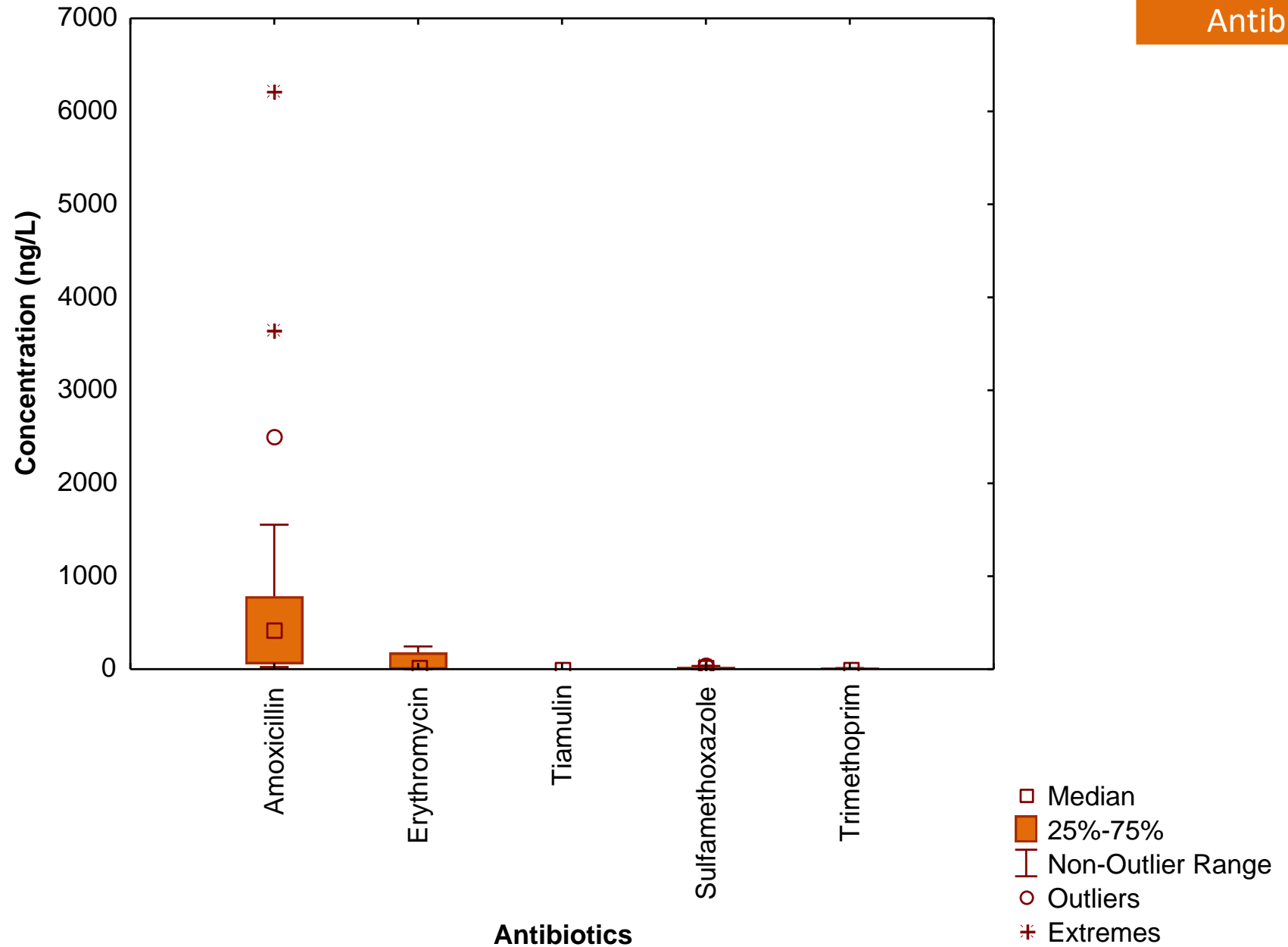
Pesticides



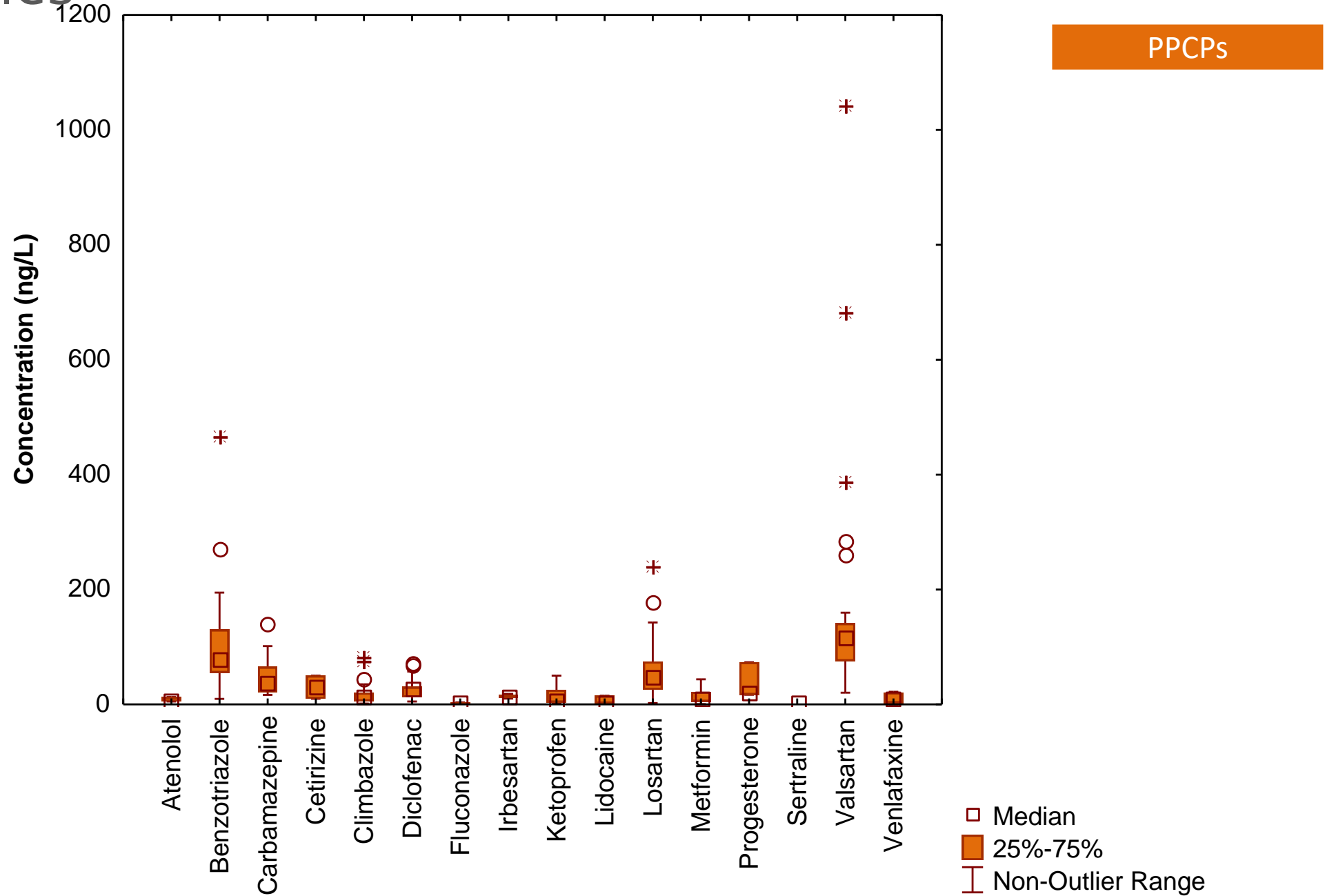
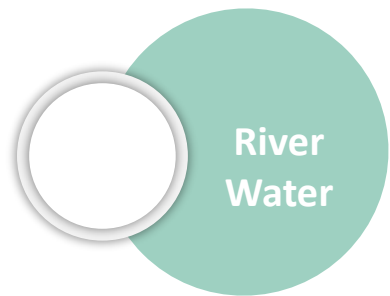
Real Samples



