# **Overview of Mine Water Classification & its Genesis**

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Outbursts of Water from the Slovenian Abandoned Mines 18 March 2010





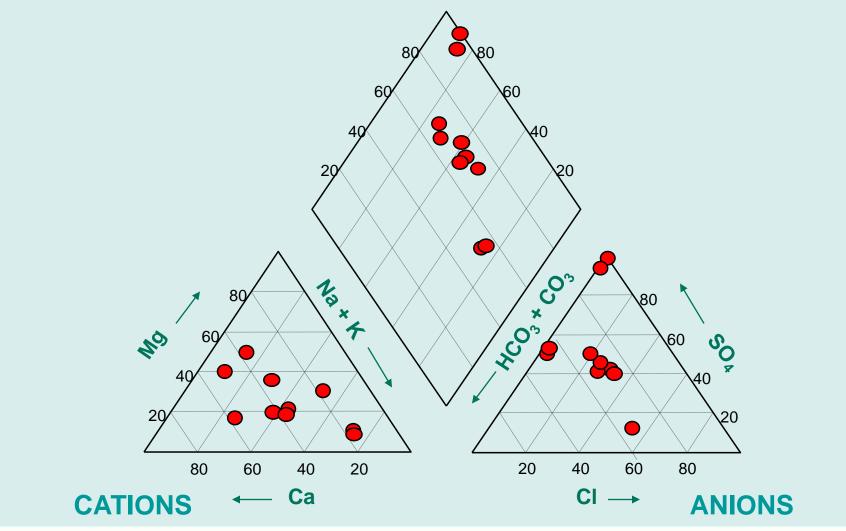
Six major schemes for classifying mine water exist:

- Facies diagrams (Piper 1944 & Durov 1948)
- Glover's scheme (1975)
- Ficklin et al. (1992)
- US Bureau of Mines' scheme (1994)
- Younger's scheme (1995)
- Azzie's scheme (2000)





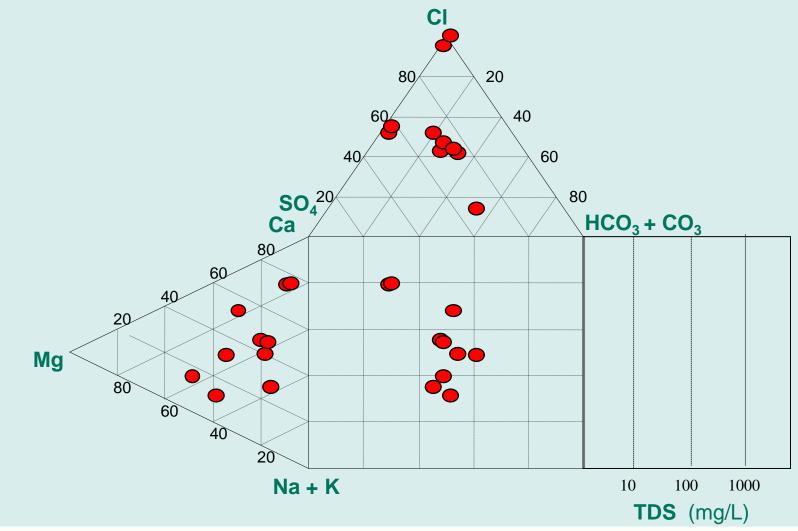
#### **Piper Diagram**







#### **Durov Diagram**





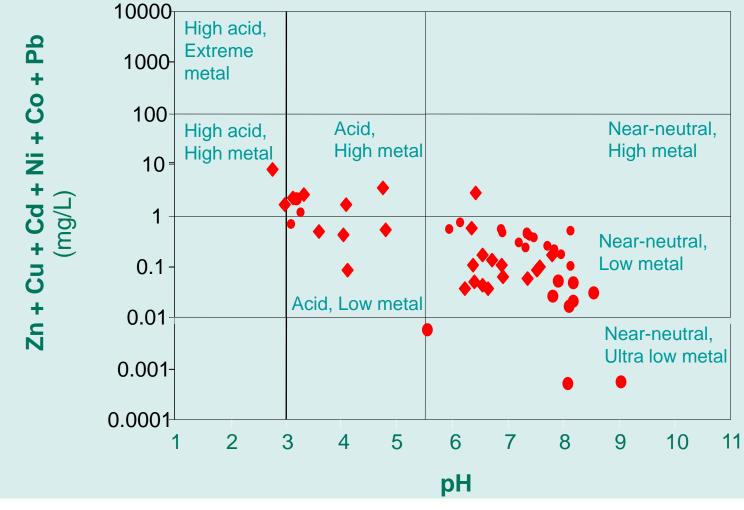


- 1. Acidic with low  $Fe_{TOTAL}$  conc.
- 2. Acidic with high  $Fe^{3+}$  conc.
- 3. Acidic with high Fe<sup>2+</sup> conc.
- 4. Neutral with high Fe<sup>2+</sup> conc.
- Suspended ferric hydroxide (combined with dissolved Fe<sup>2+</sup> or Fe<sup>3+</sup>)





#### Ficklin, Plumlee, Smith & McHugh







#### **US Bureau of Mines' Scheme**

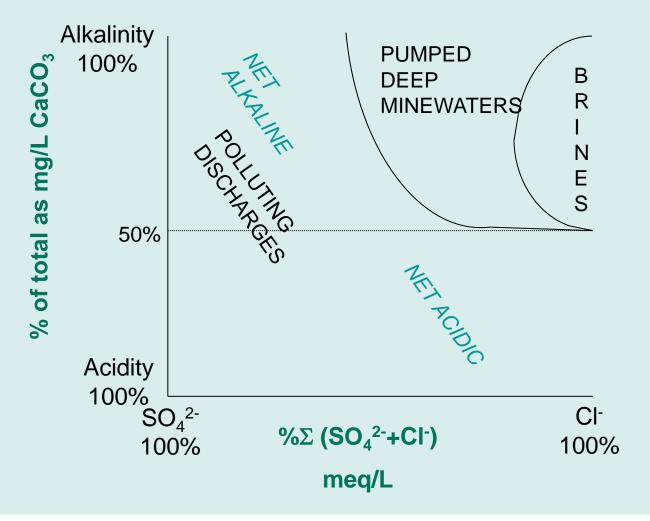
GROUP 1: Net alkaline minewaters i.e. alkalinity > acidity

GROUP 2: Net acidic mine waters i.e. acidity > alkalinity





#### Younger's Scheme







#### **Azzie's Classification**

#### **Selection Criteria**

- Alkalinity / acidity (Net)
- Salinity

 $I = \frac{1}{2} \sum m_i Z_i^2$ 

Metal ion status

SAR = Na<sup>+</sup> /  $(Ca^{2+}+Mg^{2+})^{\frac{1}{2}}$ 

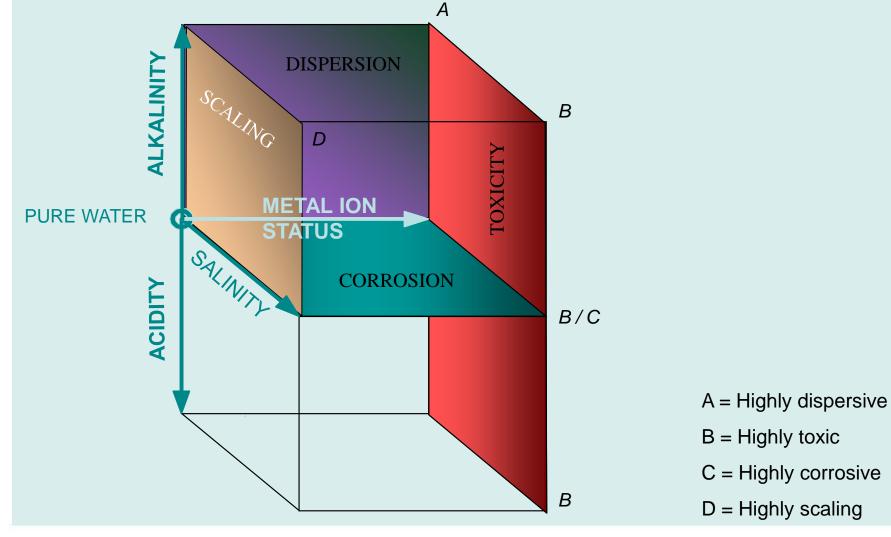
or

 $AAR = (AI^{3+} + Fe_{TOTAL}) / (Ca^{2+} + Mg^{2+})^{\frac{1}{2}}$ 





#### **Azzie's Classification**







### **Types of Drainage**

There are 3 types of drainage produced by sulphide mineral oxidation:

#### Acid Rock Drainage

- Acidic pH
- Moderate to elevated metals
- Elevated sulphate
- Treat for acid neutralization and metal & sulphate removal

#### Neutral Mine Drainage

- Near neutral to alkaline pH
- Low to moderate metals. May have elevated Zn, Cd, Mn, Sb, As, or Se
- Low to moderate sulphate
- Treat for metals & sometimes sulphate removal