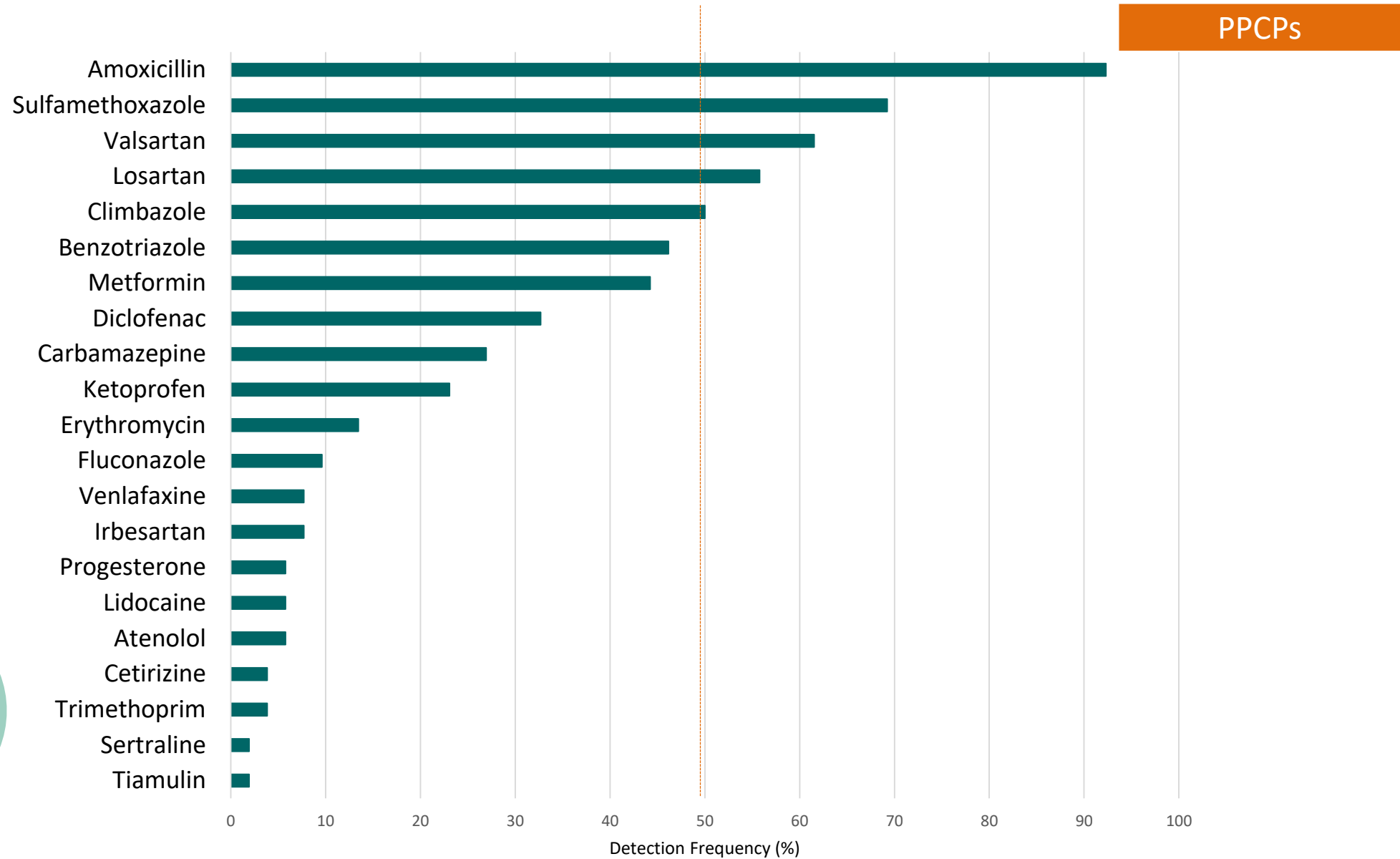
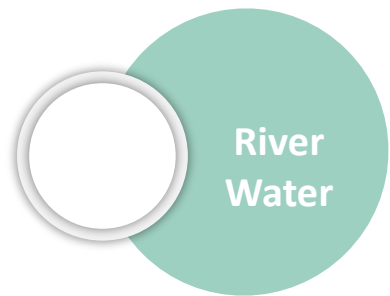
Antibiotics