#### Saline Drainage

- Neutral to alkaline pH
- Low metals. May have moderate
  Fe
- Moderate sulphate, magnesium & calcium
- Treat for sulphate and sometimes metal removal





- Metal Leaching & Acid Rock Drainage are *naturally* occurring processes.
- ML & ARD are caused when metal sulphides come into contact with both air and water.
- Rocks at most metal and some coal mines contain sulphide minerals.
- Excavating and crushing of ores greatly increases amount of rock surfaces which can be exposed to oxygen and water.
- So, mining activities can have high potential for leaching acid and metals.





### What is ML & ARD ?



- ML/ARD can occur from mining wastes (tailings & waste rock), in an open pit or along underground mine surfaces.
- Potential for environmental impacts depends on:
  - Amount of metals in the mine drainage;
  - Amount of acid-neutralizing ability in nearby rocks & water;
  - Amount of dilution available in streams; and
  - Sensitivity of the receiving environment.



#### Why are ML and ARD important ?



- They can have significant negative impacts on the environment if not adequately managed:
  - High levels of metals and/or acid can be harmful or toxic to living organisms;
  - Metals that are absorbed by plants & animals can be passed through food chain.
- Once initiated, it can persist for hundreds of years until sulphides are completely oxidized, and acid and metals are leached from rocks.
  - It can be VERY expensive to manage once it has developed
    - e.g. BC water treatment plants to treat ML/ARD have cost >\$10million (capital), with further \$1.5million/yr operating cost.



### **Mitigation Options 1**

- Proper planning of new mining developments can reduce risks, liabilities & costs associated with ML/ARD.
- Geochemical testing of rocks prior to mining can predict the likelihood of ML/ARD being an issue.
- Many strategies are available for prevention and management of ML/ARD.
- Every strategy has strengths and weaknesses, and not all strategies are applicable to all mine sites and their environments.
- For best results, a combination of strategies may be required.





#### **Mitigation Options 2**

- Basic principle behind management strategies:
  - Preventing oxygen contact with sulphide minerals
  - Reduce amount of water that comes into contact with acid generating wastes to minimize the amount of leaching.
- Most commonly used strategies include:
  - Avoidance
    - i.e. Don't mine the sulphide-bearing stuff !
    - Flooding of mine waste materials Timing is crucial ...
    - Covers
      - Susceptible to breakdown over time
    - Blending of materials

Only successful on a small scale

Drainage treatment

The last resort !





#### **Mitigation Options 3**

- Mitigation strategies must be designed to last forever !!
- Mine sites and their environments are dynamic and continue to change long after mining has ceased ..... changes can influence the effectiveness of mitigation strategies over time.
- Regular monitoring, maintenance and responsive management are key to long-term success in preventing impacts from ML/ARD.









# **Case Studies**

Implications of producing large volumes of contaminated water ... can mine water be a commodity ?

e.g. Coal mines in South Africa



# **Background to SA Coal Mining**

- Collieries exploiting the Witbank coalfields in South Africa have to continuously pump water out to reach the coal seams.
- Pyrite occurs naturally in coal formations, and when water enters the workings it becomes contaminated.
- SA environmental law requires water to be suitable for release back into the environment (which may involve management and treatment).
- Mines are located in the Upper Olifants catchment, which suffers from a chronic water shortage.
- Future mining developments are situated downstream, as is the scenic "Lowveld" and Kruger National Park.
- Water is characterized by high Ca, Mg and SO<sub>4</sub>.



#### **Case Study: Irrigation (South Africa)**

- Decided to investigate ways in which contaminated effluent could become a useful resource.
- Partnership between ACSA, WRC, Univ Pretoria & Coaltech 2020.
- Natural irrigation water varies greatly in quality.
- Some common soil problems that may develop:
  - Salinity
  - > Water infiltration rate ( $\uparrow$ Na &  $\downarrow$ Ca are problematic)
  - Specific ion toxicity (Na, Cl, B)
- SACE commissioned three centre-pivot irrigation systems, covering 82ha. Aim was to test viability of irrigating crops with saline effluent, that is high in SO<sub>4</sub> and K.





## **Case Study: Irrigation (South Africa)**



- These salts are taken up by certain crops and are highly beneficial if managed correctly.
- Irrigation of prime agricultural soils nearby, using this water, has improved productivity by 300%.