Real Samples

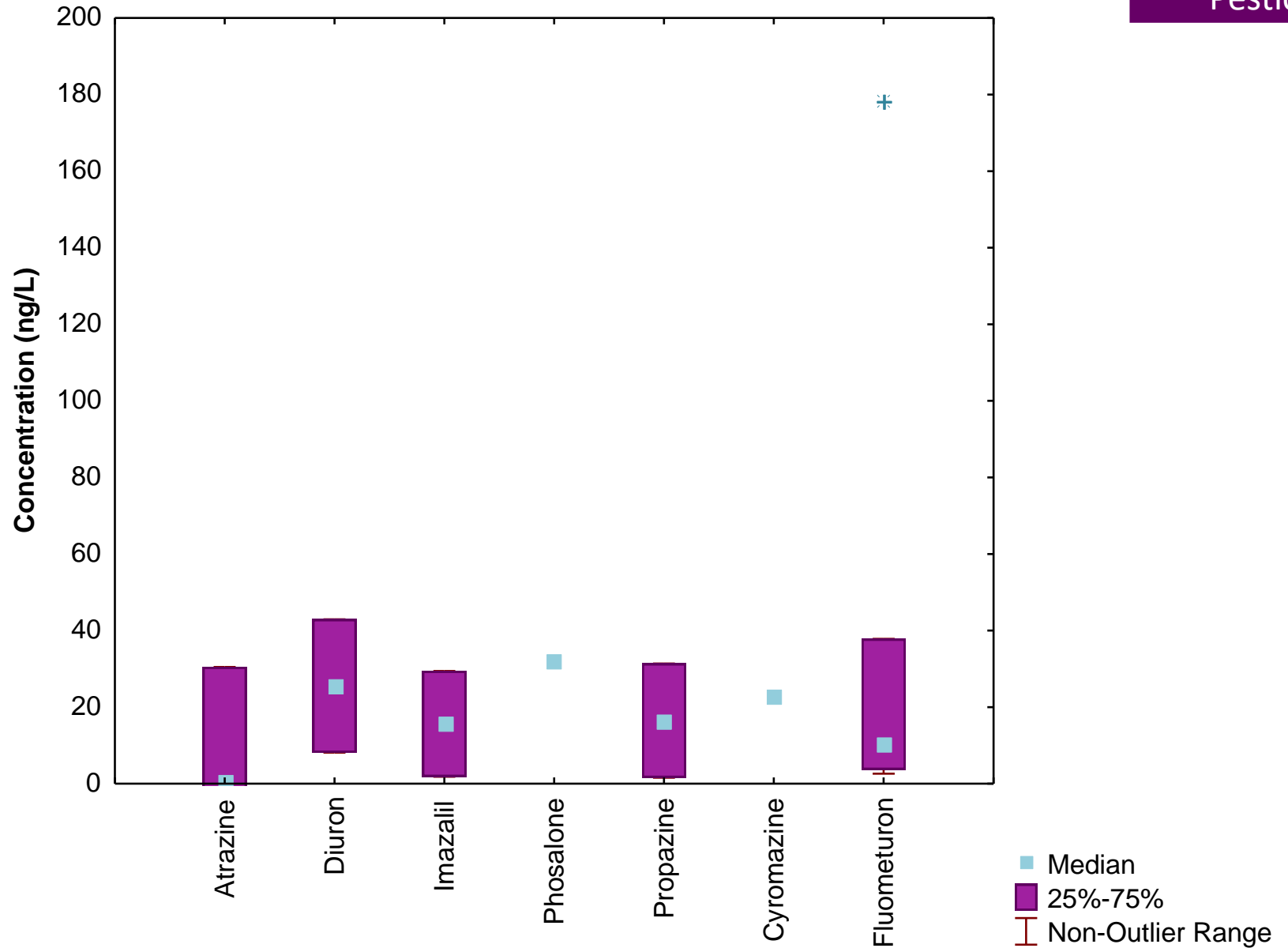


Real Samples



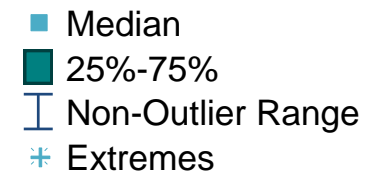
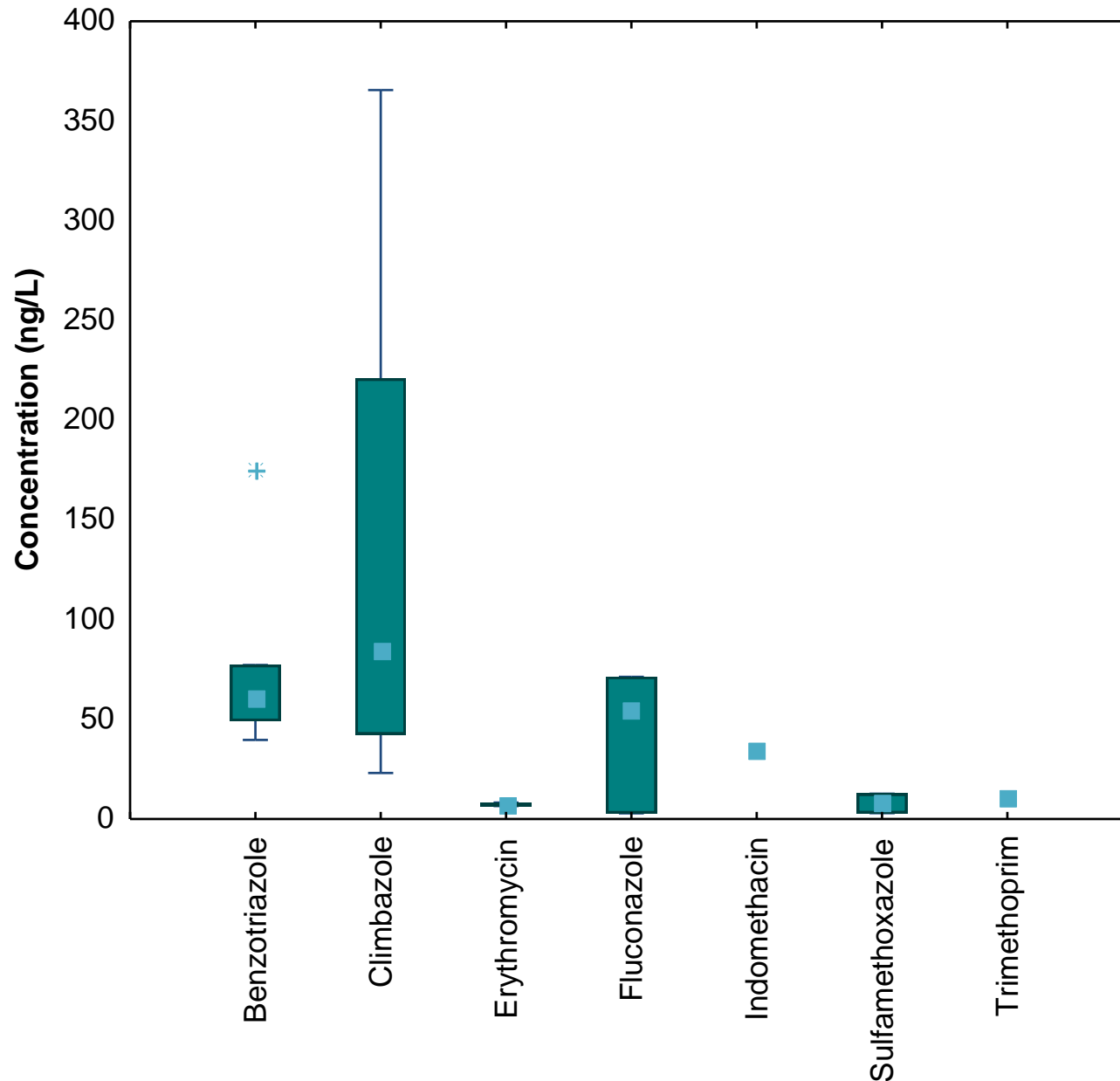
Real Samples