#### Further research required:

- Significance of crop selection
- Impact of irrigated salts on soil conditions
- Effects on groundwater
- Significant benefits for smallscale farmers in neighboring communities.





- The Emalahleni Water Reclamation Project sees the abstraction and treatment of acidic mine water from existing and old mines to a level fit for use by the local municipality.
- Sale of the water allows the mining companies involved to offset the costs of water treatment.
  - In 2005, the local municipality was drawing 80-90MLD from Witbank Dam, but this was ~20MLD short of that required.
    - A 0.120MLD demonstration plant was built and run for 3 months.

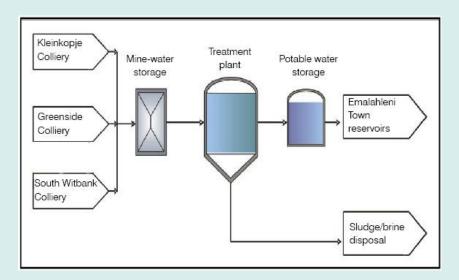




#### Results from the demonstration plant indicated that:

- ▷ pH increased from 2.9 to ~7.
- Total dissolved solids concentration reduced from 4500mg/L to 135mg/L.
- SO<sub>4</sub> concentrations reduced from 3500mg/L to 80mg/L.
- A yield of 98% was achieved.
- Treated water meets SABS 241 Class 0 Drinking Water Quality Limit.
- A full scale plant (20MLD) was commissioned in 2007, and is now fully operational.
- The full scale plant draws water from 3 mines, conveys it to a storage facility at treatment site.
- Storage facility has capacity for 46MLD, so caters for seasonal fluctuations.





- Acid water first undergoes neutralization using lime/ limestone.
- This increases pH and allows metals to precipitate out.
- Following clarification water is treated using ultrafiltration to remove remaining metals & bacteria.
- Reverse osmosis using spiral membranes then removes remaining salinity.
- 500 UF membranes and 1200 RO membranes are being used.



- Treated water is stored in 10MLD dome-shaped concrete reservoirs before being pumped 9km to the municipal reservoir for distribution to consumers.
- Approx 100m<sup>3</sup>/day of brine and 100t/day gypsiferous waste is produced.
- Brine is disposed of in 330,000m<sup>3</sup> evaporation ponds.
- An on-site laboratory monitors water quality.







# **Case Studies**

# Geochemistry to show impact of abandonment and rehabilitation

e.g. Coal mine in South Africa & Pb-Zn mine in Ireland





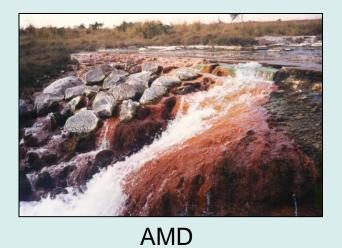
#### **Case Study: TNDBC (South Africa)**



#### Surface subsidence



#### Burning u/g workings





Polluted river





Main closure objectives include:

- Physical stability
- Chemical stability
- Biological stability
- Hydrological & hydrogeological environment
- Geographical and climatic influences
- Local sensitivities and opportunities
- Successive land use
- Funds for closure
- Socio-economic considerations





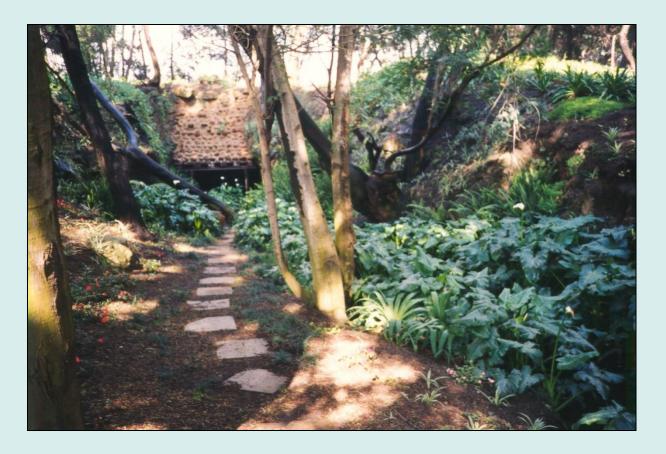
#### **Criteria for Mine Closure**

Closure plans:

- Costs included in the assessment of alternatives
- Adopt a risk assessment approach
- Are developed in Mine Planning phase
- Should be maintained during active life of a facility, and routinely updated when modifications are made
- Facilities to be designed to facilitate premature closure
- After-care design should minimize the need for active management



#### Case Study: New Largo (South Africa)



Does this qualify for a closure certificate ?