Pesticides



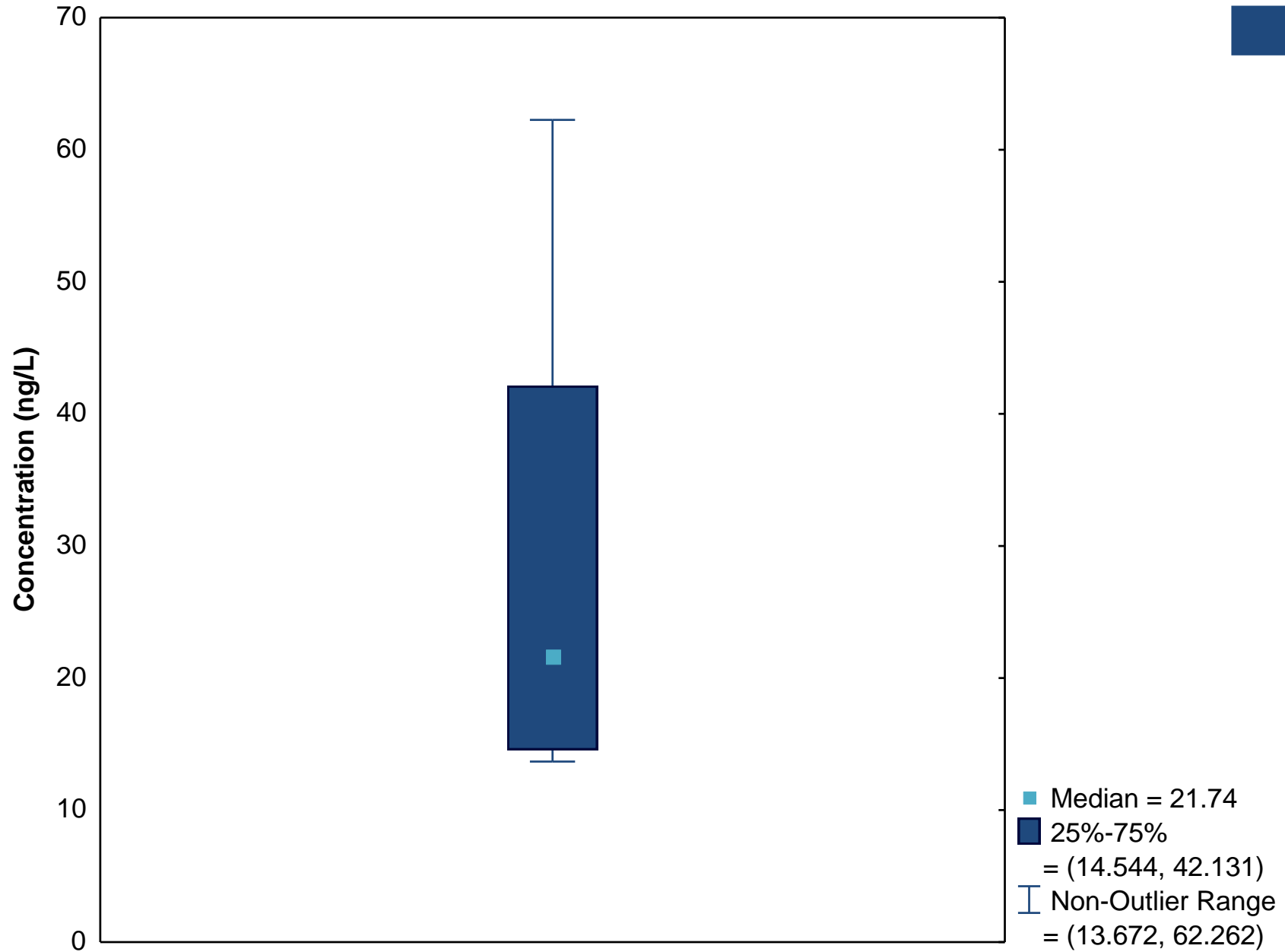
Real Samples

PPCPs

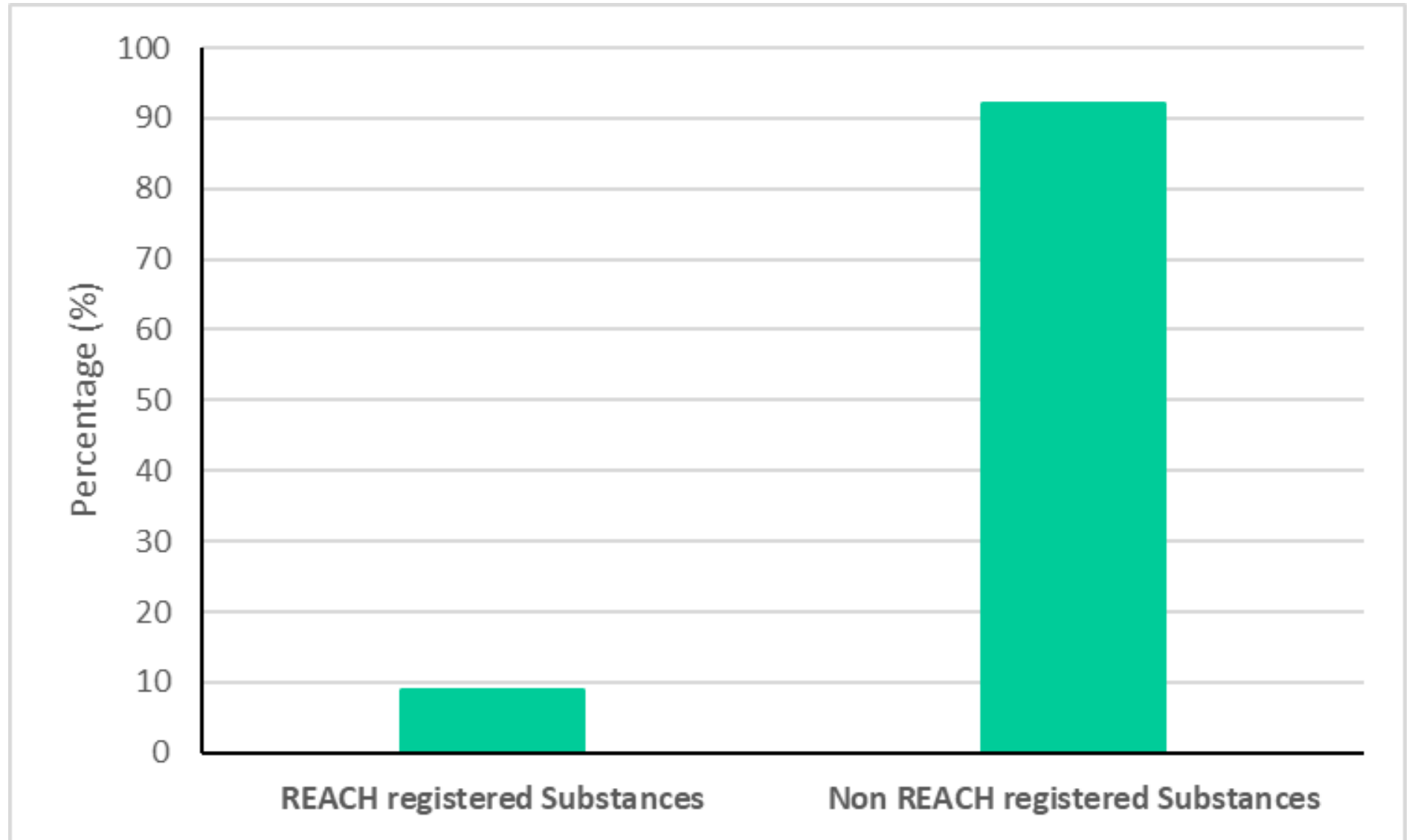


Real Samples

PFAS



Real Samples

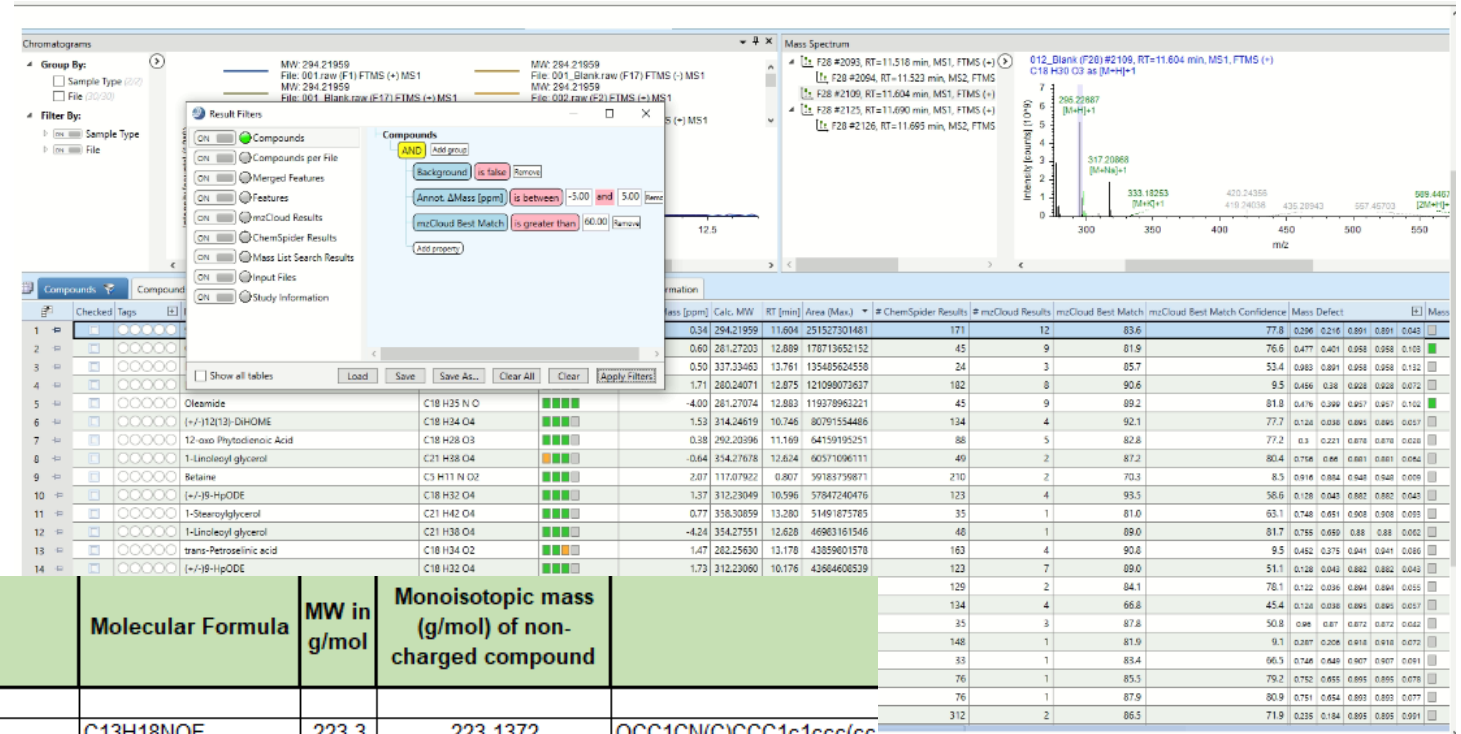




On going

Future Remarks

Suspect Screening

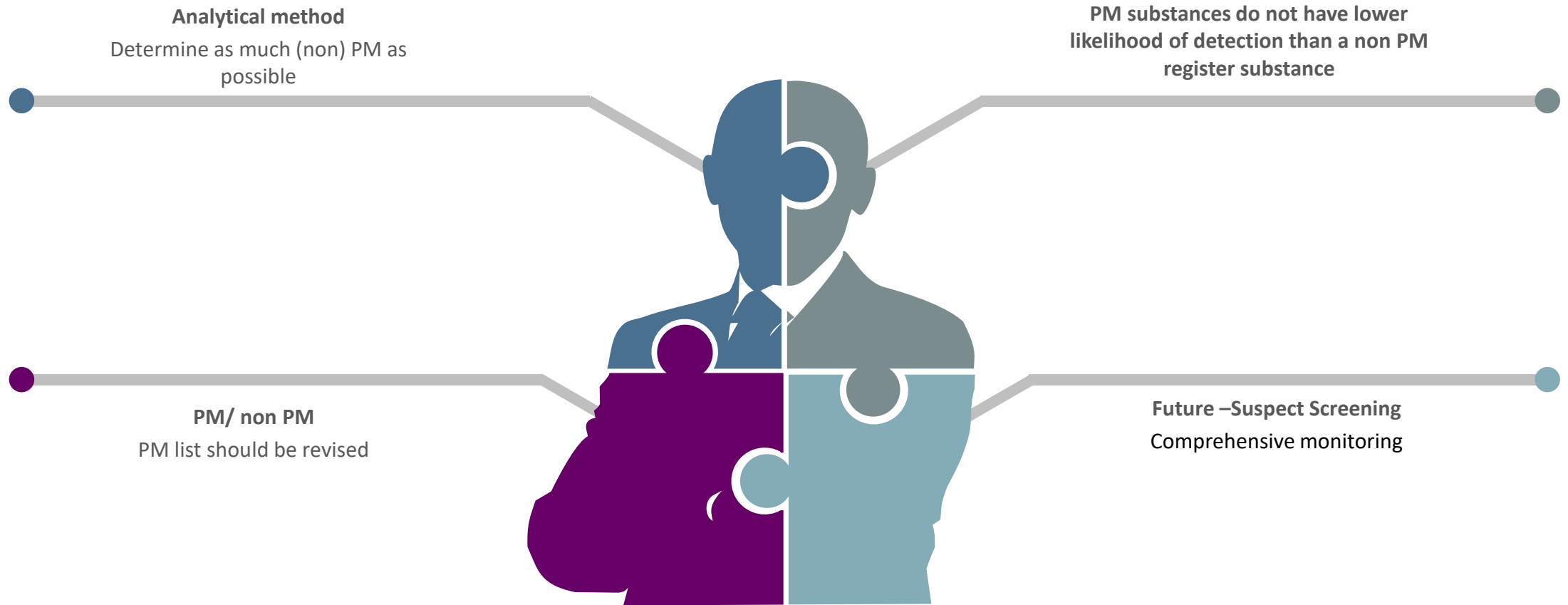


CAS	Substance Name	Molecular Formula	MW in g/mol	Monoisotopic mass (g/mol) of non-charged compound	
149-64-4	(-)-Scopolamine-N-butyl				
105812-81-5	(-)-trans-4-(4'-fluorophenyl)-3-hydroxymethyl-N-methylpiperidine	C13H18NOF	223.3	223.1372	OCC1CN(C)CCC1c1ccc(cc1)F
	(+)-menthol	C10H20O		156.151453	CC(C)C1CCC(CC1)O
107898-54-4	(+/-) trans-3,3-dimethyl-5-(2,2,3-trimethyl-cyclopent-3-en-1-yl)pent-4-en-2-ol	C15H26O	222.4	222.1984	CC(C)(C=CC1CC=C(C1(C)C)C)C(O)C
	(1,4-diazabicyclo[2.2.2]octane-2-yl)methanol	C7H14N2O		142.1106131	N12C(CO)CN(CC1)CC2
	(13Z)-Docos-13-enoic acid	C22H42O2		338.3184806	CCCCCCCC/C=C\CCCCC
342573-75-5	(1-Ethyl-3-methylimidazolium) ethyl sulfate	C2H5O4S	126.1	124.9909	CCOS(=O)(=O)[O-]
	(1-Hydroxycyclohexyl)(phenyl)methanone	C13H16O2		204.1150298	OC1(CCCCC1)C(=O)C1CCCC1
42978-66-5	(1-methyl-1,2-ethanediy)bis[oxy(methyl-2,1-ethanediy)] diacrylate	C15H24O6	300.4	300.1573	C=CC(=O)OCC(OCC(OCC(=O)C)C)C(=O)C
82203-23-4	(1R)-1-[(4R,4aR,8aS)-2,6-bis(4-chlorophenyl)tetrahydro[1,3]dioxino[5,4-d][1,3]dioxin-4-yl]ethanone	C20H20O6Cl2	427.3	426.0637	OCC(C1OC(OC2C1OC(OC2)C)C)C(=O)C
521284-22-0	(1R)-2-[[2-(4-aminophenyl)ethyl]amino]-1-phenylethanol hydrochloride	C16H20N2O	256.3	256.1576	Nc1ccc(cc1)CCNC[C@@H](O)C
119302-24-8	(1R,2R,4R,5S,7S,10R,11R,13R,14R,15S)-5-hydroxy-2,15-dimethyl-4-(morpholin-4-yl)-13-oxoheptadecan-1-ol	C29H48N2O4	488.7	488.3614	CC(=O)OC1C(CC2C1(C)CCN2)C
188199-50-0	(1R,2S,2'S,4R)-1,7,7-Trimethylbicyclo (2.2.1)heptane-2-spiro-4'-(2'-isopropyl-1',3'-dioxane)	C16H28O2	252.4	252.2089	CC1(C@H)1OCC(C@H)2C(C)CC12
847565-09-7	(1R,2S,5R)-5-methyl-2-(propan-2-yl)-N-[2-(pyridin-2-yl)ethyl]cyclohexanecarboxamide	C18H28N2O	288.4	288.2202	CC1CCC(C(C1)C(=O)NCCC2=CN=CN=C2)C
172015-79-1	(1S-cis)-4-(2-amino-6-chloro-9H-purin-9-yl)-2-cyclopentene-1-methanol hydrochloride	C11H12N5OCl	265.7	265.0730	OC[C@H]1C=C[C@H]1N2C=NC3=C(N)N=CN=C32





CONCLUSIONS





Portorož

Thank you



Thessaloniki

Join our team through social media!



ENVMS Group AUTH



@envmsgroup



ENVMS Group AUTH



@envmsgroup



<https://users.auth.gr/dlambro/contact.html>



dlambro@chem.auth.gr



+30 2310 99 7687



ENVMS Group

High-Resolution Mass Spec



Anagnostopoulou Kyriaki acknowledges a doctoral fellowship from the Hellenic Foundation for Research and Innovation (H.F.R.I.) under the “4th Call for H.F.R.I. PhD Fellowships (Fellowship Number 11182, Project Number 75685).



H.F.R.I.
Hellenic Foundation for
Research & Innovation